# Histochemistry

**VOLUME THREE** 

Edited by
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Fourth Edition

# **Histochemistry**Theoretical and Applied

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Volume 3 Enzyme histochemistry

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# **Histochemistry** Volume 3

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#### **Preface**

This volume, child of many prayers (ours) and of many talents (others), completes for 'Histochemistry, Theoretical & Applied' a span of almost 40 years.

When the first volume of the work appeared in 1953, Enzyme Histochemistry was the rising star of a resurgent discipline, resurgent only after seven decades in a wilderness made for it by Johann Friedrich Miescher's 'Guild of Dyers', namely the histological fraternity.

For at least the next two decades it formed the advancing front of the whole discipline, and particularly for the field of Diagnostic Pathology. Here it promised, and made, considerable advances in the search for its 'Philosopher's Stone', distinction between a benign cell and it's malignant counterpart.

By the early 1980s the relative importance of Enzyme Histochemistry had begun to decline, especially with respect to the younger disciplines of Immunocytochemistry and Hybridocytochemistry. Nevertheless it continues to form a significant and far from static branch of the whole subject.

For example, there have been enormous advances in quantification of enzyme activity in situ, and a far greater understanding of the fundamental mechanisms underlying many histochemical reactions. The size and complexity of the text provided in this volume is a direct response to the creation, since publication of the previous relevant volume in 1972, of a mass of new technology. This accompanies an ever widening range of histochemically demonstrable enzymes and enzyme groups.

We feel no need to apologise here for maintaining the principle enshrined in previous editions, of complete and up-to-date biochemical and histochemical coverage for each enzyme described.

We wish to acknowledge our great indebtedness to all those who have contributed to the final volume of the 4th Edition. We trust that they will feel that their time of waiting, whether patient or impatient, has been rewarded, and that this volume will stimulate and foster both the understanding and the application of the valuable techniques of Enzyme Histochemistry.

A.G.E.P., P.J.S. Exeter and Dundee 1991

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## **Enzymes reviewed**

The localisation and function of the following enzymes are described in this Volume.

Enzyme	Contributors	T	
Acetyl-CoA carboxylase	PJS	Enzyme	Contributors
Acetylcholinesterase	- 6 5	Aspartate aminotransferase	PRL
	CO, PRL	Aspartate carbamoyl	PRL
β-N-Acetylgalactosaminidase N-Acetyl-β-glucosaminidase	RG, ZL, PJS	transferase	# <u>_</u>
	RG, ZL, PJS	ATPases	JAF
β-N-Acetylhexosaminidase Acetylserotonin	RG, ZL, PJS	Benzaldehyde dehydrogenase	PJS
methyltransferase	PRL, PJS	Benzylamine oxidase	PJS, PRL
	MD AM DIG	Betaine-aldehyde	FW
Acid phosphatase	MB, AV, PJS	dehydrogenase	
Acrosin	PJS	Calcidiol 1-monooxygenase	PJS
Acyl-CoA oxidase	PJS	Carbamoyl-phosphate synthase	PJS
Acylglycerol	PRL	Carbonic anhydrase	CO, PJS
palmitoyltransferase		Carbonyl reductase	PJS
Adenylate cyclase	PJS	Carnitine acetyltransferase	PRL
Alcohol dehydrogenase	PJS, PRL	Catalase	WD, SA, HDF
Alcohol oxidase	PJS	Catechol methyltransferase	PJS
Aldehyde dehydrogenases	PJS, EW	Catechol oxidase	PJS
Aldehyde oxidase	PJS	Cathepsin B	ZL, RG, PJS
Aldehyde reductase	PJS	Cathepsin D	ZL, RG, PJS
Aldose reductase	PJS	Cathepsin G	ZL, RG, PJS
Alkaline phosphatase	MB, AV	Cathepsin H	ZL, RG, PJS
D-Amino-acid oxidase	PJS, JMR,	Cathepsin L	ZL, RG, PJS
	MJK, PRL	Cellulase	RG, ZL, PJS
4-Aminobutyrate	PRL	Cellulose 1,4-β-cellobiosidase	RG, ZL, PJS
aminotransferase		Chitinase	RG, ZL, PJS
Aminopeptidase A	ZL, RG, PJS	Cholesterol esterase	CO
Amylases ( $\alpha$ - and $\beta$ -)	RG, ZL, PJS	Cholesterol oxidase	PJS, JMR,
Apyrase	PJS	- I office of the control of the con	MJK
α-L-Arabinofuranosidase	RG, ZL, PJS	Choline acetyltransferase	PRL, PJS
Arginine aminopeptidase	ZL, RG, PJS	Choline dehydrogenase	FW
Aromatic L-amino-acid	PJS	Cholinesterases	CO, PRL
decarboxylase		Chymase	ZL, RG, PJS
Arylamine acetyltransferase	PRL	Collagenases	ZL, RG, PJS ZL, RG, PJS
Arylsulphatase	PJS	Creatine kinase	PRL
, ,		Creatific Killase	FKL

