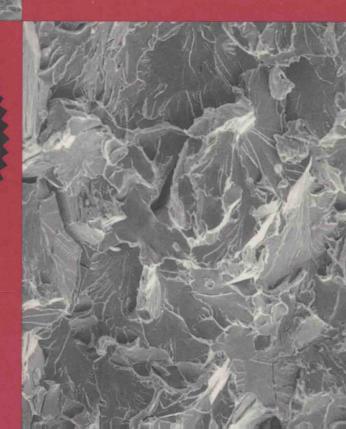
# INDUSTRIAL MATERIALS

METALS
AND Volume 1
ALLOYS

David A. Colling Thomas Vasilos



# INDUSTRIAL MATERIALS



# Metals and Alloys

David A. Colling

University of Massachusetts Lowell

Thomas Vasilos

University of Massachusetts Lowell

#### Library of Congress Cataloging-in-Publication Data

Colling, David A. Industrial materials.

Includes index.

Contents: v. 1. Metals and alloys-

- v. 2. Polymers, ceramics, and composites.
- 1. Materials. 2. Materials-Case studies.
- 3. Manufacturing processes—Case studies.

I. Vasilos, Thomas. II. Title.

TA403.C585 1995 620.1'1

.0.1

94-10077

ISBN 0-02-323560-8 (v. 1)

Cover photo: David A. Colling

Editor: Stephen Helba

Production Editor: Mary Ann Hopper Text Designer: Julia Zonneveld Van Hook Cover Designer: Julia Zonneveld Van Hook Production Buyer: Patricia A. Tonneman

Electronic Text Management: Marilyn Wilson Phelps, Matthew Williams, Jane Lopez,

Karen L. Bretz

Illustrations: Diphrent Strokes, Inc.

This book was set in Times by Prentice Hall and was printed and bound by R. R. Donnelley & Sons Company. The cover was printed by Phoenix Color Corp.



A Simon & Schuster Company Englewood Cliffs, New Jersey 07632

All rights reserved. No part of this book may be reproduced, in any form or by any means, without permission in writing from the publisher.

Printed in the United States of America

10987654321

ISBN: 0-02-323560-8

Prentice-Hall International (UK) Limited, London

Prentice-Hall of Australia Pty. Limited, Sydney

Prentice-Hall Canada Inc., Toronto

Prentice-Hall Hispanoamericana, S. A., Mexico

Prentice-Hall of India Private Limited, New Delhi

Prentice-Hall of Japan, Inc., Tokyo

Simon & Schuster Asia Pte. Ltd., Singapore

Editora Prentice-Hall do Brasil, Ltda., Rio de Janeiro

# Preface

delivery. Industrial practitioners must be well versed in all aspects of manufacture, including computer applications, manufacturing processes, quality control, production management, and organizational behavior, among others. Selection and processing of industrial materials are important, but students preparing for careers in industry do not need to first become proficient in materials science, as many of the good textbooks on materials available at present assume. To be successful, we feel that the authors should be specialists in materials first, but should also be well versed in product-oriented manufacturing processes, where making it right is the only practice that matters! For these reasons, this textbook is filled with case studies that illustrate industrial problems.

We also believe that teaching is important; the material in this textbook has been developed over many years of teaching industrial technology students, many of whom are now practicing in successful careers and some of whom have provided case studies that appear in the text. These students did not have rigorous mathematics backgrounds, so it was important to develop their understanding of concepts rather than their computational skills or theoretical knowledge. The thrust of this textbook is to define properties needed for applications, then relate these properties to the material properties for appropriate selection and control through processing.

It is impossible to cover all materials in a single one-semester course, yet curriculum demands do not always permit time for a two-course sequence. We have separated our treatment of materials into two volumes rather than including everything in a single cumbersome volume that might not be fully utilized. Where only a single course is required, your emphasis can be tailored to either metals and alloys

(Volume 1) or to polymers, ceramics, and composites (Volume 2), eliminating one volume or leaving it for an elective.

There is little new information provided in these two volumes—we have borrowed freely from other sources whose permission is acknowledged with appreciation. Our contribution is in the organization of the topics and their presentation in a logical fashion to establish a basis for optimum applications of materials in manufacturing.

