# ENCYCLOPEDIA OF CHEMICAL TECHNOLOGY VOLUME 12

# Encyclopedia OF CHEMICAL TECHNOLOGY

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VOLUME 12

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## ENCYCLOPEDIA OF CHEMICAL TECHNOLOGY VOLUME 12

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#### ABBREVIATIONS AND SYMBOLS

A.	Ångström unit(s)	A.S.M.E.	American Society of
A	anion; as, $HA$		Mechanical Engineers
abs.	absolute	A.S.T.M.	American Society for
a.c.	alternating current		Testing Materials
ac-	alicyclic; as, ac-derivatives of tetrahydro-	atm.	atmosphere(s), atmospheric
	naphthalene	at. no.	atomic number
A.C.S.	American Chemical Society	at. wt. av.	atomic weight average
addn.	addition	b. $(as, b_{11})$	boiling (at 11 mm.)
A.G.A.	American Gas Associ-	B	base; as, $B.2HCl$
	ation	bbl.	barrel(s)
A.I.Ch.E.	American Institute of	Bé.	Baumé
	Chemical Engineers	b.p.	boiling point
A.I.M.E.	American Institute of	$\mathbf{B}.\mathbf{t}.\mathbf{u}.$	British thermal unit(s)
	Mining and Metal-	bu.	bushel(s)
	lurgical Engineers	C.	centigrade
ale.	alcohol, alcoholic	C-	denoting attachment to
alk.	alkaline (not alkali)		carbon; as, C-alkyl
Alk	alkyl		derivatives of aniline
amp.	ampere(s)	cal.	calorie(s)
amphr.	$\operatorname{ampere-hour}(s)$	calcd.	$\operatorname{calculated}$
amt.	amount (noun)	c.f.m.	cubic foot (feet) per
anhyd.	anhydrous		$\mathbf{minute}$
A.P.I.	American Petroleum In-	cg.	centigram(s)
	stitute	c.g.s.	centimeter-gram-second
app.	apparatus	$\operatorname{chem}$ .	chemical
approx.	approximate (adj.), ap-	C.I.	$Colour\ Index\ { m no}.$
	proximately	cks.	centistokes
aq.	aqueous	c.l.	car lots
Ar	aryl	em.	centimeter(s)
ar-	aromatic; as, ar-deriva-	coeff.	coefficient
	tives of tetrahydro-	com.	commercial
	naphthalene	compd.	compound (noun)
as-	asymmetric; as, as-m-	compn.	composition
A C! A	xylidine	concd.	concentrated
ASA	American Standards As-	conen.	concentration
A.S.M.	sociation	cond.	conductivity
A.B.WI.	American Society for	const.	constant
	Metals	cor.	corrected

#### x ABBREVIATIONS AND SYMBOLS

	, , ,	cc	fall and a (magnet)
c.p.	chemically pure	ff.	following (pages)
cps.	centipoise	fig.	figure
erit.	critical	fl.oz.	fluid ounce(s)
cryst.	crystalline	f.o.b.	free on board
crystd.	crystallized	f.p.	freezing point
erystn.	crystallization	ft.	foot (feet)
cu.	cubic	ftlb.	foot-pound $(s)$
$d (as, d_4^{20})$	density (conveniently,	g.	$\operatorname{gram}(s)$
	specific gravity)	gal.	gallon(s)
d	differential operator	g.p.m.	gallons per minute
<i>d</i> -	dextro-, dextrorotatory	hp.	horsepower
D-	denoting configurational	hr.	hour(s)
	relationship, as to dex-	hyd.	hydrated, hydrous
	tro-glyceraldehyde	i.	insoluble
d.c.	direct current	i-	inactive; as, i-methio-
dec., decomp.	decompose(s)		nine
decompn.	decomposition	i.b.p.	initial boiling point
deriv.	derivative	I.C.C.	Interstate Commerce
detd.	determined		Commission
detn.	determination	I.D.	inner diameter
diam.	diameter	in.	inch(es)
dielec.	dielectric (adj.)	insol.	insoluble
dil.	dilute	I.P.T.	Institute of Petroleum
distd.	distilled		Technologists
distn.	distillation	I.U.	International Unit(s)
DL-, <i>dl</i> -	racemic	I.U.C.,	International Union of
dm.	decimeter	I.U.P.A.C.	Chemistry, Interna-
$e_{\perp}$	electron		tional Union of Pure
ed.	edition, editor		and Applied Chem-
elec.	electric, electrical		istry
elev.	elevated	j.	joule
e.m.f.	electromotive force	K.	Kelvin
eng.	engineering	K.	dissociation constant
eq.	equation	Kev	kilo electron volt
equil.	equilibrium	kg.	kilogram(s)
equiv.	equivalent	kgcal.	kilogram-calorie(s)
esp.	especially	ky.	kilovolt(s)
estd.	estimated	kvamp.	kilovolt-ampere(s)
estn.	estimation	kw.	kilowatt(s)
e.s.u.	electrostatic unit(s)	kwhr.	kilowatt-hour(s)
e.u.	entropy unit(s)	l.	liter(s)
e.v.	electron volt(s)	l-	levo-, levorotatory
expt.	experiment	ι- L-	denoting configurational
exptl.	experimental	L-	relationship, as to
ext.	extract		levo-glyceraldehyde
extd.	extracted	11.	•
extn.	extraction	lb.	pound(s)
F.	Fahrenheit	$LC_{50}$	concentration lethal to
Fedl.	Federal		50% of animals tested

