

Ullmann's Encyclopedia of Industrial Chemistry

Fifth, Completely Revised Edition

Volume A19:

Parkinsonism Treatment to Photoelectricity

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Parkinsonism Treatment to Photoelectricity

Editors: Barbara Elvers, Stephen Hawkins, Gail Schulz



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Symbols and Units

Symbols and units agree with SI standards (for conversion factors see pp. VIII–IX). The following list gives the most important symbols used in the encyclopedia. Articles with many specific units and symbols have a similar list as front matter.

| Symbol | Unit | Physical Quantity |
|------------|---|---|
| a_B | | activity of substance B |
| A_r | | relative atomic mass (atomic weight) |
| A | m^2 | area |
| c_B | mol/m^3 , mol/L (M) | concentration of substance B |
| C | C/V | electric capacity |
| c_p, c_v | $\text{J kg}^{-1} \text{K}^{-1}$ | specific heat capacity |
| d | cm , m | diameter |
| d | | relative density (ρ/ρ_{water}) |
| D | m^2/s | diffusion coefficient |
| D | Gy (= J/kg) | absorbed dose |
| e | | elementary charge |
| E | J | energy |
| E | V/m | electric field strength |
| E | V | electromotive force |
| E_A | J | activation energy |
| f | | activity coefficient |
| F | C/mol | Faraday constant |
| F | N | force |
| g | m/s^2 | acceleration due to gravity |
| G | J | Gibbs free energy |
| h | m | height |
| h | $\text{W} \cdot \text{s}^2$ | Planck constant |
| H | J | enthalpy |
| I | A | electric current |
| I | cd | luminous intensity |
| k | (variable) | rate constant of a chemical reaction |
| k | J/K | Boltzmann constant |
| K | (variable) | equilibrium constant |
| l | m | length |
| m | g , kg , t | mass |
| M_r | | relative molecular mass (molecular weight) |
| n_D^{20} | | refractive index (sodium D-light, 20 °C) |
| n | mol | amount of substance |
| N_A | mol^{-1} | Avogadro constant ($6.023 \times 10^{23} \text{ mol}^{-1}$) |
| p | Pa ; bar^* | pressure |
| Q | J | quantity of heat |
| r | m | radius |
| R | $\text{J K}^{-1} \text{mol}^{-1}$ | gas constant |
| R | Ω | electric resistance |
| S | J/K | entropy |
| t | s , min , h , d , month , a | time |
| t | $^{\circ}\text{C}$ | temperature |
| T | K | absolute temperature |
| u | m/s | velocity |

* The official unit of pressure is the pascal (Pa).

Symbols and units (continued from p. VII)

| Symbol | Unit | Physical Quantity |
|------------|--|--|
| U | V | electric potential |
| U | J | internal energy |
| V | m ³ , L, mL | volume |
| w | | mass fraction |
| W | J | work |
| x_B | | mole fraction of substance B |
| α | | cubic expansion coefficient |
| α | W m ⁻² K ⁻¹ | heat-transfer coefficient (heat-transfer number) |
| α | | degree of dissociation of electrolyte |
| $[\alpha]$ | 10 ⁻² deg cm ² g ⁻¹ | specific rotation |
| η | Pa · s | dynamic viscosity |
| θ | °C | temperature |
| κ | | c_p/c_v |
| λ | W m ⁻¹ K ⁻¹ | thermal conductivity |
| λ | nm, m | wavelength |
| μ | | chemical potential |
| ν | Hz, s ⁻¹ | frequency |
| ν | m ² /s | kinematic viscosity (η/ρ) |
| π | Pa | osmotic pressure |
| ρ | g/cm ³ | density |
| σ | N/m | surface tension |
| τ | Pa (N/m ²) | shear stress |
| φ | | volume fraction |
| χ | Pa ⁻¹ (m ² /N) | compressibility |

Conversion Factors

| SI unit | Non-SI unit | From SI to non-SI multiply by |
|--------------------|-----------------------|------------------------------------|
| <i>Mass</i> | | |
| kg | pound (avoirdupois) | 2.205 |
| kg | ton (long) | 9.842×10^{-4} |
| kg | ton (short) | 1.102×10^{-3} |
| <i>Volume</i> | | |
| m ³ | cubic inch | 6.102×10^4 |
| m ³ | cubic foot | 35.315 |
| m ³ | gallon (U.S., liquid) | 2.642×10^2 |
| m ³ | gallon (Imperial) | 2.200×10^2 |
| <i>Temperature</i> | | |
| °C | °F | $(^\circ\text{C} \times 1.8) + 32$ |
| <i>Force</i> | | |
| N | dyne | 1.0×10^5 |

Conversion factors (continued from p. VIII)

| SI unit | Non-SI unit | From SI to non-SI multiply by |
|---------------------|-------------|-------------------------------|
| <i>Energy, Work</i> | | |
| J | Btu (int.) | 9.480×10^{-4} |
| J | cal (int.) | 2.389×10^{-1} |
| J | eV | 6.242×10^{18} |
| J | erg | 1.0×10^7 |
| J | kW · h | 2.778×10^{-7} |
| J | kp · m | 1.020×10^{-1} |
| <i>Pressure</i> | | |
| MPa | at | 10.20 |
| MPa | atm | 9.869 |
| MPa | bar | 10 |
| kPa | mbar | 10 |
| kPa | mm Hg | 7.502 |
| kPa | psi | 0.145 |
| kPa | torr | 7.502 |

