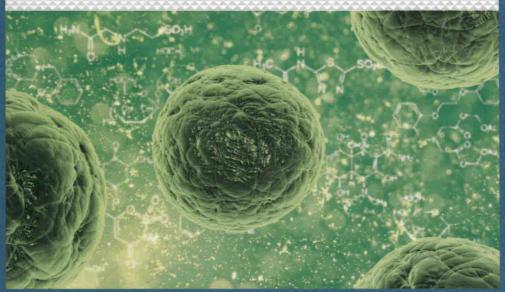
for Engineering and Applied Sciences

Polymeric Materials and Processing



Editors

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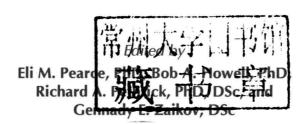




PHYSICAL CHEMISTRY RESEARCH FOR ENGINEERING AND APPLIED SCIENCES

VOLUME 2

Polymeric Materials and Processing





Apple Academic Press Inc. 3333 Mistwell Crescent Oakville, ON L6L 0A2 Canada Apple Academic Press Inc. 9 Spinnaker Way Waretown, NJ 08758

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Exclusive worldwide distribution by CRC Press, a member of Taylor & Francis Group

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Printed in the United States of America on acid-free paper

International Standard Book Number-13: 978-1-77188-057-2 (Hardcover)

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Library and Archives Canada Cataloguing in Publication

Physical chemistry research for engineering and applied sciences / edited by Eli M. Pearce, PhD, Bob A. Howell, PhD, Richard A. Pethrick, PhD, DSc, and Gennady E. Zaikov, DSc.

Includes bibliographical references and index.

Contents: Volume 2. Polymeric materials and processing.

ISBN 978-1-77188-057-2 (v. 2 : bound)

- 1. Chemistry, Physical and theoretical. 2. Chemistry, Technical. 3. Physical biochemistry.
- I. Pearce, Eli M., author, editor II. Howell, B. A. (Bobby Avery), 1942-, author, editor

III. Pethrick, R. A. (Richard Arthur), 1942-, author, editor IV. Zaikov, G. E. (Gennadii Efremovich), 1935-, author, editor

QD453.3.P49 2015

541

C2015-900409-8

Library of Congress Cataloging-in-Publication Data

Physical chemistry research for engineering and applied sciences/Eli M. Pearce, PhD, Bob A. Howell, PhD, Richard A. Pethrick, PhD, DSc, and Gennady E. Zaikov, DSc.

volumes cm

Includes bibliographical references and index.

Contents: volume 1. Principles and technological implications -- volume 2. Polymeric materials and processing -- volume 3. High performance materials and methods

ISBN 978-1-77188-053-4 (alk. paper)

1. Chemistry, Physical and theoretical. 2. Chemistry, Technical. 3. Physical biochemistry. I. Pearce, Eli M. II. Howell, B. A. (Bobby Avery), 1942- III. Pethrick, R. A. (Richard Arthur), 1942- IV. Zaikov, G. E. (Gennadii Efremovich), 1935-

QD453.3.P49 2015

541--dc23

2015000878

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VOLUME 2

Polymeric Materials and Processing



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Dr. Eli M. Pearce was the President of the American Chemical Society. He served as Dean of the Faculty of Science and Art at Brooklyn Polytechnic University in New York as well as a Professor of Chemistry and Chemical Engineering. He was the Director of the Polymer Research Institute, also in Brooklyn. At present, he consults for the Polymer Research Institute. As a prolific author and researcher, he edited the *Journal of Polymer Science* (Chemistry Edition) for 25 years and was an active member of many professional organizations.

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Physical Chemistry Research for Engineering and Applied Sciences:

Volume 3: High Performance Materials and Methods

Editors: Eli M. Pearce, PhD, Bob A. Howell, PhD,

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LIST OF ABBREVIATIONS

AFM Atomic Force Microscopy

AP Aromatic Polyesters
CNT Carbon Nanotubes
CTZ Polysaccharide Chitosan
DCV- GCMD Dual-Volume GCMD