l.c.l.	less than car lots	N.O.I.B.N.	not otherwise indexed
$LD_{50}$	dose lethal to 50% of		by name
l.	animals tested	0-	ortho; as, o-xylene
ln La sa	logarithm (natural)	<i>O</i> -	denoting attachment to
log	logarithm (common)		oxygen; as, O-acetyl-
m.	meter(s)		hydroxylamine
m-	meta; as, m-xylene	O.D.	outer diameter
M	metal	OZ.	ounce(s)
M	molar (as applied to	р., рр.	page, pages
	concn.; not molal,	p-	para; as, p-xylene
	which is written out)	pos.	positive (adj.)
ma.	$\operatorname{milliampere}(\operatorname{s})$	powd.	powdered
manuf.	manufacture	p.p.m.	parts per million
manufd.	manufactured	ppt.	precipitate
manufg.	manufacturing	pptd.	precipitated
max.	maximum	pptn.	precipitation
M.C.A.	Manufacturing Chem-	prepd.	prepared
	ists' Association	prepn.	preparation
m.e.f.	million cubic feet	Pr. no.	Foreign Prototype no.
m.e., meq.	milliequivalent(s)		(for dyes)
mech.	mechanical	p.s.i.(g.), (a.)	pound(s) per square inch
M.e.v.	million electron volts	r · · · · (g.), ()	(gage), (absolute)
mg.	$\operatorname{milligram}(\operatorname{s})$	pt.	point
m.g.d.	million gallons per day	pts.	parts
min.	minimum; minute(s)	quad. pt.	-
misc.	miscellaneous	quad. pt. qual.	quadruple point qualitative
mixt.	mixture	quant.	<del>-</del>
ml.	milliliter(s)	-	quantitative "which see"
M.L.D.	minimum lethal dose	$rac{g.v.}{ m R}$	
mm.	millimeter(s)	11	univalent hydrocarbon
mM	millimole(s)	7)	radical (or hydrogen)
mol.	molecule, molecular	$R_{r}$	Rankine
m.p.	melting point	ref.	reference
m.p.h.	miles per hour	resp.	respectively
M.R.	molar refraction	r.h.	relative humidity
mv.	$\operatorname{millivolt}(\mathbf{s})$	R.I.	Ring Index no.
$m\mu$	millimicron(s)	r.p.m.	revolutions per minute
$n \text{ (as, } n_{\mathrm{D}}^{20})$	index of refraction (for	r.p.s.	revolutions per second
	$20^{\circ}\mathrm{C}$ . and sodium	S.	soluble
	light)	S-	symmetric(al); as, s-m-
<i>n</i> -	normal; as, n-butyl		xylidine
N	normal (as applied to	S-	denoting attachment to
•	conen.)		sulfur; as, S-methyl-
N-	denoting attachment to		eysteine
	nitrogen; as, $N$ -meth-	S.A.E.	Society of Automotive
	ylaniline		Engineers
neg.	negative (adj.)	satd.	saturated
no.	number	satn.	saturation

