

Oxidation in foods and beverages and antioxidant applications

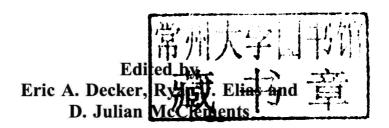
Volume 1: Understanding mechanisms of oxidation and antioxidant activity

Edited by Eric A. Decker, Ryan J. Elias and D. Julian McClements



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Oxidation in foods and beverages and antioxidant applications

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Oxidative rancidity is a major cause of food quality deterioration, leading to the formation of undesirable off-flavours as well as unhealthy compounds. The two volumes of Oxidation in foods and beverages and antioxidant applications review food quality deterioration due to oxidation and methods for its control. The second volume concentrates on oxidation and its management in different industry sectors. Part I discusses animal products, with chapters on the oxidation and protection of red meat, poultry, fish and dairy products. Part II reviews oxidation in plant-based foods and beverages, including edible oils, wine and fried products. The final chapters examine encapsulation to inhibit lipid oxidation and antioxidant active packaging and edible films.

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For a food product to be a success in the marketplace, it must be stable throughout its shelf-life. Changes due to food chemical deterioration and physical instability are not always recognised by food producers, who are more familiar with microbial spoilage, yet can be just as problematic. This book provides an authoritative review of key topics in this area. Chapters in Parts I and II focus on the chemical reactions and physical changes that negatively affect food quality. The remaining chapters outline the likely effects on different food products, for example baked goods, fruit and vegetables and beverages.

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Preface

Even though the involvement of peroxides in lipid oxidation has been known for over 100 years and the free radical mechanism of autooxidation was proposed over 60 years ago, many challenges still exist in controlling oxidative rancidity in foods. These challenges exist because the lipid oxidation process is influenced by a huge number of factors, many of which are specific for different food items. As was identified in the earliest studies on oxidative rancidity, these chemical pathways are dependent on temperature, oxygen concentrations and lipid type. In addition, the reactions often do not simply undergo autooxidation pathways, as they are also accelerated by various prooxidants which include enzymes, transition metals and reactive oxygen generators. Once the reaction starts, its rate is influenced by antioxidants that are endogenous to the food. These antioxidant systems can include free radical scavengers, metal chelators, singlet oxygen quenchers and antioxidant enzymes. Some of these antioxidant systems have been specifically designed by biological systems to inhibit lipid oxidation such as tocopherols, iron binding proteins (e.g. transferrin) and antioxidant enzymes. However, there are also numerous compounds in foods that were not specifically designed to inhibit oxidation but contain functional groups that can interact with free radicals and prooxidants. For example, proteins contain amino acids that can both scavenge free radicals and chelate transition metals. In addition, anthocyanidins and carotenoids, which are important in food color, can also scavenge free radicals, bind metals and inactivate singlet oxygen. These endogenous antioxidants are critical in the natural 'oxidative resistance' of foods.

The susceptibility of foods to lipid oxidation is increasing as food product developers attempt to make foods healthier by adding more polyunsaturated fatty acids, and by removing synthetic antioxidants that are causing consumer