Reinhard Bruckner

Advanced Organic Chemistry

Reaction Mechanisms



Sponsoring Editor Production Managers Editorial Coordinator

Promotions Manager Copyeditor Proofreader Composition

Printer

Jeremy Hayhurst

Joanna Dinsmore and Andre Cuello

Nora Donaghy Stephanie Stevens Brenda Griffing Cheryl Uppling TechBooks Maple Press

Front cover image: Computer animation of the catalytic hydrogenation of an olefin (J. Brickmann and W. Sachs, Physical Chemistry, TH Darmstadt).

This book is printed on acid-free paper. ®

English translation copyright © 2002 by HARCOURT/ACADEMIC PRESS

All Rights Reserved.

No part of this publication may be reproduced or transmitted in any form or by any means, electronic or mechanical, including photocopy, recording, or any information storage and retrieval system, without permission in writing from the publisher.

Requests for permission to make copies of any part of the work should be mailed to: Permissions Department, Harcourt, Inc., 6277 Sea Harbor Drive, Orlando Florida 32887-6777

Translation of REAKTIONMECHANISMEN, published by Spektrum Akademischer Verlag, copyright © 1996 Spektrum Akademischer Verlag.

Academic Press

A Harcourt Science and Technology Company 525 B Street, Suite 1900, San Diego, California 92101-4495, U.S.A. http://www.academicpress.com

Academic Press

Harcourt Place, 32 Jamestown Road, London NW1 7BY, UK http://www.academicpress.com

Harcourt/Academic Press

A Harcourt Science and Technology Company 200 Wheeler Road, Burlington, Massachusetts 01803 http://www.harcourt-ap.com

Library of Congress Catalog Card Number: 2001088748

International Standard Book Number: 0-12-138110-2

PRINTED IN THE UNITED STATES OF AMERICA

01 02 03 04 05 06 MM 9 8 7 6 5 4 3 2 1

Foreword

We are at the start of a revolution in molecular science that will more profoundly change our lives, our culture, indeed, our world than did the Industrial Revolution a century ago. From the human genome project, the largest natural product characterization effort ever, to the search for the molecular signatures of life on other planets, this molecular revolution is creating an ever-expanding view of ourselves and our universe. At the core of this revolution is chemistry, the quintessential molecular science within which is organic chemistry, a discipline that will surely be the source of many of the major advances in chemistry, biology, medicine, materials science, and environmental science in the 21st century.

In his text on organic chemistry, the translation of which has been impressively led by Professors Harmata and Glaser, Professor Bruckner has masterfully addressed the core concepts of the discipline, providing a rich tapestry of information and insight. The student of contemporary organic chemistry will be well-served by the depth and quality of this treatment. The underlying philosophy of this text is that much of chemistry can be understood in terms of structure, which in turn influences reactivity, ultimately defining the higher order activities of synthesis. Whether one seeks to understand nature or to create the new materials and medicines of the future, a key starting point is thus understanding structure and mechanism.

Professor Bruckner addresses the interrelationship of structure and mechanism with the rich insight of one schooled at the interface of physical organic chemistry and synthesis. His treatment is impressively rigorous, integrated, and broad. He achieves breadth through the careful selection of representative and fundamental reactive intermediates and reactions. Rigor and integration derive from his disciplined adherence to structure, orbital theory, and mechanism. The result is a powerfully coherent treatment that enables the student to address the rich subject matter at hand and importantly by analogy the far-ranging aspects of the field that lie beyond the scope of the book. Extending from his treatment of radicals, nucleophiles, carbenium ions, and organometallic agents to concerted reactions and redox chemistry, Bruckner provides an analysis that effectively merges theory and mechanism with examples and applications. His selection of examples is superb and is further enhanced by the contemporary references to the literature. The text provides clarity that is essential for facilitating the educational process.

This is a wonderfully rich treatment of organic chemistry that will be a great value to students at any level. Education should enable and empower. This text does both, providing the student with the insights and tools needed to address the tremendous challenges and opportunities in the field. Congratulations to Professors Bruckner, Harmata, and Glaser for providing such a rich and clear path for those embarking on an understanding of the richly rewarding field of organic chemistry.

Preface to the English Edition

Writing a textbook at any level is always a challenge. In organic chemistry, exciting new discoveries are being made at an ever-increasing pace. However, students of the subject still arrive in the classroom knowing only what they have been taught, often less. The challenge is to present appropriate review material, present venerable, classic chemistry while dealing with the latest results, and, most importantly, provoke thought and discussion. At the time this book was written, there was a need for an advanced text that incorporated these aspects of our science.

The German version of the text was designed for second- and third-year chemistry majors: 60–70% of the contents of this book address students before the "Diplom-chemiker-Vorexamen," while the remaining 30–40% address them thereafter. The German book is typically used one year after a standard introductory textbook such as that by Vollhardt and Schore, Streitweiser and Heathcock, or McMurry. Accordingly, in the United States this text can be used in a class for advanced undergraduates or beginning graduate students. Curricula of other English-speaking countries should allow the use of this text with optimum benefit at a similar point of progress. A good understanding of the fundamentals of organic and physical chemistry will suffice as a foundation for using this textbook to advantage.