Prefixes for Powers of Ten

| | | | |
|--------------------|------------------|-------------------------|----------------------|
| E (exa) 10^{18} | M (mega) 10^6 | d (deci) 10^{-1} | n (nano) 10^{-9} |
| P (peta) 10^{15} | k (kilo) 10^3 | c (centi) 10^{-2} | p (pico) 10^{-12} |
| T (tera) 10^{12} | h (hecto) 10^2 | m (milli) 10^{-3} | f (femto) 10^{-15} |
| G (giga) 10^9 | | μ (micro) 10^{-6} | a (atto) 10^{-18} |

Abbreviations

The following is a list of the abbreviations used in the text. Common terms, the names of publications and institutions, and legal agreements are included along with their full identities. Other abbreviations will be defined wherever they first occur in an article. For further abbreviations, see p. VII (Symbols and Units), p. XIV (Companies and Country Codes in Patent References). The names of periodical publications are abbreviated exactly as done by Chemical Abstracts Service.

| | | |
|-------|---|---|
| abs. | absolute | of dangerous goods on the Rhine and all national waterways of the countries concerned) |
| a.c. | alternating current | |
| ACGIH | American Conference of Governmental Industrial Hygienists | |
| ACS | American Chemical Society | ADP adenosine 5'-diphosphate |
| ADI | acceptable daily intake | ADR accord européen relatif au transport international des marchandises dangereuses par route (European agreement concerning the international transportation of dangerous goods by inland waterways) |
| ADN | accord européen relatif au transport international des marchandises dangereuses par voie de navigation intérieure (European agreement concerning the international transportation of dangerous goods by inland waterways) | AEC Atomic Energy Commission (United States) |
| ADNR | ADN par le Rhin (regulation concerning the transportation | AIChE American Institute of Chemical Engineers |

| | | | |
|-----------|--|-------------|---|
| AIME | American Institute of Mining, Metallurgical, and Petroleum Engineers | DAB 9 | Deutsches Arzneibuch, 9th ed., Deutscher Apotheker-Verlag, Stuttgart 1986 |
| AMP | adenosine 5'-monophosphate | d.c. | direct current |
| APhA | American Pharmaceutical Association | decomp. | decompose, decomposition |
| API | American Petroleum Institute | DFG | Deutsche Forschungsgemeinschaft (German Science Foundation) |
| ASTM | American Society for Testing and Materials | dil. | dilute, diluted |
| ATP | adenosine 5'-triphosphate | DIN | Deutsche Industrie Norm (Federal Republic of Germany) |
| BAM | Bundesanstalt für Materialprüfung (Federal Republic of Germany) | DMF | dimethylformamide |
| BAT | Biologischer Arbeitsstoff-Toleranzwert (biological tolerance value for a working material, established by MAK commission, see MAK) | DNA | deoxyribonucleic acid |
| Beilstein | Beilstein's Handbook of Organic Chemistry, Springer, Berlin Heidelberg New York | DOE | Department of Energy (United States) |
| BFT | Brunauer Emmett Teller | DOT | Department of Transportation Materials Transportation Bureau (United States) |
| BGBI. | Bundesgesetzblatt (Federal Republic of Germany) | DTA | differential thermal analysis |
| BIOS | British Intelligence Objectives Subcommittee Report (see also FIAT) | EC | European Community |
| BOD | biological oxygen demand | ed. | editor, editors, edition, edited |
| <i>bp</i> | boiling point | e.g. | for example |
| B.P. | British Pharmacopeia | emf | electromotive force |
| BS | British Standard | EN | European Standard (European Community) |
| ca. | circa | EPA | Environmental Protection Agency (United States) |
| calcd. | calculated | EPR | electron paramagnetic resonance |
| CAS | Chemical Abstracts Service | Eq. | equation |
| cat. | catalyst; catalyzed | ESCA | electron spectroscopy for chemical analysis |
| cf. | compare | esp. | especially |
| CFR | Code of Federal Regulations (United States) | ESR | electron spin resonance |
| Chap. | chapter | Et | ethyl substituent ($-C_2H_5$) |
| ChemG | Chemikaliengesetz (Federal Republic of Germany) | et al. | and others |
| C.I. | Colour Index | etc. | et cetera |
| CIOS | Combined Intelligence Objectives Subcommittee Report (see also FIAT) | EVO | Eisenbahnverkehrsordnung (Federal Republic of Germany) |
| CNS | central nervous system | exp (...) | $e^{(\dots)}$, mathematical exponent |
| Co. | Company | FAO | Food and Agriculture Organization (United Nations) |
| COD | chemical oxygen demand | FDA | Food and Drug Administration (United States) |
| conc. | concentrated | FD & C | Food, Drug and Cosmetic Act (United States) |
| const. | constant | FHSA | Federal Hazardous Substances Act (United States) |
| Corp. | Corporation | FIAT | Field Information Agency, Technical (United States reports on the chemical industry in Germany, 1945) |
| crit. | critical | Fig. | figure |
| CTFA | The Cosmetic, Toiletry and Fragrance Association (United States) | <i>fp</i> | freezing point |
| | | Friedländer | P. Friedländer, Fortschritte der Teerfarbenfabrikation und verwandter Industriezweige, |

