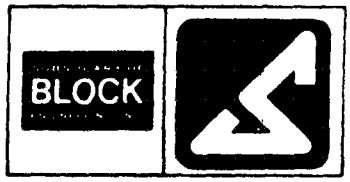
# Sadtler Standard Infrared Grating Spectra

## Volumes 47–48

## 46001–48000




SADTLER RESEARCH LABORATORIES, INC.

# SADTLER STANDARD GRATING SPECTRA

## UPDATE VOLUME -- REISSUED PRISM SPECTRA

Vol 47

46001P - 47000P

CREATIVE CHEMISTS SINCE 1924

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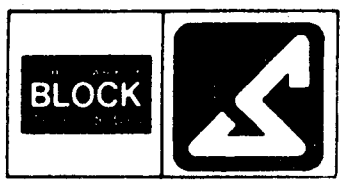


SADTLER RESEARCH LABORATORIES, INC.

# SADTLER STANDARD GRATING SPECTRA

## UPDATE VOLUME -- REISSUED PRISM SPECTRA

Vol. 48

47001P - 48000P

CREATIVE CHEMISTS SINCE 1874

3316 SPRING GARDEN ST., PHILADELPHIA, PA. 19104
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The publication of the physical data of the Sadtler Standard Spectra and the Sadtler Commercial Spectra is intended to be descriptive. The samples of the materials represented have come generally from other sources than our own laboratories and frequently without the donors' knowledge of their part in this publication.

On the other hand every effort is made by Sadtler Research Laboratories, Inc. to assure the reliability of the published spectra. When improved data is available or errors are called to our attention we revise and reissue the proper replacement spectra.

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## SADTLER STANDARD GRATING SPECTRA

## UPDATE VOLUME -- REISSUED PRISM SPECTRA

This volume contains 1000 absorption spectra recorded on a prism spectrophotometer determined in the 2.0 - 15.0 micron region and presented in a linear wavelength vs. percent transmittance format. These spectra were previously published in the Sadtler Standard Prism Spectra collection and are now re-issued and renumbered in the sequence of the Sadtler Standard Grating Spectra to integrate all available infrared data for the user of this publication. While grating spectra are generally preferred by spectroscopists, it is obviously better to have a prism spectrum if it is the only data available.

The spectra were prepared at Sadtler Research Laboratories unless otherwise noted on the spectrum heading, the name of the donor of each compound is also shown on the heading. Although some of the spectra were published over ten years ago and do not always appear to be of optimum quality, they are included in the publication to insure complete coverage of published compounds.

Standard techniques have been developed in our laboratories to insure that the spectra published are of the best possible quality and reproducible for comparison and identification purposes. The preferred sample preparation methods are the capillary cell for liquids and the KBr wafer for solids, the spectra obtained are qualitative only. The KBr method is used for solids since it is a standard technique and requires a small sample amount for preparation of good spectra, leaving the remainder for further analytical investigation. A paper describing the preparation procedure entitled Improved KBr Techniques by Traude and Philip Sadtler is available from our laboratories.

When the KBr method cannot be used for solids due to reaction with the sample, the Split Mull technique is used; the sample is mulled in mineral oil and the entire spectrum is scanned, then a perfluorinated hydrocarbon mull is prepared and scanned in the 3.0 - 3.8 and 6.6 - 7.4 micron regions. This provides a complete spectrum of the compound.

Liquid samples and low melting solids are generally determined using capillary cells.

Each spectrum is clearly labelled with the sample preparation technique used.

The following five indexes accompany the Sadtler Standard Spectra:

- Alphabetical Index
- Molecular Formula Index
- Chemical Classes Index
- Numerical Index
- Spec-Finder

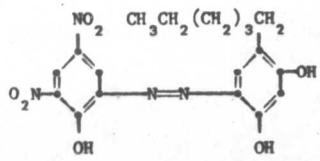
In each of the first four indexes the grating spectrum numbers of compounds are cross-referenced to their corresponding numbers in the Sadtler Standard collections of Ultraviolet and Nuclear Magnetic Resonance Spectra. A reissued prism spectrum is always signified by a P suffix to the number in the grating column of each index.

The final index, the Spec-Finder, provides a means of identifying spectra of unknown compounds by comparison with the coded peaks of the references.

WE SUGGEST THAT THE INTRODUCTIONS TO THE VARIOUS INDEXES BE READ CAREFULLY TO ASSURE THEIR BEST UTILIZATION.