The approach taken in this book conveys the message that the underlying theory of organic chemistry pervades the entire science. It is not necessary at this level to restrict the learning of reactions and mechanisms to any particular order. MO theory and formalisms such as electron pushing with arrows are powerful tools that can be applied not only to the classic chemistry that led to their development but also to the most recently developed reactions and methods, even those that use transition metals.

Theory, mechanism, synthesis, structure, and stereochemistry are discussed throughout the book in a qualitative to semiquantitative fashion. Fundamental principles such as the Hammond postulate that can be applied in the most varied contexts are reinforced throughout the book. Equations such as the Erying equation or the rate laws of all kinds of reactions are introduced with the view that they have context and meaning and are not merely formulas into which numbers are plugged.

The present text, to the best of our knowledge, does not duplicate the approach of any other treatment at a comparable level. We are convinced that this book, which has already filled a niche in the educational systems of German- and the French-speaking countries (a French translation appeared in 1999), will do the same in the textbook market of English-speaking countries now that an English edition has become available.

We hope that you enjoy many fruitful hours of insight in the course of studying this book, and we welcome your constructive comments on its content and approach.

Michael Harmata

Norman Rabjohn Distinguished Professor of Organic Chemistry Department of Chemistry University of Missouri Columbia, Missouri 65211 (for feedback: HarmataM@missouri.edu)

Reinhard Bruckner

Professor of Organic Chemistry
Institut für Organische Chemie und Biochemie
der Albert-Ludwigs-Universität
Albertstrasse 21
79104 Freiburg, Germany
(for feedback: reinhard.brueckner@organik.chemie.uni-freiburg.de)

April 16, 2001

Preface to the German Edition

To really understand organic chemistry requires three passes. First, one must familiarize oneself with the physical and chemical properties of organic chemical compounds. Then one needs to understand their reactivities and their options for reactions. Finally, one must develop the ability to design syntheses. A typical schedule of courses for chemistry students clearly incorporates these three components. Introductory courses focus on compounds, a course on reaction mechanisms follows, and a course on advanced organic chemistry provides more specialized knowledge and an introduction to retrosynthesis.

Experience shows that the *second* pass, the presentation of the material organized according to reaction mechanisms, is of central significance to students of organic chemistry. This systematic presentation reassures students not only that they can master the subject but also that they might enjoy studying organic chemistry.

I taught the reaction mechanisms course at the University of Göttingen in the winter semester of 1994, and by the end of the semester the students had acquired a competence in organic chemistry that was gratifying to all concerned. Later, I taught the same course again—I still liked its outline—and I began to wonder whether I should write a textbook based on this course. A text of this kind was not then available, so I presented the idea to Björn Gondesen, the editor of Spektrum. Björn Gondesen enthusiastically welcomed the book proposal and asked me to write the "little booklet" as soon as possible. I gave up my private life and wrote for just about two years. I am grateful to my wife that we are still married; thank you, Jutta!

To this day, it remains unclear whether Björn Gondesen used the term "little booklet" in earnest or merely to indicate that he expected *one* book rather than a series of volumes. In any case, I am grateful to him for having endured patiently the mutations of the "little booklet" first to a "book" and then to a "mature textbook." In fact, the editor demonstrated an indestructible enthusiasm, and he remained supportive when I presented him repeatedly with increases in the manuscript of yet another 50 pages. The reader has Björn Gondesen to thank for the two-color production of this book. All "curved arrows" that indicate electron shifts are shown in red so that the student can easily grasp the reaction. Definitions and important statements also are graphically highlighted.

In comparison to the preceding generation, students of today study chemistry with a big handicap: an explosive growth of knowledge in all the sciences has been accompanied in particular by the need for students of organic chemistry to learn a greater number of reactions than was required previously. The omission of older knowledge is possible only if that knowledge has become less relevant and, for this reason, the following reactions were omitted: Darzens glycidic ester synthesis, Cope elimination, S_N i reaction, iodoform reaction, Reimer–Tiemann reaction, Stobble condensation, Perkin synthesis, benzoin condensation, Favorskii rearrangement, benzil–benzilic acid rearrangement, Hofmann and Lossen degradation, Meerwein–Ponndorf reduction, and Cannizarro re-

action. A few other reactions were omitted because they did not fit into the current presentation (nitrile and alkyne chemistry, cyanohydrin formation, reductive amination, Mannich reaction, enol and enamine reactions).

This book is a highly modern text. All the mechanisms described concern reactions that are used today. The mechanisms are not just *l'art pour l'art*. Rather, they present a conceptual tool to facilitate the learning of reactions that one needs to know in any case. Among the modern reactions included in the present text are the following: Barton–McCombie reaction, Mitsunobu reaction, Mukaiyama redox condensations, asymmetric hydroboration, halolactonizations, Sharpless epoxidation, Julia–Lythgoe and Peterson olefination, *ortho*-lithiation, *in situ* activation of carboxylic acids, preparations and reactions of Gilman, Normant, and Knochel cuprates, alkylation of chiral enolates (with the methods by Evans, Helmchen, and Enders), diastereoselective aldol additions (Heathcock method, Zimmerman–Traxler model), Claisen–Ireland rearrangements, transition metal–mediated C,C-coupling reactions, Swern and Dess-Martin oxidations, reductive lithiations, enantioselective carbonyl reductions (Noyori, Brown, and Corey–Itsuno methods), and asymmetrical olefin hydrogenations.

