APPLICATIONS OF LOW ENERGY X- AND GAMMA RAYS

Edited by

CHARLES A. ZIEGLER

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Preface

THE PAPERS IN THIS VOLUME were presented at the third in a series of symposia on low energy X- and gamma-ray sources and applications held at Boston College, Chestnut Hill, Massachusetts. The symposium, which was sponsored jointly by Boston College, Panametrics, Inc., and the Atomic Energy Commission, was international in character, and broad in scope in the sense that fundamental studies were combined with applications in such diverse fields as geology and medicine as well as a wide variety of industrial applications.

The purpose of the third symposium, as was that of the first and second, was to bring together scientists engaged in the development of X- and gamma-ray technology and to provide an opportunity for an interchange of ideas. This should assist in meeting requirements placed upon us by our ever-expanding society in the fields of medicine, en-

vironmental pollution analysis, and industrial quality control.

. The general theme of the symposium was biased toward radioiso-tope-excited X-ray fluorescence analysis (XRF) and the various techniques employed to improve upon the technology. Energy determination using balanced filters and scintillation counters or semiconductor and proportional counters were treated in considerable depth. Other technologies covered were Mössbauer effect spectrometry and electron spectroscopy. It is expected that interest in each of these areas will continue to grow as the spectral capabilities of each are better understood.

It is interesting to note that radioisotope-excited XRF first became practical back in the forties after the Manhattan District Project came into being although the possibility of radioisotope X-ray excitation was demonstrated by Chadwick's experiments in 1912 when he observed that alpha particles could be used to excite fluorescent X-rays. But it has been only within the past 10 years that XRF has begun to show signs of wide acceptance by industry and by scientists working in a variety of disciplines.

Mössbauer effect spectroscopy based on recoilless resonant absorption of gamma rays has, in the 12 years since its discovery by R. L. Mössbauer, likewise made significant advances into the fields of chemistry and metallurgical research. This relatively new analytical

technique and the still newer developments of electron spectroscopy, each in its own right, appear to offer many potential benefits that are yet to be appreciated fully. The sponsorship of government programs with valuable input from industrial and other organizations will, no doubt, bring about a better understanding of these benefits.

JAMES W. HITCH
Division of Isotopes Development
U.S Atomic Energy Commission

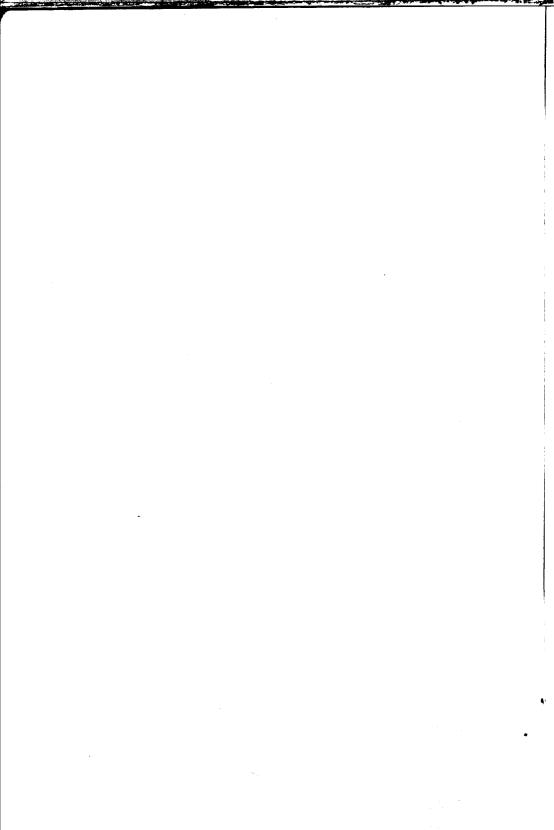
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INDUSTRIAL APPLICATIONS



Progress in Industrial Application of Low Energy Radioisotope X-Ray Techniques

J. R. RHODES

INTRODUCTION

One of the problems of a review such as this is to define its boundaries in time, space and subject matter. This paper reviews applications since 1967, with special reference to instruments and techniques in radio-isotope X-ray analysis that have found significant application in industry. Included are field and field-laboratory uses in the mining industry, and developmental equipment which is intended for industrial use.