| | | | |
|-------------|--|-------------------|--|
| | Vol. 1 - 25, Springer, Berlin 1888 - 1942 | i.m. | intramuscular |
| FT | Fourier transform | IMDG | International Maritime Dangerous Goods Code |
| (g) | gas, gaseous | IMO | Inter-Governmental Maritime Consultative Organization (in the past: IMCO) |
| GC | gas chromatography | Inst. | Institute |
| GefStoffV | Gefahrstoffverordnung (regulations in the Federal Republic of Germany concerning hazardous substances) | i.p. | intraperitoneal |
| | | IR | infrared |
| GGVE | Verordnung in der Bundesrepublik Deutschland über die Beförderung gefährlicher Güter mit der Eisenbahn (regulation in the Federal Republic of Germany concerning the transportation of dangerous goods by rail) | ISO | International Organization for Standardization |
| | | IUPAC | International Union of Pure and Applied Chemistry |
| GGVS | Verordnung in der Bundesrepublik Deutschland über die Beförderung gefährlicher Güter auf der Straße (regulation in the Federal Republic of Germany concerning the transportation of dangerous goods by road) | i.v. | intravenous |
| | | Kirk-Othmer | Encyclopedia of Chemical Technology, 3rd ed., J. Wiley & Sons, New York - Chichester - Brisbane - Toronto 1978 - 1984 |
| GGVSee | Verordnung in der Bundesrepublik Deutschland über die Beförderung gefährlicher Güter mit Seeschiffen (regulation in the Federal Republic of Germany concerning the transportation of dangerous goods by sea-going vessels) | (l) | liquid |
| | | Landolt-Börnstein | Zahlenwerte u. Funktionen aus Physik, Chemie, Astronomie, Geophysik u. Technik, Springer, Heidelberg 1950 - 1980; Zahlenwerte und Funktionen aus Naturwissenschaften und Technik, Neue Serie, Springer, Heidelberg, since 1961 |
| GLC | gas-liquid chromatography | LC ₅₀ | lethal concentration |
| Gmelin | Gmelin's Handbook of Inorganic Chemistry, 8th ed., Springer, Berlin - Heidelberg - New York | LCLo | lowest published lethal concentration |
| GRAS | generally recognized as safe | LD ₅₀ | lethal dose |
| Hal | halogen substituent (-F, -Cl, -Br, -I) | LDLo | lowest published lethal dose |
| Houben-Weyl | Methoden der organischen Chemie, 4th ed., Georg Thieme Verlag, Stuttgart | ln | logarithm (base e) |
| HPLC | high performance liquid chromatography | LNG | liquefied natural gas |
| IARC | International Agency for Research on Cancer, Lyon, France | log | logarithm (base 10) |
| IAEA | International Atomic Energy Agency | LPG | liquefied petroleum gas |
| IATA-DGR | International Air Transport Association, Dangerous Goods Regulations | M | mol L |
| ICAO | International Civil Aviation Organization | M | metal (in chemical formulas) |
| i.e. | that is | MAK | Maximale Arbeitsplatz-Konzentration (maximum concentration at the workplace in the Federal Republic of Germany); cf. Deutsche Forschungsgemeinschaft (ed.): Maximale Arbeitsplatzkonzentrationen (MAK) und Biologische Arbeitsstoff-Toleranz-Werte (BAT), VCH Verlagsgesellschaft, Weinheim (published annually) |
| | | max. | maximum |
| | | MCA | Manufacturing Chemists Association (United States) |

| | | | |
|------------|---|------------------|--|
| Me | methyl substituent ($-CH_3$) | q. v. | which see (quod vide) |
| Methodicum | Methodicum Chemicum, Georg Thieme Verlag, Stuttgart | ref. | refer, reference |
| Chimicum | | resp. | respectively |
| MIK | maximale Immissionskonzentration (maximum immission concentration) | R_f | retention factor (TLC) |
| min. | minimum | R. H. | relative humidity |
| mp | melting point | RID | règlement international concernant le transport des marchandises dangereuses par chemin de fer (international convention concerning the transportation of dangerous goods by rail) |
| MS | mass spectrum, mass spectrometry | RNA | ribonucleic acid |
| NAS | National Academy of Sciences (United States) | R phrase | risk phrase according to ChemG and GefStoffV (Federal Republic of Germany) |
| NASA | National Aeronautics and Space Administration (United States) | (R-Satz) | |
| NBS | National Bureau of Standards (United States) | rpm | revolutions per minute |
| NCTC | National Collection of Type Cultures (United States) | RTECS | Registry of Toxic Effects of Chemical Substances, edited by the National Institute of Occupational Safety and Health (United States) |
| NIH | National Institutes of Health (United States) | (s) | solid |
| NIOSH | National Institute for Occupational Safety and Health (United States) | SAE | Society of Automotive Engineers (United States) |
| NMR | nuclear magnetic resonance | s.c. | subcutaneous |
| no. | number | SI | International System of Units |
| NRC | Nuclear Regulatory Commission (United States) | SIMS | secondary ion mass spectrometry |
| NRDC | National Research Development Corporation (United States) | S phrase | safety phrase according to ChemG and GefStoffV (Federal Republic of Germany) |
| NSC | National Service Center (United States) | (S-Satz) | |
| NSF | National Science Foundation (United States) | STEL | Short Term Exposure Limit (see TLV) |
| NTSB | National Transportation Safety Board (United States) | STP | standard temperature and pressure (0 °C, 101.325 kPa) |
| OECD | Organization for Economic Cooperation and Development | T_g | glass transition temperature |
| OSHA | Occupational Safety and Health Administration (United States) | TA Luft | Technische Anleitung zur Reinhaltung der Luft (clean air regulation in Federal Republic of Germany) |
| p., pp. | page, pages | TA Lärm | Technische Anleitung zum Schutz gegen Lärm (low noise regulation in Federal Republic of Germany) |
| Patty | G. D. Clayton, F. E. Clayton (eds.): Patty's Industrial Hygiene and Toxicology, 3rd ed., Wiley Interscience, New York | TDL _o | lowest published toxic dose |
| PB report | Publication Board Report (U.S. Department of Commerce, Scientific and Industrial Reports) | THF | tetrahydrofuran |
| PEL | permitted exposure limit | TLC | thin layer chromatography |
| Ph | phenyl substituent ($-C_6H_5$) | TLV | Threshold Limit Value (TWA and STEL); published annually by the American Conference of Governmental Industrial Hygienists (ACGIH), Cincinnati, Ohio |
| Ph. Fur. | European Pharmacopoeia, 2nd. ed., Council of Europe, Strasbourg 1981 | TOD | total oxygen demand |
| phr | part per hundred rubber (resin) | TRK | Technische Richtkonzentration (lowest technically feasible level) |
| PNS | peripheral nervous system | TSCA | Toxic Substances Control Act (United States) |
| ppm | parts per million | | |

