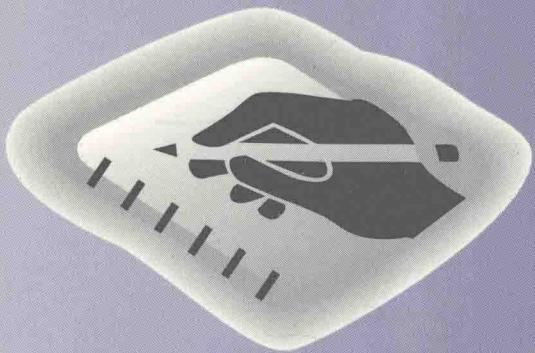
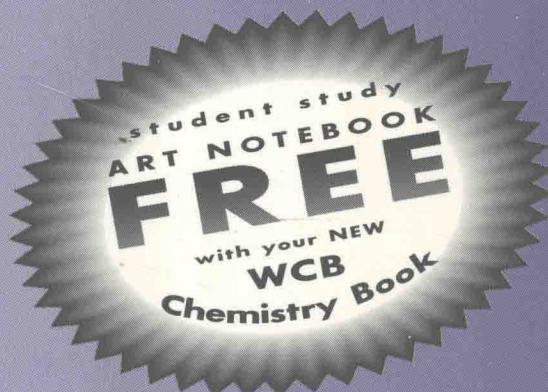


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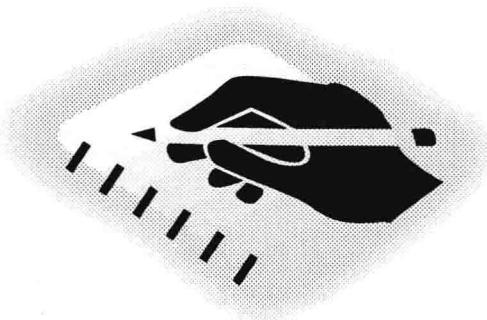


