



# Ullmann's Encyclopedia of Industrial Chemistry

Sixth, Completely Revised Edition

Volume 29

Poly (Vinyl Chloride)  
to  
Process Control  
Engineering

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# Ullmann's Encyclopedia of Industrial Chemistry

Volume 29

# Ullmann's Encyclopedia of Industrial Chemistry

Volumes 1 - 39: Alphabetically Arranged Articles

Volume 40: Index

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## Symbols and Units

Symbols and units agree with SI standards (for conversion factors see page IX). The following list gives the most important symbols used in the encyclopedia. Articles with many specific units and symbols have a similar list as front matter.

Symbol	Unit	Physical Quantity
$a_B$		activity of substance B
$A_r$		relative atomic mass (atomic weight)
$A$	$\text{m}^2$	area
$c_B$	$\text{mol}/\text{m}^3, \text{mol}/\text{L} (\text{M})$	concentration of substance B
$C$	$\text{C}/\text{V}$	electric capacity
$c_p, c_v$	$\text{J kg}^{-1}\text{K}^{-1}$	specific heat capacity
$d$	$\text{cm}, \text{m}$	diameter
$d$		relative density ( $\rho/\rho_{\text{water}}$ )
$D$	$\text{m}^2/\text{s}$	diffusion coefficient
$D$	$\text{Gy} (= \text{J/kg})$	absorbed dose
$e$	$\text{C}$	elementary charge
$E$	$\text{J}$	energy
$E$	$\text{V}/\text{m}$	electric field strength
$E$	$\text{V}$	electromotive force
$E_A$	$\text{J}$	activation energy
$f$		activity coefficient
$F$	$\text{C/mol}$	Faraday constant
$F$	$\text{N}$	force
$g$	$\text{m}/\text{s}^2$	acceleration due to gravity
$G$	$\text{J}$	Gibbs free energy
$h$	$\text{m}$	height
$\hbar$	$\text{W} \cdot \text{s}^2$	Planck constant
$H$	$\text{J}$	enthalpy
$I$	$\text{A}$	electric current
$I$	$\text{cd}$	luminous intensity
$k$	(variable)	rate constant of a chemical reaction
$k$	$\text{J/K}$	Boltzmann constant
$K$	(variable)	equilibrium constant
$l$	$\text{m}$	length
$m$	$\text{g}, \text{kg}, \text{t}$	mass
$M_r$		relative molecular mass (molecular weight)
$n_D^{20}$		refractive index (sodium D-line, 20 °C)
$n$	$\text{mol}$	amount of substance
$N_A$	$\text{mol}^{-1}$	Avogadro constant ( $6.023 \times 10^{23} \text{ mol}^{-1}$ )
$p$	$\text{Pa}, \text{bar}^*$	pressure
$Q$	$\text{J}$	quantity of heat
$r$	$\text{m}$	radius
$R$	$\text{J K}^{-1}\text{mol}^{-1}$	gas constant
$R$	$\Omega$	electric resistance
$S$	$\text{J/K}$	entropy
$t$	$\text{s}, \text{min}, \text{h}, \text{d}, \text{month}, \text{a}$	time

## Symbols and Units (Continued from p. VII)

Symbol	Unit	Physical Quantity
$t$	°C	temperature
$T$	K	absolute temperature
$u$	m/s	velocity
$U$	V	electric potential
$U$	J	internal energy
$V$	$\text{m}^3, \text{L}, \text{mL}, \mu\text{L}$	volume
$w$		mass fraction
$W$	J	work
$x_B$		mole fraction of substance B
$Z$		proton number, atomic number
$\alpha$		cubic expansion coefficient
$\alpha$	$\text{W m}^{-2}\text{K}^{-1}$	heat-transfer coefficient (heat-transfer number)
$\alpha$		degree of dissociation of electrolyte
$[\alpha]$	$10^{-2}\text{deg cm}^2\text{g}^{-1}$	specific rotation
$\eta$	$\text{Pa} \cdot \text{s}$	dynamic viscosity
$\theta$	°C	temperature
$\kappa$		$c_p/c_v$
$\lambda$	$\text{W m}^{-1}\text{K}^{-1}$	thermal conductivity
$\lambda$	nm, m	wavelength
$\mu$		chemical potential
$\nu$	Hz, s <sup>-1</sup>	frequency
$\nu$	$\text{m}^2/\text{s}$	kinematic viscosity ( $\eta/\rho$ )
$\pi$	Pa	osmotic pressure
$\varrho$	$\text{g/cm}^3$	density
$\sigma$	N/m	surface tension
$\tau$	Pa (N/m <sup>2</sup> )	shear stress
$\varphi$		volume fraction
$\chi$	$\text{Pa}^{-1} (\text{m}^2/\text{N})$	compressibility

\* The official unit of pressure is the pascal (Pa).