#### X ENZYMES REVIEWED

Enzyme	Contributors	Enzyme	Contributors
Cystathionine γ-lyase	PJS	Glucosylceramidase	RG, ZL, PJS
Cysteine lyase	PJS	β-Glucuronidase	RG, ZL, PJS
Cytochrome c oxidase	JMR, MJK	Glucuronosyltransferase	PJS
Cytochrome P450	PJS	Glutamate decarboxylase	PJS
Cytochrome P450-dependent	PJS	Glutamate dehydrogenases	FW, PJS
oxygenase		γ-Glutamyltransferase	ZL, RG, PJS
Cytosol aminopeptidase	ZL, RG, PJS	Glutathione peroxidase	PJS
α-Dextrin	RG, ZL, PJS	Glutathione reductase	PJS
endo-1,6-α-glucosidase		Glutathione transferase	PJS
Diamine oxidase	PJS	Glyceraldehyde-3-phosphate	AEFHM, PJS
Dihydroorotate dehydrogenase	PJS	dehydrogenase	
Dihydrofolate reductase	PJS, FW	Glycerol-3-phosphate	PRL
Dihydrolipoamide	PJS	acyltransferase	
dehydrogenase		Glycerol-3-phosphate	AEFHM, PJS
Dipeptidase	ZL, RG, PJS	dehydrogenase	
Dipeptidyl peptidase I	ZL, RG, PJS	Glycerol-3-phosphate	FW, PJS
Dipeptidyl peptidase II	ZL, RG, PJS	dehydrogenase (NAD <sup>+</sup> )	
Dipeptidyl peptidase IV	ZL, RG, PJS	Glycogen synthase	PRL
Dipeptidylcarboxypeptidase I	ZL, RG, PJS	Glycolate dehydrogenase	PJS
Dopamine β-monoxygenase	PJS	Glycolate oxidase	PJS
Endo-1,3-β-glucanase	RG, ZL, PJS	Glycosylceraminidase	RG, ZL, PJS
Enolase	PJS	Guanylate cyclase	PJS
Enoyl-CoA hydratase	PJS	L-Gulonolactone oxidase	PJS
Enteropeptidase	ZL, RG, PJS	Hetero-β-galactosidase	RG, ZL, PJS
Eosinophil peroxidase	PJS	Hexokinase	PRL
Ferredoxin-NADP <sup>+</sup> reductase	PJS	Histidine decarboxylase	PJS
β-Fructofuranosidase	RG, ZL, PJS	Hyaluronnidase	RG, ZL, PJS
Fructose-bisphosphate aldolase	AEFHM, PJS	L-2-Hydroxy-acid oxidase	JMR, MJK
Fructose-bisphosphatase	MB, AV, PJS,	3-Hydroxyacyl-CoA	PJS
β-D-Fucosidase	RG, ZL, PJS	dehydrogenase	DIC
α-L-Fucosidase	RG, ZL, PJS	3-Hydroxyanthranilate	PJS
Galactose oxidase	PJS, JMR,	3,4-dioxygenase	EW
a	MJK	3-Hydroxybutyrate	FW
Galactosidases ( $\alpha$ and $\beta$ -)	RG, ZL, PJS	dehydrogenase	FW, PJS
Galactosylceramide	PJS	9-Hydroxyprostaglandin	T W, 135
sulphotransferase	DIC DDI	dehydrogenase	PEH, PJS
Galactosyltransferase	PJS, PRL	3α-Hydroxycholonate dehydrogenase	1111, 135
Glucan 1,4-α-glucosidase	RG, ZL, PJS	15-Hydroxyprostaglandin	FW, PJS
1,4-α-Glucan branching enzyme		dehydrogenase (NAD <sup>+</sup> )	1 11, 133
Glucokinase	PRL	3α-Hydroxysteroid	PEH
Glucose dehydrogenase	PJS DIS IMP	dehydrogenase	1 1211
Glucose oxidase	PJS, JMR, MJK	11β-Hydroxysteroid	PEH
Glugges 6 phosphotose	MB, AV	dehydrogenase	LII
Glucose-6-phosphatase	PJS	$3\beta$ -Hydroxy- $\Delta$ <sup>5</sup> -steroid	PEH
Glucose-6-phosphate dehydrogenase	133	dehydrogenase	
Glucose-6-phosphate isomerase	PJS	L-Iditol dehydrogenase	FW, PJS
Glucosidases ( $\alpha$ -and $\beta$ -)	RG, ZL, PJS	L-Iduronidase	RG, ZL, PJS
Cracosiaases (a-ana p-)	10, 22, 130	2000 VIOLENCE CONT. TOTAL TOTA	,, 130

