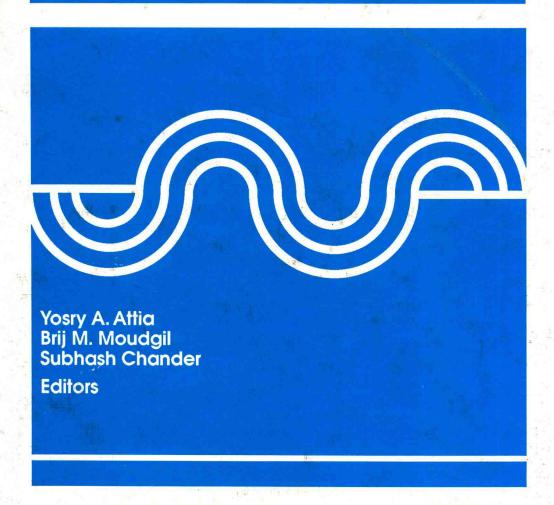
Process Technology Proceedings, 7

INTERFACIAL PHENOMENA IN BIOTECHNOLOGY AND MATERIALS PROCESSING



Elsevier

Process Technology Proceedings, 7

Interfacial Phenomena in Biotechnology and Materials Processing

Proceedings of the International Symposium on Interfacial Phenomena in Biotechnology and Materials Processing, August 3–7, 1987, Boston, Massachusetts, U.S.A.

Edited by

Yosry A. Attia

The Ohio State University, Department of Metallurgical Engineering, Mining Engineering Division, Columbus, Ohio, U.S.A.

Brij M. Moudgil

University of Florida, Department of Materials Science and Engineering, Gainesville, Florida, U.S.A.

and Associate Editor

S. Chander

The Pennsylvania State University, Mineral Processing Section, University Park, Pennsylvania, U.S.A.



ELSEVIER SCIENCE PUBLISHERS B.V. Sara Burgerhartstraat 25 P.O. Box 211, 1000 AE Amsterdam, The Netherlands

Distributors for the United States and Canada:

ELSEVIER SCIENCE PUBLISHING COMPANY INC. 52, Vanderbilt Avenue
New York, NY 10017, U.S.A.

LIBRARY OF CONGRESS Library of Congress Cataloging-in-Publication Data

```
International Symposium on Interfacial Phenomena in Biotechnology and
 Materials Processing (1987 : Boston, Mass.)
   Interfacial phenomena in biotechnology and materials processing :
 proceedings of the International Symposium on Interfacial Phenomena
  in Biotechnology and Materials Processing, August 3-7, 1987, Boston,
 Massachusetts, U.S.A. / editors, Yosry A. Attia, Brij M. Moudgil ;
  associate editor, S. Chander.
      p. cm. -- (Process technology proceedings ; 7)
    Bibliography: p.
    ISBN 0-444-42980-8 : fl 300.00

    Surface chemistry—Congresses.
    Biotechnology—Congresses.

  3. Materials--Congresses. I. Attia, Yosry A., 1945-
  II. Moudgil, Brij M. III. Chander, S., 1946-
  V. Series.
  QD506.A1I58 1987
                                                              88-16027
  660'.6--dc19
                                                                  CIP
```

ISBN 0-444-42980-8 (Vol. 7) ISBN 0-444-42382-6 (Series)

© Elsevier Science Publishers B.V., 1988

All rights reserved. No part of this publication may be reproduced, stored in a retrieval system or transmitted in any form or by any means, electronic, mechanical, photocopying, recording or otherwise, without the prior written permission of the publisher, Elsevier Science Publishers B.V./ Physical Sciences & Engineering Division, P.O. Box 330, 1000 AH Amsterdam, The Netherlands.

Special regulations for readers in the USA — This publication has been registered with the Copyright Clearance Center Inc. (CCC), Salem Massachusetts. Information can be obtained from the CCC about conditions under which photocopies of parts of this publication may be made in the USA. All other copyright questions, including photocopying outside of the USA, should be referred to the copyright owner, Elsevier Science Publishers B.V., unless otherwise specified.

No responsibility is assumed by the publisher for any injury and/or damage to persons or property as a matter of products liability, negligence or otherwise, or from any use or operation of any methods, products, instructions or ideas contained in the material herein.

pp. 171-186, 217-236: Work for the United States Government, not subject to copyright.