The presentations of many reactions integrate discussions of stereochemical aspects. Syntheses of mixtures of stereoisomers of the target molecule no longer are viewed as valuable—indeed such mixtures are considered to be worthless—and the control of the stereoselectivity of organic chemical reactions is of paramount significance. Hence, suitable examples were chosen to present aspects of modern stereochemistry, and these include the following: control of stereoselectivity by the substrate, the reagent, or an ancilliary reagent; double stereodifferentiation; induced and simple diastereoselectivity; Cram, Cram chelate, and Felkin–Anh selectivity; asymmetric synthesis; kinetic resolution; and mutual kinetic resolution.

You might ask how then, for heaven's sake, is one to remember all of this extensive material? Well, the present text contains only about 70% of the knowledge that I would expect from a really well-trained undergraduate student; the remaining 30% presents material for graduate students. To ensure the best orientation of the reader, the sections that are most relevant for optimal undergraduate studies are marked in the margin with a B on a gray background, and sections relevant primarily to graduate students are marked with an A on a red background. I have worked most diligently to show the reactions in reaction diagrams that include every intermediate—and in which the flow of the valence electrons is highlighted in color—and, whenever necessary, to further discuss the reactions in the text. It has been my aim to describe all reactions so well, that in hindsight—because the course of every reaction will seem so plausible—the readers feel that they might even have predicted their outcome. I tried especially hard to realize this aim in the presentation of the chemistry of carbonyl compounds. These mechanisms are presented in four chapters (Chapters 7-11), while other authors usually cover all these reactions in one chapter. I hope this pedagogical approach will render organic chemistry more comprehensible to the reader.

Finally, it is my pleasure to thank—in addition to my untiring editor—everybody who contributed to the preparation of this book. I thank my wife, Jutta, for typing "version 1.0" of most of the chapters, a task that was difficult because she is not a chemist and that at times became downright "hair raising" because of the inadequacy of my dicta-



Indicates relevance for undergraduate students



Indicates relevance for graduate students

tion. I thank my co-workers Matthias Eckhardt (University of Göttingen, Dr. Eckhardt by now) and Kathrin Brüschke (chemistry student at the University of Leipzig) for their careful reviews of the much later "version .10" of the chapters. Their comments and corrections resulted in "version .11" of the manuscript, which was then edited professionally by Dr. Barbara Elvers (Oslo). In particular, Dr. Elvers polished the language of sections that had remained unclear, and I am very grateful for her editing. Dr. Wolfgang Zettelmeier (Laaber-Waldetzenberg) prepared the drawings for the resulting "version .12," demonstrating great sensitivity to my aesthetic wishes. The typsesetting was accomplished essentially error-free by Konrad Triltsch (Würzburg), and my final review of the galley pages led to the publication of "version .13" in book form. The production department was turned upside-down by all the "last minute" changes—thank you very much, Mrs. Notacker! Readers who note any errors, awkward formulations, or inconsistencies are heartily encouraged to contact me. One of these days, there will be a "version .14."

It is my hope that your reading of this text will be enjoyable and useful, and that it might convince many of you to specialize in organic chemistry.

Reinhard Brückner Göttingen, August 8, 1996

Acknowledgments

My part in this endeavor is over. Now, it is entirely up to the staff at Harcourt/Academic Press to take charge of the final countdown that will launch Advanced Organic Chemistry: Reaction Mechanisms onto the English-speaking market. After three years of intense trans-Atlantic cooperation, it is my sincere desire to thank those individuals in the United States who made this enterprise possible. I am extremely obliged to Professor Michael Harmata from the University of Missouri at Columbia for the great determination he exhibited at all phases of the project. It was he who doggedly did the legwork at the 1997 ACS meeting in San Francisco, that is, cruised from one science publisher's stand to the next, dropped complimentary copies of the German edition on various desks, and talked fervently to the responsibles. David Phanco from Academic Press was immediately intrigued and quickly set up an agreement with the German publisher. David Phanco was farsighted enough to include Mike Harmata on board as a "language polisher" (of the translation) before he passed on the torch to Jeremy Hayhurst in what then was to become Harcourt/Academic Press. The latter's sympathetic understanding and constant support in the year to follow were absolutely essential to the final success of the project: Mike Harmata, at that time a Humboldt Fellow at the University of Göttingen, and I needed to develop a very Prussian sense of discipline when doing our best to match the first part of the translation to the quality of the original. I am very much indebted to Professor Rainer Glaser, who reinforced the Missouri team and, being bilingual, finished the second half of the translation skillfully and with amazing speed. He also contributed very valuably to improving the galley proofs, as did Joanna Dinsmore, Production Manager at Harcourt/Academic Press. It is she who deserves a great deal of gratitude for her diligence in countless hours of proofreading, and for her patience with an author who even at the page proof stage felt that it was never too late to make all sorts of small amendments for the future reader's sake. It is my sincere hope, Ms. Dinsmore, that in the end you, too, feel that this immense effort was worth the trials and tribulations that accompanied it.

