

Templated Organic Synthesis

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Preface

Templated synthesis is a timely research area where molecular and supramolecular sciences meet. Moreover, it represents an emerging interface between chemistry, biology, and materials sciences. This multi-author monograph includes authoritative contributions from ten leading research laboratories involved in the development of novel template-assisted organic synthesis at this interface. It demonstrates how non-covalent and covalent templates are successfully applied to control the rate and the regio- and stereo-selectivity of molecular and supramolecular organic reactions.

In view of the broad, often indiscriminate use of the words “templates” and “tethers”, the scope of the monograph requires careful definition. According to *Webster's Dictionary*, a template is “a gauge, pattern, or mold (as a thin plate or board) used as a guide to the form of a piece being made”. The same source defines a tether as “something (as a rope or chain), by which an animal is fastened so that it can range only within a set radius”. “Mold” is perhaps the best single word to define a template in the frame of this monograph. Since clear definitions are important, it is appropriate that Chapter 1 provides definitions and describes the roles of templates in organic synthesis.

Either non-covalent or covalent templates need to be absent in the original substrate as well as in the product or, at least, should be removable after having accomplished their task. Tethers should be effective in the transition states of reactions and, by acting as a mold, control the rate and the regio- and stereochemistry. Thus the monograph focuses on reactivity rather than on static structural chemistry. It is intended to promote the use of intermolecular non-covalent interactions, besides steric discrimination, to control reactivity and selectivity. Its content was selected to provide useful template- or tether-based methodology to both expert and novice practitioners in both the molecular and supramolecular sciences.

This book is focused on organic synthesis and reactivity, naturally combining and merging molecular and supramolecular aspects. It includes one contribution (Chapter 5) on templated synthesis of biological polymers and their mimics. A profound discussion of biological templated reactions or inorganic templated synthesis (such as biomineralization) is beyond the scope of this monograph. Likewise, templates (better: scaffolds) in medicinal chemistry are excluded. Thus, carbohydrates or other central scaffolds (or cores), from which functional groups for protein recognition diverge, have sometimes been named as templates. Furthermore, the concept of reversibility, i.e., the removal of a non-covalent or covalent tether after the function of providing a “mold” has been accomplished, is important. This eliminates a variety of supramolecular self-assembly processes as well as “structural templates” such as those used to organize four-helix bundle proteins.

Although all metal-catalyzed reactions such as the Pauson–Khand reaction, Co-catalyzed cyclotrimerization, and metal-catalyzed cross-coupling reactions are templated

reactions in which the metal center acts as template, they are excluded since they have been the subject of the previous monographs, *Modern Acetylene Chemistry* and *Metal-catalyzed Cross-coupling Reactions* from the same editors and publishers as this book.

The requirement for ultimate removal of the covalent template precludes examples in which the reaction has been changed from an intermolecular to an intramolecular one by connecting the reaction partners by some kind of permanent tether which remains in the molecule. Even with this restriction, organic synthesis provides an extremely rich variety of reversible templating and tethering, as illustrated in Chapter 10; here, we would only like to mention the "silicon connection" chemistry introduced by G. Stork. Claimed template effects sometimes deserve critical mechanistic examination, and this is nicely illustrated in Chapter 9.

The increasingly successful use of intermolecular interactions by non-covalent and covalent molecular and polymeric tethers provides an important link between organic synthesis and supramolecular chemistry in this monograph. Synthesis in Nature abundantly takes advantage of non-covalent bonding and templating and, therefore, the chemistry described herein can be rightly qualified as biomimetic. Emphasis is placed upon key developments and important advances, which are illustrated with attractive and useful examples. Areas covered range from molecular imprinting (Chapter 2), molecular recognition and supramolecular self-assembly (Chapters 3 and 4), biomimetic catalysis (Chapter 6), templated synthesis of biopolymers and unnatural oligomers and polymers (Chapters 1, 5, and 8), regio- and stereoselective synthesis of covalent fullerene derivatives (Chapter 7), templated macrocyclizations (Chapters 1 and 9), to the use of temporary connections in organic synthesis (Chapter 10). Carefully chosen references guide the reader through the extensive literature. A useful feature is the inclusion of key synthetic protocols, in experimental format, in six chapters. We are hopeful that the monograph will stimulate the further development of efficient and exciting new templated synthesis methodology in the molecular and supramolecular sciences.

