Further Scientific English Practice

G. C. Thornley MA PH D

Further Scientific English Practice

G. C. Thornley MAPHD



Longman Group Limited London

Associated companies, branches, and representatives throughout the world

© Longman Group Limited 1972

All rights reserved. No part of this publication may be reproduced, stored in a retrieval system, or transmitted in any form or by any means, electronic, mechanical, photocopying, recording, or otherwise, without prior permission of the Copyright owner

First published 1972 Third impression 1974

ISBN 0 582 52436 9

Printed in Great Britain by Western Printing Services Ltd, Bristol

Foreword

This book provides reading material of scientific interest that is harder than the pieces to be found in *Scientific English Practice*. The passages come from the writings of scientists in the forefront of modern thinking, and it is hoped that students will find them interesting.

Vocabulary

More difficult words are needed in this book than in Scientific English Practice. At the end of each passage there is a glossary, which explains the words and phrases printed in italics in the texts. An important point to be noted is that the glossaries explain words as they are used in the particular passage; this may not always be the word's most common meaning. The words included are those that are not among the 5,000 most common English words, as well as more common words that are used with a special scientific meaning. At the end of the book, a word index lists the words from all the glossaries in alphabetical order, and shows in which glossary a word can be found.

Pronunciation

The pronunciation of every word in the glossaries is given in a simple international transcription. British pronunciation is used, and the transcription is similar to that in, for example, An International Reader's Dictionary by Dr Michael West (Longman, 1965 and later reprints). The symbols used are as follows (in alphabetical order as far as possible):

VOWELS				CONSONANTS		
[a]	as in bad	[bad]	[b]	as in big	[big]	
[a:]	as in far	[faː]	[d]	as in day	[dei]	
ai	as in mine	[main]	$[d_3]$	as in jump	[dʒʌmp]	
[au]	as in now	[nau]	[f]	as in for	[foː]	
[e]	as in bed	[bed]	[g]		[get]	
[ei]	as in say	[sei]	[h]		[him]	
[eə]	as in where	[weə]	[j]	as in yes	[jes]	
[ə]	as in above	[ə'bʌv]	[k]		[kʌm]	
[ə:]	as in her	[həː]	[1]		[laik] '	
[i]	as in h <i>i</i> t	[hit]	[m]	as in my	[mai]	
[iː]	as in meet	[mi:t]	[u]	as in no	[nou]	
[iə]	as in hear	[hiə]	[ŋ]	as in bring	[briŋ]	
[o]	as in not	[not]	[p]		[put]	
[o:]	as in for	[fo:]	[r]	as in run	[rʌn]	
[oi]	as in boy	[boi]	[s]		[sou]	
[ou]	as in go	[gou]	[t]		[top]	
[u]	as in put	[put]	[t]]	as in <i>ch</i> air	[t∫eə]	
[u:]	as in boot	[buːt]	[v]		['veri]	
[uə]	as in sure	[ʃuə]	[w]		[wiː]	
[A]	as in cut	[kʌt]	[z]		[iz]	
			[0]	as in through		
			[ð]		[ðis]	
			[0]		[ʃiː]	
			[3]	as in pleasure	['pleʒə]	

Syllabic consonants have the mark, below them, as in table ['teib]]. The mark ' is written before the syllable that has the main stress, as in ['veri].

Exercises

The exercises that follow each passage are probably harder than those in the earlier books, because more difficult words and structures are needed for these subjects. The Index to Language Questions, at the end of the book, shows where exercises on a particular structure can be found.

vii

Acknowledgements

We are grateful to the following for permission to reproduce copyright material:

Aldus Books Ltd for an extract from Man, Nature and History by W. M. S. Russell, copyright 1967, Aldus Books Ltd. London: Batsford (B.T.) Ltd for an extract from Collecting Rocks & Fossils (1966) by J. B. Delair; G. Bell & Sons Ltd for a slight adaptation of An Approach to Modern Physics by Prof. Andrade: Ernest Benn Ltd for an extract from Discovery II (1962) by Dr H. M. Finniston: Blackie & Son Ltd for an extract from Digital Computers by Marchant & Pegg; Adam and Charles Black for an adapted extract from Peace or Atomic War by Albert Schweitzer; David & Charles for an extract from Animal Senses by Robert Burton; Garnstone Press for an extract from Oddities by Lt Comdr. Rupert Gould, @ Geoffrey Bles; extracts reproduced by permission of The Hamlyn Group Ltd from Microscopes and Microscopic Life by Peter Healey and Rapid Car-Fault Diagnosis by G. C. Sneed; George G. Harrap & Co. Ltd for an adapted extract from Chemistry for the Modern World by Prof. George Porter; Heinemann Educational Books Ltd for an adapted extract from Frontiers of Astronomy by F. Hoyle; Prof. R. C. Jennison for an extract from Introduction to Radio Astronomy by Prof. R. C. Jennison; Macmillan, London and Basingstoke, for an extract from Water and Waste by J. H. Stephens and an adapted extract from A Star Called the Sun by George Gamow; Longman Group Ltd and the authors for an adapted extract from Roads by Prof. R. G. Batson and an extract from Roads (second edition) by J. A. Proudlove: Frederick Muller Ltd for an extract from Television (Mechanical Age Library) by T. J. Morgan; Penguin

