

Ullmann's Encyclopedia of Industrial Chemistry

Sixth. Completely Revised Edition

Volume 6

Butenes

to

Cellulose Ethers

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Ullmann's Encyclopedia of Industrial Chemistry

Volume 6

Ullmann's Encyclopedia of Industrial Chemistry

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Volume 40: Index

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Symbols and Units

Symbols and units agree with SI standards (for conversion factors see page IX). The following list gives the most important symbols used in the encyclopedia. Articles with many specific units and symbols have a similar list as front matter.

Symbol	Unit	Physical Quantity
a_B		activity of substance B
A_r		relative atomic mass (atomic weight)
A	m^2	area
c_B	mol/m^3 , mol/L (M)	concentration of substance B
C	C/V	electric capacity
c_p , c_v	$\text{J kg}^{-1} \text{K}^{-1}$	specific heat capacity
d	cm, m	diameter
d		relative density (ρ/ρ_{water})
D	m^2/s	diffusion coefficient
D	Gy (= J/kg)	absorbed dose
e	C	elementary charge
E	J	energy
E	V/m	electric field strength
E	V	electromotive force
E_A	J	activation energy
f		activity coefficient
F	C/mol	Faraday constant
F	N	force
g	m/s^2	acceleration due to gravity
G	J	Gibbs free energy
h	m	height
$\hbar$	$\text{W} \cdot \text{s}^2$	Planck constant
H	J	enthalpy
I	A	electric current
I	cd	luminous intensity
k	(variable)	rate constant of a chemical reaction
k	J/K	Boltzmann constant
K	(variable)	equilibrium constant
l	m	length
m	g, kg, t	mass
M_r		relative molecular mass (molecular weight)
n_D^{20}		refractive index (sodium D-line, 20 °C)
n	mol	amount of substance
N_A	mol^{-1}	Avogadro constant ($6.023 \times 10^{23} \text{ mol}^{-1}$)
p	Pa, bar *	pressure
Q	J	quantity of heat
r	m	radius
R	$\text{J K}^{-1} \text{mol}^{-1}$	gas constant
R	Ω	electric resistance
S	J/K	entropy
t	s, min, h, d, month, a	time

Symbols and Units (Continued from p. VII)

Symbol	Unit	Physical Quantity
t	$^{\circ}\text{C}$	temperature
T	K	absolute temperature
u	m/s	velocity
U	V	electric potential
U	J	internal energy
V	m^3 , L, mL, μL	volume
w		mass fraction
W	J	work
x_B		mole fraction of substance B
Z		proton number, atomic number
α		cubic expansion coefficient
α	$\text{W m}^{-2}\text{K}^{-1}$	heat-transfer coefficient (heat-transfer number)
α		degree of dissociation of electrolyte
$[\alpha]$	$10^{-2}\text{deg cm}^2\text{g}^{-1}$	specific rotation
η	$\text{Pa} \cdot \text{s}$	dynamic viscosity
θ	$^{\circ}\text{C}$	temperature
$\varkappa$		c_p/c_v
λ	$\text{W m}^{-1}\text{K}^{-1}$	thermal conductivity
λ	nm, m	wavelength
μ		chemical potential
ν	Hz , s^{-1}	frequency
ν	m^2/s	kinematic viscosity (η/ρ)
π	Pa	osmotic pressure
ρ	g/cm^3	density
σ	N/m	surface tension
τ	$\text{Pa (N/m}^2\text{)}$	shear stress
φ		volume fraction
χ	$\text{Pa}^{-1} (\text{m}^2/\text{N})$	compressibility

* The official unit of pressure is the pascal (Pa).

Conversion Factors

SI unit	Non-SI unit	From SI to non-SI multiply by
<i>Mass</i>		
kg	pound (avoirdupois)	2.205
kg	ton (long)	9.842×10^{-4}
kg	ton (short)	1.102×10^{-3}
<i>Volume</i>		
m ³	cubic inch	6.102×10^4
m ³	cubic foot	35.315
m ³	gallon (U.S., liquid)	2.642×10^2
m ³	gallon (Imperial)	2.200×10^2
<i>Temperature</i>		
°C	°F	$^{\circ}\text{C} \times 1.8 + 32$
<i>Force</i>		
N	dyne	1.0×10^5
<i>Energy, Work</i>		
J	Btu (int.)	9.480×10^{-4}
J	cal (int.)	2.389×10^{-1}
J	eV	6.242×10^{18}
J	erg	1.0×10^7
J	kW · h	2.778×10^{-7}
J	kp · m	1.020×10^{-1}
<i>Pressure</i>		
MPa	at	10.20
MPa	atm	9.869
MPa	bar	10
kPa	mbar	10
kPa	mm Hg	7.502
kPa	psi	0.145
kPa	torr	7.502