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DNA	deoxyribonucleic acid
DNOE	difference nuclear Overhauser effect
L-DOPA	3-hydroxy-L-tyrosine
dppe	bis(diphenylphosphino)ethane
DSC	differential scanning calorimetry
DTBP	2,6-di- <i>tert</i> -butylpyridine
EDTA	ethylenediaminetetraacetic acid
e.e.	enantiomeric excess
EM	effective molarity
Fmoc	(9 <i>H</i> -fluoren-9-ylmethoxy)carbonyl
GC	gas chromatography
GDS	guest determining step
HETCOR	{ ¹ H- ¹³ C} heteronuclear shift correlation
HMDS	hexamethyldisilazane
HMPA	hexamethylphosphoric triamide
HOBT	1-hydroxy-1 <i>H</i> -benzotriazole
HPLC	high-performance liquid chromatography; high-pressure liquid chromatography (Chapter 8)
h-RNA	homo-RNA
IDCP	iodonium collidine perchlorate
IDCT	iodonium collidine triflate
IMDA	intramolecular Diels–Alder
<i>i</i> -PrOH	isopropanol
KHMDS	potassium hexamethyldisilazide
LDA	lithium diisopropylamide
LNA	locked-nucleic acid
L-selectride	lithium tri- <i>sec</i> -butylborohydride
LUMO	lowest unoccupied molecular orbital
MC	Monte Carlo
<i>m</i> CPBA	<i>m</i> -chloroperoxybenzoic acid
MMA	methyl methacrylate
MMTR	4-methoxytriphenylmethyl
MOM	methoxymethyl
ms	molecular sieves
NBD	norbornadiene
NBS	<i>N</i> -bromosuccinimide
NIS	<i>N</i> -iodosuccinimide
NFP	<i>N</i> -formylpiperidine
NMO	<i>N</i> -methylmorpholine <i>N</i> -oxide
NMP	<i>N</i> -methyl-2-pyrrolidinone
OPNA	oxy-peptide-RNA
OTf	trifluoromethane sulfonate
PCC	pyridinium chlorochromate
PDC	pyridinium dichromate

PEG	poly(ethylene glycol)
phen	1,10-phenanthroline
Phth	phthalyl
Piv	pivaloyl
PM3	parametric method 3
PMB	<i>p</i> -methoxybenzyl
pMMA	poly(methyl methacrylate)
PNA	peptide-nucleic acid
Pr	propyl
pRNA	pyranosyl-RNA
<i>p</i> TSA	<i>p</i> -toluenesulfonic acid
py	pyridine
Ra-Ni	Raney nickel
RCM	ring-closing metathesis
RNA	ribonucleic acid
RORCM	ring-opening and ring-closing metathesis
rSNA	dimethylene sulfone RNA
rt	room temperature
RVC	reticulated vitreous carbon
SDDS	sodium dodecylsulfate
T	thymine
TBAF	tetrabutylammonium fluoride
TBDMS	<i>tert</i> -butyldimethylsilyl
TBDPS	<i>tert</i> -butyldiphenylsilyl
TBS	<i>tert</i> -butyldimethylsilyl
Tf	trifluoromethane sulfonyl
TFA	trifluoroacetic acid
THF	tetrahydrofuran
THP	3,4,5,6-tetrahydro-2 <i>H</i> -pyran-2-yl
TLC	thin layer chromatography
TMP	tetramethylpiperidine
TMS	trimethylsilyl
TMSI	trimethylsilyl iodide
Tr	triphenylmethyl
Ts	<i>p</i> -toluenesulfonyl
TsOH	<i>p</i> -toluenesulfonic acid
T. S.	transition state
TSA	<i>p</i> -toluenesulfonic acid

Contents

1 Templates in Organic Synthesis: Definitions and Roles

Sally Anderson, Harry L. Anderson

1.1	Introduction – Early Templates	1
1.2	The Definition of a Molecular Template	4
1.3	Roles of Templates	5
1.3.1	Thermodynamic and Kinetic Templates	5
1.3.2	Covalent and Non-covalent Template-Substrate Interactions	8
1.3.3	Topology of Reaction	9
1.3.3.1	Cyclization templates	10
1.3.3.2	Linear templates	14
1.3.3.3	Interweaving templates	18
1.3.4	Scavenger Templates	19
1.3.5	Negative Templates	20
1.4	Measuring Template Effects	21
1.4.1	Qualitative Detection of Template Effects	21
1.4.2	Quantification of Kinetic Template Effects in Terms of Effective Molarity, Substrate Affinity, and Maximum Rate Enhancement	22
1.4.2.1	Linear templates	23
1.4.2.2	Quantitative analysis of template effects in tethered reactions	29
1.4.2.3	Cyclization templates	29
1.5	Conclusion	33
	Appendix 1a: Equations for Figure 1-5	34
	Appendix 1b: Equations for Figure 1-10	34
	References	35