***Inorganic, Organic, &
Biological Chemistry***



Caret Denniston Topping

s t u d e n t s t u d y
A R T N O T E B O O K



***Inorganic, Organic, &
Biological Chemistry***

**Robert L. Caret / Katherine J. Denniston / Joseph J. Topping
Towson State University**



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TO INSTRUCTORS AND STUDENTS

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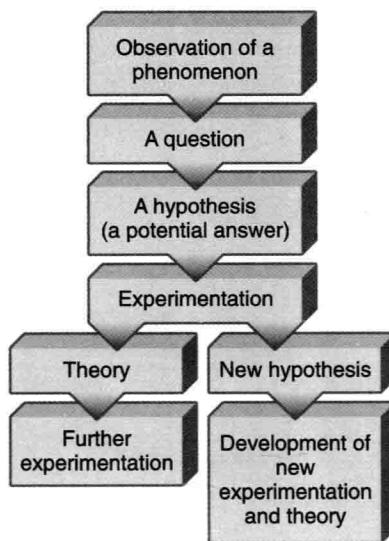
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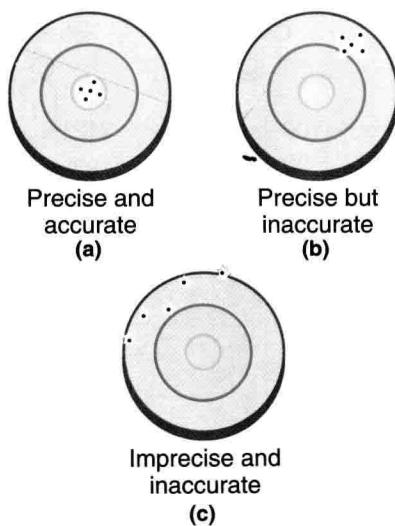
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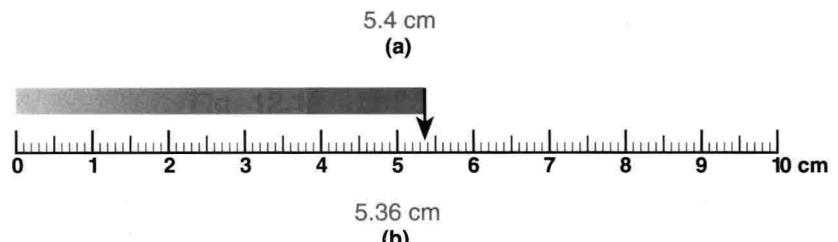
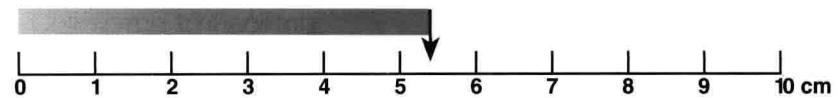
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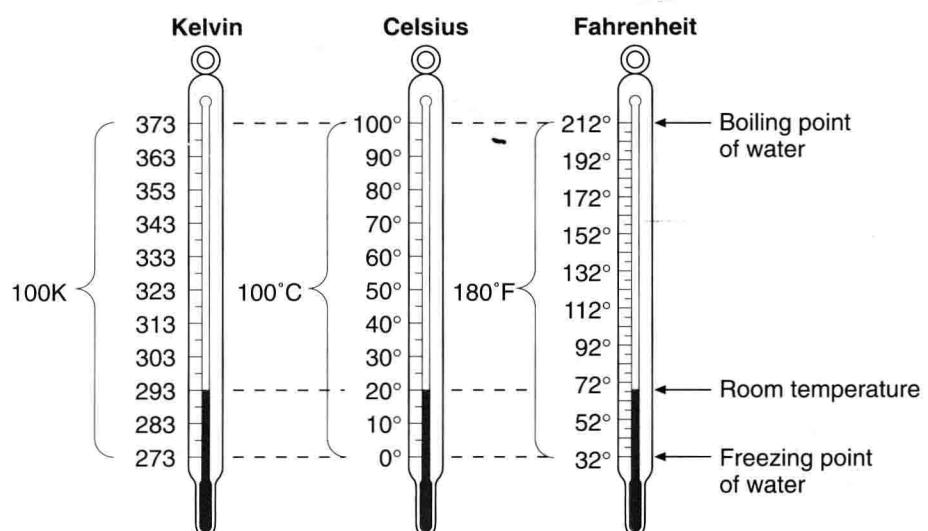
The Scientific Method



Precision and Accuracy



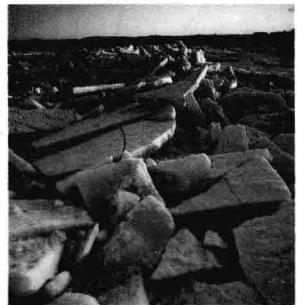
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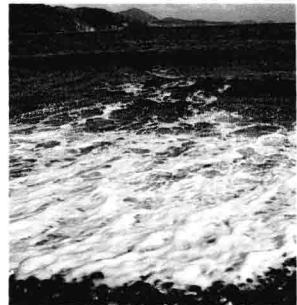
Three Common Units of Temperature



Density—A Unique Property of a Material



(a)

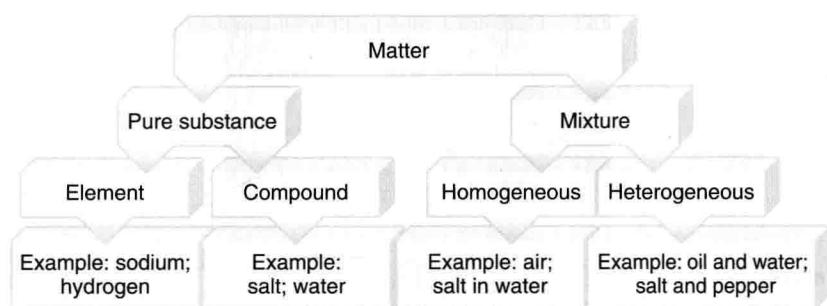


(b)