46001 P

4-[(3,5-DINITRO-2-HYDROXYPHENYL)AZO]-6-HEXYLRESORCINOL

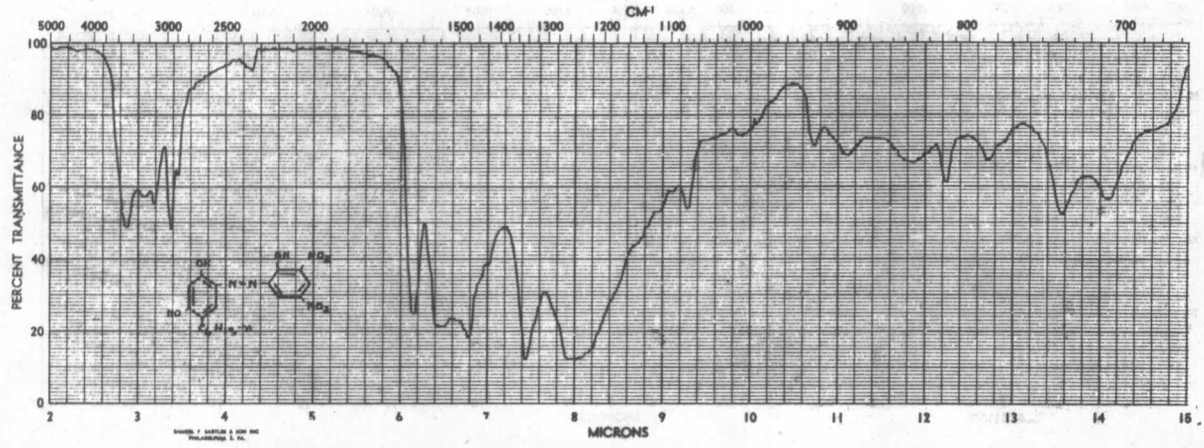


$C_{18}H_{20}N_4O_7$ Mol. Wt. 404.38

M.P. 204°C

Source of Sample: E. Koshimura, Kanazawa University
Kanazawa, Japan

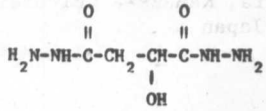
KBr Wafer






46002 P

MALIC ACID, DIHYDRAZIDE

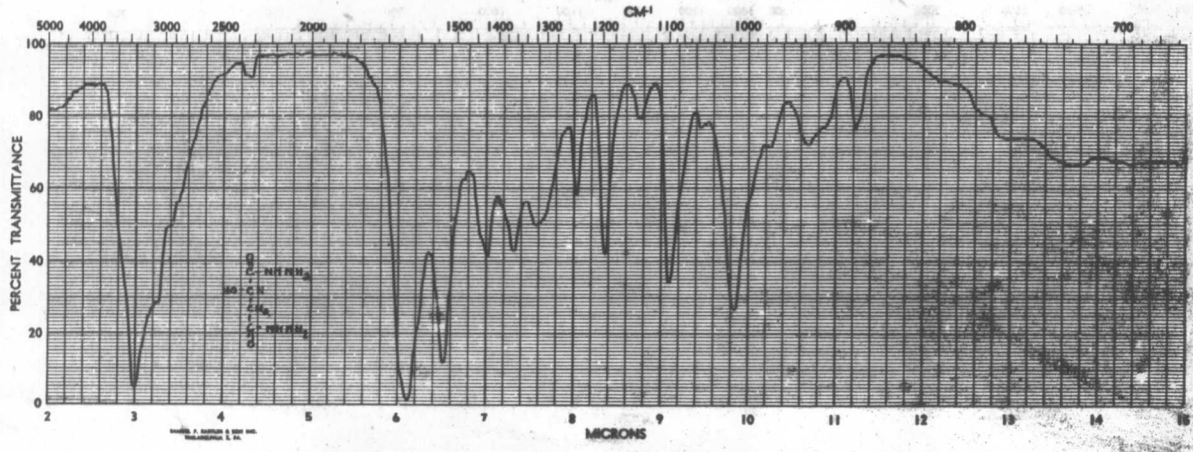


$C_4H_{10}N_4O_3$ Mol. Wt. 162.15

M.P. 172.5-174°C

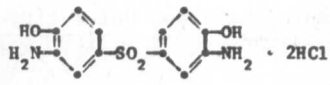
Source of Sample: E. Koshimura, Kanazawa University
Kanazawa, Japan

KBr Wafer



# 46003 P

4,4'-SULFONYLBIS[2-AMINOPHENOL], DIHYDROCHLORIDE

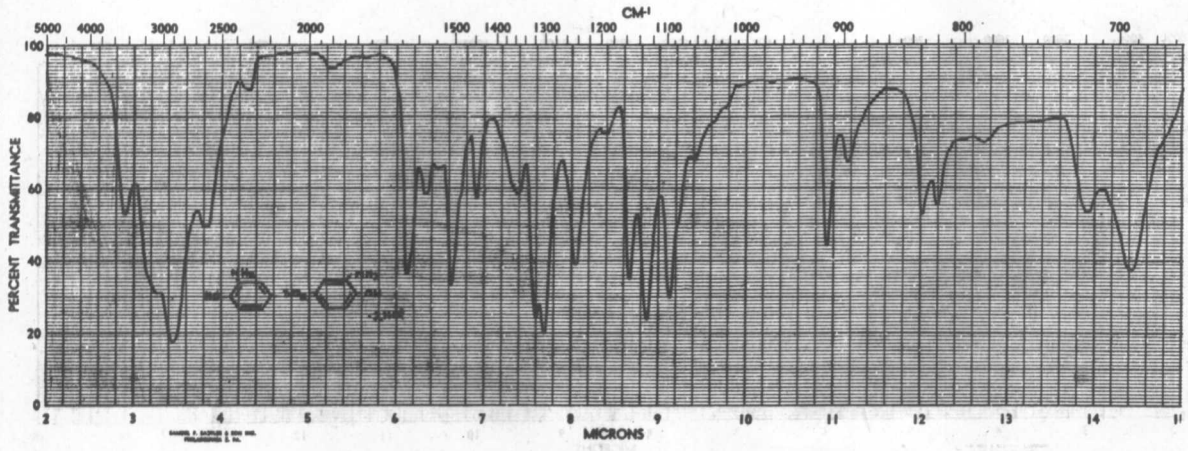


$C_{12}H_{12}N_2O_4S \cdot 2HCl$ Mol. Wt. 353.23

M.P. 262-266°C (dec.)

Source of Sample: E. Koshimura, Kanazawa University
Kanazawa, Japan

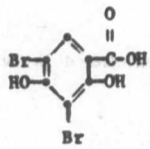
KBr Wafer






# 46004 P

3,5-DIBROMO-β-RESORCYLIC ACID

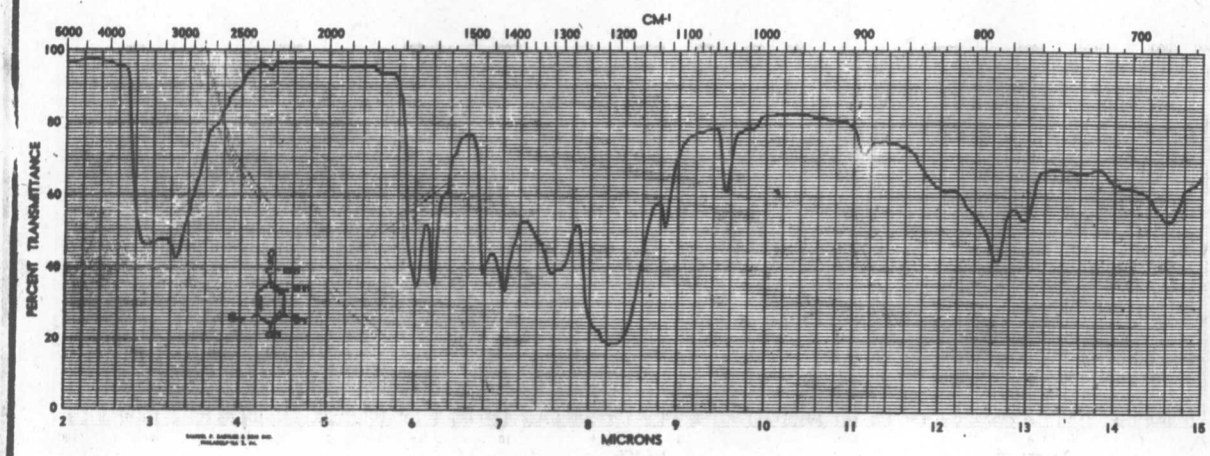


$C_7H_4Br_2O_4$ Mol. Wt. 311.93

M.P. 214°C

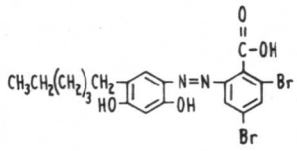
Source of Sample: E. Koshimura, Kanazawa University
Kanazawa, Japan