> Reinhard Bruckner Freiburg, April 25, 2001

Contents

Forew	ord		xv	
Prefa	ce to the	English Edition	xvii	
Prefa	Preface to the German Edition			
Ackn	owledgm	ents	xxiii	
1	Radica	al Substitution Reactions at the Saturated C Atom	1	
1.1	Bondi	ng and Preferred Geometries in C Radicals, Carbenium		
	Ions a	nd Carbanions	1	
	1.1.1	Preferred Geometries	2	
	1.1.2	Bonding	3	
1.2	Stabili	ity of Radicals	5	
	1.2.1	Reactive Radicals	5	
	1.2.2	Unreactive Radicals	8	
1.3	Relati	ve Rates of Analogous Radical Reactions	9	
	1.3.1	J 1	10	
	1.3.2	The Hammond Postulate	11	
1.4	Radic	al Substitution Reactions: Chain Reactions	13	
1.5	Radic	al Initiators	15	
1.6		al Chemistry of Alkylmercury(II) Hydrides	16	
1.7	Radic	al Halogenation of Hydrocarbons	19	
	1.7.1	Simple and Multiple Chlorinations	19	
	1.7.2	Regioselectivity of Radical Chlorinations	22	
	1.7.3	Regioselectivity of Radical Brominations Compared		
		to Chlorinations	24	
	1.7.4	Rate Law for Radical Halogenations;		
		Reactivity/Selectivity Principle	26	
	1.7.5	Chemoselectivity of Radical Brominations	28	
1.8		ridations	32	
1.9	Defur	nctionalizations via Radical Substitution Reactions	34	
	1.9.1	Simple Defunctionalizations	34	
	1.9.2	Defunctionalization via 5-Hexenyl Radicals:		
		Competing Cyclopentane Formation	37	
	Refer	ences	40	
2	Nucle	ophilic Substitution Reactions at the Saturated C Atom	43	
2.1	Nucle	ophiles and Electrophiles; Leaving Groups	43	
2.2		and Poor Nucleophiles	44	

2.3	Leaving Gr	oups and the Quality of Leaving Groups	46
2.4	S _N 2 Reaction	ons: Kinetic and Stereochemical Analysis—	
	Substituent	Effects on Reactivity	49
	2.4.1 Ene	ergy Profile and Rate Law for S _N 2 Reactions:	
		action Order	49
	2.4.2 Ster	reochemistry of S _N 2 Substitutions	51
		Refined Transition State Model for the S _N 2 Reaction;	
		rossover Experiment and Endocyclic Restriction Test	52
		ostituent Effects on S _N 2 Reactivity	54
2.5		ons: Kinetic and Stereochemical Analysis;	31
		Effects on Reactivity	57
		ergy Profile and Rate Law of S _N 1 Reactions;	37
		ady State Approximation	57
		reochemistry of S _N 1 Reactions; Ion Pairs	61
		vent Effects on S_N 1 Reactivity	62
		ostituent Effects on S _N 1 Reactivity	65
2.6		-	0.5
2.0		S _N Reactions at Saturated C Atoms Take Place	
		to the S _N 1 Mechanism and When Do They Take	(0
2.7		rding to the S _N 2 Mechanism?	69
2.7		lar S _N Reactions That Do Not Take Place via Simple	
		Ion Intermediates: Neighboring Group Participation	69
		nditions for and Features of S _N Reactions with	
		ghboring Group Participation	69
		reased Rate through Neighboring Group Participation	71
		reoselectivity through Neighboring Group Participation	72
2.8	_	ely Useful S _N 2 Reactions: Alkylations	76
	References		81
3	Additions t	to the Olefinic C=C Double Bond	85
3.1	The Concep	pt of cis and trans Addition	86
3.2	Vocabulary	of Stereochemistry and Stereoselective Synthesis I	87
	3.2.1 Ison	merism, Diastereomers/Enantiomers, Chirality	87
		emoselectivity, Diastereoselectivity/Enantioselectivity,	
		reospecificity/Stereoconvergence	88
3.3		That Take Place Diastereoselectivity as cis Additions	91
		Cycloaddition Forming Three-Membered Rings	92
		ditions to C=C Double Bonds That Are Related to	, -
		cloadditions and Form Three-Membered Rings, Too	93
		Hydration of Olefins via the	75
		droboration/Oxidation/Hydrolysis Reaction Sequence	95
		terogeneously Catalyzed Hydrogenation	104
3.4		ective cis Additions to C=C Double Bonds	
		cabulary of Stereochemistry and Stereoselective	105
		ithesis II: Topicity, Asymmetric Synthesis	107
	Syn	tenesis 11. Topicity, Asymmetric Symmesis	106