Powers of Ten

E (exa)	10 ¹⁸	d (deci)	10 ⁻¹
P (peta)	10 ¹⁵	c (centi)	10 ⁻²
T (tera)	10 ¹²	m (milli)	10 ⁻³
G (giga)	10 ⁹	μ (micro)	10 ⁻⁶
M (mega)	10 ⁶	n (nano)	10 ⁻⁹
k (kilo)	10 ³	p (pico)	10 ⁻¹²
h (hecto)	10 ²	f (femto)	10 ⁻¹⁵
da (deca)	10	a (atto)	10 ⁻¹⁸

Abbreviations

The following is a list of the abbreviations used in the text. Common terms, the names of publications and institutions, and legal agreements are included along with their full identities. Other abbreviations will be defined wherever they first occur in an article. For further abbreviations, see page VII, Symbols and Units; page XIV, Frequently Cited Companies (Abbreviations), and page XV, Country Codes in patent references. The names of periodical publications are abbreviated exactly as done by Chemical Abstracts Service.

abs.	absolute	BAM	Bundesanstalt für Materialprüfung (Federal Republic of Germany)
a.c.	alternating current	BAT	Biologischer Arbeitsstoff-Toleranz-Wert (biological tolerance value for a working material, established by MAK Commission, see MAK)
ACGIH	American Conference of Governmental Industrial Hygienists	Beilstein	Beilstein's Handbook of Organic Chemistry, Springer, Berlin – Heidelberg – New York
ACS	American Chemical Society	BET	Brunauer – Emmett – Teller
ADI	acceptable daily intake	BGA	Bundesgesundheitsamt (Federal Republic of Germany)
ADN	accord européen relatif au transport international des marchandises dangereuses par voie de navigation interieure (European agreement concerning the international transportation of dangerous goods by inland waterways)	BGBI.	Bundesgesetzblatt (Federal Republic of Germany)
ADNR	ADN par le Rhin (regulation concerning the transportation of dangerous goods on the Rhine and all national waterways of the countries concerned)	BIOS	British Intelligence Objectives Subcommittee Report (see also FIAT)
ADP	adenosine 5'-diphosphate	BOD	biological oxygen demand
ADR	accord européen relatif au transport international des marchandises dangereuses par route (European agreement concerning the international transportation of dangerous goods by road)	<i>bp</i>	boiling point
AEC	Atomic Energy Commission (United States)	B.P.	British Pharmacopeia
a.i.	Active ingredient	BS	British Standard
AICHe	American Institute of Chemical Engineers	ca.	circa
AIME	American Institute of Mining, Metallurgical, and Petroleum Engineers	calcd.	calculated
ANSI	American National Standards Institute	CAS	Chemical Abstracts Service
AMP	adenosine 5'-monophosphate	cat.	catalyst, catalyzed
APhA	American Pharmaceutical Association	CEN	Comité Européen de Normalisation
API	American Petroleum Institute	cf.	compare
ASTM	American Society for Testing and Materials	CFR	Code of Federal Regulations (United States)
ATP	adenosine 5'-triphosphate	cfu	colony forming units
		Chap.	chapter
		ChemG	Chemikaliengesetz (Federal Republic of Germany)
		C.I.	Colour Index
		CIOS	Combined Intelligence Objectives Subcommittee Report (see also FIAT)
		CNS	central nervous system
		Co.	Company
		COD	chemical oxygen demand
		conc.	concentrated
		const.	constant
		Corp.	Corporation
		crit.	critical