Radioisotope low energy X-ray fluorescence, absorption and backscatter are used in two main categories of measurement, coating thickness and elemental analysis. A convenient and self-consistent set of formulae and recipes suitable for calculating feasibility of these measurements by the above techniques, and by beta particle backscatter, has recently been published.²

This review is confined to the two most significant areas of industrial application that have developed in the past three years, the use of portable radioisotope X-ray fluorescence analyzers and the installation of a number of on-stream process analyzers.

Later in this volume Cameron³ will be discussing the state of the art in portable, laboratory and on-line methods of coating thickness measurement. Also Chen and Cahill⁴ will describe industrial application of X-ray fluorescence and absorption to the measurement of thickness.

The range of applications of X-ray absorption and backscatter for analysis is not great enough to warrant special review. The main applications are to effectively two-component systems such as coal/ash⁵ and meat/fat,⁶ and to the determination of heavy elements in light matrices such as sulphur or cobalt in hydrocarbons.⁷

APPLICATIONS

Half of all the papers published in the last three years describe portable analyzers and applications of them in various industrial and field en-

vironments. Half of the rest describe applications to industrial process control. These include "on-stream" analyzers where a continuously moving sample of the plant stream is measured, and "off-stream" analyzers which measure discrete samples taken automatically.

It is not known how many individual instruments are in use, but it is probable that the number of portable analyzers in use is much greater than the number of papers written on their use, since they have been commercially available for five years. However, other analyzers are not yet generally available, and the number in use is not likely to greatly exceed the number of papers written.

Table 1 classifies, alphabetically by element, the significant analyses reported in the past few years in the above-mentioned areas. A study of the published feasibility investigations reveals that in about 60% of them the objective was to optimize the accuracy of major component determinations, and in the rest it was to obtain the best possible sensitivity. In the former case, the main sources of error are matrix effects. In applications of portable analyzers, much ingenuity has gone into devising methods of minimizing these errors without compromising the essential simplicity of the equipment and measurement procedures. We expect to see more development along these lines, as matrix effects appear to be one of the main obstacles to much more widespread field use.

Sensitivity is limited by the basic considerations of counting statistics, and the fluorescent to scatter ratio. Using scintillation and proportional counters, detection limits of about 0.005% (and in favorable cases, 0.001%), can usually be obtained by one of two methods. The first is to use a source whose backscattered radiation is just resolved from the required characteristic X-rays. The second is to excite the required fluorescence with maximum possible efficiency using an energy just above the absorption edge of the element to be determined. Resolution limitations prevent employment of both these methods simultaneously with proportional or scintillation counters. They do not. however, when Si(Li) or Ge(Li) detectors are used, and in this case sensitivities of a few p.p.m. can be obtained even with heterogeneous ore samples. 8.9 It seems that the development potential of proportional and scintillation counters in this respect has been thoroughly investigated and largely exhausted. 10 This is not the case with Si(li) and Ge(Li) spectrometers where many studies using obvious combinations of high resolution detectors, monochromatic line sources and thin samples¹¹ (for example) have not yet been performed. We expect to see a rapid increase in the number of investigations in this area.