Printed in The Netherlands

Interfacial Phenomena in Biotechnology and Materials Processing

Process Technology Proceedings

Vol. 1 Multi-Phase Flow and Heat Transfer III (Proceedings of the Third Multi-Phase Flow and Heat Transfer Symposium-Workshop, Miami Beach, Florida, U.S.A., April 18–20, 1983), edited by T.N. Veziroğlu and A.E. Bergles (Parts A and B)

Vol. 2 Industrial Crystallization (Proceedings of the 9th Symposium on Industrial Crystallization, The Hague, The Netherlands, September 25-28,

1984), edited by S.J. Jančić and E.J. de Jong

Vol. 3 Supercritical Fluid Technology, edited by J.M.L. Penninger, M. Radosz,

M.A. McHugh and V.J. Krukonis

Vol. 4 Flocculation in Biotechnology and Separation Systems (Proceedings of the International Symposium on Flocculation in Biotechnology and Separation Systems, San Francisco, California, U.S.A., July 28-August 1, 1986), edited by Y.A. Attia

Vol. 5 Preconcentration and Drying of Food Materials (Thijssen Memorial Symposium — Proceedings of the International Symposium on Preconcentration and Drying of Foods, Eindhoven, The Netherlands, November 5-

6, 1987), edited by S. Bruin

Vol. 6 Industrial Crystallization 87 (Proceedings of the 10th Symposium on Industrial Crystallization, Bechyne, Czechoslovakia, September 21-25,

1987), edited by J. Nývlt and S. Žáček

Vol. 7 Interfacial Phenomena in Biotechnology and Materials Processing (Proceedings of the International Symposium on Interfacial Phenomena in Biotechnology and Materials Processing, Boston, Massachusetts, U.S.A., August 3-7, 1987), edited by Y.A. Attia, B.M. Moudgil and S. Chander

INTERNATIONAL SYMPOSIUM ON INTERFACIAL

PHENOMENA IN BIOTECHNOLOGY AND

MATERIALS PROCESSING

18th Annual Meeting of the Fine Particle Society Boston, Massachusetts, August 3-7, 1987

EXECUTIVE COMMITTEE

Y.A. Attia Ohio State University B.M. Moudgil University of Florida S. Chander Pennsylvania State University

ORGANIZING COMMITTEE

Yosry A. Attia, Chairman Ohio State University Columbus, Ohio

K.P. Anatha Union Carbide Corporation Terrytown, New York

G. Barbery Laval University Sainte-Foy, Quebec, Canada

A. Bleier Oak Ridge National Laboratory Oak Ridge, Tennessee

S. Chander Pennsylvania State University University Park, Pennsylvania

D.W. Fuerstenau University of California Berkeley, California

R. Hogg Pennsylvania State University University Park, Pennsylvania

F.T. Hong Wayne State University Detroit, Michigan

W. Hu University of Utah Salt Lake City, Utah R.G. Jenkins Pennsylvania State University University Park, Pennsylvania

M. Labib David Sarnoff Research Center SRI International Princeton, New Jersey

A. Marabini
Institute per il trattamento
dei minerali
Rome, Italy

B. Moudgil University of Florida Gainesville, Florida

S. Rao Rutgers University Piscataway, New Jersey

D.O. Shah University of Florida Gainesville, Florida

R.H. Yoon Virginia Polytechnic Institute Blacksburg, Virginia

Cover design: Elsa Drake

PREFACE

The importance of interfacial phenomena in the processing of biological organic and inorganic materials has been increasingly recognized in recent years. Many processes such as separations, stabilization, aggregation, preparation, formation, transport, lubrication, adhesion, etc. depend primarily on interfacial behavior of the materials involved. The utilization of interfacial and colloidal properties of materials are numerous and include such applications as protein separation, biomembranes, contact lenses, bioengineering of surface-active materials, bacterial transport, bioleaching, superconducting materials, high-performance ceramics, advanced minerals separations, metals processing and composite materials. In materials processing, for example, the development of high-performance ceramics and superconductors requires component reliability with low cost. To meet these goals, refinement of existing forming processes or the development of new processes is required. An understanding of the interactions between particles constituting high-performance materials is recognized as a critical factor in the development of the required technology. In advanced materials processing via the chemical route, synthesis of inorganic, organic, polymer and surface chemistry is involved. It is recognized that controlling the interfacial chemistry during the early stages would lead to the production of materials with the desired properties. Interfacial phenomena are of major significance also in mineral processing. The efficiency of techniques such as selective flocculation for the beneficiation of finely disseminated low grade ores depends critically on the surface chemistry of the particles and solution chemistry of the polymers employed. Understanding the dispersion and aggregation behavior of fine particle suspensions with and without chemical additives is therefore a prerequisite to developing suitable solid-solid separation technology.