Enzyme	Contributors	Enzyme	Contributors
IMP dehydrogenase	PJS	Phosphoenolpyruvate	PJS
Indoleamine 2,3-dioxygenase	PJS	carboxykinase	
Inorganic pyrophosphatase	MB, AV	6-Phosphofructokinase	PRL
Isocitrate dehydrogenases	AEFHM	Phosphoglucomutase	PRL, PJS
Ketohexokinase	PJS	Phosphogluconate	ES, CJFVN,
Lactase	RG, ZL, PJS	dehydrogenase	PJS
Lactate dehydrogenase	AEFHM,	Phosphorylase	PRL
	PRL, PJS	Plasmin	ZL, RG, PJS
Leucocyte elastase	ZL, RG, PJS	Plasminogen activator	ZL, RG, PJS
Lysophospholipase	CO	Platelet peroxidase	PJS
Lysozyme	RG, ZL, PJS	Polyphenol oxidases	PJS, JMR,
Lysyl oxidase	PJS		MJK
Malate dehydrogenases	AEFHM, PJS	Prolyl hydroxylase	PJS
Malate synthase	PJS	Prostacyclin synthase	PJS
Mannosidase	RG, ZL, PJS	Prostaglandin synthase	PJS
Membrane	ZL, RG, PJS	Protochlorophyllide reductase	PJS
metallo-endopeptidase		Protein-disulphide reductase	PJS
Microsomal aminopeptidase	ZL, RG, PJS	Purine-nucleoside	PRL
Monoamine oxidase	PJS, JMR,	phosphorylase	
	MJK, PRL	Pyruvate dehydrogenase	PJS
NAD(P)H oxidases	PJS, JMR,	complex	
	MJK	Pyruvate kinase	PRL, PJS
NAD <sup>+</sup> kinase	PJS	Renin	ZL, RG, PJS
NADH dehydrogenases	PJS	Retinol dehydrogenase	PJS
NADH-ferrihemoprotein	PJS	Ribose-5-phosphate isomerase	PJS
reductase		Ribulose bisphosphate	PJS
NADP(H) phosphohydrolases	MB, AV, PJS	carboxylase	
NADPH dehydrogenases	PJS	Secondary alcohol	PJS
Nicotinamide-nucleotide	PRL	dehydrogenase	
adenyltransferase		Serine-pyruvate	PJS
Nicotinate-nucleotide	PJS	aminotransferase	
pyrophosphorylase		Sialidase	RG, ZL, PJS
Nitrate reductase	PJS	Submandibular protease A	ZL, RG, PJS
Non-specific esterases	CO, PJS, PRL	Succinate dehydrogenases	PJS
Nucleoside-diphosphatase	PJS, MB, AV	Succinate-semialdehyde	PJS
Nucleoside-triphosphatase	MB, AV	dehydrogenase	
3'-Nucleotidase	MB, AV	Sucrose-α-glucosidase	RG, ZL, PJS
5'-Nucleotidase	MB, AV	Sulphinoalanine decarboxylase	PJS
Oligo-1,6-glucosidase	RG, ZL, PJS	Superoxide dismutase	PJS
Ornithine aminotransferase	PJS	Terminal deoxynucleotidyl	PJS
Ornithine carbamoyltransferase	PRL	transferase	
Ornithine decarboxylase	PJS	Tetrametaphosphatase	PJS
Peroxidases	WD, SA, HDF	Thiamine monophosphatase	MB, AV
Phenolases	PJS	Thiamine pyrophosphatase	MB, AV, PJS
Phenylethanolamine	PJS	Thioglucosidase	RG, ZL, PJS
N-methyltransferase		Thioredoxin reductase	PJS
Phospho <i>enol</i> pyruvate	PJS	Thiosulphate sulphurtransferase	PJS, PRL
carboxylase		Tissue kallikrein	ZL, RG, PJS