KBr Wafer



46005 P

2,4-DIBROMO-6-[(2,4-DIHYDROXY-5-HEXYLPHENYL)AZO]BENZOIC ACID

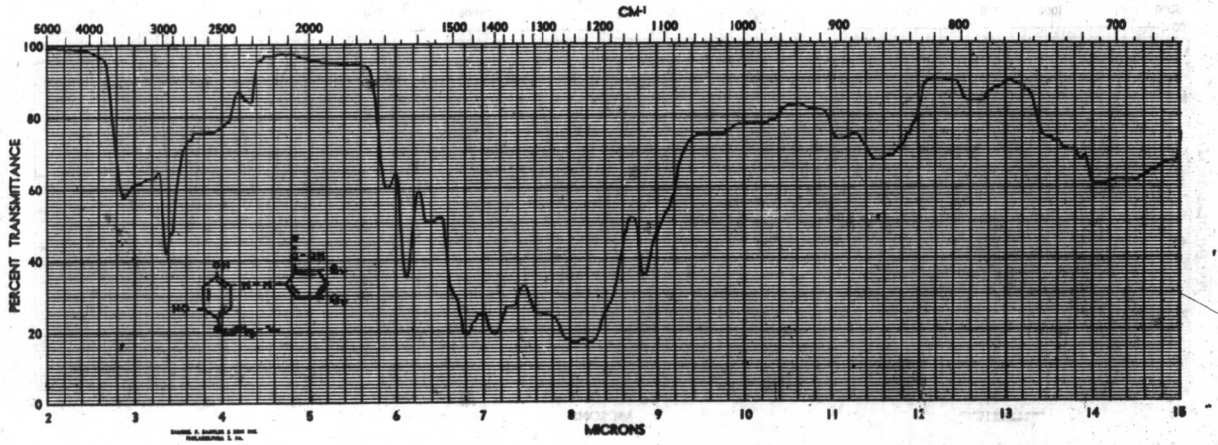


$C_{19}H_{20}Br_2N_2O_4$ Mol. Wt. 500.20

M.P. 181-182°C

Source of Sample: E. Koshimura, Kanazawa University
Kanazawa, Japan

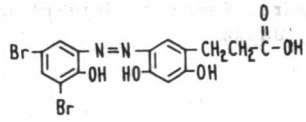
KBr Wafer






46006 P

3-{5-[(3,5-DIBROMO-2-HYDROXYPHENYL)AZO]-2,4-DIHYDROXYPHENYL}-PROPIONIC ACID

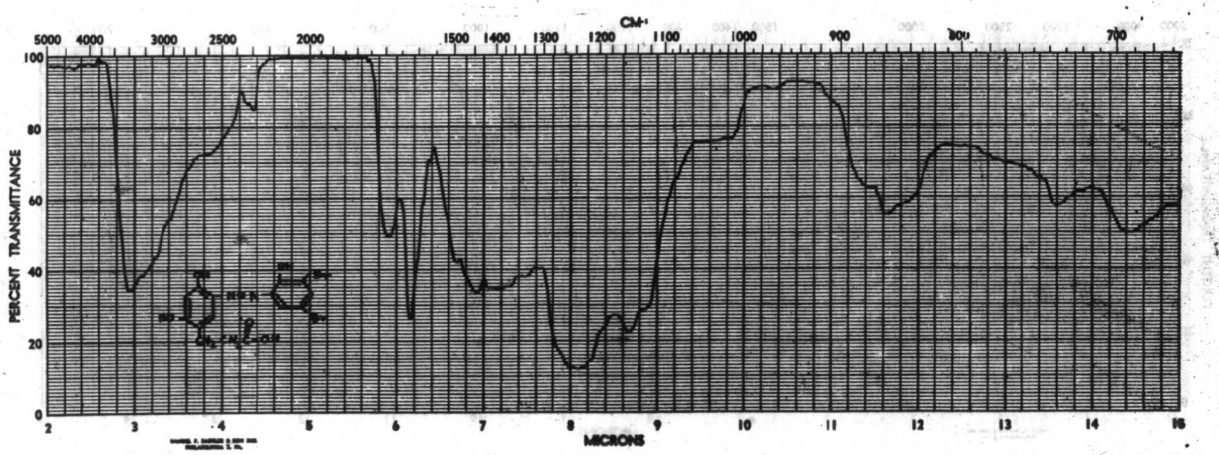


$C_{15}H_{12}Br_2N_2O_5$ Mol. Wt. 460.09

M.P. 258°C (dec.)

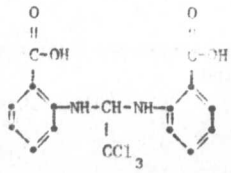
Source of Sample: E. Koshimura, Kanazawa University
Kanazawa, Japan