| | | | |
|---------|---|-----------------------|---|
| TÜV | Technischer Überwachungsverein (Technical Control Board of the Federal Republic of Germany) | | derung brennbarer Flüssigkeiten (regulation in the Federal Republic of Germany concerning the con- struction and operation of plants for storage, filling, and transpor- tation of flammable liquids, clas- sification according to the flash point of liquids, recently in ac- cordance with the classification in the United States) |
| TWA | Time Weighted Average | | |
| Ullmann | Ullmanns Encyklopädie der Tech- nischen Chemie, 4th ed., Verlag Chemie, Weinheim 1972-1984; 3rd ed., Urban und Schwarzen- berg, München 1951-1970 | | |
| USAEC | United States Atomic Energy Commission | VDE | Verband Deutscher Elektro- ingenieure (Federal Republic of Germany) |
| USAN | United States Adopted Names | VDI | Verein Deutscher Ingenieure (Federal Republic of Germany) |
| USD | United States Dispensatory | vol | volume |
| USDA | United States Department of Agriculture | vol. | volume (of a series of books) |
| U.S.P. | United States Pharmacopeia | vs. | versus |
| UV | ultraviolet | WHO | World Health Organization (United Nations) |
| UVV | Unfallverhütungsvorschriften der Berufsgenossenschaft (workplace safety regulations in the Federal Republic of Germany) | Winnacker- Küchler | Chemische Technologie, Carl Hanser Verlag, München |
| VbF | Verordnung in der Bundesrepublik Deutschland über die Errichtung und den Betrieb von Anlagen zur Lagerung, Abfüllung und Beför- | wt | weight |
| | | \$ | U.S. dollar, unless otherwise stated |

Abbreviations for the Names of Frequently Cited Companies

| | | | |
|----------------|---|---------------|--|
| Air Products | Air Products and Chemicals | ICI | Imperial Chemical Industries |
| Akzo | Algemene Koninklijke Zout | IFP | Institut Français du Pétrole |
| | Organon | INCO | International Nickel Company |
| Alcoa | Aluminum Company of America | 3M | Minnesota Mining and Manufacturing Company |
| Allied | Allied Corporation | Mitsubishi | Mitsubishi Chemical Industries |
| Amer. Cyanamid | American Cyanamid Company | Chemical | |
| BASF | BASF Aktiengesellschaft | Monsanto | Monsanto Company |
| Bayer | Bayer AG | Nippon | Nippon Shokubai Kagaku Kogyo |
| BP | British Petroleum Company | Shokubai | |
| Celanese | Celanese Corporation | PCUK | Pechiney Ugine Kuhlmann |
| Daicel | Daicel Chemical Industries | PPG | Pittsburg Plate Glass Industries |
| Dainippon | Dainippon Ink and Chemicals Inc | Searle | G.D. Searle & Company |
| | | SKF | Smith Kline & French Laboratories |
| Dow Chemical | The Dow Chemical Company | SNAM | Società Nazionale Metandotti |
| DSM | Dutch Staats Mijnen | | |
| Du Pont | E.I. du Pont de Nemours & Company | Sohio | Standard Oil of Ohio |
| | | Stauffer | Stauffer Chemical Company |
| Exxon | Exxon Corporation | Sumitomo | Sumitomo Chemical Company |
| FMC | Food Machinery & Chemical Corporation | Toray | Toray Industries Inc. |
| | | UCB | Union Chimique Belge |
| GAF | General Aniline & Film Corporation | Union Carbide | Union Carbide Corporation |
| | | UOP | Universal Oil Products Company |
| W.R. Grace | W.R. Grace & Company | | |
| Hoechst | Hoechst Aktiengesellschaft | VEBA | Vereinigte Elektrizitäts- und Bergwerks-AG |
| IBM | International Business Machines Corporation | Wacker | Wacker Chemie GmbH |