This book presents the edited proceedings of an international symposium at which researchers reported the latest discoveries and theoretical and experimental developments leading to the recent advances in the scientific principles and applications of interfacial phenomena in many fields of biotechnology and materials processing. All the papers included in this book, which were selected from the symposium proceedings of over 60 papers, have been peer-reviewed and revised accordingly. Major technical areas covered in this book are: Interfacial Phenomena in Biotechnology; Interfacial Phenomena in Advanced Materials Processing; Interfacial Phenomena in Minerals Processing; and Colloid Formation and Characterization. This book will be of particular interest to researchers, graduate students, and all persons involved in biotechnology, ceramics, superconducting materials, biomedical applications, minerals and energy conservation, as well as investors in industry and new technology.

The success of the symposium and the realization of this book was due to the great cooperation received from the authors, reviewers and the organizing committee, all of whom we would like to thank very much. The financial support of the Ohio State University, which made possible the editing of this book, is gratefully acknowledged. The book cover design was made possible by the University of Florida and we are indebted to them. We would like to thank Patty Permar, Toni DiGeranimo and Chris Schulz as well as Dr. Attia's graduate students, Catherine Dentan, Shanning Yu and Farshad Bavarian, for their help and assistance in editing the manuscripts in a timely manner.

February 5, 1988 Columbus, Ohio Yosry A. Attia Brij M. Moudgil Subhash Chander

TABLE OF CONTENTS

Symposium Organizing Committee	٧
Foreword	vii
DADT 1 INTERCACIAL DUENOMENA IN DIOTECTION CON	
PART 1. INTERFACIAL PHENOMENA IN BIOTECHNOLOGY	1
Bacterial Cell Attachment and Sulfur Leaching in Microbial Coal Desulfurization by Sulfolobus Acidocaldarius C.C.Y. Chen and D.R. Skidmore	3
Surface Tension Effects on Bacteria Transport through Porous Media J.R. Chen, D. Momeni, J.F. Kuo and T.F. Yen	17
The Electric Potential Profile due to Discrete Charges in the Inhomogeneous Interfacial Regions V.S. Vaidhyanathan	27
Nonlinear Effects of Interfacial Electrical Fluctuations and Oscillations on Membrane Enzymes R.D. Astumian and T.Y. Tsong	
Use of a Hydrophobic Molecular Sieve for the Separation of Alcohol from Dilute Aqueous Solutions C.D. Chriswell and R. Markuszewski	57
A Microbial Biosurfactant: Genetic Engineering and Applications W.R. Finnerty and M.E. Singer	75
Interfacial Phenomena in Pigment-containing Biomembranes F.T. Hong	
The Utilization of Electrical Diffuse Double Layer Theory in Understanding Transport Phenomena in Synthetic and in Biological Membrane Systems M. Bender	105
Adsorption of Polymers on Contact Lens Surfaces	
K. Kumar and D.O. Shah	117
PART 2: INTERFACIAL PHENOMENA IN ADVANCED MATERIALS	131
Generation of Aerosol Particles by Bubbles R. Williams and J.R. Nelson	133

The Surface Chemistry of High-Tc Superconducting YBa ₂ Cu ₃ O ₇ Material M.E. Läbib ^X and P.J. Zanzucchi	9
The Dynamics of Growth of Silica Particles from Alkoxides T. Matsoukas and E. Gulari	9
Suspension Properties of Alumina and Titania System: Effect of Added Surface Active Agents A.S. Rao	3
Colloidal Stability of Oxidized Silicon Particles in Ethanolic and Aqueous Media E.M. DeLiso and A.A. Bleier	1
Electrokinetic Behavior of Molybdenum, Silicon and Combustion-Synthesized Molybdenum Disilicide Powders Dispersed in Aqueous Media S.C. Deevi, C.K. Law and A.S. Rao	37
Electrokinetic Behavior of Silicon Carbide, Aluminum Nitride, Titanium Diboride, Titanium Carbide and Boron Carbide Dispersed in Dilute Aqueous Dispersions S.C. Deevi, C.K. Law and A.S. Rao	01
Effects of Adsorption of Polyacrylic Acid on the Stabilization of α -Al $_2$ 0 $_3$, m-Zr0 $_2$ and their Binary Suspension Systems A. Bleier and C.G. Westmoreland	17
Surface Chemical and Adsorption Characteristics of Magnetic Ferrite and Oxide Particles in Nonaqueous Systems G.F. Hudson, S. Raghavan and M.A. Mathur	37
Dispersion Stability of Alumina in the Presence of Jeff Amines A.S. Rao	47
PART 3: INTERFACIAL PHENOMENA IN MINERALS PROCESSING	59
The Interaction of Copper Silicate and Copper Hydroxide Surfaces with Aqueous Octylhydroxamate R. Herrera-Urbina and D.W. Fuerstenau	61
Adsorption of Metallic Ions and Surfactants onto the Coal Surface Q. Yu, W. Hu and M. Guo	75
Adsorpotion of Oleate on Dolomite and Apatite B.M. Moudgil, T.V. Vasudevan, D. Ince and M. May	85