Volume 1 is confined to metals and alloys. We begin the first four chapters by discussing properties related to these materials and their applications. We then build a basis for understanding the selection and control needed to satisfy the demands of these materials, beginning with brief treatments of atomic structure and bonding to form crystalline solids. Defects and their consequences provide the groundwork for single phase alloys, then binary and higher order alloys. Chapter 5 describes the making of metals and alloys and the microstructure and properties resulting from solidification. Chapter 6 emphasizes the proper heat treatment to use in making the material right for its application. In Chapters 7 and 8, the important ferrous and nonferrous alloys are described, along with ideas for their use. In Chapter 9, we emphasize what happens to metals during deformation and the metalworking methods for shaping metals by deforming them. Although we emphasize proper selection and control throughout the text, we describe metal joining in Chapter 10 because many products and properties are altered by joining or during application, particularly corrosion, which is presented in Chapter 11.

We would like to thank our students for their inspiration and our colleagues at the University of Massachusetts Lowell for their encouragement and support. Particular thanks are due Professor V. E. Simms, Jr., for his invaluable discussions and assistance in locating some of the photomicrographs. We would also like to thank all those who reviewed the final manuscript: David H. Devier, Ohio Northern University; Peter Stracener, South Plains College; C. J. Law, Western New Mexico University; Bill G. Cullins, Aims Community College; Thomas F. Kilduff, Thomas Nelson Community College; and Boyd Larson, University of Wisconsin–Platteville. Finally, we could not have completed the text without the sacrifice of our families and friends, particularly our wives, Dr. Jane Dreskin and Mrs. Helen Vasilos.

# **Contents**

**Properties of Metals** 

1.1

1.1.1

1.1.2

1.1.3

1.1.4

1.1.5

1.1.6

1.2

1.3

1.4

	Summary	20		
	Terms to	Remember	21	
	Problems	21		
2				
Atom.	s, Bondi	ing, and C	rystal Structur	es 23
	2.1 A	Atomic Bondin	g 24	
	2.2	Crystal Structur	e 26	

**Mechanical Properties** 

8

13

19

**Electrical and Thermal Properties** 

**Tensile Properties** 

**Stress Concentration** 

14

**Magnetic Properties** 

Thermal Effects

Hardness

Toughness

Wear

Metal Fatigue

3

10

17

16

-	0 1 1
VI	Contents
v.,	00//10//10

VI	on to the
	Summary 33 Terms to Remember 34 Problems 34
<b>3</b> Pure N	letals and Single-Phase Alloys 35
	3.1 Characteristics of Pure Metals 36 3.2 Defects in Materials 36 3.3 Single-Phase Alloys 42 3.4 Influence of Alloying on Properties 45 Summary 48 Terms to Remember 49 Problems 50
<b>4</b> Rinary	Alloys and Phase Diagrams 51
Dillai y	,
	4.1 The Gibbs Phase Rule 52 4.2 Invariant Reactions 55 4.3 Phase Diagrams of Common Commercial Alloys 59 4.4 The Iron–Iron Carbide (Fe-Fe <sub>3</sub> C) Phase Diagram 61 4.5 Microstructure and Invariant Reactions 62 4.6 Polynary Alloys and Phase Diagrams 64 Summary 65 Terms to Remember 65 Problems 66
<b>5</b> Meltin	g and Solidification 69
	5.1 Production of Metal Alloys 69 5.2 Solidification of Alloys 74 5.2.1 Cast Microstructure 75 5.2.2 Chemical Segregation 76 5.2.3 Porosity Formed in Solidification 78 5.3 Microstructure Resulting from Solidification of Eutectic Alloys 81 Summary 82 Terms to Remember 83 Problems 84

0	4	β	۹
1		6	i
и	7	٦	
V.	ı.	u	

## Heat Treatment of Metals and Alloys 85

- 6.1 Atomic Diffusion 85
- 6.2 Heat Treatment of Cast Metals 86
- 6.3 Furnaces and Atmospheres for Heat Treatment 88
- 6.4 The Decomposition of Austenite 89
- 6.4.1 Hardenability 93
- 6.4.2 Alloying Effects on Transformation 93
- 6.4.3 Heat Treatment of Steel 95
- 6.4.4 Quenching Media 95
- 6.4.5 Surface Hardening 99
- 6.4.6 Stress Relief Annealing 101
- 6.5 Precipitation Hardening, or Age Hardening 102

Summary 104

Terms to Remember 105

Problems 106

# 7

#### Ferrous Alloys 107

- \_\_\_\_
- 7.1 Steel 108
- 7.2 Cast Iron 111
- 7.3 Stainless Steel 113
- 7.4 Tool Steel 117
- 7.5 Maraging Steels 120
- 7.6 Ferrous Magnetic Alloys 123
- 7.6.1 Soft Magnetic Electrical Steels 124
- 7.6.2 Permanent Magnetic Alloys 124