Country Codes

The following list contains a selection of standard country codes used in the patent references

| | | | |
|----|---|----|--|
| AT | Austria | ID | Indonesia |
| AU | Australia | IL | Israel |
| BE | Belgium | IT | Italy |
| BG | Bulgaria | JP | Japan * |
| BR | Brazil | LU | Luxembourg |
| CA | Canada | MA | Morocco |
| CH | Switzerland | NL | Netherlands * |
| CS | Czechoslovakia | NO | Norway |
| DD | German Democratic Republic | NZ | New Zealand |
| DE | Federal Republic of Germany (and Germany before 1949) * | PL | Poland |
| | | PT | Portugal |
| DK | Denmark | SE | Sweden |
| ES | Spain | SU | Soviet Union |
| FI | Finland | US | United States of America |
| FR | France | YU | Yugoslavia |
| GB | United Kingdom | ZA | South Africa |
| GR | Greece | EP | European Patent Office * |
| HU | Hungary | WO | World Intellectual Property Organization |

* For Europe, Federal Republic of Germany, Japan, and the Netherlands, the type of patent is specified: EP (patent), EP-A (application), DE (patent), DE-OS (Offenlegungsschrift), DE-AS (Auslegeschrift), JP (patent), JP-Kokai (Kokai tokkyo koho), NL (patent), and NL-A (application).

Periodic Table of Elements

1A ("European" group designation according to old IUPAC recommendation)

1 ("group designation according to 1985 IUPAC proposal)

1A ("American" group designation, also used by the Chemical Abstracts Service until the end of 1986)

0
18
VIII A

| | | | | | | | | | | | | | | | | | |
|----------------|----------------|-----------------|----------------|----------------|----------------|-----------------|----------------|----------------|------------------|----------------|-----------------|------------------|-----------------|-----------------|-----------------|------------------|--------------------|
| 1.0079 1H | 2A 2 IIA | 3A 3 IIIA | 4A 4 IVA | 5A 5 VA | 6A 6 VIA | 7A 7 VIIA | 8 8 VIII | 9 9 VIII | 10 10 VIII | 11 11 IB | 12 12 IIB | 13 13 IIIA | 14 14 IVA | 15 15 VA | 16 16 VIA | 17 17 VIIA | 18 18 VIII A |
| 6.941 3Li | 9.0122 4Be | 24.305 9B | 47.88 12Mg | 50.942 13Al | 51.996 14Si | 54.938 15P | 55.847 16S | 58.933 17Cl | 58.69 18Ar | 63.546 19K | 65.39 20Ca | 69.723 21Sc | 72.61 22Ti | 74.922 23V | 78.96 24Cr | 79.904 25Mn | 83.80 26Fe |
| 22.990 11Na | 24.305 12Mg | 27.97 13Al | 28.086 14Si | 30.974 15P | 32.066 16S | 35.453 17Cl | 36.46 18Ar | 39.098 19K | 40.078 20Ca | 44.956 21Sc | 47.88 22Ti | 50.942 23V | 51.996 24Cr | 54.938 25Mn | 55.847 26Fe | 58.933 27Co | 58.69 28Ni |
| 39.098 19K | 40.078 20Ca | 44.956 21Sc | 47.88 22Ti | 50.942 23V | 51.996 24Cr | 54.938 25Mn | 55.847 26Fe | 58.933 27Co | 58.69 28Ni | 63.546 29Cu | 65.39 30Zn | 69.723 31Ga | 72.61 32Ge | 74.922 33As | 78.96 34Se | 79.904 35Br | 83.80 36Kr |
| 85.468 37Rb | 87.62 38Sr | 88.906 39Y | 91.224 40Zr | 92.906 41Nb | 95.94 42Mo | 98.906 43Tc | 101.07 44Ru | 102.91 45Rh | 106.42 46Pd | 107.87 47Ag | 112.41 48Cd | 114.82 49In | 118.71 50Sn | 121.75 51Sb | 127.60 52Te | 126.90 53I | 131.29 54Xe |
| 132.91 55Cs | 137.33 56Ba | 140.91 57La | 140.12 58Ce | 140.91 59Pr | 144.24 60Nd | 146.92 61Pm | 150.36 62Sm | 151.97 63Eu | 157.25 64Gd | 158.93 65Tb | 162.50 66Dy | 164.93 67Ho | 167.26 68Er | 168.93 69Tm | 173.04 70Yb | 174.97 71Lu | 222.02 86Rn |
| 223.02 87Fr | 226.03 88Ra | 227.03 89Ac | 232.04 90Th | 231.04 91Pa | 238.03 92U | 237.05 93Np | 244.06 94Pu | 243.06 95Am | 247.07 96Cm | 247.07 97Bk | 251.08 98Cf | 252.08 99Es | 257.10 100Fm | 258.10 101Md | 259.10 102No | 260.11 103Lr | |

* Elements with unstable isotopes; mass of most important isotope given

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| Pectin → Polysaccharides | Phosphanes → Phosphorus Compounds, Inorganic |
| Pencils → Drawing and Writing Materials | Phosphates, Inorganic → Phosphoric Acid and Phosphates |
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Parkinsonism Treatment

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1. Introduction

Parkinson's disease (PD) is one of the major degenerative neurological diseases afflicting humans, with a worldwide prevalence exceeding 1 per 1000 population. It is, however, primarily a disease of middle-aged and elderly individuals [1]. Thus, 1% of persons over 50 years of age is a sufferer, and more than 5% of those older than 60 years.