CTFA	The Cosmetic, Toiletry and Fragrance Association (United States)	FIAT	Field Information Agency, Technical (United States reports on the chemical industry in Germany, 1945)
DAB 9	Deutsches Arzneibuch, 9th ed., Deutscher Apotheker-Verlag, Stuttgart 1986	Fig.	figure
d.c.	direct current	<i>fp</i>	freezing point
decomp.	decompose, decomposition	Friedländer	P. Friedländer, Fortschritte der Teerfarbenfabrikation und verwandter Industriezweige, Vol. 1 – 25, Springer, Berlin 1888 – 1942
DFG	Deutsche Forschungsgemeinschaft (German Science Foundation)	FT	Fourier transform
dil.	dilute, diluted	(g)	gas, gaseous
DIN	Deutsche Industrie Norm (Federal Republic of Germany)	GC	gas chromatography
DMF	dimethylformamide	GefStoffV	Gefahrstoffverordnung (regulations in the Federal Republic of Germany concerning hazardous substances)
DNA	deoxyribonucleic acid	GGVE	Verordnung in der Bundesrepublik Deutschland über die Beförderung gefährlicher Güter mit der Eisenbahn (regulation in the Federal Republic of Germany concerning the transportation of dangerous goods by rail)
DOE	Department of Energy (United States)	GGVS	Verordnung in der Bundesrepublik Deutschland über die Beförderung gefährlicher Güter auf der Straße (regulation in the Federal Republic of Germany concerning the transportation of dangerous goods by road)
DOT	Department of Transportation – Materials Transportation Bureau (United States)	GGVSee	Verordnung in der Bundesrepublik Deutschland über die Beförderung gefährlicher Güter mit Seeschiffen (regulation in the Federal Republic of Germany concerning the transportation of dangerous goods by sea-going vessels)
DTA	differential thermal analysis	GLC	gas-liquid chromatography
EC	effective concentration	Gmelin	Gmelin's Handbook of Inorganic Chemistry, 8th ed., Springer, Berlin – Heidelberg – New York
EC	European Community	GRAS	generally recognized as safe
ed.	editor, edition, edited	Hal	halogen substituent (–F, –Cl, –Br, –I)
e.g.	for example	Houben-Weyl	Methoden der organischen Chemie, 4th ed., Georg Thieme Verlag, Stuttgart
emf	electromotive force	HPLC	high performance liquid chromatography
EmS	Emergency Schedule	IAEA	International Atomic Energy Agency
EN	European Standard (European Community)	IARC	International Agency for Research on Cancer, Lyon, France
EPA	Environmental Protection Agency (United States)		
EPR	electron paramagnetic resonance		
Eq.	equation		
ESCA	electron spectroscopy for chemical analysis		
esp.	especially		
ESR	electron spin resonance		
Et	ethyl substituent (–C ₂ H ₅)		
et al.	and others		
etc.	et cetera		
EVO	Eisenbahnverkehrsordnung (Federal Republic of Germany)		
exp (...)	e ^(...) , mathematical exponent		
FAO	Food and Agriculture Organization (United Nations)		
FDA	Food and Drug Administration (United States)		
FD & C	Food, Drug and Cosmetic Act (United States)		
FHSA	Federal Hazardous Substances Act (United States)		

IATA-DGR	International Air Transport Association, Dangerous Goods Regulations	Federal Republic of Germany); cf. Deutsche Forschungsgemeinschaft (ed.): Maximale Arbeitsplatzkonzentrationen (MAK) und Biologische Arbeitsstoff-Toleranz-Werte (BAT), WILEY-VCH Verlag, Weinheim (published annually)
ICAO	International Civil Aviation Organization	
i.e.	that is	
i.m.	intramuscular	
IMDG	International Maritime Dangerous Goods Code	max. maximum
IMO	Inter-Governmental Maritime Consultative Organization (in the past: IMCO)	MCA Manufacturing Chemists Association (United States)
Inst.	Institute	Me methyl substituent ($-\text{CH}_3$)
i.p.	intraperitoneal	Methodicum Chimicum Methodicum Chimicum, Georg Thieme Verlag, Stuttgart
IR	infrared	MFAG Medical First Aid Guide for Use in Accidents Involving Dangerous Goods
ISO	International Organization for Standardization	MIK maximale Immissionskonzentration (maximum immission concentration)
IUPAC	International Union of Pure and Applied Chemistry	min. minimum
i.v.	intravenous	<i>mp</i> melting point
Kirk-Othmer	Encyclopedia of Chemical Technology, 3rd ed., J. Wiley & Sons, New York – Chichester – Brisbane – Toronto 1978 – 1984; 4th ed., J. Wiley & Sons, New York – Chichester – Brisbane – Toronto 1991 – 1998	MS mass spectrum, mass spectrometry
(l)	liquid	NAS National Academy of Sciences (United States)
Landolt-Börnstein	Zahlenwerte u. Funktionen aus Physik, Chemie, Astronomie, Geophysik u. Technik, Springer, Heidelberg 1950 – 1980; Zahlenwerte und Funktionen aus Naturwissenschaften und Technik, Neue Serie, Springer, Heidelberg, since 1961	NASA National Aeronautics and Space Administration (United States)
LC ₅₀	lethal concentration for 50 % of the test animals	NBS National Bureau of Standards (United States)
LCLo	lowest published lethal concentration	NCTC National Collection of Type Cultures (United States)
LD ₅₀	lethal dose for 50 % of the test animals	NIH National Institutes of Health (United States)
LDLo	lowest published lethal dose	NIOSH National Institute for Occupational Safety and Health (United States)
ln	logarithm (base e)	NMR nuclear magnetic resonance
LNG	liquefied natural gas	no. number
log	logarithm (base 10)	NOEL no observed effect level
LPG	liquefied petroleum gas	NRC Nuclear Regulatory Commission (United States)
M	mol/L	NRDC National Research Development Corporation (United States)
M	metal (in chemical formulas)	NSC National Service Center (United States)
MAK	Maximale Arbeitsplatz-Konzentration (maximum concentration at the workplace in the	NSF National Science Foundation (United States)
		NTSB National Transportation Safety Board (United States)
		OECD Organization for Economic Cooperation and Development
		OSHA Occupational Safety and Health Administration (United States)