#### xii ABBREVIATIONS AND SYMBOLS

S.C.F.	standard cubic foot	t.s.i.	tons per square inch
	(feet)	Twad.	Twaddell
$\operatorname{Sch}$ .	Schultz no. (for dyes)	u.v.	ultraviolet
sec.	second(s)	v.	volt(s)
sec-	secondary; as, sec-butyl	var.	variety
S.F.s.	Saybolt Furol second(s)	vic-	vicinal; as, vic-m-xyli-
sl.s.	slightly soluble		$\operatorname{dine}$
sol.	soluble	vol.	volume(s) (not volatile)
soln.	solution	v.s.	very soluble
soly.	solubility	w.	watt(s)
$\operatorname{sp}$ .	specific	wt.	weight
sp., spp.	species	X.U. $(10^{-10}$	X-unit
spec.	specification	mm.)	
$\mathrm{sp.gr.}$	specific gravity	yd.	yard(s)
$\operatorname{sq}$ .	square	yr.	year(s)
S.T.P.	standard temperature	$[lpha]_{ m D}^{20}$	optical rotation (for
	and pressure		20°C. and sodium
subl.	sublime(s), subliming		light)
S.U.s.	Saybolt Universal	$\gamma$	microgram(s)
	$\mathbf{second}(\mathbf{s})$	9	differential operator
sym-	symmetric(al); as, $sym$ -		(partial)
	m-xylidine	$\Delta$	finite difference
T.A.P.P.I.	Technical Association of	η	viscosity
	the Pulp and Paper	λ	wave length
	Industry	$\mu$	micron(s)
tech.	technical	$\Omega$	ohm(s)
temp.	temperature	<	less than
tert-	tertiary; as, tert-butyl	>	more than
theoret.	theoretical	~	cycle(s)
t.p.h.	tons per hour	≈	approximately equal to

Other letter symbols may be found in "Standard System of Nomenclature for Chemical Engineering Unit Operations" adopted by the American Institute of Chemical Engineers.

#### SHIPPING REGULATIONS

Complete information for the U.S. is given in "Tariff No. 9 Publishing Interstate Commerce Commission Regulations for Transportation of Explosives and Other Dangerous Articles by Land and Water in Rail Freight Service and by Motor Vehicle (Highway) and Water Including Specifications for Shipping Containers," with supplements, issued by H. A. Campbell, Agent, 30 Vesey Street, New York 7, N.Y. (1954). The following terms for labeling explosives and other dangerous articles have been used in the Encyclopedia:

Red label (for inflammable liquids)
Yellow label (for inflammable solids and oxidizing materials)
White label (for acids and corrosive liquids)
Red label (for inflammable compressed gases)
Green label (for noninflammable compressed gases)
N.O.I.B.N. (not otherwise indexed by name)

In the text of the Encyclopedia the preferred terms "flammable" and "nonflammable" are used in place of "inflammable" and "noninflammable," respectively.

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Chimie & industrie CIOS Repts.

Compt. rend.

FIAT Repts. Fortschr. chem. Forsch. Gazz. chim. ital. Helv. Chim. Acta

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Angewandte Chemie

Annalen der Chemie, Justus Liebigs Archives of Biochemistry and Biophysics

Archives of Industrial Hygiene and Occupational Medicine Biochemical Journal, The Biochemische Zeitschrift Biochimica et Biophysica Acta British Intelligence Objectives Subcommittee Reports Bulletin of the Chemical Society of Japan Bulletin de la société chimique de France Chemical Abstracts Canadian Journal of Research Chemische Berichte Chemical Engineering with Chemical & Metallurgical Engineering Chemical and Engineering News

Chemical Engineering Progress with Transactions of American Institute of Chemical Engineers Chemical Engineering Science Chemische Industrie Chemistry & Industry

Chemical Reviews Chemische Technik, Die (Berlin)

Chemical Week Chemisches Zentralblatt Chemiker-Zeitung mit dem Sonderteil, Die Chemische Praxis und der Beilage, Chemisch-technische Übersicht Chimica, La, e l'industria (Italy) or (Milan)

Chimie & industrie Combined Intelligence Objectives Subcommittee Reports Comptes rendus hebdomadaires des séances de l'académie des sciences Field Information Agency Technical Reports Fortschritte der chemischen Forschung Gazzetta chimica italiana Helvetica Chimica Acta

Ind. Chemist Ind. Eng. Chem. (superseding J. Ind. Eng. Chem.) J. Agr. Food Chem. J. Am. Chem. Soc. J. Am. Med. Assoc. J. Am. Pharm. Assoc. J. Appl. Chem. (U.S.S.R.) (see also Zhur. Priklad. Khim.)
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Zeitschrift für Elektrochemie und angewandte physikalische Chemie

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Zeitschrift für physikalische Chemie

## S

**SABADINE,**  $C_{29}H_{51}NO_8$ . See Alkaloids, Vol. 7, p. 502.

**SABINANE**, C<sub>10</sub>H<sub>18</sub>. See "Thujane" under Hydrocarbons, Vol. 7, p. 609; Terpenes.

**SABINENE**,  $C_{10}H_{16}$ . See Terpenes.

SABINIC ACID, CH<sub>2</sub>OH(CH<sub>2</sub>)<sub>10</sub>COOH. See Fatty acids (hydroxy and keto), Vol. 6, p. 288.