	3.4.2	Asymmetric Hydroboration of Achiral Olefins	107
	3.4.3	Thought Experiment I on the Hydroboration of Chiral Olefins	
		with Chiral Boranes: Mutual Kinetic Resolution	109
	3.4.4	Thought Experiments II and III on the Hydroboration of	
		Chiral Olefins with Chiral Boranes: Reagent Control	
		of Diastereoselectivity, Matched/Mismatched Pairs,	
		Double Stereodifferentiation	110
	3.4.5	Thought Experiment IV on the Hydroboration of Chiral	
		Olefins with Chiral Dialkylboranes: Kinetic Resolution	112
	3.4.6	Catalytic Asymmetric Synthesis: Sharpless Oxidations	
		of Allyl Alcohols	113
3.5	Additi	ions That Take Place Diastereoselectively as trans	
		ions (Additions via Onium Intermediates)	116
	3.5.1	Addition of Bromine	117
	3.5.2	The Formation of Halohydrins; Halolactonization	
		and Haloetherification	118
	3.5.3	Solvomercuration of Olefins: Hydration of C=C	
		Double Bonds through Subsequent Reduction	121
3.6	Additi	ions That Take Place or Can Take Place without Stereocontrol	
		nding on the Mechanism	123
	3.6.1	Additions via Carbenium Ion Intermediates	123
	3.6.2		126
	Refere		126
4	8-Elin	ninations	129
	·		
4.1	Conce	epts of Elimination Reactions	129
4.1	Conce 4.1.1	The Concept of α,β - and $1,n$ -Elimination	129 129
4.1	Conce 4.1.1 4.1.2	The Concept of α,β - and $1,n$ -Elimination The Terms syn - and $anti$ -Elimination	
4.1	Conce 4.1.1	The Concept of α,β - and $1,n$ -Elimination	129
4.1	Conce 4.1.1 4.1.2 4.1.3	The Concept of α,β - and $1,n$ -Elimination The Terms syn - and $anti$ -Elimination When Are Stereogenic syn - and $anti$ -Selective Eliminations Stereoselective?	129
4.1	Conce 4.1.1 4.1.2	The Concept of α,β - and $1,n$ -Elimination The Terms syn - and $anti$ -Elimination When Are Stereogenic syn - and $anti$ -Selective	129 130
4.1	Conce 4.1.1 4.1.2 4.1.3	The Concept of α,β - and $1,n$ -Elimination The Terms syn - and $anti$ -Elimination When Are Stereogenic syn - and $anti$ -Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β -Elimination: Saytzeff and Hofmann Product(s)	129 130
4.1	Conce 4.1.1 4.1.2 4.1.3	The Concept of α,β - and $1,n$ -Elimination The Terms syn - and $anti$ -Elimination When Are Stereogenic syn - and $anti$ -Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β -Elimination:	129 130 131
4.1	Conce 4.1.1 4.1.2 4.1.3 4.1.4 4.1.5	The Concept of α,β - and $1,n$ -Elimination The Terms syn - and $anti$ -Elimination When Are Stereogenic syn - and $anti$ -Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β -Elimination: Saytzeff and Hofmann Product(s) The Synthetic Value of Het^1/Het^2 in Comparison to H/Het Eliminations	129 130 131
4.1	Conce 4.1.1 4.1.2 4.1.3 4.1.4 4.1.5	The Concept of α,β - and $1,n$ -Elimination The Terms syn - and $anti$ -Elimination When Are Stereogenic syn - and $anti$ -Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β -Elimination: Saytzeff and Hofmann Product(s) The Synthetic Value of Het^1/Het^2 in Comparison to	129 130 131 134
	Conce 4.1.1 4.1.2 4.1.3 4.1.4 4.1.5 β-Elin β-Elin	The Concept of α,β - and $1,n$ -Elimination The Terms syn - and $anti$ -Elimination When Are Stereogenic syn - and $anti$ -Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β -Elimination: Saytzeff and Hofmann Product(s) The Synthetic Value of Het^1/Het^2 in Comparison to H/Het Eliminations minations of H/Het via Cyclic Transition States minations of H/Het via Acyclic Transition States:	129 130 131 134 135
4.2	Conce 4.1.1 4.1.2 4.1.3 4.1.4 4.1.5 β-Elin β-Elin The M	The Concept of α,β- and 1,n-Elimination The Terms syn- and anti-Elimination When Are Stereogenic syn- and anti-Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β-Elimination: Saytzeff and Hofmann Product(s) The Synthetic Value of Het¹/Het² in Comparison to H/Het Eliminations minations of H/Het via Cyclic Transition States minations of H/Het via Acyclic Transition States: Mechanistic Alternatives	129 130 131 134 135
4.2	Conce 4.1.1 4.1.2 4.1.3 4.1.4 4.1.5 β-Elin β-Elin The M	The Concept of α,β - and $1,n$ -Elimination The Terms syn - and $anti$ -Elimination When Are Stereogenic syn - and $anti$ -Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β -Elimination: Saytzeff and Hofmann Product(s) The Synthetic Value of Het^1/Het^2 in Comparison to H/Het Eliminations minations of H/Het via Cyclic Transition States minations of H/Het via Acyclic Transition States:	129 130 131 134 135 136
4.2 4.3	Conce 4.1.1 4.1.2 4.1.3 4.1.4 4.1.5 β-Elin β-Elin The M	The Concept of α,β- and 1,n-Elimination The Terms syn- and anti-Elimination When Are Stereogenic syn- and anti-Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β-Elimination: Saytzeff and Hofmann Product(s) The Synthetic Value of Het¹/Het² in Comparison to H/Het Eliminations minations of H/Het via Cyclic Transition States minations of H/Het via Acyclic Transition States: Mechanistic Alternatives	129 130 131 134 135 136
4.2 4.3	Conce 4.1.1 4.1.2 4.1.3 4.1.4 4.1.5 β-Elin β-Elin The M E2 Eli	The Concept of α,β- and 1,n-Elimination The Terms syn- and anti-Elimination When Are Stereogenic syn- and anti-Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β-Elimination: Saytzeff and Hofmann Product(s) The Synthetic Value of Het¹/Het² in Comparison to H/Het Eliminations minations of H/Het via Cyclic Transition States minations of H/Het via Acyclic Transition States: Mechanistic Alternatives miniminations of H/Het and the E2/S _N 2 Competition	129 130 131 134 135 136 140 141
4.2 4.3	Conce 4.1.1 4.1.2 4.1.3 4.1.4 4.1.5 β-Elin β-Elin The M E2 Eli 4.4.1	The Concept of α,β- and 1,n-Elimination The Terms syn- and anti-Elimination When Are Stereogenic syn- and anti-Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β-Elimination: Saytzeff and Hofmann Product(s) The Synthetic Value of Het¹/Het² in Comparison to H/Het Eliminations minations of H/Het via Cyclic Transition States minations of H/Het via Acyclic Transition States: Mechanistic Alternatives minations of H/Het and the E2/S _N 2 Competition Substrate Effects on the E2/S _N 2 Competition	129 130 131 134 135 136 140 141 141
4.2 4.3	Conce 4.1.1 4.1.2 4.1.3 4.1.4 4.1.5 β-Elin β-Elin The M E2 Eli 4.4.1 4.4.2	The Concept of α,β- and 1,n-Elimination The Terms syn- and anti-Elimination When Are Stereogenic syn- and anti-Selective Eliminations Stereoselective? Formation of Regioisomeric Olefins by β-Elimination: Saytzeff and Hofmann Product(s) The Synthetic Value of Het¹/Het² in Comparison to H/Het Eliminations minations of H/Het via Cyclic Transition States minations of H/Het via Acyclic Transition States: Mechanistic Alternatives iminations of H/Het and the E2/S _N 2 Competition Substrate Effects on the E2/S _N 2 Competition Base Effects on the E2/S _N 2 Competition	129 130 131 134 135 136 140 141 141 143