p., pp.	page, pages		regulation in Federal Republic of Germany)
Patty	G. D. Clayton, F. E. Clayton (eds.): Patty's Industrial Hygiene and Toxicology, 3rd ed., Wiley Interscience, New York	TA Lärm	Technische Anleitung zum Schutz gegen Lärm (low noise regulation in Federal Republic of Germany)
PB	Publication Board Report (U.S. Department of Commerce, Scientific and Industrial Reports)	TDLo	lowest published toxic dose
PEL	permitted exposure limit	THF	tetrahydrofuran
Ph	phenyl substituent ($-\text{C}_6\text{H}_5$)	TLC	thin layer chromatography
Ph. Eur.	European Pharmacopoeia, 2nd. ed., Council of Europe, Strasbourg 1981	TLV	Threshold Limit Value (TWA and STEL); published annually by the American Conference of Governmental Industrial Hygienists (ACGIH), Cincinnati, Ohio
phr	part per hundred rubber (resin)	TOD	total oxygen demand
PNS	peripheral nervous system	TRK	Technische Richtkonzentration (lowest technically feasible level)
ppm	parts per million	TSCA	Toxic Substances Control Act (United States)
q. v.	which see (quod vide)	TÜV	Technischer Überwachungsverein (Technical Control Board of the Federal Republic of Germany)
ref.	refer, reference	TWA	Time Weighted Average
resp.	respectively	UBA	Umweltbundesamt (Federal Environmental Agency)
R_f	retention factor (TLC)	Ullmann	Ullmann's Encyclopedia of Industrial Chemistry, 5th ed., VCH Verlagsgesellschaft, Weinheim, 1985 – 1996; Ullmanns Encyklopädie der Technischen Chemie, 4th ed., Verlag Chemie, Weinheim 1972 – 1984; 3rd ed., Urban und Schwarzenberg, München 1951 – 1970
R. H.	relative humidity	USAEC	United States Atomic Energy Commission
RID	règlement international concernant le transport des marchandises dangereuses par chemin de fer (international convention concerning the transportation of dangerous goods by rail)	USAN	United States Adopted Names
RNA	ribonucleic acid	USD	United States Dispensatory
R phrase	risk phrase according to ChemG and GefStoffV (Federal Republic of Germany)	USDA	United States Department of Agriculture
(R-Satz)		U.S.P.	United States Pharmacopeia
rpm	revolutions per minute	UV	ultraviolet
RTECS	Registry of Toxic Effects of Chemical Substances, edited by the National Institute of Occupational Safety and Health (United States)	UVV	Unfallverhütungsvorschriften der Berufsgenossenschaft (workplace safety regulations in the Federal Republic of Germany)
(s)	solid	VbF	Verordnung in der Bundesrepublik Deutschland über die Errichtung und den Betrieb von Anlagen zur Lagerung, Abfüllung und Beförderung brennbarer Flüssigkeiten (regulation in the Federal Republic of Germany)
SAE	Society of Automotive Engineers (United States)		
s.c.	subcutaneous		
SI	International System of Units		
SIMS	secondary ion mass spectrometry		
S phrase	safety phrase according to ChemG and GefStoffV (Federal Republic of Germany)		
(S-Satz)			
STEL	Short Term Exposure Limit (see TLV)		
STP	standard temperature and pressure (0°C , 101.325 kPa)		
T_g	glass transition temperature		
TA Luft	Technische Anleitung zur Reinhaltung der Luft (clean air		