Books Ltd for an adapted extract from Sledge by Martin Lindsay: Author's agent and The Viking Press Inc. for an adapted extract from The Story of Eyes by S. Sutton-Vane: Author's Agents, Methuen & Co. Ltd and Holt. Rinehart & Winston Inc. for an extract from Radiation Genes and Man by Bruce Wallace and Theodore Dobzhansky; Souvenir Press Ltd for an extract from The Story of the Laser by J. M. Carroll; University of London Press Ltd for extracts from the following: Teach Yourself Botany by Dr J. H. Elliott, Engineering Metallurgy Part 1 by Raymond Higgins, Electricity and Magnetism by C. G. Wilson, Teach Yourself Chemistry revised by James Knight and The Depth of Cold by Dr A. R. Meetham: George Weidenfeld & Nicolson Ltd for an extract from The World of Rome by Michael Grant and Winant Towers Ltd for an adapted extract from All About Nuclear Power by David Le Roi.

Contents

I	The Interior of the Earth, Sir Fred Hoyle	I
2	Radioactivity, Albert Schweitzer	7
3	Land of Ice, Martin Lindsay	. 13
4	Fuels, George Porter	21
5	Nuclear Energy, David Le Roi	27
6	Binary System for Computers, Marchant and Pegg	32
7	Fermat's Last Theorem, Rupert T. Gould	38
8	Electric Charges, C. G. Wilson	44
9	Superconductivity, A. R. Meetham	49
10	Neutrons, H. M. Finniston	56
IL	Radio Carbon, E. N. da C. Andrade	62
12	Television, T. J. Morgan	68
13.	The Purification of Water, J. H. Stephens	74
14	The Story of Eyes, S. Sutton-Vane	80
15	The Surface Temperature of the Sun, G. Gamow	86
16	Photosynthesis, John H. Elliott	91
17	Lasers, John M. Carroll	97
18	Aluminium and its Alloys, Raymond Higgins	103
19	The Electron Microscope, Peter Healey	109
20	Steam Turbines, Rayner Joel	116
21	The Control of Population, W. M. S. Russell	123
22	Flexible Road Bases, R. G. Batson	130
23	Animal Senses, Robert Burton	137
24	Sparking-Plug Polarity in Car Engines, G. Sneed	143
25	Concrete in Roman Architecture, Michael Grant	150
26	Synthetic Products, Knight and MacAlpine	157
27.		163
28	Radio Telescopes, R. C. Jennison	170
29	Chromosomes, Wallace and Dobzhansky	176
30	Fossils and Oil, Justin B. Delair	183
	Glossary Index	189
	Index to Language Questions	197

1 The Interior of the Earth Sir Fred Hoyle

Sir Fred Hoyle (born 1915) is one of the world's foremost astronomers. He is also a great mathematician, and was lecturer in mathematics at Cambridge University before he became a professor. One of his many books is Frontiers of Astronomy (1955), from which the piece below is taken.

The planet we live on is not just a ball of *inert* material. During past ages *dramatic* changes have taken place inside the earth. Indeed, it is likely that without these changes life could never have *originated* on the earth. And changes are still going on today. They show themselves in the occurrence of *earthquakes*, in the outbursts of *volcanoes*, and in the uplift of mountain *ranges*.

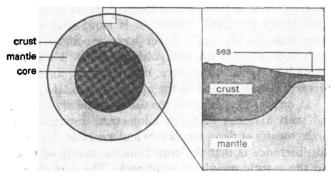
In outward appearance, the earth is a nearly spherical ball with a radius of 6,350 kilometres. Internally the earth consists of two parts, a core and a mantle. An essential difference is that the core consists mainly of liquid and the mantle mainly of solid rock. The core extends outwards from the centre to a distance of some 3,450 kilometres. The mantle, as its name shows, is an outer covering extending from the core to the surface of the earth.