## Conversion Factors

SI unit	Non-SI unit	From SI to non-SI multiply by
<i>Mass</i>		
kg	pound (avoirdupois)	2.205
kg	ton (long)	$9.842 \times 10^{-4}$
kg	ton (short)	$1.102 \times 10^{-3}$
<i>Volume</i>		
m <sup>3</sup>	cubic inch	$6.102 \times 10^4$
m <sup>3</sup>	cubic foot	35.315
m <sup>3</sup>	gallon (U.S., liquid)	$2.642 \times 10^2$
m <sup>3</sup>	gallon (Imperial)	$2.200 \times 10^2$
<i>Temperature</i>		
°C	°F	$^{\circ}\text{C} \times 1.8 + 32$
<i>Force</i>		
N	dyne	$1.0 \times 10^5$
<i>Energy, Work</i>		
J	Btu (int.)	$9.480 \times 10^{-4}$
J	cal (int.)	$2.389 \times 10^{-1}$
J	eV	$6.242 \times 10^{18}$
J	erg	$1.0 \times 10^7$
J	kW · h	$2.778 \times 10^{-7}$
J	kp · m	$1.020 \times 10^{-1}$
<i>Pressure</i>		
MPa	at	10.20
MPa	atm	9.869
MPa	bar	10
kPa	mbar	10
kPa	mm Hg	7.502
kPa	psi	0.145
kPa	torr	7.502

## Powers of Ten

E (exa)	$10^{18}$	d (deci)	$10^{-1}$
P (peta)	$10^{15}$	c (centi)	$10^{-2}$
T (tera)	$10^{12}$	m (milli)	$10^{-3}$
G (giga)	$10^9$	μ (micro)	$10^{-6}$
M (mega)	$10^6$	n (nano)	$10^{-9}$
k (kilo)	$10^3$	p (pico)	$10^{-12}$
h (hecto)	$10^2$	f (femto)	$10^{-15}$
da (deca)	10	a (atto)	$10^{-18}$

## Abbreviations

The following is a list of the abbreviations used in the text. Common terms, the names of publications and institutions, and legal agreements are included along with their full identities. Other abbreviations will be defined wherever they first occur in an article. For further abbreviations, see page VII, Symbols and Units; page XIV, Frequently Cited Companies (Abbreviations), and page XV, Country Codes in patent references. The names of periodical publications are abbreviated exactly as done by Chemical Abstracts Service.

abs.	absolute	BAM	Bundesanstalt für Materialprüfung (Federal Republic of Germany)
a.c.	alternating current	BAT	Biologischer Arbeitsstoff-Toleranz-Wert (biological tolerance value for a working material, established by MAK Commission, see MAK)
ACGIH	American Conference of Governmental Industrial Hygienists	Beilstein	Beilstein's Handbook of Organic Chemistry, Springer, Berlin – Heidelberg – New York
ACS	American Chemical Society	BET	Brunauer – Emmett – Teller
ADI	acceptable daily intake	BGA	Bundesgesundheitsamt (Federal Republic of Germany)
ADN	accord européen relatif au transport international des marchandises dangereuses par voie de navigation interieure (European agreement concerning the international transportation of dangerous goods by inland waterways)	BGBI.	Bundesgesetzblatt (Federal Republic of Germany)
ADNR	ADN par le Rhin (regulation concerning the transportation of dangerous goods on the Rhine and all national waterways of the countries concerned)	BIOS	British Intelligence Objectives Subcommittee Report (see also FIAT)
ADP	adenosine 5'-diphosphate	BOD	biological oxygen demand
ADR	accord européen relatif au transport international des marchandises dangereuses par route (European agreement concerning the international transportation of dangerous goods by road)	bp	boiling point
AEC	Atomic Energy Commission (United States)	B.P.	British Pharmacopeia
a.i.	Active ingredient	BS	British Standard
AIChE	American Institute of Chemical Engineers	ca.	circa
AIME	American Institute of Mining, Metallurgical, and Petroleum Engineers	calcd.	calculated
ANSI	American National Standards Institute	CAS	Chemical Abstracts Service
AMP	adenosine 5'-monophosphate	cat.	catalyst, catalyzed
APhA	American Pharmaceutical Association	CEN	Comité Européen de Normalisation
API	American Petroleum Institute	cf.	compare
ASTM	American Society for Testing and Materials	CFR	Code of Federal Regulations (United States)
ATP	adenosine 5'-triphosphate	cfu	colony forming units
		Chap.	chapter
		ChemG	Chemikaliengesetz (Federal Republic of Germany)
		C.I.	Colour Index
		CIOS	Combined Intelligence Objectives Subcommittee Report (see also FIAT)
		CNS	central nervous system
		Co.	Company
		COD	chemical oxygen demand
		conc.	concentrated
		const.	constant
		Corp.	Corporation
		crit.	critical