	concerning the construction and operation of plants for storage, filling, and transportation of flammable liquids; classification according to the flash point of liquids, in accordance with the classification in the United States)	vs.	versus
VDE	Verband Deutscher Elektroingenieure (Federal Republic of Germany)	WVG	Wassergefährdungsklasse (water hazard class)
VDI	Verein Deutscher Ingenieure (Federal Republic of Germany)	WHO	World Health Organization (United Nations)
vol	volume	Winnacker-Küchler	Chemische Technologie, 4th ed., Carl Hanser Verlag, München, 1982-1986;
vol.	volume (of a series of books)		Winnacker-Küchler, Chemische Technik: Prozesse und Produkte, Wiley-VCH, Weinheim, from 2003
		wt	weight
		\$	U.S. dollar, unless otherwise stated

Frequently Cited Companies (Abbreviations)

Air Products	Air Products and Chemicals	ICI	Imperial Chemical Industries
Akzo	Algemene Koninklijke Zout Organon	IFP	Institut Français du Pétrole
Alcoa	Aluminum Company of America	INCO	International Nickel Company
Allied	Allied Corporation	3M	Minnesota Mining and Manufacturing Company
Amer.	American Cyanamid	Mitsubishi	Mitsubishi Chemical Industries
Cyanamid	Company	Chemical	
BASF	BASF Aktiengesellschaft	Monsanto	Monsanto Company
Bayer	Bayer AG	Nippon	Nippon Shokubai Kagaku Kogyo
BP	British Petroleum Company	Shokubai	
Celanese	Celanese Corporation	PCUK	Pechiney Ugine Kuhlmann
Daicel	Daicel Chemical Industries	PPG	Pittsburg Plate Glass Industries
Dainippon	Dainippon Ink and Chemicals Inc.	Searle	G.D. Searle & Company
		SKF	Smith Kline & French Laboratories
Dow Chemical	The Dow Chemical Company	SNAM	Società Nazionale Metandotti
DSM	Dutch Staats Mijnen	Sohio	Standard Oil of Ohio
Du Pont	E.I. du Pont de Nemours & Company	Stauffer	Stauffer Chemical Company
Exxon	Exxon Corporation	Sumitomo	Sumitomo Chemical Company
FMC	Food Machinery & Chemical Corporation	Toray	Toray Industries Inc.
GAF	General Aniline & Film Corporation	UCB	Union Chimique Belge
		Union Carbide	Union Carbide Corporation
W.R. Grace	W.R. Grace & Company	UOP	Universal Oil Products Company
Hoechst	Hoechst Aktiengesellschaft	VEBA	Vereinigte Elektrizitäts- und Bergwerks-AG
IBM	International Business Machines Corporation	Wacker	Wacker Chemie GmbH

Country Codes

The following list contains a selection of standard country codes used in the patent references.

AT	Austria	ID	Indonesia
AU	Australia	IL	Israel
BE	Belgium	IT	Italy
BG	Bulgaria	JP	Japan *
BR	Brazil	LU	Luxembourg
CA	Canada	MA	Morocco
CH	Switzerland	NL	Netherlands *
CS	Czechoslovakia	NO	Norway
DD	German Democratic Republic	NZ	New Zealand
DE	Federal Republic of Germany (and Germany before 1949) *	PL	Poland
DK	Denmark	PT	Portugal
ES	Spain	SE	Sweden
FI	Finland	SU	Soviet Union
FR	France	US	United States of America
GB	United Kingdom	YU	Yugoslavia
GR	Greece	ZA	South Africa *
HU	Hungary	EP	European Patent Office *
		WO	World Intellectual Property Organization

* For Europe, Federal Republic of Germany, Japan, and the Netherlands, the type of patent is specified:
 EP (patent), EP-A (application), DE (patent), DE-OS (Offenlegungsschrift), DE-AS (Auslegeschrift),
 JP (patent), JP-Kokai (Kokai tokkyo koho), NL (patent), and NL-A (application).