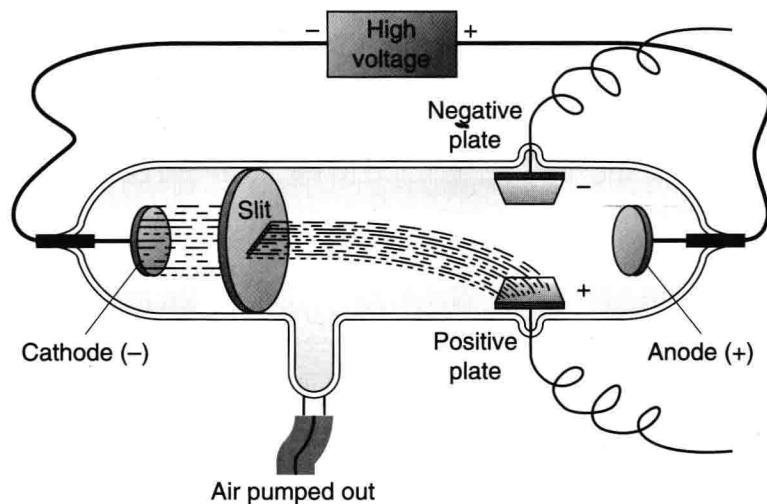


(c)

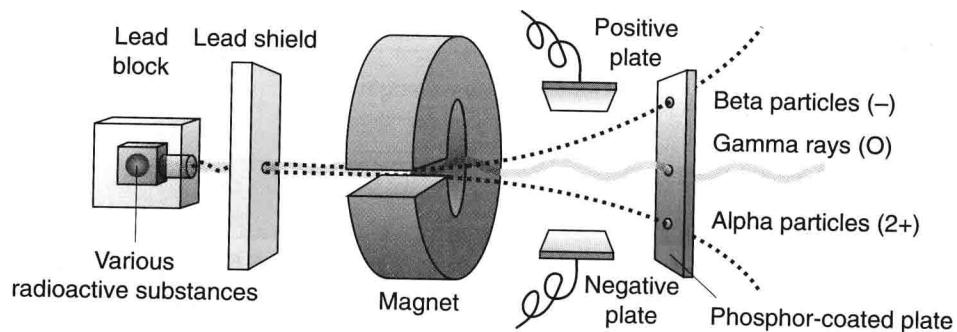
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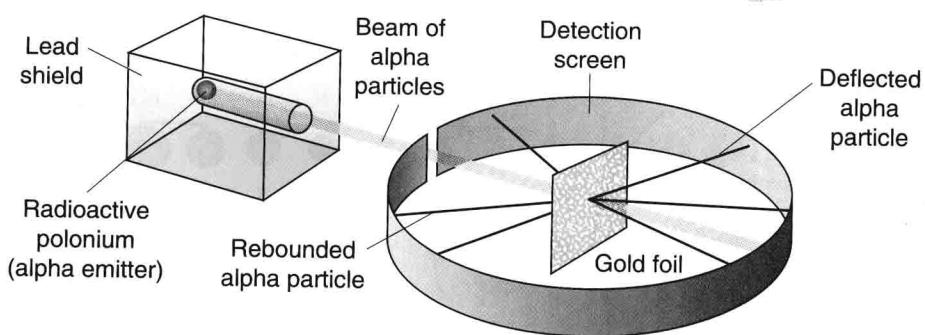
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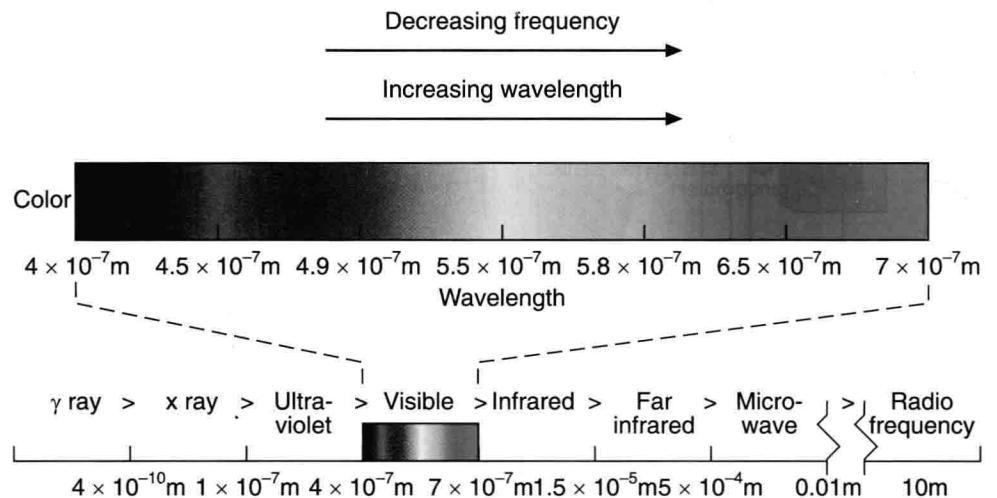
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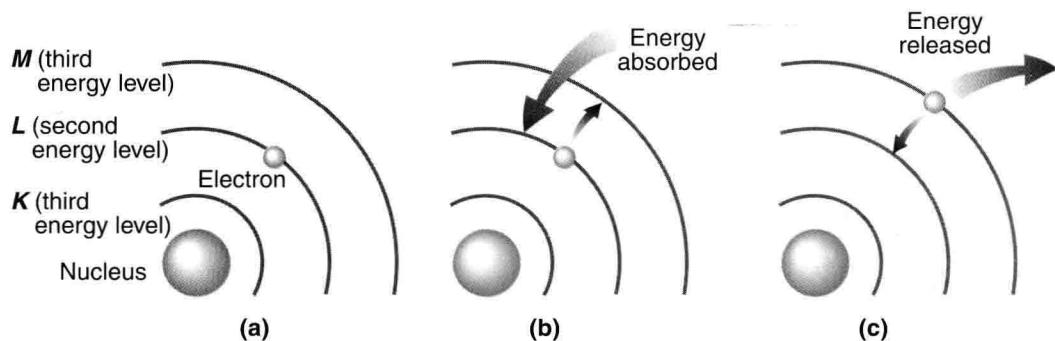
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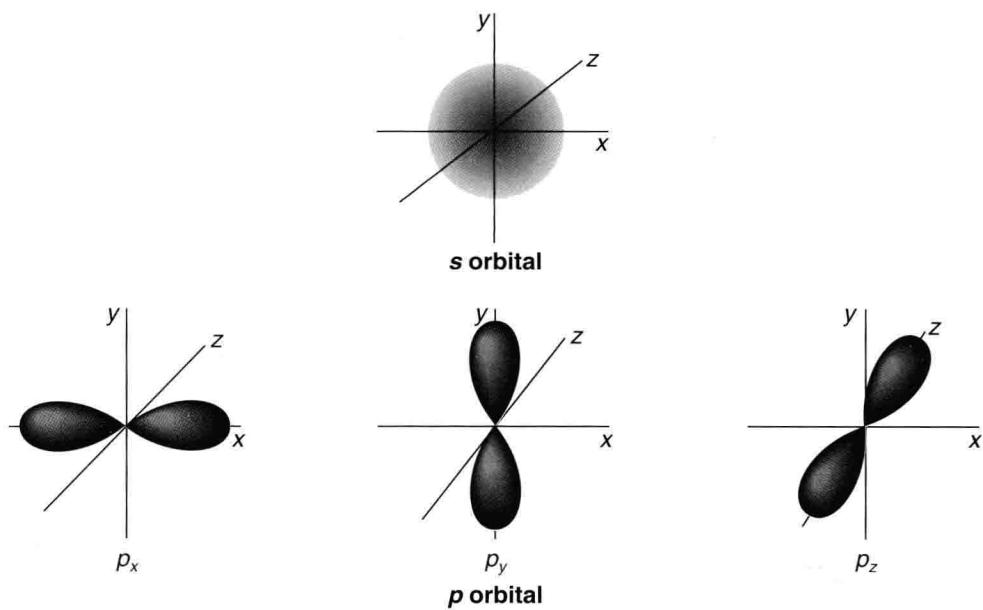