CTFA	The Cosmetic, Toiletry and Fragrance Association (United States)	FIAT	Field Information Agency, Technical (United States reports on the chemical industry in Germany, 1945)
DAB 9	Deutsches Arzneibuch, 9th ed., Deutscher Apotheker-Verlag, Stuttgart 1986	Fig.	figure
d.c.	direct current	fp	freezing point
decomp.	decompose, decomposition	Friedländer	P. Friedländer, Fortschritte der Teerfarbenfabrikation und verwandter Industriezweige, Vol. 1 – 25, Springer, Berlin 1888 – 1942
DFG	Deutsche Forschungsgemeinschaft (German Science Foundation)	FT	Fourier transform
dil.	dilute, diluted	(g)	gas, gaseous
DIN	Deutsche Industrie Norm (Federal Republic of Germany)	GC	gas chromatography
DMF	dimethylformamide	GefStoffV	Gefahrstoffverordnung (regulations in the Federal Republic of Germany concerning hazardous substances)
DNA	deoxyribonucleic acid	GGVE	Verordnung in der Bundesrepublik Deutschland über die Beförderung gefährlicher Güter mit der Eisenbahn (regulation in the Federal Republic of Germany concerning the transportation of dangerous goods by rail)
DOE	Department of Energy (United States)	GGVS	Verordnung in der Bundesrepublik Deutschland über die Beförderung gefährlicher Güter auf der Straße (regulation in the Federal Republic of Germany concerning the transportation of dangerous goods by road)
DOT	Department of Transportation – Materials Transportation Bureau (United States)	GGVSee	Verordnung in der Bundesrepublik Deutschland über die Beförderung gefährlicher Güter mit Seeschiffen (regulation in the Federal Republic of Germany concerning the transportation of dangerous goods by sea-going vessels)
DTA	differential thermal analysis	GLC	gas-liquid chromatography
EC	effective concentration	Gmelin	Gmelin's Handbook of Inorganic Chemistry, 8th ed., Springer, Berlin – Heidelberg – New York
EC	European Community	GRAS	generally recognized as safe
ed.	editor, edition, edited	Hal	halogen substituent (-F, -Cl, -Br, -I)
e.g.	for example	Houben-Weyl	Methoden der organischen Chemie, 4th ed., Georg Thieme Verlag, Stuttgart
emf	electromotive force	HPLC	high performance liquid chromatography
EmS	Emergency Schedule	IAEA	International Atomic Energy Agency
EN	European Standard (European Community)	IARC	International Agency for Research on Cancer, Lyon, France
EPA	Environmental Protection Agency (United States)		
EPR	electron paramagnetic resonance		
Eq:	equation		
ESCA	electron spectroscopy for chemical analysis		
esp.	especially		
ESR	electron spin resonance		
Et	ethyl substituent (-C <sub>2</sub> H <sub>5</sub> )		
et al.	and others		
etc.	et cetera		
EVO	Eisenbahnverkehrsordnung (Federal Republic of Germany)		
exp (...)	e <sup>(...)</sup> , mathematical exponent		
FAO	Food and Agriculture Organization (United Nations)		
FDA	Food and Drug Administration (United States)		
FD & C	Food, Drug and Cosmetic Act (United States)		
FHSA	Federal Hazardous Substances Act (United States)		