Judged by ordinary standards, the core is made of rather dense stuff. The material at the centre of the earth is at least thirteen times as heavy as ordinary water, while in the outer parts of the core the material is about ten times as heavy as ordinary water.

The mantle possesses a thin outer crust that is exceptional in being composed of a particularly light kind of rock, with a density about 2.7 times that of water. (Compare this with a density of 13 at the centre of the earth.) Over

r The Interior of the Earth

the continents of the world this crustal rock is about thirty-five kilometres thick, while over the oceans it is at most only two or three kilometres thick. Below the crustal layer comes a different, denser rock, probably of a basic silicate variety. Indeed it seems likely that, apart from the thin outer crust, the rocks of the whole mantle are of a basic silicate variety right down to the junction with the core, at a depth below the surface of about 2,900 kilometres.



Cross-section of the earth.

We must now introduce the idea that the pressures occurring inside the earth are very considerable. It is well known that at sea level our atmosphere exerts a pressure of about fifteen pounds per square inch. This in itself is no mean pressure³, as we all soon come to realise⁴ if we have to pump up an automobile tyre. But the pressure inside the earth is vastly greater than this, amounting to tens of millions of pounds per square inch. At such enormous pressures, ordinary rock becomes appreciably squashed. Therefore, as we go inwards to greater and greater depths, the density of the rocks of the earth's

r The Interior of the Earth

mantle increases. The density immediately below the outer crust is about 3.3 times that of water. We may compare this with a density of 4.0 at a depth of 500 kilometres, 4.5 at 1,000 kilometres, about 5.0 at 2,000 kilometres, and with about 5.6 at the surface of the core, which is at a depth of 2,000 kilometres.

The last of these values is important. We are now saying that in the part of the mantle immediately outside the core, the density is about 5.6 times that of water. On the other hand immediately inside the core the density is about 9.7. This means that at the surface of the core there is not only a change from liquid on the inside to solid on the outside, but there is also a very considerable change in the density of the material, from 9.7 on the inside to 5.6 on the outside. This change gives an important clue to the nature of the material⁵ in the core.

Explanation of words in italic

WORDS appreciably [ə'pri:siəbli] considerably astronomer [2'stronama] someone who studies the stars basic ['beisik] (here) general clue [klu:] something that helps one to find the answer to a problem core [ko:] the most central part crust [krast] the outer surface dense [dens] thick density ['densiti] how dense something is dramatic [dra'matik] very exciting, as in a play or drama earthquake ['a:0kweik] movement of the earth's crust inert [i'nə:t] not chemically active—does not combine with other chemicals junction ['dʒʌŋkʃn] place where a join occurs layer ['leia] a thin covering mantle ['mant]] a cover originate [a'ridzineit] to begin range [reindz] in mountain range, a chain of mountains

The Interior of the Earth

silicate ['silikit] like glass (adjective for a certain kind of rock); see phrases

spherical ['sferik]] round, like a ball or sphere [sfia]

squash [skwof] press together very hard

value ['valju:] (here) measurement, quantity

variety [vəˈraiəti] type

volcano [vol'keinou] a kind of mountain or hill that sometimes throws out hot liquid rock from the inside of the earth

PHRASES

in being composed of: because it is composed of

- ² of a basic silicate variety: (rock) that belongs to the general silicate (glass-like) type
- ⁸ no mean pressure: not at all a small (insignificant) pressure
- 4 come to realise: to be made to realise (understand)
- 5 the nature of the material: what the material is like

1 Exercises

COMPREHENSION

- How do changes inside the earth show themselves?
- 2 What variations are there in the thickness of the crustal rock?
- 3 Give some examples of the densities of the material inside the earth.
- 4 What is the pressure of our atmosphere at sea level? What is the earth's diameter?
- What great change takes place at the outer part of the core?
 WRITE A SHORT DESCRIPTION, in your own words, of the interior of the earth.

ESSAY

- The results of earthquakes.
- 2 Will men ever go to the centre of the earth? (Give reasons for your answer.)

LANGUAGE

Fill each space with one of these words in its right form: inert; spherical; radius; dense; outer; crust; considerable; tyre; occur; clue.