KBr Wafer



# 46007 P

N,N'-(2,2,2-TRICHLOROETHYLIDENE)DIANTHRANILIC ACID

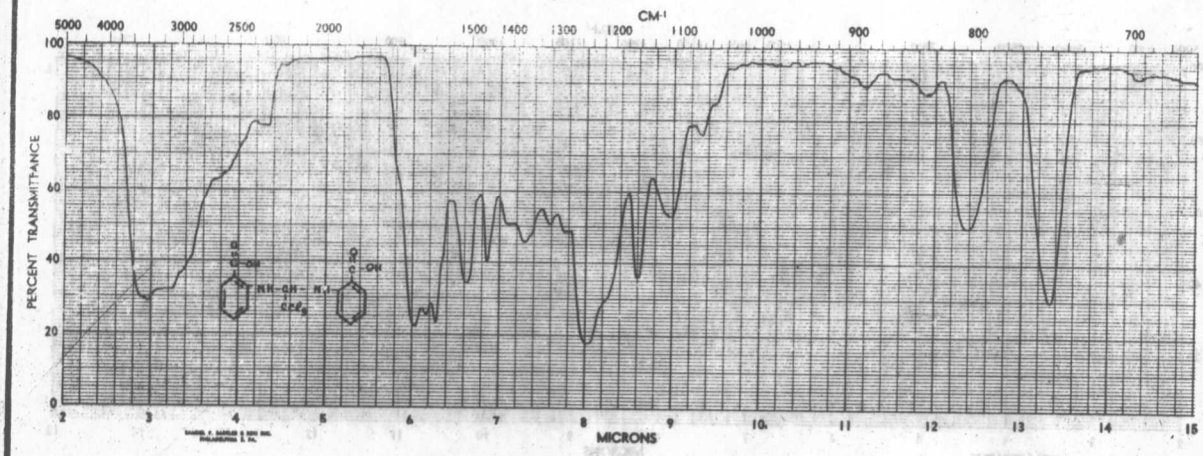


$C_{16}H_{13}Cl_3N_2O_4$ Mol. Wt. 403.65

M.P. 163-164°C

Source of Sample: E. Koshimura, Kanazawa University
Kanazawa, Japan

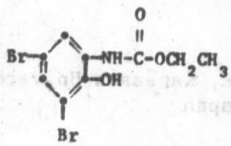
KBr Wafer






# 46008 P

3,5-DIBROMO-2-HYDROXYCARBANILIC ACID, ETHYL ESTER

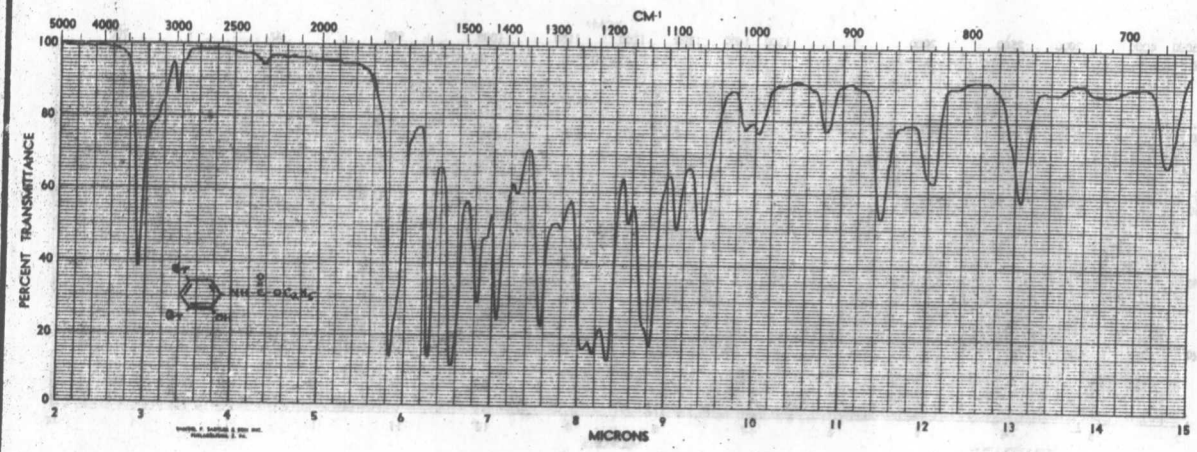


$C_9H_9Br_2NO_3$ Mol. Wt. 339.00

M.P. 124-125°C

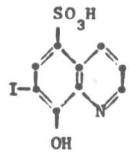
Source of Sample: E. Koshimura, Kanazawa University
Kanazawa, Japan

KBr Wafer



46009 P

8-HYDROXY-7-IODO-5-QUINOLINESULFONIC ACID

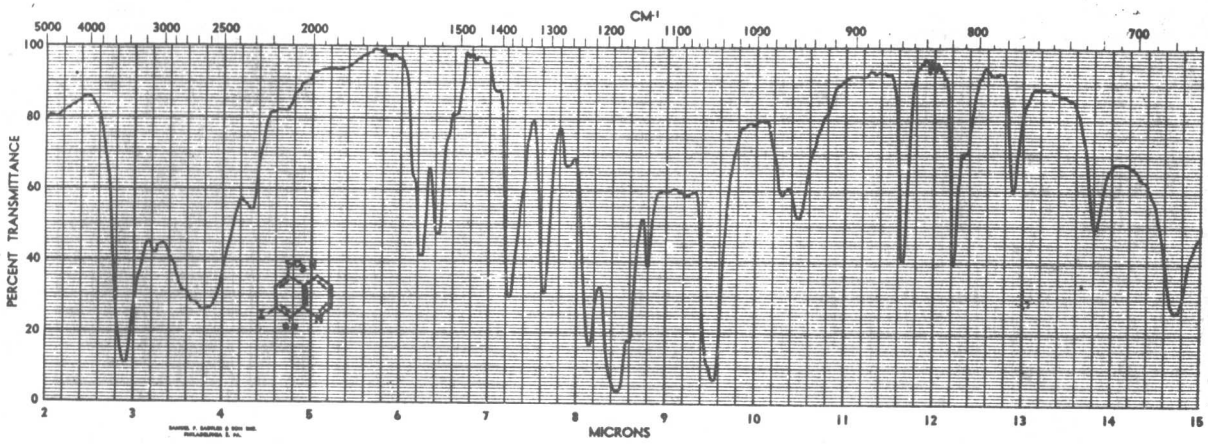


$C_9H_6INO_4S$ Mol. Wt. 351.12

M.P. 260°C (dec.)

Source of Sample: E. Koshimura, Kanazawa University
Kanazawa, Japan

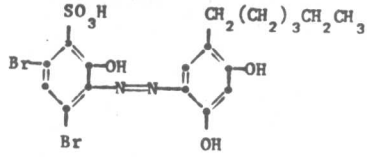
KBr Wafer






46010 P

4,6-DIBROMO-2-HYDROXY-3-[(2,4-DIHYDROXY-5-HEXYLPHENYL)AZO]-BENZENESULFONIC ACID

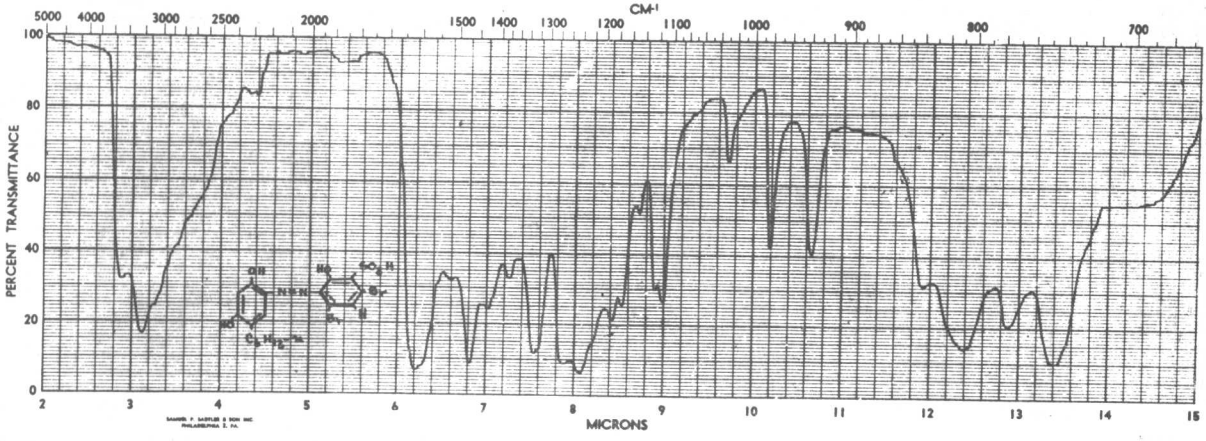


$C_{18}H_{20}Br_2N_2O_6S$ Mol. Wt. 552.25

M.P. 238°C (dec.)

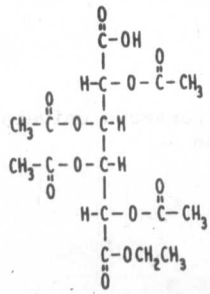
Source of Sample: E. Koshimura, Kanazawa University
Kanazawa, Japan