#### XII ENZYMES REVIEWED

Enzyme	Contributors	Enzyme	Contributors
Tissue-endopeptidase	ZL, RG, PJS	Tryptophan 2,3-dioxygenase	PJS
degrading		Tyrosine 3-monoxygenase	PJS
collagenase-synthetic		UDPglucose dehydrogenase	PJS
substrate		Urate oxidase	PJS, JMR,
Transglutaminase	PRL		MJK, PRL
Triglyceride lipase	CO	Xanthine dehydrogenase	ES
Trimetaphosphatase	MB, AV, PJS	Xanthine oxidase	PJS, JMR,
Tryptase	ZL, RG, PJS		MJK

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### 21. Principles of oxidoreductase histochemistry

P. J. Stoward F. P. Altman E. Seidler

The commonest method of localizing oxidoreductases histochemically is based on one simple principle — the reduction, by the electrons released in the reaction catalysed by the enzyme, of a colourless, soluble tetrazolium salt to an intensely-coloured, highly insoluble derivative called a formazan. This is an oversimplification of what actually occurs. In 1972, when the previous edition was published, it was not widely appreciated how serious this oversimplification was. Several techniques recommended at that time have since been shown to be grossly inadequate and to give false results. For example, it was not realised that soluble enzymes diffused very rapidly out of sections into aqueous incubation media; consequently, the amount and distribution of the final reaction product (i.e. staining) subsequently seen in sections bore little relation to the original activity of the enzyme in situ. Unfortunately, many workers have continued to use these techniques despite the wealth of evidence of their shortcomings published in the two decades since 1970.

This chapter is, therefore, devoted to the principles on which the modern histochemistry of oxidoreductases is based. It is particularly concerned with the special factors that should be taken into account in order to obtain a reliable, accurate and specific localization. The most important concerns are; the reaction mechanisms underlying oxido-reductions, the characteristics of tetrazolium salts, intermediate electron acceptors and cofactors; and the suppression of the leakage of enzymes from unfixed cells and tissues during incubation. The account that follows is based partly on Chapter 20 of the last edition, but to a greater extent on texts by Altman (1972,

1976a), Wohlrab et al, (1979), Van Noorden (1984) and Stoward (1991).

### OXIDOREDUCTASE-CATALYSED REACTIONS

Oxidoreductases (EC class 1) catalyse reactions between *two* substrates in which the principle substrate (S.H<sub>2</sub>), called the donor, is oxidized by the other, the acceptor (A). Such reactions are generally written in the form

$$S.H_2 + A \longrightarrow S + AH_2$$
  
Donor Acceptor

The donor is not always dehydrogenated as shown in this equation, but when it is, the catalysing enzyme is known as a dehydrogenase. However, two electrons are always removed from the donor. The electrons are transferred to, and thus accepted by, either the enzyme itself or separate molecules such as coenzymes or special proteins within the cell. The vast majority of dehydrogenases fall into two groups, *flavincontaining dehydrogenases* and *coenzymelinked* (or pyridine-linked) dehydrogenases.

The flavin-containing dehydrogenases contain electron acceptors bound to the enzyme molecule itself. These acceptors are known as prosthetic groups, and are either a flavin adenine nucleotide (FAD) or a flavin mononucleotide. Succinate dehydrogenase belongs to this group.

The second group of dehydrogenases have no integrally-bound acceptors, and function only in the presence of separate electron acceptors or carriers known as coenzymes, usually NAD<sup>+</sup> or NADP<sup>+</sup>.

Apart from these two groups, there is a small group of oxidoreductases which require other

acceptors, of which the most important are hydrogen peroxide (the peroxidases) and molecular oxygen (the oxidases).

The initial products of any biochemical reaction catalysed by a dehydrogenase are either a reduced flavin (for example, FADH2 in the case of succinate dehydrogenase) or a reduced coenzyme (for example, NADH in the case of NAD+-dependent lactate dehydrogenase) plus, in both instances, the appropriate oxidized substrate. The reduced acceptors normally must be immediately re-oxidized to allow the enzymes concerned to go on functioning. Cells and tissues have a number of pathways to achieve this re-oxidation, the most well-defined ones being the respiratory chain in mitochondria and the cytochrome P<sub>450</sub> system. They are discussed in detail later in connection with their relationships to the reduction of tetrazolium salts.