TABLE 1 Analysis

Element and % conc. range	Material (see Note 1)	Source and activity	X-ray excit.	Filter	Analyzer type (sec Note 2)	Application	Detection limit happlication (or 1 \sigma prec.)*	Remarks
Aluminum 2.4 Al ₃ O ₃ Barium 1-15	Cement raw mix 13, 49 Barytes ores 41, 53	*H/Zr, 3 Ci; *10 PO; 10 mCi *11 Am-Sm; *11 Am-Ba,	AIK BaK	IA	On-line P (fused pellets) On-stream S (slurry)	Kiln-feed control trol Barytes flotation pilot plant	Kiln-feed con- 0.09% Al ² O ³ rol Barytes flota- 0.01% ion pilot plant	Also see under Ca, Fe, Si Balanced tar- get method
Calcium 40-50 1-5	Cement raw mix ⁸⁶ Fe, 2 mGi 22 Silicate rocks ⁸⁶ Fe, 10 mG	66Fe, 2 mCi 66Fe, 10 mCi	CaK CaK	— K/Ca	Portable S Portable SS	Feasibility study Soil and rock	0.2% CaO* 0.5% CaO*	Also see under K
15–19 CaO	Sinter mix 44, 54	66Fe, 7 mCi	CaK	1	On-line P (pressed	Blast furnace control	0.15%	Also see under Si; Fe
75-85 CaCO ₃ 40-50 CaO	Cement raw mix ⁶⁶ Fe, 2 mCi 13, 48, 54 Cement raw mix ⁸ H/Zr, 3 Ci 13, 49 ⁹¹⁰ Po, 10 m(⁵⁶ Fe, 2 mCi ³ H/Zr, 3 Ci; ³¹⁰ Po, 10 mCi	CaK	1 1	On-stream P (slurry) On-line P (fused pellets)	Kiln-feed control	Kiln-feed con- 0.35% CaCO ^{33*} — trol Kiln-feed con- 0.16% CaCO* Also see trol	Also see under Al, Fe;
40–50 CaO	Cement raw mix 53, 56	*H/Ti, 5 Ci	CaK			Grinder control	0.4% CaO*	des cel
40-45 CaO	Cement raw mix "H/11, 3 Ci 45 Cement raw mix 109Cd, 1 mCi 47, 57	³ H/11, 3 Cl ¹⁰⁰ Cd, 1 mCi	CaK	1 1	On-stream F Alin-teed (powder) trol Powder sample Feasibility analyzer P study	Kun-teed con- trol Feasibility study		under Fe Reduced par- ticle size effects

TABLE 1 Analysis—continued

Element and % conc. range	Material (see Note 1)	Source and activity	X-ray excit.	Filter	Analyzer type (see Note 2)	Application	Detection limit (or 1 \sigma prec.)*	Remarks
Chromium	Steels 36	⁸⁶ Fe, 7 mCi;	CrK	Ti/V	Portable S	Alloy analysis	0.1% Cr*	
0.5–8 0.5–5	Steels 22 Electroplating	338Pu, 20 mCi 34/Zr, 4 Ci	CrK CrK	Ti/V Ti/V	Portable S Portable S	Alloy analysis Feasibility	0.15% Cr* 0.15% Cr	
Copper 0.1–15	bath solutions Copper ore pulps; *H/Zr, 12 Circockfaces 12,	*H/Zr, 12 Ci	CuK	Co/Ni	Portable S	study Field assay	0.2% Cu*	
0.4-4	14, 34 Cu ore pulps 55	³H/Zr	CuK	Co/Ni	Portable S	Draw control; 0.1% Cu*sub-level	0.1% Cu*	
0.03-2.5	Cu ore pulps 22 109Cd, 1 mCi	109Cd, 1 mCi	CuK	Co/Ni	Portable S	caving Feasibility	0.03% Cu	
0.1–5	Cu ore pulps 50, 60	³ H/Zr, 2.5 Ci	CuK	Cr/Mn, Ni/Co	Portable S	study Feasibility study	0.2% Cu*	Special method eliminates
0.1–10 0.4–1.6	Core samples 58 Mn nodule	288Pu, 30 mCi 286Pu, 20 mCi	CuK CuK	Co/Ni Co/Ni	Portable S Portable S	Core analysis Shipboard	0.05% Cu*	matrix effects Also see under
60–95 0.1–26	pulps 24 Cu alloys 36 Various Cu-Pb-Zn ores 39	109Cd 2 mCi 288Pu-Ga, 30 mCi	CuK CuK	Co/Ni Cu	Portable S On-stream (slurry) S	assay Alloy sorting Mill control	0.45% Cu* 0.08-1%*	Ni, Fe and Mn See also under Pb, Sn and Zn
Gold 0.01-1	Simulated ores 23	⁶⁷ Co, 1 mCi	AuK	M/JH	Portable S	Feasibility study	0.03% Au	