Application of Fluorosurfactants in Selective Flotation of Coal R.W. Lai and M.L. Gray	293
Flotation of Oxidized Chalcopyrite with Hydroxamate Collectors K.K. Das and Pradip	305
Effects of Process Parameters on the Selective Flocculation Cleaning of Upper Freeport Coal Y.A. Attia and K.H. Driscoll	317
Induction Time Measurements for a Coal Flotation System J.L. Yordan and R.H. Yoon	333
Possibility of Using Starches in Selective Flocculation of a Rutile Ore A. Marabini, M. Barbaro and A. Falbo	345
Microbubble Flotation of Fine Particles R.H. Yoon, G.T. Adel, G.H. Luttrell, M.J. Mankosa and A.T. Weber	363
Effect of Water Recycling on Selective Flocculation of Coal H. Soto, P. Dauphin, G. Barbery	375
Surfactant Adsorption and Wetting Behavior of Freshly Ground and Aged Coal B.R. Mohal and S. Chander	385
Interfacial and Colloidal Effects in Liquid-liquid Separation of Ultrafine Coal T.C. Hsu and S. Chander	
Surfactant Enhanced Electro-Osmotic Dewatering in Mineral Processing C.S. Grant and E.J. Clayfield	
PART 4: COLLOID FORMATION AND CHARACTERIZATION	427
Assessment of Film Flotation Efficacy for the Characterization of Solid Surfaces D.W. Fuerstenau, J.L. Diao, K.S. Narayanan and R. Herrera-Urbina	429
Formation and Characterization of Ultrafine Silver Halide Particles M.J. Hou and D.O. Shah	
New Developments in Knowledge of Aluminum Colloids J.Y. Bottero, D. Tchoubar, J.M. Cases, J.J. Fripiat and F. Fiessinger	
Hydrosols from Low-Rank Coals: Low Temperature Oxidations E.S. Olson, J.W.Diehl and M.L. Froehlich	

Entrapment and Entrainment in the Selective Flocculation Process, Part 1: Mechanisms and Process Parameters Y.A. Attia and S. Yu	491
Entrapment and Entrainment in the Selective Flocculation Process, Part 2: Possible Methods for the Minimization of Entrapment and Entrainment S. Yu and Y.A. Attia	503
Use of Flocculant for Filtration and Dewatering of Fine Coals S.R. Fang, Y.S. Cheng and S.H. Chiang	
Detection of Low Concentration of Proteins by Solubilization and Desolubilization with the Anionic Surfactant Dodecyl Sulfate A. Lopez-Valdivieso, E.J. Platt, K. Karlsen and G.L. Firestone	531
Polymers in Flocculation and Agglomerate Bonding R. Hogg and D.T. Ray	543

n	٨	D	т	1

INTERFACIAL PHENOMENA IN BIOTECHNOLOGY

BACTERIAL CELL ATTACHMENT AND SULFUR LEACHING IN MICROBIAL COAL DESULFURIZATION BY Sulfolobus Acidocaldarius

Charles C. Y. Chen, and Duane R. Skidmore Department of Chemical Engineering, The Ohio State University, Columbus, Ohio 43210

SUMMARY

Pre-leaching cell attachment and bacterial leaching were studied on finely divided Kentucky #9 coal and an Ohio coal mixture. Rates and extents of Sulfolobus acidocaldarius cell attachment on the coals were determined. Batch leaching of the coals by the microorganism were investigated by monitoring sulfate and iron ions released to the leaching solution. The Kentucky #9 coal adsorbed more cells and yielded higher sulfur and iron leaching rates. The results were interpreted in terms of mechanisms of microbial sulfur oxidation, surface areas of the coals and sulfur-containing grain sizes in the coals.