Thus, oxidation and reduction reactions catalysed by oxidoreductases do not necessarily involve oxygen and hydrogen. The fundamental event is the exchange of electrons; the passage or transfer of hydrogen or oxygen atoms is incidental. Nonetheless, it is often more convenient to visualize oxido-reduction reactions as 'hydrogen transport' rather than as 'electron transfer'.

## PRINCIPLES UNDERLYING THE HISTOCHEMICAL LOCALIZATION OF DEHYDROGENASES

In most techniques, cell smears or sections are incubated in a suitably buffered medium containing:

- 1. the substrate of the enzyme to be localized
- 2. any necessary cofactor (NAD<sup>+</sup> or NADP<sup>+</sup>)
- 3. an intermediate electron acceptor
- 4. a tetrazolium salt or some other final electron acceptor.

In addition, if the dehydrogenase is soluble, an inert tissue stabilizer, such as suitable grade of polyvinyl alcohol (PVA), should be dissolved in the medium to prevent the leakage of the enzyme from the section into the medium during incubation.

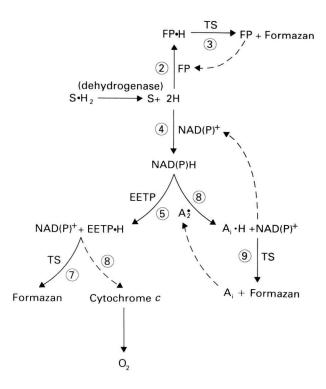
The dehydrogenase being localized oxidises the substrate  $(S \cdot H_2)$  in the medium to a reduced

intermediate reaction product (S), releasing electrons and hydrogen which pass either to the flavin component (FP) of the enzyme molecule or, if a coenzyme-linked dehydrogenase is being demonstrated, to exogenous oxidised coenzymes (NAD<sup>+</sup> or NADP<sup>+</sup>) in the substrate medium. These are shown in Fig. 21.1 as reactions 2 and 4.

When a flavin-linked dehydrogenase, such as succinate dehydrogenase, is being localized, the reduced flavoprotein (FADH<sub>2</sub> or FADH) usually has a sufficiently high redox potential to reduce the tetrazolium salt in the medium to form an insoluble formazan, which then becomes deposited on the subcellular site of the enzyme.

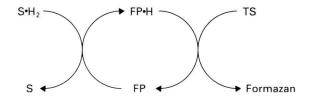
In contrast, the reaction paths followed during the localization of a coenzyme-linked dehydrogenase are more complex. The reduced coenzymes (NADH or NADPH) formed in reaction 4 (Fig. 21.1) are, unless steps are taken to prevent it, almost immediately oxidized by the endogenous electron-transfer protein (EETP) systems present in most cells. The reducing equivalents thus formed are either passed onto endogenous acceptors, such as cytochrome c (reaction 7), or onto exogenous acceptors such as molecular oxygen (reaction 8), or they reduce the tetrazolium salt in the substrate medium. If reaction 5 occurs to a significant extent, the insoluble formazan arising from the reduction of the tetrazolium salt is deposited on the subcellular site of the endogenous electron-transfer system, and not on the site of the enzyme. Thus, the resulting localization is 'false'.

In order to achieve a specific localization of a coenzyme-linked dehydrogenase, the endogenous electron-transfer systems of the tissue under study must be bypassed. This is achieved by adding an intermediate electron acceptor (A<sub>i</sub>), such as phenazine methosulphate, to the substrate medium. The acceptor is reduced by the reducing equivalents (i.e. the electrons and protons released by the enzyme) to form a reduced intermediate which, in turn, immediately reduces the tetrazolium salt in the substrate medium. This results, in theory at least, in the deposition of an insoluble formazan on the subcellular site of the initial substrate-specific dehydrogenase (reactions 8 and 9 in Fig. 21.1). In this situation, the



Alternatively the above reaction sequences may be written in the form normally drawn in biochemistry textbooks for coupled oxidation-reduction reactions as follows

For flavin-containing enzymes:



For pyridine-linked dehydrogenases:

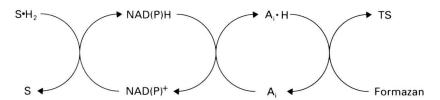


Fig. 21.1 Reaction sequences involved in the localization of coenzyme-linked and flavoprotein-linked dehydrogenases.