viii Contents

4.5	E1 Eli	mination of H/Het from R_{tert} —X and the E1/S _N 1 Competition	149
	4.5.1	Energy Profiles and Rate Laws for E1 Eliminations	150
	4.5.2	The Regioselectivity of E1 Eliminations	152
	4.5.3	E1 Eliminations in Protecting Group Chemistry	155
4.6	E1 _{cb} E	Eliminations	156
	4.6.1	Unimolecular E1 _{cb} Eliminations: Energy Profile and	
		Rate Law	156
	4.6.2	Nonunimolecular E1 _{cb} Eliminations: Energy Profile and	
		Rate Law	157
	4.6.3	E1 _{cb} Eliminations in Protecting Group Chemistry	158
4.7	β-Elin	ninations of Het ¹ /Het ²	160
	4.7.1	Fragmentation of β -Heterosubstituted	
		Organometallic Compounds	161
	4.7.2	Julia-Lythgoe Synthesis of trans-Olefins	161
	4.7.3	Peterson Olefination	163
	4.7.4	Oxaphosphetane Fragmentation, Last Step of Wittig and	
		Horner-Wadsworth-Emmons Reactions	163
	4.7.5	Corey-Winter Reaction	165
	Refere		166
5	Substi	tution Reactions on Aromatic Compounds	169
5.1	Electr	ophilic Aromatic Substitutions via Wheland Complexes	
	("Ar-S	SE Reactions")	169
	5.1.1	Mechanism: Substitution of H ⁺ vs ipso-Substitution	169
	5.1.2	Thermodynamic Aspects of Ar-S _E Reactions	171
	5.1.3	Kinetic Aspects of Ar-S _E Reactions: Reactivity	
		and Regioselectivity in Reactions of Electrophiles with	
		Substituted Benzenes	176
5.2	$Ar-S_E$	Reactions via Wheland Complexes: Individual Reactions	182
	5.2.1	Ar—Hal Bond Formation by Ar-S _E Reaction	182
	5.2.2	Ar—SO ₃ H Bond Formation by Ar-S _E Reaction	184
	5.2.3	Ar-NO ₂ Bond Formation by Ar-S _E Reaction	186
	5.2.4	Ar—N=N Bond Formation by Ar-S _E Reaction	188
	5.2.5	Ar—Alkyl Bond Formations by Ar-S _E Reaction	189
	5.2.6	Ar—C(OH) Bond Formation by Ar-S _E Reactions	
		and Associated Secondary Reactions	196
	5.2.7	Ar—C(=O) Bond Formation by Ar-S _E Reaction	197
	5.2.8	Ar—C(=O)H Bond Formation through Ar-S _E Reaction	199
5.3	Electro	ophilic Substitution Reactions on Metallated	
	Aroma	atic Compounds	200
	5.3.1	Electrophilic Substitution Reactions of ortho-Lithiated	
		Benzene and Naphthalene Derivatives	201