2 Templated Synthesis of Polymers – Molecularly Imprinted Materials for Recognition and Catalysis

Günter Wulff

2.1	Introduction	39
2.2	Preparation of Optically Active Linear Vinyl Polymers by Templated Synthesis	40

2.3	Exact Placement of Functional Groups on the Surfaces of Rigid Polymeric Materials Using Template Molecules	42
2.4	Molecular Imprinting in Polymeric Materials Using Template Molecules	45
2.4.1	The Principle	45
2.4.2	The Optimization of the Structure of the Polymer Network	50
2.4.3	The Role of the Binding-site Interactions	52
2.4.4	Chiroptical Properties of the Crosslinked Polymers	57
2.4.5	Chromatography Using Molecularly Imprinted Polymers	58
2.4.6	Catalysis With Molecularly Imprinted Polymers	60
2.4.7	Outlook	64
2.5	Experimental Procedures	65
2.5.1	Polymer from Scheme 2-5	65
2.5.1.1	Preparation of template monomer 7 [34]	65
2.5.1.2	Preparation of the polymer [35]	66
2.5.2	Polymer from Scheme 2-6	66
2.5.2.1	Thermally initiated polymerization [36]	66
2.5.2.2	Photochemically initiated polymerization [50]	66
2.5.3	Polymer from Scheme 2-9 [136, 137]	67
2.5.3.1	<i>N</i> -Ethyl-4-vinylbenzamide	67
2.5.3.2	<i>N</i> -Ethyl-4-vinylbenzocarbonylimide acid ethyl ester	67
2.5.3.3	<i>N,N'</i> -Diethyl-4-vinylbenzamidine (10a)	67
2.5.4	Preparation of the polymer [112]	68
	References	68

3 Templated Synthesis of Catenanes and Rotaxanes

Françisco M. Raymo, J. Fraser Stoddart

3.1	Introduction	75
3.2	Metal-Templated Syntheses	76
3.3	Hydrogen Bonding-assisted Syntheses	78
3.4	Hydrophobically Driven Syntheses	81
3.5	Aromatic Templates	84
3.6	Dialkylammonium-containing Rotaxanes	91
3.7	Conclusions	95
3.8	Experimental Procedures	95
3.8.1	[2]Catenane 4 [13]	95
3.8.2	[2]Catenane 12 [16]	96

3.8.3	[2]Catenane 43 [31]	96
3.8.4	[2]Rotaxane 51 [38]	96
3.8.5	[2]Rotaxane 56 [38]	97
3.8.6	[2]Rotaxane 68 [45]	97
	References	97

4 **Templated Synthesis of Carceplexes, Hemicarceplexes, and Capsules**

Darren Makeiff, John C. Sherman

4.1	Introduction	105
4.2	Carceplexes	106
4.2.1	The First Soluble Carceplex	106
4.2.1.1	Template ratios in the formation of an acetal-bridged carceplex	106
4.2.1.2	Formation of a charged hydrogen bonded complex	107
4.2.1.3	Mechanism of formation for an acetal-bridged carceplex	108
4.2.2	Large Carceplexes from Cyclic Arrays of Cavitands	109
4.2.2.1	Synthesis of a trimer carceplex containing three DMF molecules	110
4.2.2.2	Synthesis of a bis(carceplex)	110
4.2.3	Benzylthia-bridged Carceplex	111
4.2.4	Calix[4]arene–Cavitand Hybrid Carceplexes	112
4.2.5	Metal-bridged Carceplexes	113
4.3	Hemicarceplexes	114
4.3.1	Template Effects in the Formation of a Trismethylene-Bridged Hemicarceplex	114
4.3.2	Hemicarceplexes Containing Four Slotted Portals	115
4.4	Capsules	118
4.4.1	Capsules Composed of Cavitands Linked via Covalent Bonds or Charged Hydrogen Bonds	118
4.4.2	Rebek's "Tennis Balls", "Softballs", and "Wiffle Balls"	119
4.4.2.1	Template effects of solvent in the synthesis of "softball" dimers	122
4.4.3	Other Hydrogen Bonded Capsules	123
4.4.4	Capsules that do not Involve Hydrogen Bonding	125
4.4.5	Other Capsules	126
4.5	Conclusions	127
	References	127