Periodic Table of Elements

element symbol, atomic number, and relative atomic mass (atomic weight)

1A "European" group designation and old IUPAC recommendation

1 group designation to 1986 IUPAC proposal

IA "American" group designation, also used by the Chemical Abstracts Service until the end of 1986

0
18
VIII A

1 H 1.0079	2A	3 Li 6.941	4 Be 9.0122	5A	6A	7A	8	8	8	1B	2B	3B	4B	5B	6B	7B	12 Hg 200.59
11 Na 22.990	12 Mg 24.305	13 Al 26.982	14 Si 28.086	15 P 30.974	16 S 32.06	17 Cl 35.453	18	18	18	18	18	18	18	18	18	18	18
19 K 39.098	20 Ca 40.078	21 Sc 44.956	22 Ti 47.867	23 V 50.942	24 Cr 51.996	25 Mn 54.938	26 Fe 55.845	27 Co 58.933	28 Ni 58.693	29 Cu 63.546	30 Zn 65.409	31 Ga 69.723	32 Ge 72.61	33 As 74.922	34 Se 78.96	35 Br 79.904	36 Kr 83.80
37 Rb 85.468	38 Sr 87.62	39 Y 88.906	40 Zr 91.224	41 Nb 92.906	42 Mo 95.94	43 Tc 98.906	44 Ru 101.07	45 Rh 102.91	46 Pd 106.42	47 Ag 107.87	48 Cd 112.41	49 In 114.82	50 Sn 118.71	51 Sb 121.76	52 Te 127.60	53 I 126.90	54 Xe 131.29
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87 Fr 223.02	88 Ra 226.03	89 Ac 227.03	90 Th 232.04	91 Pa 231.04	92 U 238.03	93 Np 237.05	94 Pu 244.06	95 Am 243.06	96 Cm 247.07	97 Bk 247.07	98 Cf 251.08	99 Es 252.08	100 Fm 257.10	101 Md 258.10	102 No 259.10	103 Lr 260.11	104 Uuh 260.11

* provisional IUPAC symbol

* radioactive element; mass of most important isotope given.

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Butenes

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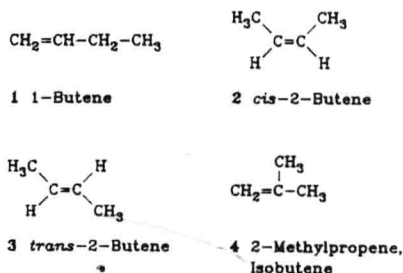
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1. Introduction

Butenes are unsaturated olefinic hydrocarbons, C_4H_8 , M_r 56.1080. There are four isomers:



"Butylenes", the older name for "butenes", is still used today; 4 is frequently referred to as "isobutylene". The designation "*n*-butenes" refers to mixtures of 1, 2, and 3.

All the butenes, which do not exist as natural products, have been known for more than 100 years, but remained in very limited use and importance. Scarce availability has been for long the reason. However with the growth of cracking processes in crude oil refining and for ethylene production, butenes have been obtained as coproducts in huge quantities. Since then the complicated nature of the raw C_4 streams, containing besides the four butene isomers also the two butane isomers and multiple unsaturated C_4 hydrocarbons, has been the main barrier for specific chemical use of the butenes.

Until the mid 1980s availability of butene mixtures exceeds by far worldwide the production of the single butene isomers and their deriva-

tives. But during the 1980s the sharp price increase of hydrocarbons accelerated the development of economic separation processes, which opened for the butenes access to appropriate upgraded use.

2. Physical Properties

Butenes are colorless, flammable gases at room temperature and atmospheric pressure. They are completely miscible with alcohols, ethers, and hydrocarbons [1]. Butenes are only slightly water soluble and water is only slightly butene soluble [2]. Important physical properties are summarized in Table 1. Other thermodynamic properties and transport regulations are reported in [3–5].

3. Chemical Properties

Butenes behave as typical olefins. The main reactions are acid-catalyzed addition reactions, isomerization, and polymerization. The four isomers, including two isomers with nonterminal double bonds and one branched-chain olefin, show differences in their chemical behavior: while the 2-butenes as the lowest olefins with nonterminal double bonds generally show a minor chemical activity, isobutene, the lowest branched-chain olefin, exhibits higher reactivity, especially in addition and polymerization reactions. This difference in reactivity permits, e.g.,