Summary 126

Terms to Remember 129

Problems 129

# 8

## Nonferrous Metals and Alloys 131

- 8.1 Copper and Copper Alloys 132
  - 8.1.1 Pure Copper 132
  - 8.1.2 Brass Alloys 133
  - 8.1.3 Bronze Alloys 133

#### viii Contents

	8.1.4 8.1.5 8.2 8.2.1 8.2.2 8.2.3 8.3 8.3.1 8.3.2 8.3.3 8.4 8.5 8.6 Summa Terms t	o Remember 151
<b>9</b> Deform	nation	Processing of Metals 153
	9.1	Plastic Deformation 153
	9.1.1	Slip 155
	9.1.2	Dislocations and Slip 156
	9.1.3	Stored Energy of Cold Work, Recovery, Recrystallization, and Grain Growth 158
	9.1.4	Texture in Deformation Processing 163
	9.2	Fundamentals of Metalworking 166
	9.2.1	Deformation Processes 166
	9.2.2	Metalworking Relationships 168
	9.2.3	Hot Working 168
	9.2.4	Cold Working 169
	9.3	Forging 170
	9.3.1	Forgeability 172
	9.4	Rolling Processes 172
	9.5	Extrusion 174
	9.6	Wire Drawing 176
	9.7	Forming Processes 179
	Summa	-
		Remember 182
	Problem	

4	
7	U

# Metal-Joining Processes 185

Problems 217

10.1	Mechanical Joining 186	
10.1.1	Riveted and Bolted Joints	186
10.1.2	Removable Fasteners 189	
10.2	Adhesive Bonding 191	
10.2.1	Types of Adhesives 192	
10.3	Metal Joining 193	
10.3.1	Joint Design 193	
10.3.2	Soldered and Brazed Joints	194
10.3.3	Welding 205	
10.3.4	<b>Electrical Welding Processes</b>	207
10.3.5	Welding Metallurgy 211	
10.3.6	Welding Defects 212	
10.3.7	Weld Testing 214	
Summa	ry 215	
Terms to	Remember 216	

# 11

# Corrosion and Corrosion Protection 219

11.1	The Nature of Corrosion 219	
11.2	Corrosion Rate 225	
11.3	Passivity 225	
11.4	Corrosion Protection 229	
11.4.1	Material Selection 230	
11.4.2	Material Design 233	
11.4.3	Inspection 234	
11.4.4	Coatings 236	
11.4.5	Cathodic Protection 237	
11.4.6	Alteration of Corrosive Environment	238
Summa	ry 239	
Terms t	o Remember 240	
Probler	ns 240	

# Glossary 243

#### Index 255

# 1

# **Properties of Metals**

The history of man can be measured by the development and applications of materials. It is with good reason that we refer to periods of human culture as the Stone Age, the Bronze Age, or the Iron Age because materials developments during these ages led to the production of weapons for hunting or warfare and to the production of cooking and storage utensils. Even today, our sophisticated technology is dependent on materials developments at work, at home, and at play. Enterprises such as the transportation, computer, electronics, communications, and aerospace industries are a result of our being able to study and learn about the materials needed to develop the dreams of entrepreneurs.

Our understanding of materials did not really begin until the late nineteenth century when the microscope and methods for testing materials were first developed. Up until then, materials development was purely empirical, thus limiting the technology of the time. For example, both the social and economic development of the United States was made possible by railroads. Prior to improved steelmaking by the Bessemer process, however, rails were too weak to sustain the constant travel of steam engines.

Today's materials can be classified as metals and alloys, as polymers or plastics, as ceramics, or as composites; composites, most of which are man-made, actually are combinations of different materials. Applications of these materials depend on their properties; therefore, we need to know what properties are required by the application and to be able to relate those specifications to the material. For example, a ladder must withstand a design load, the weight of a person using the ladder. However, the material property that can be measured is strength, which is affected by the load and design dimensions. Strength values must therefore be applied to determine the ladder dimensions to ensure safe use.

The properties that we will be using throughout this textbook are those of metals and alloys. They include **physical properties** such as density and melting point, **mechanical properties** such as strength and ductility, electrical and thermal properties such as conductivity, and magnetic properties. (Other properties of materials, such as optical properties, will be discussed in the second volume of this textbook on polymers, ceramics, and composites.)

The units for measurement of properties are supposedly uniform, with the International System of Units (SI units) universally acceptable. Nevertheless, conventional usage of British units in the United States has persisted in many disciplines. This mix really does not present a problem, however, because we can readily convert to SI units when measuring in British units. Table 1.1 compares the units of measurement and lists conversion factors.