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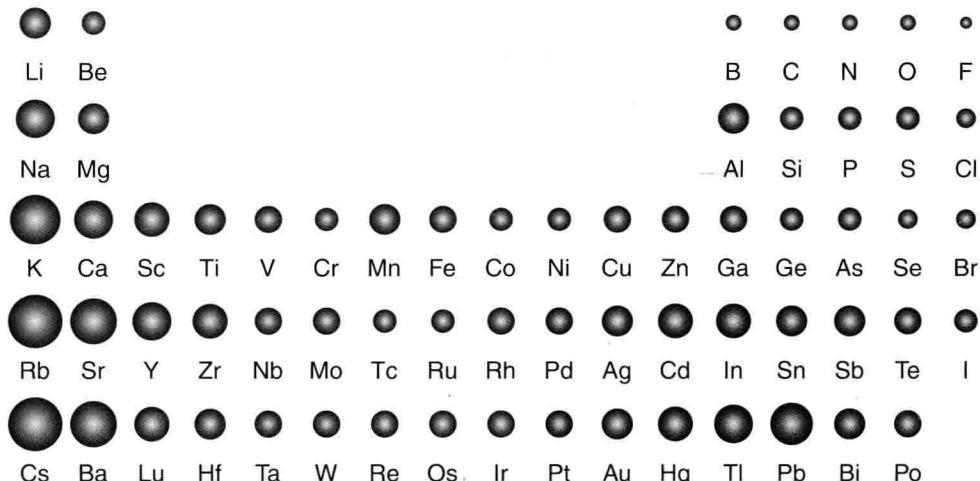
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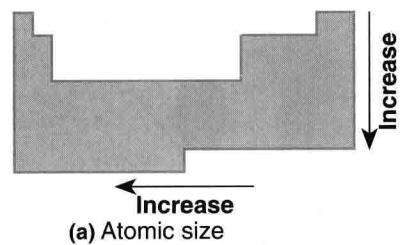
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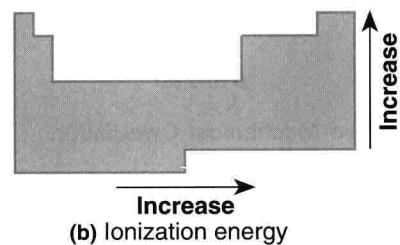
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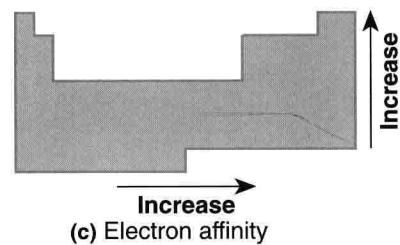
Variation in Size of Atoms