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		X-ray back- scatter also	Also see under Ca; Si	Also see under Ca, Fe: Si	Also see			Matrix; particle		Used nomograms		Absorption edge analysis. See also under Cu. Sn. Zn	Comparison of sources
0.5% Fe*	0.24% Fe*	~ 1% Fe*	0.2% Fe*	0.05% Fe		0.003% Fe	0.01% Fe	0.08% Pb	2% Pb* 0.25% Pb	I	0.05% Pb	0.04-0.34% Pb*	0.4% Pb
Field assay, mine control	Shipboard assay	Strip mine control	Blast furnace control	Kiln-feed con- 0.05% Fe trol	Kiln-feed con- trol	Pilot plant	Wear	Field analysis	Core analysis Mine control	Field assay	Alloy sorting	Mill control	Feasibility study
Portable S	Portable S	Portable S	On-line (pressed	On-line (pressed	On-stream (powder) P	On-stream (slurry) P	Assay G-M	Portable S	Portable S	Portable S	Portable S	On-stream (slurry) S	On-stream S
Cr/Mn	Cr/Mn	Cr/Mn	Cr/Mn	1	Cr/Mn	Cr/Mn	Fe	Ga/Ge	Ga/Ge		W/Ir	1	ı
FeK	FeK	FeK	FeK	FeK	FeK	FeK	FeK	PbL	PbL PbL PbK	PbK	PbK	1	PbK
286Pu, 30 mCi	889Pu, 20 mCi	288Pu, 10 mCi	⁸ H/Zr, 4 Gi	³ H/Zr, 3 Gi and ³¹⁰ Po, 10 mGi	*H/Ti, 5 Ci	328Pu, 10 mCi	*H/Zr, 2 Ci	***Pu, 30 mCi	100Cd 288Pu, 30 mCi 57Co	78Se	168Gd, 0.6 mCi	²⁴¹ Am, 5 mCi ⁵⁷ Co, 1.2 mCi	¹⁸⁸ Gd, 1 mCi ¹⁸⁷ Cs, 1 Ci
Iron ores (pulps; cores) 15, 57, 61 Lead-zinc ores	Manganese nodule pulps 24	Iron ores (crushed rock) 57, 62-64	Sinter mix 44, 54 sH/Zr, 4 Gi	Cement raw mix 13, 49	Cement raw mix	Sand (slurry) 13	Lubric, oil 57	Ore pulps 66	Lead ores 27 Core samples 58 Lead ores; mine walls 28	Lead ores; core samples 29, 30, 32	Leaded brass; steels 36	Ores, zinc, concentrates, tailings 38, 39	Solutions; slurries 65
Iron 20-70 1-10	2-12	5-50	25–45	0.5-3	0.5-3	0.1–5	0.1	Lead 0.05–30	1-8		0.05-3	0.1–32	0.5-5