KBr Wafer



# 46011 P

D-GALACTARIC ACID, MONOETHYL ESTER, TETRAACETATE

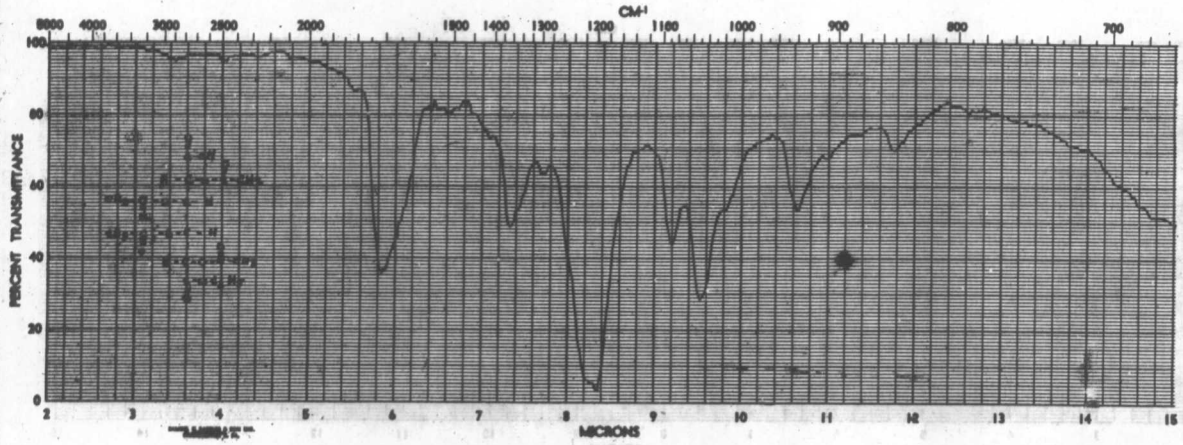


$C_{16}H_{22}O_{12}$ Mol. Wt. 406.35

M.P. 181-183°C

Source of Sample: Bealor, Niagara Food & Machinery Corp. Buffalo, New York

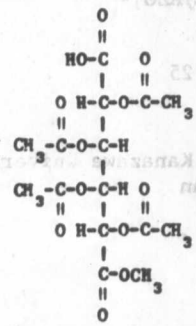
KBr Wafer






# 46012 P

D-GALACTARIC ACID, MONOMETHYL ESTER, TETRAACETATE

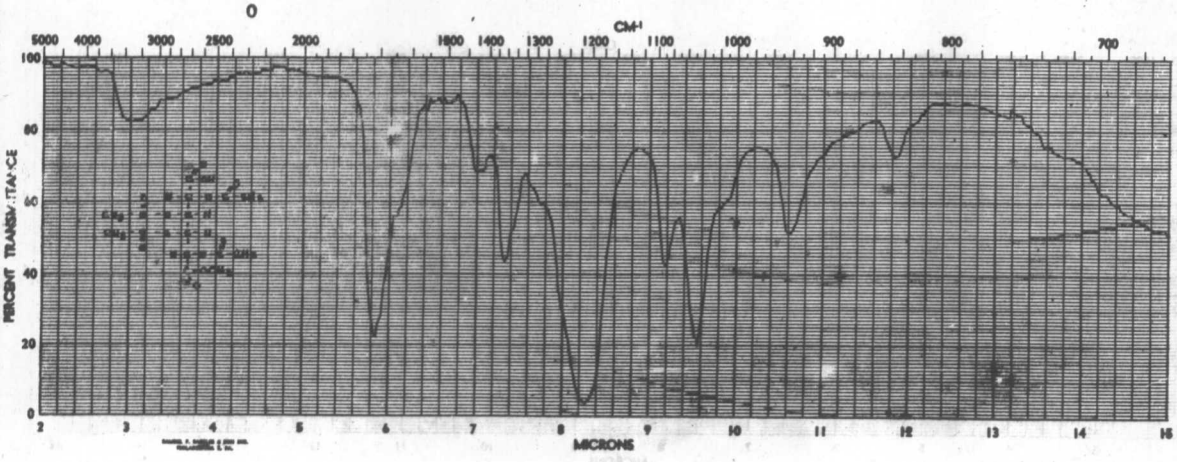


$C_{15}H_{20}O_{12}$ Mol. Wt. 392.32

M.P. 169-172°C

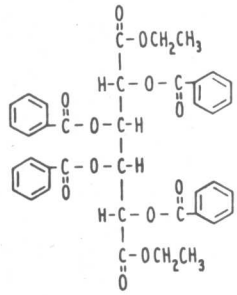
Source of Sample: Bealor, Niagara Food & Machinery Corp. Buffalo, New York

KBr Wafer



46013 P

D-GALACTARIC ACID, DIETHYL ESTER, TETRABENZOATE

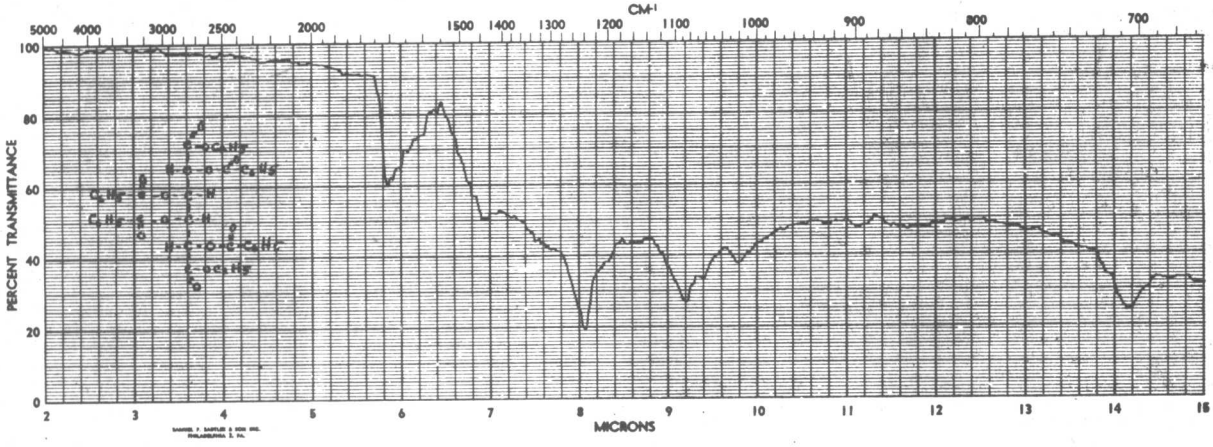


$C_{38}H_{34}O_{12}$ Mol. Wt. 682.69

M.P. 151-152.5°C

Source of Sample: Bealor, Niagara Food & Machinery Corp. Buffalo, New York

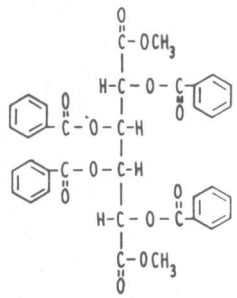
KBr Wafer






46014 P

D-GALACTARIC ACID, DIMETHYL ESTER, TETRABENZOATE

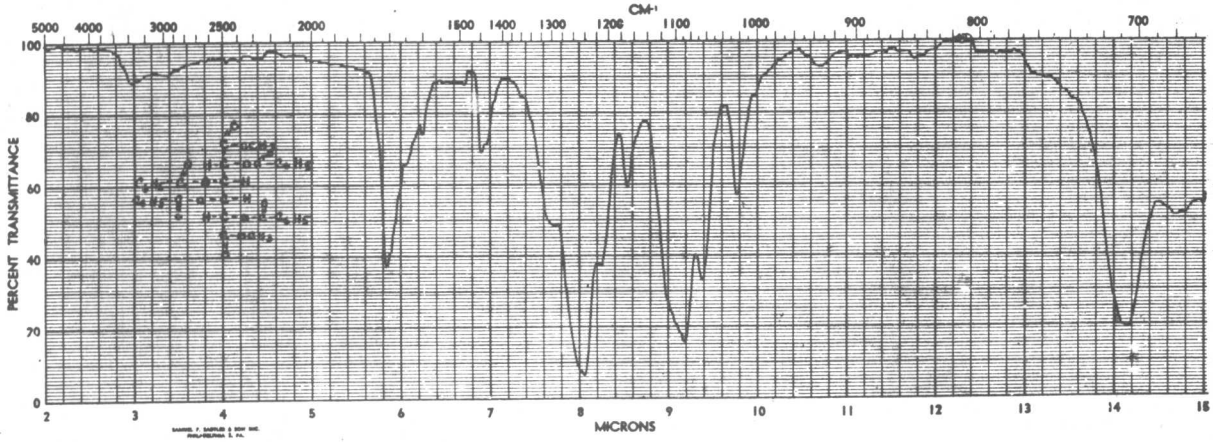


$C_{36}H_{30}O_{12}$ Mol. Wt. 654.63

M.P. 169.8-170.8°C

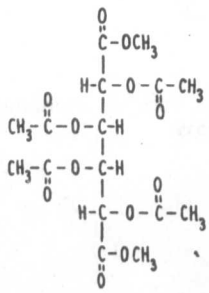
Source of Sample: Bealor, Niagara Food & Machinery Corp. Buffalo, New York