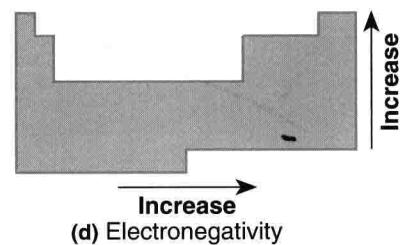
(a) Atomic size



(b) Ionization energy

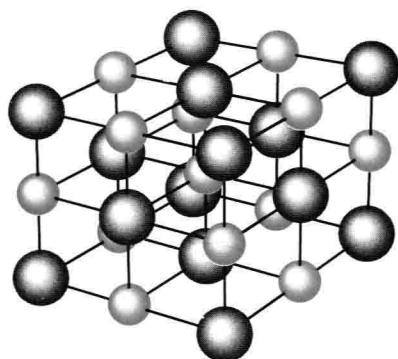


(c) Electron affinity



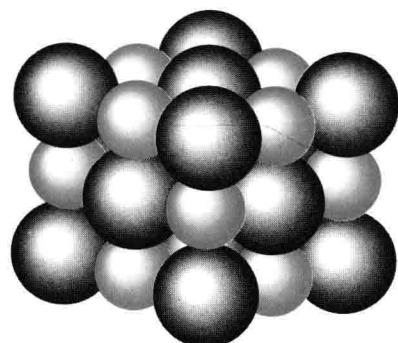
(d) Electronegativity

**Periodic Trends for Atomic Size, Ionization Energy,
Electron Affinity, and Electronegativity**



Cl^- Na^+

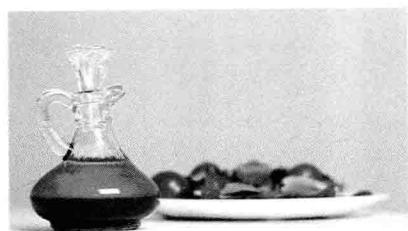
(a)



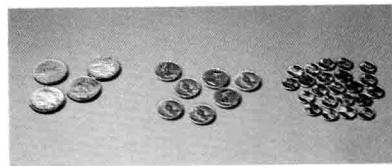
Cl^- Na^+

(b)

Arrangement of Ions in a Crystal of NaCl



**Interaction of Polar Water Molecules and Nonpolar
“Oil” Molecules**



One Mole of Atoms