5 **Template-Directed Ligation: Towards the Synthesis of Sequence Specific Polymers**

Yahaloma Gat, David G. Lynn

5.1	Introduction	133
5.2	Template-directed Ligation: Minimalist Scheme	134
5.3	Catalytic RNA and DNA	136
5.4	Molecular Recognition	137
5.4.1	Sugar Substitution	137
5.4.2	Phosphate Substitution	138
5.4.3	Neutral Acyclic Backbones	142
5.4.4	Peptide–Peptide Association	144
5.5	Ligation	146
5.6	Product Dissociation	149
5.7	Summary	152
	References	153

6 **Biomimetic Reactions Directed by Templates and Removable Tethers**

Ronald Breslow

6.1	Introduction	159
6.2	Biomimetic Reactions Using Covalently Linked Tethers and Templates	160
6.2.1	Photochemical Functionalizations by Tethered Benzophenones . . .	160
6.2.2	Free-radical Halogenations by Tethered Reagents	163
6.2.3	Free-radical Reactions Directed by Tethered Templates – the Radical Relay Mechanism	164
6.2.4	Selective Intramolecular Epoxidations Directed by Removable Tethers	171
6.3	Selective Reactions Directed by Non-covalently Linked Templates	172
6.3.1	Selective Aromatic Substitution Directed by Cyclodextrin Complexing	172
6.3.2	Photochemical Functionalizations by Complexed Benzophenones and Chlorinations Directed by Ion-paired Templates	174
6.3.3	Oxidations Directed by Metalloporphyrin and Metallosalen Templates	177
6.3.4	Other Reactions Catalyzed by Coordinated Template Catalysts . . .	183
	References	185

7 Regio- and Stereoselective Multiple Functionalization of Fullerenes

François Diederich, Roland Kessinger

7.1	Introduction	189
7.2	Anthracenes as Reversible Covalent Templates	189
7.3	Tether-directed Remote Functionalizations of C ₆₀	192
7.3.1	The First Tether-directed Multiple Functionalization of C ₆₀ : Formation of Higher Adducts with Novel Addition Patterns	193
7.3.2	The Bingel Macrocyclization	196
7.3.2.1	Formation of <i>cis</i> -2 bis-adducts	200
7.3.2.2	Formation of <i>cis</i> -3 bis-adducts	202
7.3.2.3	Other bis-functionalization patterns	204
7.3.3	Formation of Bis-adducts by Double Diels–Alder Addition of Tethered Bis(buta-1,3-diene)s to C ₆₀	205
7.3.4	Double [3+2] Cycloaddition of Tethered Vinylcarbenes	209
7.3.5	Double [3+2] Cycloadditions of Tethered Bis-azides	210
7.4	Conclusions	211
7.5	Experimental Procedures	212
7.5.1	Preparation of Hexakis-adduct 11 using DMA as a Template (Scheme 7-2) [20]	212
7.5.2	Preparation of Bis-adduct 24 by Reaction of Methano[60]fullerenecarboxylic Acid 21 with the Tether–Reactive-group Conjugate 22 (Scheme 7-4) [47]	212
7.5.3	Preparation of Tris-adduct 28 by Addition of the Anchor–Tether– Reactive-group Conjugate 27 to C ₆₀ (Scheme 7-4) [47]	213
7.5.4	Formation of Tetrakis-adduct 20 by Removal of the Tether in Hexakis-adduct 30 and Transesterification (Scheme 7-5) [2]	213
7.5.5	Bingel Macrocyclization: Synthesis of <i>cis</i> -2 Bis-adduct 42 Starting from Benzene-1,2-dimethanol (Scheme 7-8) [25]	213
7.5.6	Enantioselective Synthesis of <i>cis</i> -3 Bis adduct (¹ C)-2 by Bingel Macrocyclization of (<i>S,S</i>)-66 with C ₆₀ Followed by Transesterification (Scheme 7-11) [25]	214
	References	214