Early diagnosis is important if patients are to be relieved of both early and later symptoms with currently available therapies. Early diagnosis may also postpone development of the more distressing later manifestations of the disease. Although therapy alleviates symptoms, as well as restoring and sustaining a worthwhile quality of life for the patient, it does not appear to retard progression of the disease [2]–[5].

Clinical Diagnostic Criteria. The major clinical diagnostic criteria for PD are tremor, rigidity, akinesia, and bradyphrenia [6], [7]. Depression is the most frequent psychopathologic disorder. Reports on its frequency range from ca. 30 to 90%, strongly supporting the hypothesis that depression in idiopathic PD is due to endogenous neurophysiological defects, particularly a

reduction in the level of the neurotransmitter serotonin (5-hydroxytryptamine) in the brain.

Tremor is a major symptom; it usually begins unilaterally, often as the typical "pill-rolling" hand and arm activity. Tremor is present at rest, can be decreased or blocked by conscious will, but is aggravated by stress especially when directed motor actions are attempted.

Rigidity is reported as stiffness and cramplike pains in the limbs. A plastic resistance to passive movement is apparent on examination that, when coexisting with tremor, may appear as a "cogwheel effect".

Akinesia is the inability to translate an intended movement into real activity. The execution of movement is slow and is associated with postural changes in the neck and trunk, as well as a characteristic shuffling gait consisting of small hesitant steps that quicken to prevent falling. Coordinated simultaneous movement of the arms and legs is impossible.

Bradyphrenia is a slowness of cognitive processing associated with apathy, impaired concentration, and finally dementia. It may correlate positively with depression.

Other characteristics of PD are an expressionless masklike face with greasy skin, increased salivation, and impaired swallowing. Handwrit-

ing may become small, and speech difficulties produce a soft, monotonous voice. Sweating attacks, constipation, urinary frequency, and incontinence are not uncommon.

Pathophysiology of Parkinson's Disease. The central nervous system (CNS) defects associated with the development of PD are centered in the paired basal ganglia, including the striatum (caudate nucleus plus putamen), the globus pallidus (pallidum), the subthalamic nucleus, and the substantia nigra (Fig. 1) [8]–[10]. The latter is divided into two regions, the pars compacta and pars reticulata.

The basal ganglia belong to the extrapyramidal tract and are responsible for the transmission of cortical motor signals associated with voluntary movement (→ Neuropharmacology, A 16, pp. 140–141). Parkinson's disease is one of a number of extrapyramidal disorders (i.e., neurological disorders) that arise from lesions specifically affecting the basal ganglia. Other extrapyramidal disorders include Huntington's chorea, Wilson's disease, dystonia, dyskinesia, and torticollis. Clinically, they represent combinations of abnormal voluntary movements, and disturbances in postural stability and muscle tone, and are collectively termed movement disorders.

The substantia nigra is severely damaged in PD. The basal ganglia appear to operate together with the cerebellum as interfaces between the cognitive and motor systems of the brain, converting intentions into motor actions. Destruction of any of the components of the basal gan-

glia, like that which occurs in the substantia nigra in PD, consequently impairs muscle function and control.

The nerve fibers of the basal ganglia are parts of input, processing, and output circuits. The input pathway from the cortex to the basal ganglia contains the largest number of neuronal fibers in the basal ganglia system. Input connections run from the cortex to the striatum (Fig. 2). The striatum in turn sends nerve fibers to dopamine-secreting nerve cells (neurons) in the substantia nigra (pars compacta). The dopamine-secreting neurons connect back to cholinergic interneurons in the striatum. The output structures of the striatum include the pallidum and the subthalamic nucleus. From here, nerve fibers project to the motor cortex via the ventrolateral thalamus to complete the overall cortex–basal ganglia–cortex loop. This sophisticated feedback control system works through a finely tuned balance of excitatory and inhibitory neurotransmitters. All the nerve fibers leading from the striatum to the pallidum and from there to the ventrolateral thalamus are in effect inhibitory pathways that secrete the inhibitory transmitter γ -aminobutyric acid (GABA). The circular pathway cortex–basal ganglia–cortex is a feedback loop that is needed to stabilize motor control. Neurotransmission from the striatum to the substantia nigra also involves inhibitory GABA secretion at the substantia nigra nerve terminals.

The darkly staining melanophore cells in the substantia nigra send out nerve axons to the striatum to form the nigrostriatal pathway and employ dopamine as an inhibitory transmitter. Both GABA and dopamine normally control their own respective brain structures unless some disease or lesion specifically damages the pathways in which they are involved.

Another significant source of input to the striatum is the corticostriatal pathway derived from the neocortex that uses glutamate as its neurotransmitter. Glutamate is an important excitatory transmitter within the basal ganglia complex and is thought to be responsible for about half of the total input to the striatum. Acetylcholine, another excitatory neurotransmitter, is secreted from interneurons within the striatum. This interconnected system is maintained in balance via neuronal feedback control loops.