Contents ix

	5.3.2	Electrophilic Substitution Reactions in Aryl Grignard and Aryllithium Compounds That Are Accessible from	
		Aryl Halides	203
	5.3.3	Electrophilic Substitutions on Arylboronic Acids and	203
	0.0.0	Arylboronic Esters	206
5.4	Nucle	ophilic Substitution Reactions in Aryldiazonium Salts	207
5.5		ophilic Substitution Reactions via	
		nheimer Complexes	211
	5.5.1	Mechanism	211
	5.5.2	Examples of Reactions of Preparative Interest	213
	5.3.3	A Special Mechanistic Case: Reactions of Aryl Sulfonates	
		with NaOH/KOH in a Melt	215
5.6	Nucle	ophilic Aromatic Substitution via Arynes, cine Substitution	216
	Refere	· -	217
6		ophilic Substitution Reactions on the Carboxyl Carbon	221
	(Exce	pt through Enolates)	
6.1		-Containing Substrates and Their Reactions with Nucleophiles	221
6.2		anisms, Rate Laws, and Rate of Nucleophilic	
		tution Reactions at the Carboxyl Carbon	224
	6.2.1	Mechanism and Rate Laws of S _N Reactions on the	
		Carboxyl Carbon	224
	6.2.2	S _N Reactions on the Carboxyl Carbon: The Influence of	
		Resonance Stabilization of the Attacked C=O Double Bond	
		on the Reactivity of the Acylating Agent	230
	6.2.3	S _N Reactions on the Carboxyl Carbon: The Influence of the	
		Stabilization of the Tetrahedral Intermediate on the Reactivity	234
6.3		ation of Carboxylic Acids and of Carboxylic Acid Derivatives	236
	6.3.1	Activation of Carboxylic Acids and Carboxylic Acid	
	(22	Derivatives in Equilibrium Reactions	237
	6.3.2	Conversion of Carboxylic Acids into Isolable Acylating Agents	238
<i>c</i>	6.3.3	Complete in Situ Activation of Carboxylic Acids	240
6.4		ed S _N Reactions of Heteroatom Nucleophiles on the	
		oxyl Carbon	244
	6.4.1	Hydrolysis of Esters	246
	6.4.2	Lactone Formation from Hydroxycarboxylic Acids	250
	6.4.3	Forming Peptide Bonds	254
	6.4.4	S _N Reactions of Heteroatom Nucleophiles with	25.
65	6 D.	Carbonic Acid Derivatives	256
6.5		eactions of Hydride Donors, Organometallics, and	2.50
		oatom-Stabilized "Carbanions" on the Carboxyl Carbon	260
	6.5.1	When Do Pure Acylations Succeed, and When Are	0
		Alcohols Produced?	260

	6.5.2	Acylation of Hydride Donors: Reduction of Carboxylic Acid Derivatives to Aldehydes	262
	6.5.3	Aculation of Organometallic Compounds and	263
	0.5.5	Heteroatom-Stabilized "Carbanions": Synthesis of Ketones	265
	Refere		268
	Refere		208
7	Additi	ons of Heteroatom Nucleophiles to Heterocumulenes.	
		ons of Heteroatom Nucleophiles to Carbonyl Compounds	
	and Fo	ollow-up Reactions	271
7.1	Additi	ons of Heteroatom Nucleophiles to Heterocumulenes	271
	7.1.1	Mechanism of the Addition of Heteroatom Nucleophiles to	
		Heterocumulenes	271
	7.1.2	Examples of the Addition of Heteroatom Nucleophiles to	
7.2	A 44iti	Heterocumulenes	272
1.2	7.2.1	ons of Heteroatom Nucleophiles to Carbonyl Compounds On the Equilibrium Position of Addition Reactions of	279
	7.2.1	Heteroatom Nucleophiles to Carbonyl Compounds	270
	7.2.2	Hemiacetal Formation	279
	7.2.3	Oligomerization/Polymerization of Carbonyl Compounds	281 286
7.3		on of Heteroatom Nucleophiles to Carbonyl Compounds in	200
		nation with Subsequent S _N 1 Reactions: Acetalizations	288
	7.3.1	Mechanism	288
	7.3.2	Formation of O,O-Acetals	290
	7.3.3	Formation of S,S-Acetals	295
	7.3.4	Formation of N,N-Acetals	297
7.4	Additi	on of Nitrogen Nucleophiles to Carbonyl Compounds in	
	Combi	nation with Subsequent E1 Eliminations: Condensation	
	Reaction	ons of Nitrogen Nucleophiles with Carbonyl Compounds	299
	Refere	nces	303
8	Additio	on of Hydride Donors and Organometallic Compounds	
		bonyl Compounds	305
8.1	Suitabl	e Hydride Donors and Organometallic Compounds and	
	a Surve	ey of the Structure of Organometallic Compounds	305
8.2	Chemo	selectivity of the Addition of Hydride Donors to	-
	Carbor	nyl Compounds	307
8.3		reoselectivity of the Addition of Hydride Donors to	
		yl Compounds	309
	8.3.1	Diastereoselectivity of the Addition of Hydride Donors	
		to Cyclic Ketones	310