Table 1.1
Measurements and material properties

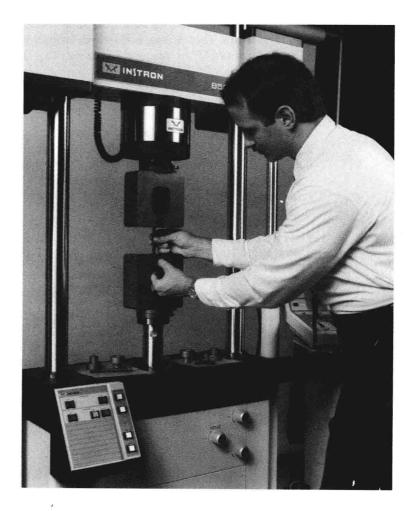
Property	SI unit	British unit	Conversion factors
Length	meter	inch, foot	1 in. = 2.54 cm = 25.4 mm 1 m = 39.37 in. 1 Å = 10 <sup>-8</sup> cm 1 mil (.001 in.) = .0394 mm
Mass	kilogram	pound mass (lbm)	1 kg = 2.204 lbm 1 lbm = 453.7 g
Force	newton (N)	pound force (lbf)	1 lbf = 4.44 N
Stress	pascal (Pa)	lbf/in. <sup>2</sup> or psi	1 Pa = 1 N/m <sup>2</sup> = .145 × 10 <sup>-3</sup> lbf/in. <sup>2</sup> 1 lbf/in. <sup>2</sup> = $6.89 \times 10^3$ Pa
Temperature	°C K (absolute)	°F °R (absolute)	°F = %°C + 32 K = °C + 273 °R = °F + 460

# 1.1 Mechanical Properties

# 1.1.1 Tensile Properties

Mechanical properties are always specified in material selection for structural applications. Structural design, in turn, must provide the size that is appropriate for these properties. It is typical to specify tensile properties, which simply refer to the applied forces that stretch a shape. Tensile tests are performed in universal machines such as that shown in Figure 1.1. These machines can test materials under compression, shear, and flexure. In most cases, we will specify engineering properties, which are determined from a stress-strain curve of test results.

Figure 1.1
Universal testing machine for mechanical property measurement (Courtesy of Instron Corporation.)



Engineering stress,  $\sigma$ , is defined as

$$\sigma = \frac{P}{A_o}$$

where  $A_o$  is the original cross-sectional area and P is the force that is applied. This applied force will extend or elongate the metal, causing a **strain**,  $\varepsilon$ , which is given by

$$\varepsilon = \frac{\ell - \ell_o}{\ell_o}$$

where  $\ell$  is specimen length after force is applied and  $\ell_o$  is the original specimen length.

When we load the specimen to failure, the stress-strain curve can be plotted, giving us useful data for specifications. Figure 1.2 shows a typical stress-strain curve for a metal sample tested to failure. Initially, there is a large *linear* increase in stress with little strain. This change is **elastic deformation** because the metal shape is completely recovered if the force is removed. The linear relation is described by the equation

$$\sigma = E\varepsilon$$

where E is the **elastic modulus**, or Young's modulus. This equation is known as Hooke's law and introduces us to the most common example of a structure independent property. No matter what we do to the material, E remains unchanged. We have to change the temperature or composition to alter it. Structure sensitive properties, on the other hand, can be altered by either heat treatment or deformation.

Figure 1.2
Typical stress-strain curve for metal sample tested to failure

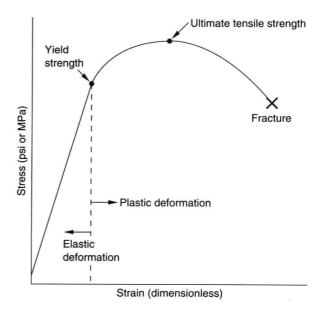
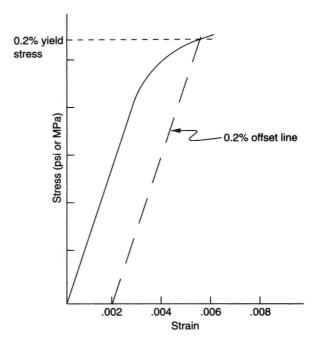


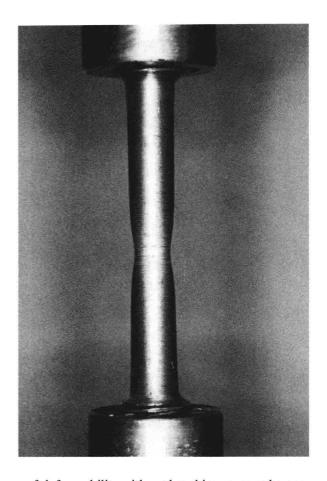
Figure 1.3 Method for determining 0.2% offset yield strength