The clinical symptoms of parkinsonism are caused by destruction of the dopaminergic fibers of the darkly staining neuromelanin neurons in

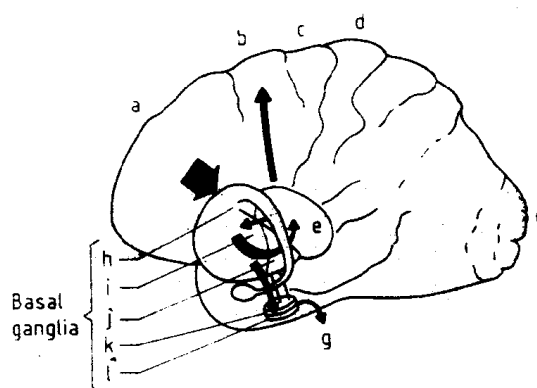


Figure 1. The basal ganglia and some of their connections [11]

a) Prefrontal cortex; b) Premotor cortex; c) Motor cortex; d) Sensory cortex; e) Thalamus; f) Visual cortex; g) Brainstem; h) Striatum (caudate nucleus); i) Striatum (putamen); j) Pallidum; k) Substantia nigra (pars compacta); l) Substantia nigra (pars reticulata)

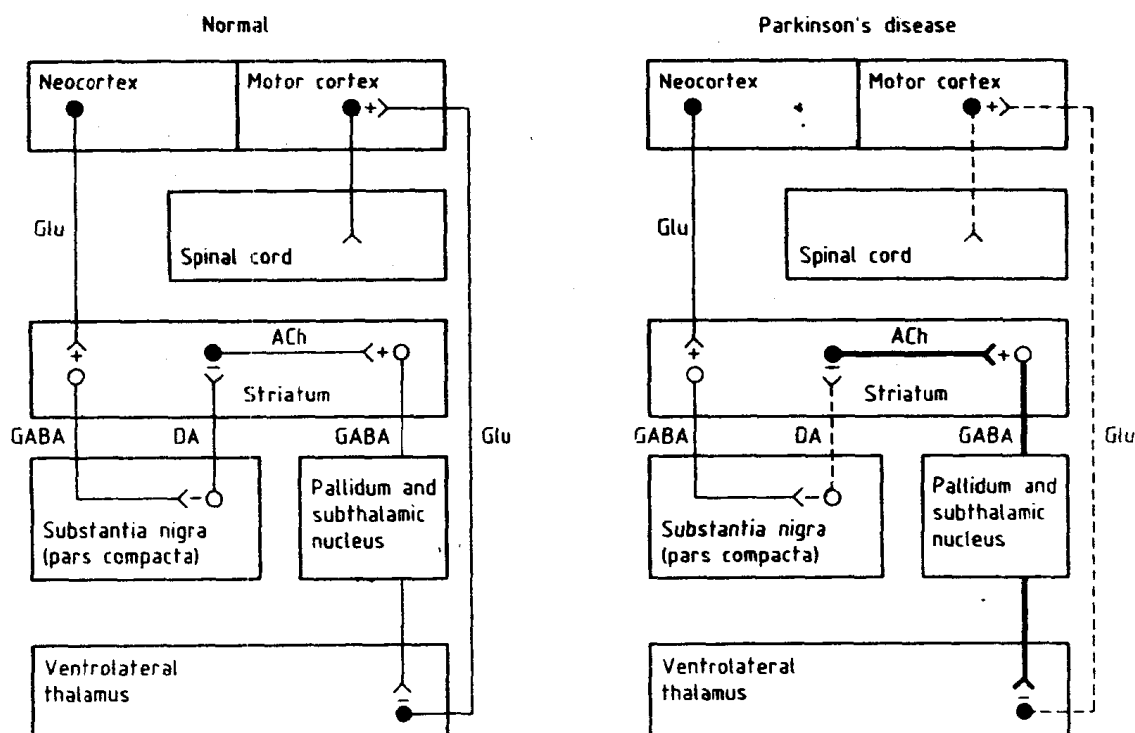


Figure 2. Schematic diagram illustrating the basal ganglia circuitry in normal humans and in patients suffering from Parkinson's disease (PD).

Excitatory (●) and inhibitory (○) neurons are shown with their axons, which form synaptic contacts with other nerve cells (+ excitatory synapse, - inhibitory synapse). Neurotransmitters secreted at the nerve terminals include γ -aminobutyric acid (GABA), dopamine (DA), glutamate (Glu), and acetylcholine (ACh). Broken and thick lines indicate reduced and increased activity, respectively. In PD, dopaminergic neurons of the substantia nigra (pars compacta) are destroyed, resulting in disinhibition (excitation) of cholinergic cells in the striatum. The overactivity of the striatal complex and its increased inhibitory output lead to a marked reduction of the activity of the thalamocortical feedback loop. These mechanisms can explain the symptoms of PD such as tremor, hypokinesia, or akinesia.

the substantia nigra (pars compacta) that project to the striatum (see Figs. 1 and 2). This destruction results in a dopamine deficiency. The cholinergic function in the striatum is simultaneously maintained, leading to an imbalance between inhibitory and excitatory neurotransmitters. A reduction in dopamine levels leads to excessive excitation by acetylcholine, which in turn augments the inhibitory (GABA) output of the striatum. As a consequence, the activity of the thalamocortical feedback loop is markedly reduced, leading to impaired motor function. Parkinson's disease can therefore be regarded as an overactivity of the striatal complex resulting from the loss of the fine control normally exercised by the neuromelanin-producing dopaminergic cells. By the time the first clinical symptoms (e.g., tremor or hypokinesia) become apparent, some 80% of these pigmented substantia nigral cells have already been destroyed.