KBr Wafer



# 46015 P

D-GALACTARIC ACID, DIMETHYL ESTER, TETRAACETATE

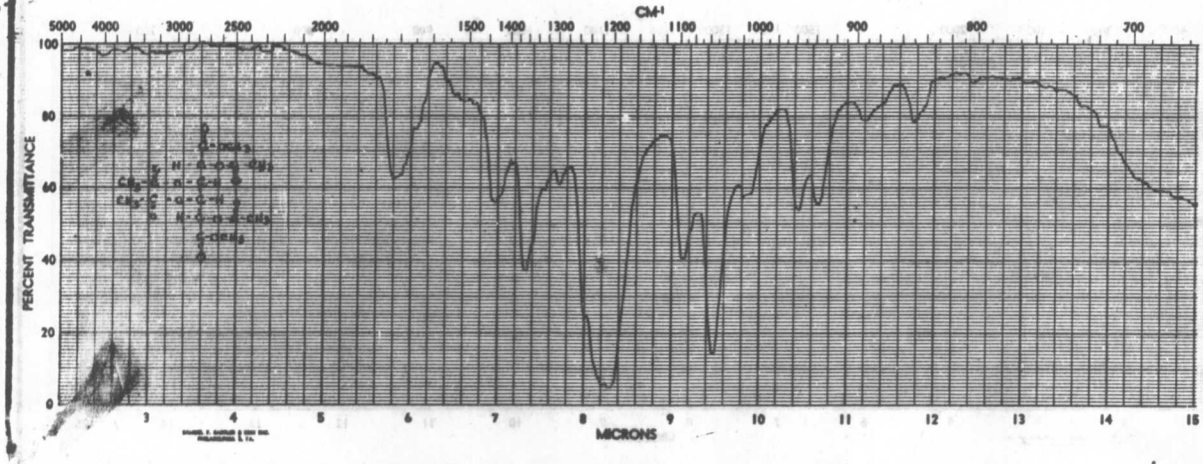


$C_{16}H_{22}O_{12}$ Mol. Wt. 406.35

M.P. 195.8-196.6°C

Source of Sample: Bealor, Niagara Food & Machinery Corp. Buffalo, New York

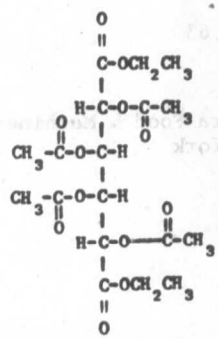
KBr Wafer






# 46016 P

D-GALACTARIC ACID, DIETHYL ESTER, TETRAACETATE

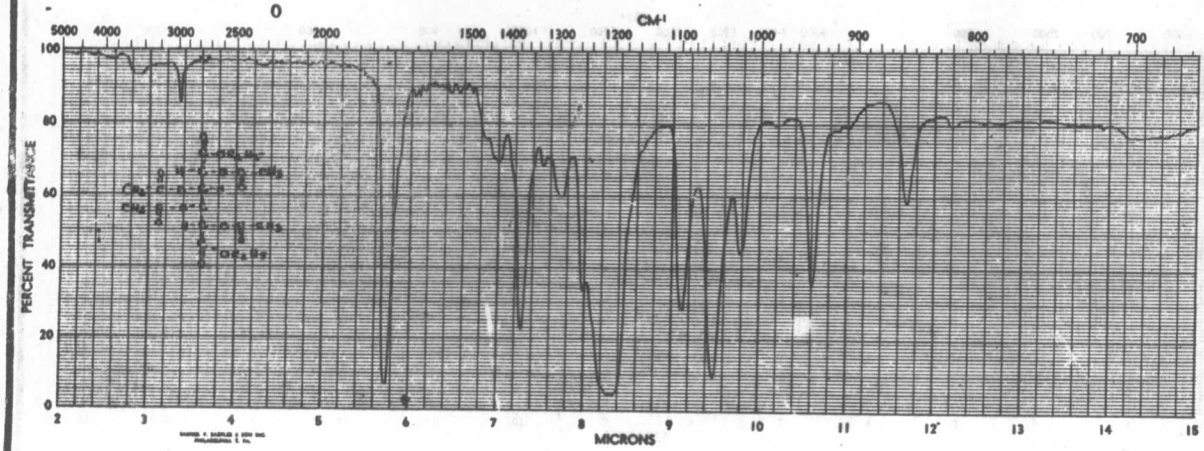


$C_{18}H_{26}O_{12}$ Mol. Wt. 434.40

M.P. 190-191°C

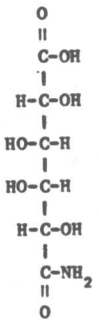
Source of Sample: Bealor, Niagara Food & Machinery Corp. Buffalo, New York

KBr Wafer



46017 P

D-GALACTARAMIC ACID

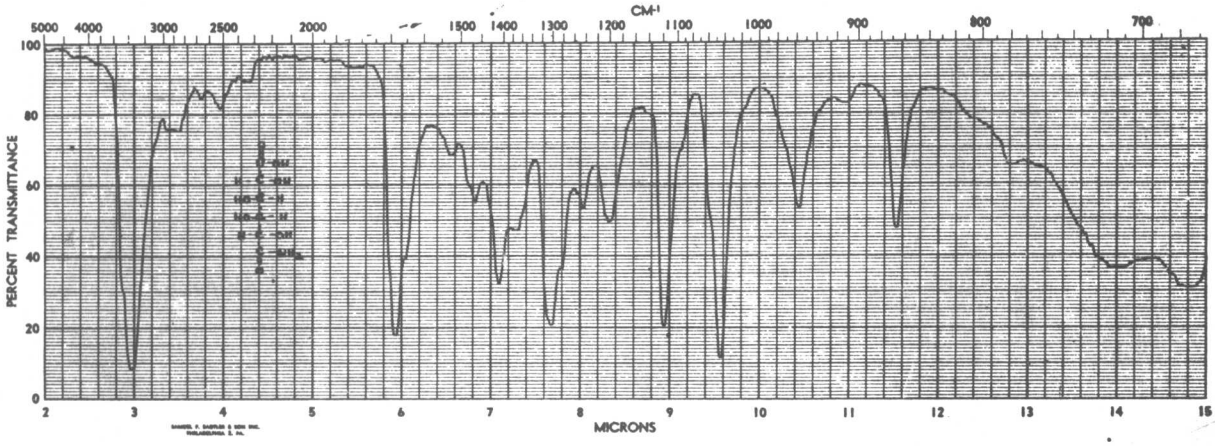


$C_6H_{11}NO_7$ Mol. Wt. 209.16

M.P. 186.5-188°C (dec.)