IATA-DGR	International Air Transport Association, Dangerous Goods Regulations	Federal Republic of Germany); cf. Deutsche Forschungsgemeinschaft (ed.): Maximale Arbeitsplatzkonzentrationen (MAK) und Biologische Arbeitsstoff-Toleranz-Werte (BAT), WILEY-VCH Verlag, Weinheim (published annually)
ICAO	International Civil Aviation Organization	
i.e.	that is	
i.m.	intramuscular	
IMDG	International Maritime Dangerous Goods Code	max.
IMO	Inter-Governmental Maritime Consultive Organization (in the past: IMCO)	MCA
Inst.	Institute	Manufacturing Chemists Association (United States)
i.p.	intraperitoneal	Me
IR	infrared	Methodicum Chemicum Methodicum Chemicum, Georg Thieme Verlag, Stuttgart
ISO	International Organization for Standardization	MFAG
IUPAC	International Union of Pure and Applied Chemistry	Medical First Aid Guide for Use in Accidents Involving Dangerous Goods
i.v.	intravenous	MIK
Kirk-Othmer	Encyclopedia of Chemical Technology, 3rd ed., J. Wiley & Sons, New York – Chichester – Brisbane – Toronto 1978 – 1984; 4th ed., J. Wiley & Sons, New York – Chichester – Brisbane – Toronto 1991 – 1998	maximale Immissionskonzentration (maximum immission concentration)
(l)	liquid	min.
Landolt-Börnstein	Zahlenwerte u. Funktionen aus Physik, Chemie, Astronomie, Geophysik u. Technik, Springer, Heidelberg 1950 – 1980; Zahlenwerte und Funktionen aus Naturwissenschaften und Technik, Neue Serie, Springer, Heidelberg, since 1961	mp
LC <sub>50</sub>	lethal concentration for 50 % of the test animals	MS
LCL <sub>0</sub>	lowest published lethal concentration	NAS
LD <sub>50</sub>	lethal dose for 50 % of the test animals	NASA
LDLo	lowest published lethal dose	NBS
ln	logarithm (base e)	NCTC
LNG	liquefied natural gas	NIH
log	logarithm (base 10)	NIOSH
LPG	liquefied petroleum gas	NMR
M	mol/L	no.
M	metal (in chemical formulas)	NOEL
MAK	Maximale Arbeitsplatz-Konzentration (maximum concentration at the workplace in the	NRC
		NRDC
		NSC
		NSF
		NTSB
		OECD
		OSHA

p., pp.	page, pages		regulation in Federal Republic of Germany)
Patty	G. D. Clayton, F. E. Clayton (eds.): Patty's Industrial Hygiene and Toxicology, 3rd ed., Wiley Interscience, New York		TA Lärm
PB	Publication Board Report (U.S.	Technische Anleitung zum Schutz gegen Lärm (low noise regulation in Federal Republic of Germany)	
report	Department of Commerce, Scientific and Industrial Reports)		
PEL	permitted exposure limit	TDLo	
Ph	phenyl substituent ( $-C_6H_5$ )	THF	
Ph. Eur.	European Pharmacopoeia, 2nd. ed., Council of Europe, Strasbourg 1981	TLC	
phr	part per hundred rubber (resin)	TLV	
PNS	peripheral nervous system		
ppm	parts per million	TOD	
q. v.	which see (quod vide)	TRK	
ref.	refer, reference	TSCA	
resp.	respectively		
$R_f$	retention factor (TLC)	TÜV	
R. H.	relative humidity		
RID	règlement international concernant le transport des marchandises dangereuses par chemin de fer (international convention concerning the transportation of dangerous goods by rail)	TWA	
RNA	ribonucleic acid	UBA	
R phrase	risk phrase according to		
(R-Satz)	ChemG and GefStoffV (Federal Republic of Germany)	Ullmann	
rpm	revolutions per minute		
RTECS	Registry of Toxic Effects of Chemical Substances, edited by the National Institute of Occupational Safety and Health (United States)	USAEC	
(s)	solid		
SAE	Society of Automotive Engineers (United States)	USAN	
s.c.	subcutaneous	USD	
SI	International System of Units	USDA	
SIMS	secondary ion mass spectrometry		
S phrase	safety phrase according to	U.S.P.	
(S-Satz)	ChemG and GefStoffV (Federal Republic of Germany)	UV	
STEL	Short Term Exposure Limit (see TLV)	UVV	
STP	standard temperature and pressure (0° C, 101.325 kPa)		
$T_g$	glass transition temperature	VbF	
TA Luft	Technische Anleitung zur Reinhaltung der Luft (clean air		