8 Template Controlled Oligomerizations

Ken S. Feldman, Ned A. Porter, Jennifer R. Allen

8.1	Introduction	219
8.2	Controlled Oligomerizations with Tethered Monomers	221

8.2.1	Templates for $n=2$ Telomers	221
8.2.2	Templates for $n=4$ Telomers	223
8.2.3	Ether-based Templates	228
8.3	Controlled Oligomerizations with a Polynorbornane-based Template	230
8.3.1	Template Design and Synthesis	231
8.3.2	Measuring the Initiator–Terminator Gap in Molecular Terms	235
8.3.3	Model Studies for Optimizing Terminator Performance	236
8.3.4	Template Controlled Oligomerization Studies	237
8.3.5	Mechanistic Studies: Probing Bimolecular Termination and Solvent Effects	239
8.4	Concluding Remarks	243
8.5	Experimental Procedures	244
8.5.1	(2 <i>R</i> ,4 <i>S</i>)- and (2 <i>S</i> ,4 <i>S</i>)-2-(5-Bromopentyl)-2-methyl-3-acryloyl- 4- <i>tert</i> -butyl oxazolidine (23)	244
8.5.2	1,2-Bis-[3,4-bis-(5-(2 <i>R</i> -methyl-3-acryloyl-4 <i>S</i> - <i>tert</i> -butyl- 2-oxazolidinyl)-pentyloxy)]-ethane (24)	244
8.5.3	Typical conditions for ACT reaction ($2.5^{\text{Å}}/80^{\text{l}}/200^{\text{Sn}}$)	245
8.5.4	Telomer assay	245
8.5.5	Synthesis of Template Diester 48	245
8.5.6	Controlled Oligomerization of Methyl Methacrylate with Template 63	246
	References	247
9	Templated or Not Templated, That is the Question: Analysis of Some Ring Closure Reactions	
	<i>Alois Fürstner</i>	
9.1	Introduction	249
9.2	Possible Pitfalls	251
9.3	Titanium-induced Intramolecular Carbonyl Coupling Reactions	253
9.4	Zinc versus Samarium Mediated Reformatsky Reactions	258
9.5	Macrocyclic Syntheses by Ring Closing Metathesis (RCM)	259
9.6	Macrocycles via Tsuji–Troost Allylation	265
9.7	Summary	268
9.8	Experimental Procedures	269
9.8.1	Synthesis of 15,16-Diphenyl-1,4,7,10-tetraoxa(10.2)[20]- paracyclophan-15-ene (14, Scheme 9-8) by an Intramolecular McMurry Coupling under “Salt Free” Conditions [19]	269

9.8.2	Synthesis of Oxacyclohexadec-11-en-2-one (22, Scheme 9-14) via RCM [32]	269
9.8.3	Palladium-catalyzed Alkenyl Epoxide Cyclization: Synthesis of Macrocyclic 39 (Scheme 9-22) [42]	269
	References	270
10	Use of the Temporary Connection in Organic Synthesis	
	<i>Liam R. Cox, Steven V. Ley</i>	
10.1	Introduction	275
10.2	Cycloaddition Reactions	277
10.2.1	[4+2] The Diels–Alder Reaction	277
10.2.1.1	Type I IMDA reactions	278
10.2.1.2	Type II IMDA reactions	295
10.2.2	Other Cycloadditions	299
10.2.2.2	An efficient synthesis of (–)-detoxinine using a tandem [4+2]/[3+2] reaction	300
10.2.2.3	Sulfur tethers in [5+2] cycloaddition	303
10.2.2.4	Azomethine ylides	304
10.3	Temporary Tethering Strategies in Radical Cyclization Reactions	307
10.3.1	Introduction	307
10.3.2	Silyl Ethers Containing the Radical Precursor	308
10.3.2.1	The α -(bromomethyl)dimethylsilyl ether in radical cyclizations	308
10.3.2.2	Other (haloalkyl)dimethylsilyl ethers	318
10.3.2.3	Dimethylsilyl ethers possessing alkenyl and aryl radical precursors	321
10.3.3	Silyl Ethers Containing the Radical Acceptor	323
10.3.4	Silyl Acetals	327
10.3.5	Acetal Tethers in Radical Cyclizations	331
10.3.6	Tether-directed Radical Cyclization Approaches to the Synthesis of C-Glycosides	333
10.4	Use of the Temporary Connection in O-Glycosylation and Nucleosidation	337
10.4.1	Tether-Directed O-Glycosylation	337
10.4.2	Directed Nucleosidations	354
10.5	Tether-mediated Nucleophile Delivery	356
10.5.1	Synthesis of (+)-Hydantocidin	356
10.5.2	Synthesis of Lincomycin	357
10.5.3	Intramolecular Allylation – Use in the Synthesis of Tricyclic β -Lactam Antibiotics	358
10.5.4	Synthesis of Corticosteroids	359
10.5.5	Synthesis of (–)- α -Kainic Acid	361