Source of Sample: Bealor, Niagara Food & Machinery Corp. Buffalo, New York

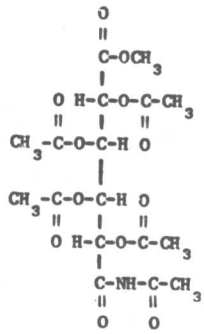
KBr Wafer






46018 P

D-GALACTARAMIC ACID, METHYL ESTER, PENTAACETATE

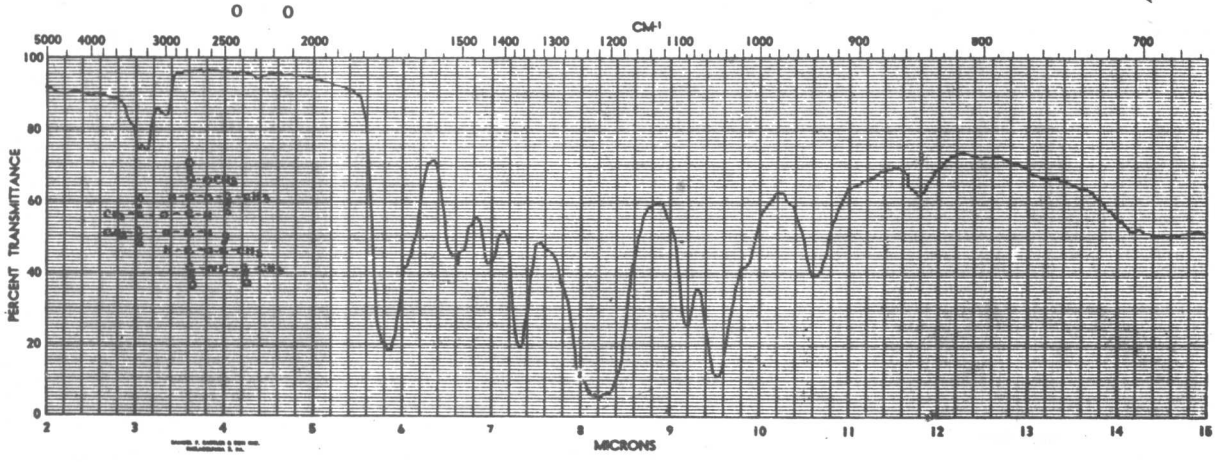


$C_{17}H_{23}NO_{12}$ Mol. Wt. 433.37

M.P. 219-222°C

Source of Sample: Bealor, Niagara Food & Machinery Corp. Buffalo, New York

KBr Wafer



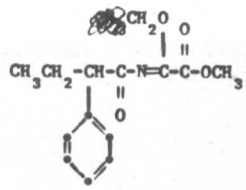
## 46019 P

ETHOXY[(2-PHENYLBUTYRYL)IMINO]ACETIC ACID, METHYL ESTER

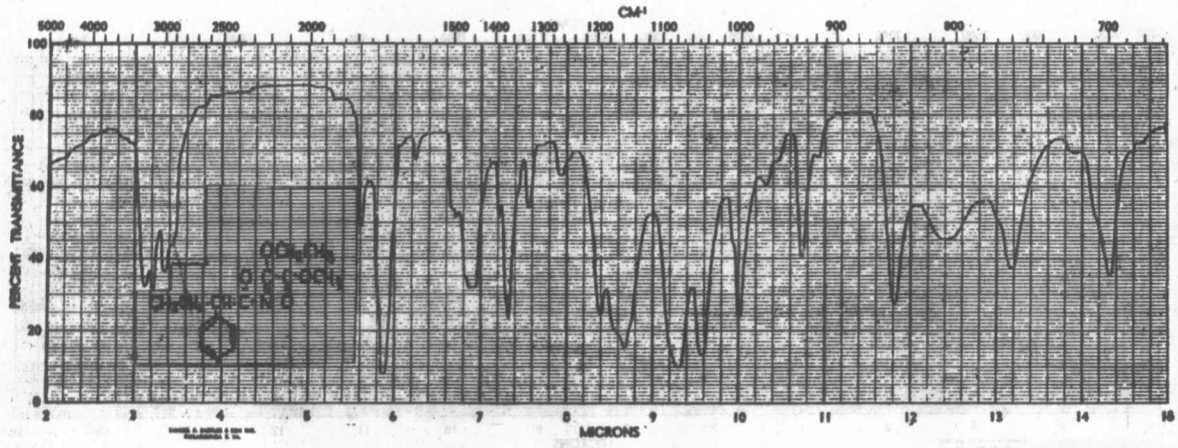
$C_{14}H_{17}NO_4$ Mol. Wt. 263.30

M.P. 120-121°C

Spectrum determined at: University of Delaware
Newark, Delaware



KBr Wafer





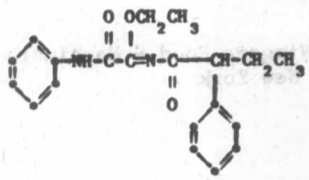

## 46020 P

N-(2-PHENYLBUTYRYL)OXANILIMIDIC ACID, ETHYL ESTER

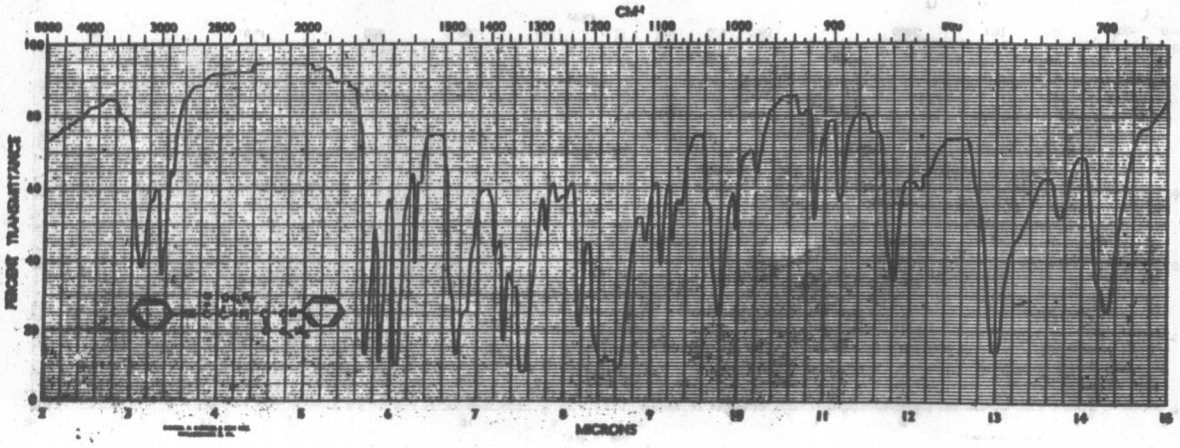
$C_{20}H_{22}N_2O_3$ Mol. Wt. 338.41

M.P. 120-121°C

Spectrum determined at: University of Delaware
Newark, Delaware

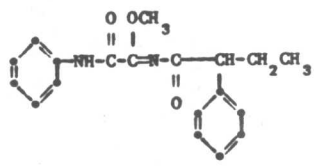


KBr Wafer



46021 P