	concerning the construction and operation of plants for storage, filling, and transportation of flammable liquids; classification according to the flash point of liquids, in accordance with the classification in the United States)	vs.	versus
VDE	Verband Deutscher Elektroingenieure (Federal Republic of Germany)	WGK	Wassergefährdungsklasse (water hazard class)
VDI	Verein Deutscher Ingenieure (Federal Republic of Germany)	WHO	World Health Organization (United Nations)
vol	volume		Winnacker-Küchler Chemische Technologie, 4th ed., Carl Hanser Verlag, München, 1982-1986;
vol.	volume (of a series of books)	wt	Winnacker-Küchler, Chemische Technik: Prozesse und Produkte, Wiley-VCH, Weinheim, from 2003
		\$	weight U.S. dollar, unless otherwise stated

## Frequently Cited Companies (Abbreviations)

Air Products	Air Products and Chemicals	ICI	Imperial Chemical Industries
Akzo	Algemene Koninklijke Zout	IFP	Institut Français du Pétrole
	Organon	INCO	International Nickel Company
Alcoa	Aluminum Company of America	3M	Minnesota Mining and Manufacturing Company
Allied	Allied Corporation	Mitsubishi	Mitsubishi Chemical Industries
Amer.	American Cyanamid	Chemical	Monsanto Company
Cyanamid	Company	Monsanto	Nippon Shokubai Kagaku Kogyo
BASF	BASF Aktiengesellschaft	Nippon	Pechiney Ugine Kuhlmann
Bayer	Bayer AG	Shokubai	Pittsburg Plate Glass Industries
BP	British Petroleum Company	PCUK	G.D. Searle & Company
Celanese	Celanese Corporation	PPG	Smith Kline & French Laboratories
Daicel	Daicel Chemical Industries	Searle	Societá Nazionale Metandotti
Dainippon	Dainippon Ink and Chemicals Inc.	SKF	Standard Oil of Ohio
Dow Chemical	The Dow Chemical Company	SNAM	Stauffer Chemical Company
DSM	Dutch Staats Mijnen	Sohio	Sumitomo Chemical Company
Du Pont	E.I. du Pont de Nemours & Company	Stauffer	Toray Industries Inc.
Exxon	Exxon Corporation	Sumitomo	Union Chimique Belge
FMC	Food Machinery & Chemical Corporation	Toray	Union Carbide Corporation
GAF	General Aniline & Film Corporation	UCB	Universal Oil Products Company
W.R. Grace	W.R. Grace & Company	Union Carbide	Vereinigte Elektrizitäts- und Bergwerks-AG
Hoechst	Hoechst Aktiengesellschaft	UOP	Wacker Chemie GmbH
IBM	International Business Machines Corporation	VEBA	
		Wacker	

## Country Codes

The following list contains a selection of standard country codes used in the patent references.

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AT	Austria	ID	Indonesia
AU	Australia	IL	Israel
BE	Belgium	IT	Italy
BG	Bulgaria	JP	Japan *
BR	Brazil	LU	Luxembourg
CA	Canada	MA	Morocco
CH	Switzerland	NL	Netherlands *
CS	Czechoslovakia	NO	Norway
DD	German Democratic Republic	NZ	New Zealand
DE	Federal Republic of Germany (and Germany before 1949) *	PL	Poland
DK	Denmark	PT	Portugal
ES	Spain	SE	Sweden
FI	Finland	SU	Soviet Union
FR	France	US	United States of America
GB	United Kingdom	YU	Yugoslavia
GR	Greece	ZA	South Africa
HU	Hungary	EP	European Patent Office *
		WO	World Intellectual Property Organization

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\* For Europe, Federal Republic of Germany, Japan, and the Netherlands, the type of patent is specified:  
EP (patent), EP-A (application), DE (patent), DE-OS (Offenlegungsschrift), DE-AS (Auslegeschrift),  
JP (patent), JP-Kokai (Kokai tokkyo koho), NL (patent), and NL-A (application).