The stress-strain curve of Figure 1.2 contains a great deal of information that we will use when specifying mechanical properties of materials used in manufacturing. Most applications cannot permit any permanent deformation, much less fracture of the material, so we must design for a stress below that where **plastic deformation** first occurs. We call this stress the **yield strength** or flow strength (also referred to as yield or flow stress). The exact value of the yield strength cannot be determined with any accuracy; therefore, we use the convention denoting the 0.2% offset yield strength. This is determined by selecting the strain value of 0.002 in./in. or mm/mm (0.2%) and drawing a line parallel to the elastic or linear portion of the stress-strain curve. The intersection of this line with the stress-strain curve, illustrated in Figure 1.3, is defined as the 0.2% yield strength.

The highest stress that a metal can withstand without breaking is termed the **ultimate tensile strength**, as indicated in Figure 1.2. When the ultimate tensile strength has been reached, the sample cross section is reduced by additional loading; we call this reduction **necking**. Figure 1.4 demonstrates the necking of a mild steel sample just prior to fracture. Therefore the reduction in engineering stress beyond the ultimate tensile strength is really an artifact because we continue to divide the load by the original cross-sectional area even though the cross section has been reduced by the necking (in Chapter 9, we will examine behavior in terms of true stress and true strain). We now have identified the elastic modulus and two strengths, the yield strength used for design purposes and the ultimate tensile strength, which is the maximum strength. Both yield strength and ultimate tensile strength are structure sensitive properties that can be affected by deformation or heat treatment.

Figure 1.4 Necking of mild steel sample during testing



**Ductility**, which is a measure of deformability without breaking, cannot be easily determined from the stress-strain diagram. The high stresses and low strains of the elastic region are experimentally measured with extensometers or strain gauges; these cannot be used for high elongation values. Also, necking that occurs is a form of ductility that is not recorded on the stress-strain diagram. We therefore express ductility in terms of the percentage elongation to fracture or the percentage **reduction in area** (% RA). These are defined as

% Elongation = 
$$\frac{\ell_f - \ell_o}{\ell_o} \times 100$$
 and % RA = 
$$\frac{A_o - A_f}{A_o} \times 100$$

A standard length is marked and measured both before  $(\ell_o)$  and after  $(\ell_f)$  testing, and the area before  $(A_o)$  and the area in the necked region after  $(A_f)$  testing are also determined to ascertain the property values. Both % elongation to fracture and the % RA are structure sensitive properties.

#### Sample Problem 1.1

Steel has a modulus of elasticity, E, of  $29 \times 10^6$  psi ( $20 \times 10^4$  MPa).

- a. What is the stress on a steel rod 0.125 in. in diameter and 12 in. long that is stretched 0.02 in.?
  - b. What is the force causing this stress?

#### Solution

a. 
$$\varepsilon = \frac{\ell - \ell_o}{\ell_o} = \frac{0.02 \text{ in.}}{12 \text{ in.}} = 1.67 \times 10^{-3} \text{ in./in.}$$

$$\sigma = E\varepsilon = 29 \times 10^6 \text{ psi} \times 1.67 \times 10^{-3} \text{ in./in.}$$

$$= 48.43 \times 10^3 \text{ psi}$$

$$= 48,400 \text{ psi } (333.8 \text{ MPA}) *$$
b. 
$$\sigma = \frac{P}{A}$$

$$48,400 \text{ psi} = \frac{P}{\pi (0.125 \text{ in.})^2 / 4}, \text{ since } A = \frac{\pi d^2}{4}$$

$$P = 48,400 \text{ psi} \times \frac{\pi (0.125 \text{ in.})^2}{4}$$

$$= 594 \text{ lbf } (133.8 \text{ N})$$

#### Sample Problem 1.2

A yellow brass rod (Cu-30% Zn) 0.25 in. in diameter has a yield strength of 16,000 psi (110 MPa) and tensile strength of 48,000 psi (330 MPa). What is the maximum force that can be applied

- a. without breaking the rod?
- b. without deforming the rod permanently?

#### Solution

Using the ultimate tensile strength,

$$\frac{P}{A}$$
 = 48,000 psi  

$$P = \frac{48,000 \text{ psi}}{\pi (0.25 \text{ in.})^2 / 4}$$
= 2360 lb (532 N)

<sup>\*</sup> Values with more than three significant numbers cannot be supported by testing accuracy, so convention limits the answer to the rounded value of 48,400 psi.