## N-(2-PHENYLBUTYRYL)OXANILIMIDIC ACID, METHYL ESTER

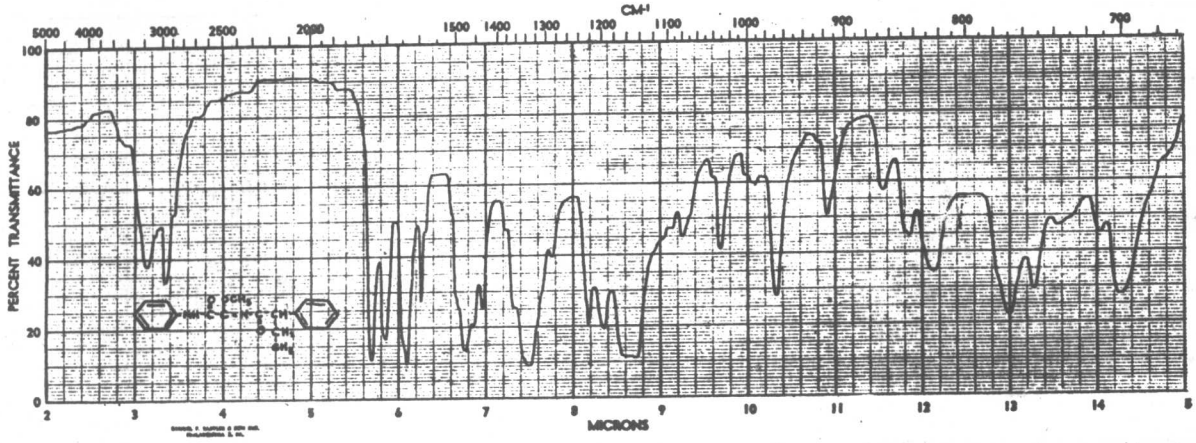


$C_{19}H_{20}N_2O_3$ Mol. Wt. 324.38

M.P. 73-74°C

Spectrum determined at: University of Delaware
Newark, Delaware

KBr Wafer






46022 P

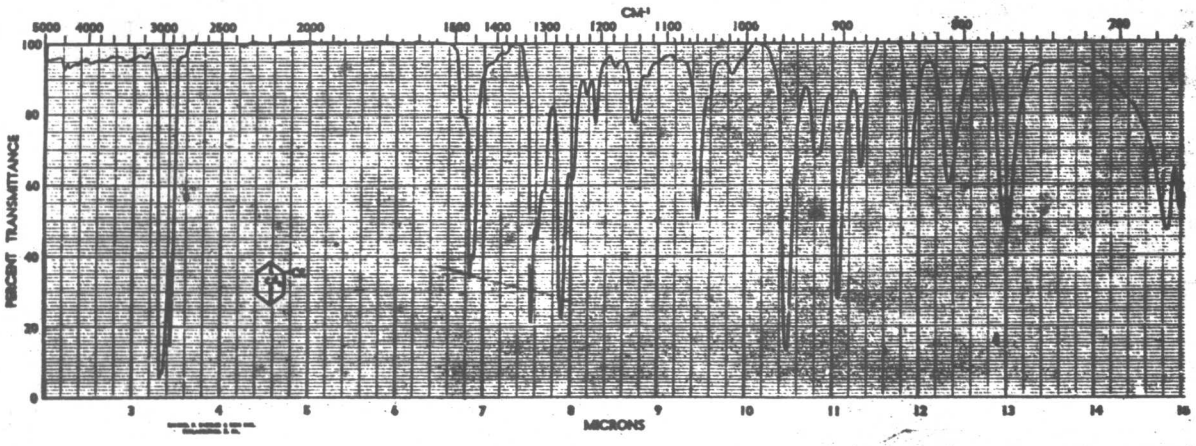
## exo-2-CHLORONORBORNANE

$C_7H_{11}Cl$ Mol. Wt. 130.62

B.P. 54-55.5°C/16mm

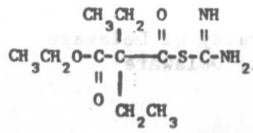
Spectrum determined at: University of Delaware
Newark, Delaware

Capillary Cell: .03mm



# 46023 P

DIETHYLTHIOMALONIC ACID, O-ETHYL ESTER, ANHYDROSULFIDE
WITH 2-THIOPSEUDOUREA

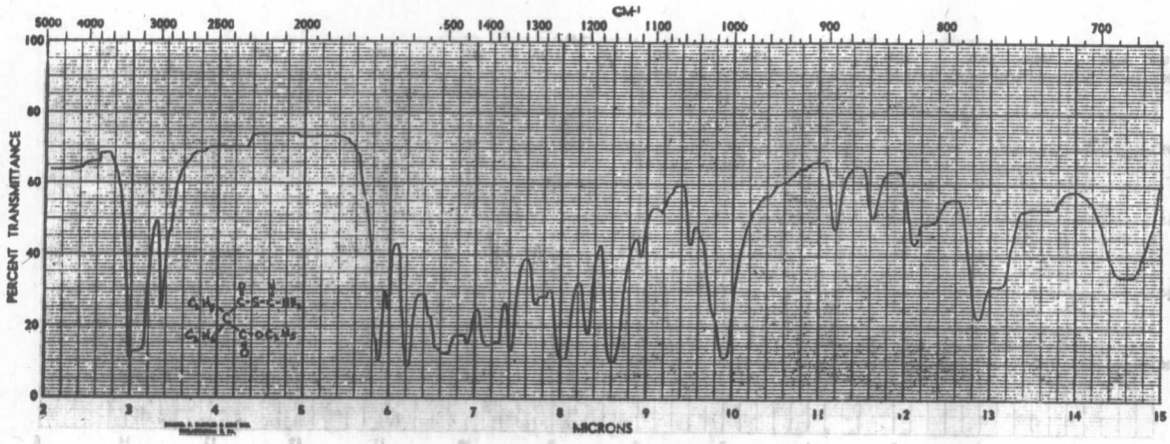


$C_{10}H_{18}N_2O_3S$ Mol. Wt. 246.33

M.P. 107-107.5°C

Spectrum determined at: University of Delaware
Newark, Delaware

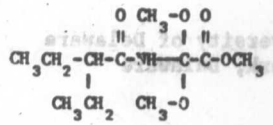
KBr Wafer






# 46024 P

2,2-DIMETHOXY-N-(2-ETHYLBUTYRYL)GLYCINE, METHYL ESTER



$C_{11}H_{21}NO_5$ Mol. Wt. 247.29

M.P. 99-100°C

Spectrum determined at: University of Delaware
Newark, Delaware

KBr Wafer