Contents xi

	8.3.2	Diastereoselectivity of the Addition of Hydride Donors	
		to α-Chiral Acyclic Carbonyl Compounds	313
	8.3.3	Diastereoselectivity of the Addition of Hydride Donors to	
		β-Chiral Acyclic Carbonyl Compounds	322
8.4	Enanti	oselective Addition of Hydride Donors to	
		nyl Compounds	323
8.5	Additi	on of Organometallic Compounds to Carbonyl Compounds	327
	8.5.1	Simple Addition Reactions of Organometallic Compounds	328
	8.5.2	Enantioselective Addition of Organozinc Compounds	
		to Carbonyl Compounds: Chiral Amplification	333
	8.5.3	Diastereoselective Addition of Organometallic Compounds to	
		Carbonyl Compounds	335
8.6	1,4-Ad	ditions of Organometallic Compounds to	
	α, β -Ui	nsaturated Ketones	339
	Refere	ences	342
9		on of Ylides with Saturated or $lpha,eta$ -Unsaturated nyl Compounds	347
	Carbo	iyi compounds	347
9.1	Ylides	/Ylenes	347
9.2	Reacti	ons of S Ylides with Saturated Carbonyl Compounds or	
		Michael Acceptors: Three-Membered Ring Formation	349
	9.2.1	Mechanism for the Formation of Cyclopropanes	
		and Epoxides	349
	9.2.2	Stereoselectivity and Regioselectivity of	
		Three-Membered Ring Formation from S Ylides	351
9.3	Conde	nsation of P Ylides with Carbonyl Compounds:	
	Wittig	Reaction	353
	9.3.1	Nomenclature and Preparation of P Ylides	354
	9.3.2	Mechanism of the Wittig Reaction	355
9.4	Horne	r-Wadsworth-Emmons Reaction	360
	9.4.1	Horner-Wadsworth-Emmons Reactions with	
		Achiral Substrates	361
	9.4.2	Horner-Wadsworth-Emmons Reactions between	
		Chiral Substrates: A Potpourri of Stereochemical Specialties	364
	Refere	ences	372
10	Chemi	stry of the Alkaline Earth Metal Enolates	373
10.1	Rocio (Considerations	272
10.1	10.1.1	Notation and Structure of Englates	373
	10.1.1	Preparation of Enolates by Deprotonation	373
	10.1.2	Other Methods for the Generation of Enolates	377
	10.1.3	one memors for the deheration of Englates	386

	10.1.4 Survey of Reactions between Electrophiles and Enolates	
	and the Issue of Ambidoselectivity	388
10.2	Alkylation of Quantitatively Prepared Enolates and Aza-Enolates;	
	Chain-Elongating Syntheses of Carbonyl Compounds and	
	Carboxylic Acid Derivatives	391
	10.2.1 Chain-Elongating Syntheses of Carbonyl Compounds	391
	10.2.2 Chain-Elongating Syntheses of Carboxylic Acid Derivatives	s 400
10.3	Hydroxyalkylation of Enolates with Carbonyl Compounds ("Aldol	
	Addition"): Synthesis of β -Hydroxyketones and β -Hydroxyesters	406
	10.3.1 Driving Force of Aldol Additions and Survey of	
	Reaction Products	406
	10.3.2 Stereocontrol	408
10.4	Condensation of Enolates with Carbonyl Compounds:	
	Synthesis of Michael Acceptors	414
	10.4.1 Aldol Condensations	414
	10.4.2 Knoevenagel Reaction	418
	10.4.3 A Knoevenagel Reaction "with a Twist"	419
10.5	Acylation of Enolates	422
	10.5.1 Acylation of Ester Enolates	422
	10.5.2 Acylation of Ketone Enolates	425
10.6	Michael Additions of Enolates	428
	10.6.1 Simple Michael Additions	428
	10.6.2 Tandem Reactions Consisting of Michael Addition	
	and Consecutive Reactions	430
	References	432
-1-1		10.0
11	Rearrangements	435
11.1	Nomenclature of Sigmatropic Shifts	435
11.2	Molecular Origins for the Occurrence of [1,2]-Rearrangements	436
11.3	[1,2]-Rearrangements in Species with a Valence Electron Sextet	438
	11.3.1 [1,1]-Rearrangements of Carbenium Ions	438
	11.3.2 [1,2]-Rearrangements in Carbenes or Carbenoids	453
11.4	[1,2]-Rearrangements without the Occurrence of a Sextet Intermed	
	11.4.1 Hydroperoxide Rearrangements	459
	11.4.2 Baeyer-Villiger Rearrangements	459
	11.4.3 Oxidation of Organoborane Compounds	462
	11.4.4 Beckmann Rearrangement	464
	11.4.5 Curtius Rearrangement	464
11.5	Claisen Rearrangement	467
	11.5.1 Classical Claisen Rearrangement	467
	11.5.2 Claisen-Ireland Rearrangements	468
	References	474