# Periodic Table of Elements

element symbol, atomic number, and relative atomic mass (atomic weight)

		VIIIA																				
		1A								1B										18		
IA		"European" group designation to 1986 IUPAC proposal								"American" group designation, also used by the Chemical Abstracts Service until the end of 1986										0		
1	2A																	3B	4B	5B	6B	7B
1.0079	H																13	14	15	16	17	
3	4	B	e														5	6	7	8	9	
Li	Mg																10	11	O	F	Ne	
6.941	9.0122																10.811	12.011	14.007	15.999	18.998	
22.990	24.305																				20.180	
11	12	3A	4A	5A	6A	7A	8	9	10	11	12	13	14	15	16	17	18					
Na	Mg						VIB	VIB	VIII	VIII	VIII	IB	IB	28.086	30.974	32.066	35.453	39.948				
22.990	24.305																					
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36					
K	Ca	S	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr					
39.098	40.078	44.956	47.867	50.942	51.996	54.938	55.845	58.933	58.693	63.546	65.409	69.723	72.61	74.922	78.96	79.904	83.80					
Rb	Sr	Y	Zr	Nb	Mo	Tc*	Ru	Rh	Pd	Ag	Cd	In	Sn	Tb	I	Xe						
85.468	87.62	88.906	91.224	92.906	95.94	98.906	101.07	102.91	106.42	107.87	112.41	114.82	118.71	121.76	127.60	126.90	131.29					
55	56	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86						
Cs	Ba	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po*	At*	Rn*						
132.91	137.33	178.49	180.95	183.84	186.21	190.23	192.22	195.08	196.97	200.59	204.38	207.2	208.98	208.98	209.99	222.02						
87	88	104	105	106	107	108	109	110	111	112	114	116	118	120	122	124						
Ft*	Ra*	Rf*	Db*	Sg	Bh	Hs	Mt	Uun	Uuu	Uuu	Uuuq											
223.02	226.03	261.11	262.11																			

\* provisional IUPAC symbol

		VIIIA																				
		1A								1B										18		
IA		"European" group designation to 1986 IUPAC proposal								"American" group designation, also used by the Chemical Abstracts Service until the end of 1986										0		
1	2A																					
La	Ce	Pr	Nd	Prm*	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu								
138.91	140.12	140.91	144.24	146.92	150.36	151.97	157.25	158.93	162.50	164.93	167.26	168.93	173.04	174.97								
89	90	91	92	93	94	95	96	97	98	99	100	101	102	103								
Ac*	Th*	Pa*	U*	Np*	Pu*	Am*	Cm*	Bk*	Cf*	Es*	Fm*	Md*	No*	Lr*								
227.03	232.04	231.04	238.03	237.05	244.06	243.06	247.07	247.07	251.08	252.08	257.10	258.10	259.10	260.11								

\* radioactive element; mass of most important isotope given.

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# Poly(Vinyl Chloride)

*General aspects of polymers, properties and testing, processing, additives, and analysis are discussed in Plastics, General Survey; Plastics, Properties and Testing; Plastics, Processing; Plastics, Additives; and Plastics, Analysis, respectively. States of order are treated in Plastics, Properties and Testing. Fundamental aspects of polymerization reactions are treated in Polymerization Processes.*

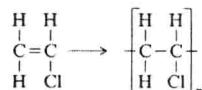
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## 1. Introduction

Poly(vinyl chloride), PVC, a polymer prepared from vinyl chloride monomer (VCM),



where  $n = 700–1500$ , holds a unique position amongst all the polymers produced today. It is relatively inexpensive and is used in such a wide range of applications that its versatility is almost unlimited. However, if it were discovered today it would probably be shelved as a somewhat intractable and thermally unstable material. How can this apparent contradiction be explained?

The uniqueness of PVC can be considered under three headings: morphology, versatility, and molecular structure.

### 1.1. Morphology

As made, PVC is particulate in nature and comes in two main sizes depending on the process used. Suspension and mass polymerizations give grains (particles) of  $100–180 \mu\text{m}$  in diameter, whereas the emulsion process affords a latex of particle size  $0.1–3.0 \mu\text{m}$ . The latter is dried to yield friable grain-like structures of  $5–50 \mu\text{m}$ .

Because of this unique particulate structure the most frequently used word in the vocabulary