INTRODUCTION

Bacterial cell attachment is closely related to sulfur leaching in microbial coal desulfurization. The mechanism for pyritic sulfur oxidation was first proposed by Silverman (ref. 1) for Thiobacillus species as comprised of direct and indirect oxidative components. The direct mechanism requires direct contact between bacteria and pyrite since no extracellular enzymes are involved. In the indirect mechanism, the ferric iron ion chemically reacts with pyrite to give ferrous ions and elemental sulfur. The bacteria then oxidize ferrous ions to ferric ions and oxidize elemental sulfur to sulfate ions. The chemical and biological reactions are summarized as follows (ref. 1):

$$2 \text{ FeS}_2 + 7 \text{ 0}_2 + 2 \text{ H}_2 \text{0} \xrightarrow{\text{Bacteria}} 2 \text{ FeSO}_4 + 2 \text{ H}_2 \text{SO}_4$$
 (1)

$$2 \text{ FeSO}_4 + 1/2 \text{ O}_2 + \text{H}_2 \text{SO}_4 \xrightarrow{\text{Bacteria}} \text{Fe}_2(\text{SO}_4)_3 + \text{H}_2 \text{O}$$
 (2)

Overall reaction:

$$2 \text{ FeS}_2 + \text{H}_2\text{O} + 15/2 \text{ O}_2 \xrightarrow{\text{Bacteria}} \text{Fe}_2(\text{SO}_4)_3 + \text{H}_2\text{SO}_4$$
 (3)

Indirect mechanism:

$$FeS_2 + Fe_2(SO_4)_3 \longrightarrow 3 FeSO_4 + 2 S$$
 (4)

$$2S + 3 O_2 + 2 H_2O \xrightarrow{Bacteria} 2 H_2SO_4$$
 (5)

$$2 \text{ FeSO}_4 + 1/2 \text{ O}_2 + \text{H}_2 \text{SO}_4 \xrightarrow{\text{Bacteria}} \text{Fe}_2(\text{SO}_4)_3 + \text{H}_2 \text{O}$$
 (2)

In the absence of bacteria, the regeneration of ferric iron ions is the rate limiting step for pyrite oxidation. The bacteria increase the pyrite oxidation rate by oxidizing ferrous iron to ferric iron ions.

The application of the bacterial mechanisms derived for Thiobacillus
ferrooxidans to Sulfolobus is not well documented. One of the reasons is that the cell wall structure of Sulfolobus is different from that of Thiobacillus by the absence of a peptidoglycan layer. However, direct or indirect mechanisms, which are the result of bacterial activities, can be assumed to be valid for Sulfolobus species because:

- 1) Sulfolobus oxidizes Fe $^{2+}$, S 0 and pyrite at acidophilic conditions with production of Fe $^{3+}$ and S0 $_4^{2-}$ (refs. 2-4),
- 2) Selective cell attachment of \underline{S} . $\underline{brierleyi}$ and of \underline{S} . $\underline{acidocaldarius}$ to pyrite have been observed (refs. 5-6)
- 3) Evidence of Fe and S at the cell attaching sites of \underline{S} . $\underline{acidocaldarius}$ on coal was reported (ref. 7)

Irreversible adhesion of <u>Thiobacillus</u> ferrooxidans on substrate surfaces has been studied by many investigators (refs. 8-9). It is generally agreed that a wetting agent is responsible for cell adhesion. Murr and Berry studied cell attachment on mineral ores by scanning electron microscopy and concluded that adhesion is selective on sulfide surfaces (ref. 5). Adsorption of <u>T. ferrooxidans</u> on coal and other particles was studied by some investigators (refs. 10-11). Some researchers even found that surfactants enhanced contact between cells and the sulfide surface and therefore increased sulfur removal rates (ref. 12). The pyrite-selective adsorption characteristics were applied to improve physical separation of coal and pyrite as described in the literature (refs. 13-14). In those studies, the properties of pyrite particle surfaces were modified by cell adsorption. Then the coal was separated by oil agglomeration, selective flocculation or froth flotation.

The attachment of <u>Sulfolobus</u> to solid particles has been reported by some investigators (ref. 3, ref. 5, ref. 15). In their study of sulfur oxidation by <u>S. acidocaldarius</u>, Shivvers and Brock observed that the cells were attached to sulfur crystals until the late exponential stage and stationary stage were achieved (ref. 16). Weiss also studied the phenomenon of cell attachment for <u>S. acidocaldarius</u> and found that the cells attached to sulfur crystals by means of pili (ref. 15). He indicated that cell