**SACCHARATED IRON OXIDE.** See *Iron preparations*, Vol. 8, p. 70.

SACCHARIC ACID, COOH(CHOH)4COOH. See Sugar derivatives.

**SACCHARIDES.** See Carbohydrates; Polysaccharides; Sugars.

**SACCHARIMETRY.** See Polarimetry; Sugar analysis.

**SACCHARIN, C<sub>6</sub>H<sub>4</sub>.SO<sub>2</sub>.NH.CO.** See Sweetening agents.

S ACID. See Amino naphthols and amino naphtholsulfonic acids, Vol. 1, p. 733; Naphthylamines and naphthylaminesulfonic acids, Vol. 9, p. 267.

#### SAFETY

At one time it was assumed that injuries in industries were a matter of chance and were both nonpredictable and nonpreventable. They quite obviously arose from industrial accidents and the dictionary definition of an accident is: "An event which takes place without one's foresight or expectation; an undesigned, sudden, and unexpected event; often an undesigned and unforeseen occurrence of an afflictive or unfortunate character, etc." It is the general experience of safety men that they are constantly predicting that if processes are carried on in a specified way there will be an injury eventually and that it is actually rare to have an accidental injury according to the dictionary definition of an accident. When personal injuries occur under these conditions they are not unforeseen or unpredictable or unusual and they could have been and should have been prevented. Harold Miner, after twenty-five years as head of the safety organization of a large chemical corporation, stated that he had never investigated a single personal injury which could not have been prevented by a little more foresight and a little more constructive thinking.

#### 2 SAFETY

The prevention of accidents and accidental injuries is largely, if not entirely, a matter of attitude and approach to the problem. If industrial management truly accepts responsibility for accident prevention, and accepts as a fact that accidents can be and should be prevented, then methods will be found and the problem will be solved rather simply. It should also be accepted that the solution must be rather simple, as there are too many people involved for any very complicated solutions to be practical.

If any justification is needed for the acceptance of this type of responsibility there are several compelling reasons:

- (1) All states now have laws which require managements not only to pay financially for injuries to employees but also require the maintenance of a certain minimum of safety on the operation and provide for inspection to enforce the requirements.
- (2) Economically, prevention of accidents costs less than the costs of injuries. This has been amply demonstrated many times. One of the outstanding recent examples on a fairly large scale is the program of the Pacific Coast Association of Pulp and Paper Manufacturers which has saved the member companies in the states of Oregon, Washington, and California, in the six years between 1947 and 1952, at least \$800,000 annually in an industry with only about 40,000,000 man-hours annual exposure. These savings are in direct medical and compensation costs. They are certainly no greater than the indirect savings, which are just as real although not so easily measured.
- (3) Long before either of the other reasons was of pressing importance, progressive managements recognized the prevention of accidental injuries as a social obligation and took steps to discharge the responsibility. Thus the United States Steel Corporation organized a central safety committee in 1906, just five years after the formation of the Corporation, and placed a staff man of the rank of plant superintendent in charge of safety activities. By 1910 the Corporation was voluntarily paying workmen's compensation at rates which were fixed and published. Both of these events preceded the first safety or compensation laws in this country. The compensation rates set up in some of the early laws were based upon those in the voluntary plans of the International Harvester Company and the U.S. Steel Corporation.

Many detailed aspects of industrial safety, as well as hygiene, are discussed in other articles. See Air conditioning; Allergens, industrial; Carcinogens; Dust; Explosions (gaseous); Fire-resistant textiles; Fire prevention and extinction; First aid; Industrial hygiene and toxicology; Laboratories; Lead poisoning; Pilot plant. For protection against hazardous radioactivity see Isotopes; Nucleonics; Radioactive elements, natural; Radiography, industrial.

Several organizations, such as those listed in references (1,2,5,8) issue both general and detailed publications on industrial safety. These include many of direct application to chemical industries.

Fundamentals of Safety. There are certain very fundamental requirements for any successful industrial safety program. The details of organization vary, even among the most successful programs, depending mainly upon the variations in the corporate organization, but the following must be done:

- (1) Both mechanical safeguards and personal protective equipment must be provided in sufficient amounts and appropriate kinds for the operations, and they must be used.
- (2) Employees must be taught and constantly encouraged to work safely. They must be made to understand that it is the considered attitude of all levels of manage-