THE CHEMICAL FORMULAR

BENNETT

VOL.XV



CHEMICAL PUBLISHING COMPANY

The Chemical Formulary

A Collection of Commerical Formulas, Collected During 1968, for Making Thousands of Products in Many Fields

VOLUME XV

Editor-in-Chief
H. BENNETT, F.A.I.C.

Director, B. R. Laboratory (Formula Consultants) Miami Beach, Florida, 33140 © 1970 BY H. BENNETT

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PREFACE TO VOLUME XV

This volume contains formulas collected in 1968. The only repetitious formulas are these in the Introduction (Chapter 1) which are given for those who have no technical training and experience in compounding. These will serve as a guide for beginners and students. They should read the introduction carefully and even make a few preparations before attempting more complicated formulas that follow.

H. BENNETT

NOTE: All the formulas in Volumes I to XV (except in the introduction) are different. Thus, if you do not find what you want in this voume, you may find it in one of the others.

NOTE: This book is the result of cooperation of many chemists and engineers who have given freely of their time and knowledge. It is their business to act as consultants and to give advice on technical matters for a fee. As publishers, we do not maintain a laboratory or consulting service to compete with them. Therefore, please do not ask us for advice or opinions, but confer with a chemist.

Formulas for which patent numbers are listed can be manufactured only after obtaining a license from the patentees.

PREFACE

Chemistry, as taught in our schools and colleges, concerns chiefly synthesis, analysis, and engineering — and properly so. It is part of the right foundation for the education of the chemist.

Many a chemist entering an industry soon finds that most of the products manufactured by his concern are not synthetic or definite chemical compounds, but are mixtures, blends, or highly complex compounds of which he knows little or nothing. The literature in this field, if any, may be meager, scattered, or obsolete.

Even chemists with years of experience in one or more industries spend considerable time and effort in acquainting themselves with any new field which they may enter. Consulting chemists similarly have to solve problems brought to them from industries foreign to them. There was a definite need for an up-to-date compilation of formulae for chemical compounding and treatment. Since the fields to be covered are many and varied, an editorial board of chemists and engineers engaged in many industries was formed.

Many publications, laboratories, manufacturing firms, and individuals have been consulted to obtain the latest and best information. It is felt that the formulae given in this volume will save chemists and allied workers much time and effort.

Manufacturers and sellers of chemicals will find, in these formulae, new uses for their products. Nonchemical executives, professional men, and interested laymen will make through this volume a "speaking acquaintance" with products which they may be using, trying, or selling.

It often happens that two individuals using the same ingredients in the same formula get different results. This may be due to slight deviations in the raw materials or unfamiliarity with the intricacies of a new technique. Accordingly, repeated experiments may be

necessary to get the best results. Although many of the formulae given are being used commercially, many have been taken from the literature and may be subject to various errors and omissions. This should be taken into consideration. Wherever possible, it is advisable to consult with other chemists or technical workers regarding commercial production. This will save time and money and help avoid trouble.

A formula will seldom give exactly the results which one requires. Formulae are useful as starting points from which to work out one's own ideas. Also, formulae very often give us ideas which may help us in our specific problems. In a compilation of this kind, errors of omission, commission, and printing may occur. I shall be glad to receive any constructive criticism.

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ABBREVIATIONS

amp	ampere
amp /dm²	amperes per square decimeter
amp/sq ft	amperes per square foot
anhydr	anhydrous
avoir	
bbl	barrel
Bé	
B.P.	boiling point
°C	degrees Centigrade
œ	cubic centimeter
od	
cm	centimeter
cm3	cubic centimeter
conc	concentrated
c.p.	chemically pure
ср	centipoise
cu. ft	cubic foot
cu in	cubic inch
cwt	hundredweight
d	density
dil	
dm	decimeter
$dm^2\ \dots\dots\dots$	square decimeter
dr	dram
E	Engler
°F	degrees Fahrenheit
ffc	free from chlorine
ffpa	free from prussic acid
fl dr	fluid dram
fl oz	fluid ounce
fl pt	flash point
F.P.	freezing point
ft	
ft²	
σ	

ABBREVIATIONS

galgallon
grgrain
hlhectoliter
hrhour
ininch
kgkilogram
1liter
lbpound
liqliquid
mmeter
min minim, minute
mlmilliliter (cubic centimeter)
mmmillimeter
M.Pmelting point
NNormal
N.FNational Formulary
ozounce
pHhydrogen-ion concentration
p.p.mparts per million
ptpint
pwtpennyweight
q.sa quantity sufficient to make
qtquart
r.p.mrevolutions per minute
secsecond
spspirits
Sp. Grspecific gravity
sq. dmsquare decimeter
techtechnical
tinctincture
trtincture
Tw Twaddell
U.S.PUnited States Pharmacopeia
vvolt
viscviscosity
volvolume
wt weight

CHAPTER I

INTRODUCTION

The following introductory matter has been included at the suggestion of teachers of chemistry and home economics.

This section will enable anyone, with or without technical education or experience, to start making simple products without any complicated or expensive machinery. For commercial production, however, suitable equipment is necessary.

Chemical specialties are composed of pigments, gums, resins, solvents, oils, greases, fats, waxes, emulsifying agents, dyestuffs, perfumes, water, and chemicals of great diversity. To compound certain of these with some of the others requires definite and wellstudied procedures, any departure from which will inevitably result in failure. The steps for successful compounding are given with the formulae. Follow them rigorously. If the directions require that (a) is added to (b), carry this out literally, and do not reverse the order. The preparation of an emulsion is often quite as tricky as the making of mayonnaise. In making mayonnaise, you add the oil to the egg, slowly, with constant and even stirring. If you do it correctly, you get mayonnaise. If you depart from any of these details: If you add the egg to the oil, or pour the oil in too quickly, or fail to stir regularly, the result is a complete disappointment. The same disappointment may be expected if the prescribed procedure of any other formulation is violated.

The point next in importance is the scrupulous use of the proper ingredients. Substitutions are sure to result in inferior quality, if not in complete failure. Use what the formula calls for. If a cheaper