Etiology of Parkinson's Disease. The cause of parkinsonism is not known with certainty. However, it is probably triggered by some environmental insult in early to middle life. As a result, the pigmented neurons are destroyed much sooner than in normal age-dependent degeneration. Pesticides used in rural areas (e.g., paraquat) may constitute such an environmental insult. However, the disease was known long before it was described by JAMES PARKINSON in 1817; thus, other environmental factors must also be responsible.

A recent hypothesis links PD to lipid oxidation involving excessive formation of oxygen-derived free radicals in the basal ganglia. The free radicals then progressively destroy dopamine-producing neurons under participation of the immune system. Thus, Parkinson's disease is regarded as a failure of scavenger molecules produced by the basal ganglia cells to

neutralize the harmful free radical products. Further acceleration may be caused by the enzyme monoamine oxidase-B (MAO-B), which catalyzes the metabolism of catecholamine neurotransmitters and simultaneously produces reactive oxygen species. Therefore, therapeutic inhibition of MAO-B may also inhibit neurodegeneration in PD. Of major interest in this context is parkinsonism induced by 1-methyl-4-phenyl-1,2,3,6-tetrahydropyridine [28289-54-5] (MPTP) because of its importance for new in vitro and in vivo pharmacological research models. This comparatively simple molecule selec-

tively destroys dopaminergic cells in the substantia nigra and induces advanced parkinsonian symptoms in humans within a few weeks. Thus, the possibility that an environmental toxin triggers the development of PD appears reasonable.

2. Therapy of Parkinson's Disease

Modern therapy started with the discovery that the brains of Parkinson patients were deficient in dopamine. Dopamine was originally re-

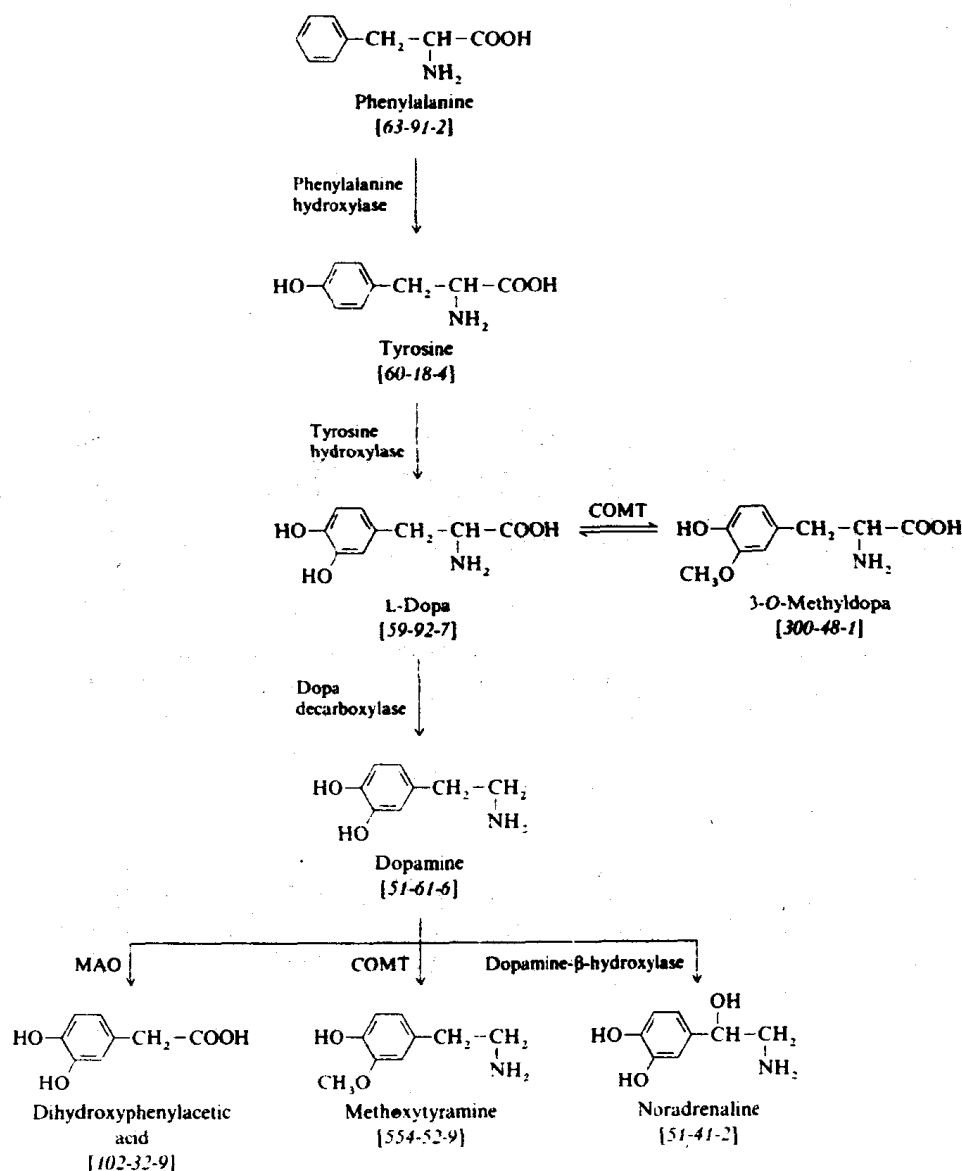


Figure 3. Biosynthesis and metabolism of dopamine
COMT = Catechol-*O*-methyl transferase; MAO = Monoamine oxidase