DM Dibenzothiazole Disulphide

DP Diamond Pore

DSC Differential Scanning Calorimetry

EB Electron Beam
ER Epoxy Resin
F Flexible
FG Fiber Glass

GCMD Grand Canonical Molecular Dynamics

HR Heat Radiation

HTS High Temperature Shearing

IUPAC International Union of Pure and Applied Chemis-

try

LDPE Low Density Polyethylene LQPS Liquid-Crystal Polyesters

MC Monte Carlo

MD Molecular Dynamics
MF Microfiltration

MSD Mean-Square Displacement
MWCO Molecular Weight Cut-Off
MWNT Multi-Walled Carbon Nanotube
NBR Acrylonitrile-Butadiene Rubber
NCN National Science Centre Poland

NF Nanofiltration
NR Natural Rubber

OIT Oxidation Induction Time
OOT Oxidation Onset Temperature

PEG Polyethylene Glycols

PFR Phenoloformaldehyde Resin PMS Polymethyl-Silsesquioxane

PP Polypropylene

PUE Polyurethane Elastomers

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R Rigid

RESPA Reference System Propagator Algorithm

RHR Rate of Heat Release RO Reverse Osmosis

ROA Rheometrics Optical Analyzer
SALS Small Angle Light Scattering
SBR Styrene-Butadiene Rubber
SE Secondary Electron Signal
SEM Scanning Electron Microscope

SFE Surface Free Energy

SP Straight Path

SWNT Single-Walled Carbon Nanotube TPES Thermoplastic Elastomers

TS Tensile Strength
UF Ultrafiltration
US Ultrasound

VACF Velocity Autocorrelation Function

ZP Zigzag Path

LIST OF SYMBOLS

α_{\circ}	amplitude of the initial disturbance		
$\alpha_{\scriptscriptstyle B}$	amplitude of the instability		
$\frac{u_{\scriptscriptstyle B}}{u}$	average molecular speed		
(6.6)	coefficients of the Fourier expansion		
$A_{m's}$ P_{S}^{*}	constant		
$c_{j\sigma}^{+}$	creation operators of electrons		
	electron hopping integral		
t_{Δ}			
$\mathcal{E}_{l\sigma}$	energy of the electron by the impurity Fermi annihilation		
$c_{j\sigma}$			
$f_{i}(\mathbf{p},\mathbf{r})$	Fermi distribution function		
J_A	fugacity		
V_{ij}	matrix element of hybridization		
$egin{aligned} f_A \ V_{lj} \ \xi_i \ C_h^* \ D_A^* \end{aligned}$	random numbers generated for each trial		
C_h	saturation constant		
	self-diffusion coefficient		
Γ	surface tension		
$\eta_{\scriptscriptstyle m}$	viscosity of the matrix material		
A	proportionality constant		
a_1 and a_2	unit vectors		
aq	energy barrier		
B	hole affinity constant		
c(x)	concentration		
c _A	concentration of diffusant A		
D	pore diameter		
e_1 , e_2 and e_3	coordinates in the current configuration		
eV	electron energy		
F	applied external force		
G_1, G_2	material coordinates of a point in the initial configuration		
J	molecular flux		
K	temperature dependent Henry's law coefficient		
$K_{\rm n}$	Knudsen number		
$K_0^{"}$	proportionality constant		
K _P	henry's law constant		
L	membrane thickness		
M	molecular mass		

xviii List of Symbols

m_{sample} sample weight M_n molecular weight

N number of carbon atoms in the lattice

quantization numbernthe number of molecules

 N_{imp} number of adsorbed hydrogen atoms

P permeability

 $p_{\rm x}$ parallel component of the graphene sheet

Q heat of adsorption

R radius of the modeled SWCNT

 R_0 initial radius of the undisturbed fibril

r pore radius

S solubility coefficient

T thickness of the adsorbate film

T₀ constants depending on some quantum mechanical values

T_{melt} melting point of polyamide

U constant of the Coulomb repulsionV center-of-mass velocity component

V hybridization potential V'', V' volumes of polysaccharide

 V_L molecular volume of the condensate X position across the membrane

 Γ interfacial tension

 Δp pressure drop across the membrane

 $\begin{array}{ll} \Delta S & \text{entropy change} \\ \Delta H & \text{fusion heat} \\ \Theta & \text{time-lag} \end{array}$

mean free path of molecules

 $\rho(x)$ an arbitrary probability distribution function

 ρ'', ρ' density of polysaccharide

T pore tortuosity

 Ω known function of the viscosity ratio