Table 1 Analysis-continued

		J						
Remarks	Also see under Fe, Cu, Ni				Also see under Nb; Sn	Also see un- der Fe, Cu, Mn	Also see under Mo; Sn	Also see under Ca
Detection limit (or 1 \alpha prec.)*	0.75% Mn* 0.14% Mn* 0.04% Mn	0.01% Мо	0.003% Mo	0.04% Mo 0.1% Mo* 0.002 to 0.01% Mo		0.05% Ni* 0.3% Ni	0.01% Nb 0.06% Nb	~ 0.5% K*
Application	Shipboard assay Alloy sorting Feasibility study	Mine control	Feasibility study	Alloy sorting Alloy analysis Feasibility study	Pilot plan control	Shipboard assay Alloy sorting	Pilot plant control Alloy analysis	Soil; rock type identity
Analyzer type (see Note 2)	Portable S Portable S Portable S	Portable S	Portable S	Portable S Portable S Laboratory; on-stream SS	On-stream S	Portable S Portable S	On-stream S Portable S	Portable SS
Filter	V/Cr V/Cr V/Cr	Y/Zr	Y/Zr	Y/Zr Y/Zr —	Y/Zr	Fe/Co Fe/Co	Sr/Y Sr/Y	CI/K
X-ray excit.	MnK MnK MnK	MoK	MoK	MoK MoK MoK	MoK	NiK NiK	NbK NbK	KK
Source and activity	²³⁸ Pu, 20 mCi ³ H/Zr, 214 Ci ³³⁸ Pu, 20 mCi	100Cd, 2 mCi	109Cd, 1 mCi	109Cd, 2 mCi 109Cd, 1 mCi 125I, 2 mCi	147Pm/Al-Ag 1 Ci	³⁸ Pu, 20 mCi ³ H/Zr, 2.5 Ci	¹⁴⁷ Pm/Al-Ag, 1 Ci ¹⁰⁹ Cd, 2 mCi	⁵6Fe, 10 mCi
Material (see Note 1)	Manganese nodule pulps 24 Steels 36 Plating bath sol. 23	Ore pulps; Cu conc. 15, 61	Ore pulps 23	Steels 36 Steels 22 Mine samples, Cu conc. slags 67	Ore slurries 41	Manganese nodule pulps 24 Cu alloys 36	Ore slurries 41 Steels 36	Silicate rocks 18, 86Fe, 10 mCi 19
Element and % conc. range	Manganese 20-40 0.2-1.5 0.5-5	Molybdenum 0.01-1 1-20	0.001-0.3	0.01–5 0.1–10 0.002–0.5	0.01-7 MoO ₃	Nickel 0.8-1.6 3-30	Niobium 0.01-5 0.06-2	Potassium 1–5

Also see under Fe; Ca	Also see under Al, Ca, Fe		Low noise	Paper com-	nuclear and non-nuclear techniques					Feasibility	study See also under Mo, Nb	
0.17% SiO ₈	Kiln-feed con- 0.13% SiOs trol	0.003% Ag	0.17% S	0.03% Sos		0.03% Sn	0.1% Sn	0.1% Sn	0.1% Sn	0.005% Sn	0.005% Sn	0.2% Sn*
Blast furnace control	Kiln-feed control	Field assay	Feasibility	Feasibility study		Mine control	In situ assay	In situ assay	Core analysis	Geochemical	Process control	Alloy sorting
On-line (pressed	penets) r On-line (fused pellets) P	Portable S	Portable S	On-stream P		Portable S	Portable S	Portable S	Portable S	Portable S	On-stream (slurry) S	Portable S
I	I	Mo/Rh	P/S	P/S		Ag/Pd	Ag/Pd	l	Ag/Pd	Ag/Pd	Ag/Pd	Ag/Pd
SiK	SiK	AgK	SK	SK		SnK	SnK	SnK	\mathbf{SnK}	SnK	SnK	$\mathbf{S}_{\mathbf{n}}\mathbf{K}$
³H/Zr, 4 Gi	³ H/Zr, 3 Ci; ²¹⁹ Po, 10 mCi	¹⁴⁷ Pm/Al, 0.5 Ci	³H./Ti, 5 Ci	⁶⁵ Fe, 10 mCi		147Pm/Al, 0.5 Ci	¹⁴⁷ Pm/Al, 0.5 Ci	seS/Ba	¹⁴⁷ Pm/Al, 0.5 Ci	¹⁴⁷ Pm/Al, 0.5 Ci	²⁴¹ Am-Ba, ²⁴¹ Am-Sn, 14 mCi	147Pm/Al, 0.5 Ci
Sinter mix 44, 57	Cement raw mix 13, 49	Ag ores 8	Pulverized coal	Stack gases 70		Tin ore pulps 12, 147Pm/Al, 33, 34 . 0.5 Ci	Tin mineralization 14, 34, 68	Tin mineraliza- tion 28	Tin ores, cores 39, 58	Tin mineraliza- tion 69	Tin ore slurries 35, 39, 41	Bronzes; gun metals 36
Silicon 12–16 SiO _s	14 SiO ₂	Silver 0.01-0.1	Sulphur 0.1–5 S	0.01-0.5 SO ₂		Tin 0.05-25	0.5–5	0.5-5	0.5-5	0.002-0.1	0.01-2	0.1-11