The Interior of the Earth

- a. The material inside the earth is near the core than near the surface.
- b. The detective found a which showed him who the thief was.
- c. The diameter of a circle is twice its
- d. Argon is a gas that does not combine with other chemicals; it is
- e. No one ever pumps up the spare of my car.
- f. Accidents rarely when one is careful.
- g. I still have a amount of work to do.
- h. A ball is in shape.
- i. You need good teeth to bite a hard of bread.
- The walls of the building were not strong enough, and it fell down.
- 2 Give the adjectives formed from the following nouns: sphere; crust; drama; exception; importance. Give the nouns formed from the following adjectives and verbs: occur; dense; vary; deep; press.
- 3 Rewrite the following sentences. Start in each case with the words given in brackets:
 - a. Compare this with a density of 13 at the centre. (This should be)
 - b. We soon realise this if we have to pump up an automobile tyre. (If an automobile tyre)
 - c. This change gives an important clue to the nature of the material in the core. (An important clue)
 - d. We must now introduce a new idea. (A new idea)
- e. They show themselves in the occurrence of earthquakes. (They are)
- 4 a. Change the phrase in italics to one that begins with If, and make any other necessary changes: Without these changes, life could never have originated on the earth.
 - b. Use other words for the phrase in italics without changing the meaning: Apart from the outer crust the rocks of the whole mantle are of a basic silicate variety.
 - c. Put which in the following and make any necessary changes: The pressures occurring inside the earth are very considerable.

The Interior of the Earth

- d. How do we pronounce the dot in 3.3? What is .3 of 100? e. What is a junction on a railway?
- The word kilometre has two parts. Kilo- is a Greek prefix that means 1,000, and a metre is a certain length. Many other prefixes from Greek and Latin are used in scientific words. Here are some:

 centi-: one hundredth. A centimetre is one hundredth of a metre. milli-: one thousandth. What is a millimetre?

These two prefixes may also be used for having a hundred, having a thousand. What is a centipede and what is a millipede? What is a centigrade thermometer scale?

Dec- (from the Latin DECEM = ten) is also used. A decade is ten years. December used to be the tenth month. Decapoda are creatures with ten legs, such as crabs. A decagon is a geometrical figure having ten angles and ten sides.

Some more prefixes are: Tri- 3 (triangle). Quadr- 4 (quadrangle). Pent- 5 (pentagon). Hex- 6 (hexagon). Hept- or Sept- 7 (heptagon). Oct- 8 (octagon). Non- or Novem- 9 (nonane; November).

Give some other scientific words starting with such pre-

fixes and say what the words mean.

- 6 Notice 10 times as heavy as water. Write sentences containing the following:
 - a. ten times as long as;
 - b. three times as dense as;
 - c. half as difficult as.

2 Radioactivity

Albert Schweitzer

Albert Schweitzer was born on January 14th, 1875, at Kayserberg, Alsace. He spent eight years in the Lycée at Mulhouse, where he learnt to play the organ extremely well. At the age of eighteen, in 1893, he left the Lycée and went to study at Strasbourg. He was unusually clever, and he was already well known when he decided, at the age of thirty, to add medicine to his subjects of study. In that year he became a medical student at Strasbourg, and he remained there for six more years. He married in 1912. On 20th March, 1913, he set out for Africa, where he became world-famous as a doctor and a humanitarian. He died in 1965.

The piece given below is from a talk which he made from Oslo radio station on April 24th, 1957. It may be found complete in Norman Cousins' book, Dr Schweitzer of Lambaréné (published in 1961). Its subject is the danger to everyone in the world from radiation.

My thanks to the radio station in Oslo, the city of the Nobel Peace Prize, for making it possible for me to speak to far-off places.

What is radioactivity?

Radioactivity consists of rays differing from those of light in being invisible, 1 and able to pass not only through glass but also through thin metal discs and layers of cell tissue in human and animal bodies. Rays of this kind were first discovered in 1895 by the physicist, Wilhelm Roentgen of Munich, and named after him.

In 1896 the French physicist Henri Becquerel demonstrated that rays of this kind occur in nature. They are emitted from uranium, an element known since 1786.

In 1898 Pierre Curie and his wife discovered in the *mineral* pitchblende, which is a uranium *ore*, the strongly radioactive element *radium*.

2 Radioactivity

The joy that such rays were at the disposal of humanity² was at first unmixed³. For the rays appeared to destroy the fast-growing cells of terrible growths in the body. But after a time it was found that the destruction of these cells does not always mean a cure, and that the normal cells of the body may be seriously damaged if they are exposed to radioactivity for a long time.

When Mme Curie, after handling uranium ore for four years, finally held the first gramme of radium in her hand, abrasions that no treatment could cure appeared in her skin. With the years, she grew more and more ill from a disease caused by radioactive rays. They damaged her bone marrow and, through this, her blood. In 1934 death ended her suffering.

Even so, for many years we were not aware of the grave risks *involved* in X-rays to those who are constantly exposed to them. Through operating X-ray apparatus, thousands of doctors and nurses have incurred incurable diseases.

Radioactive rays are material things. Through these rays, the radioactive element constantly emits tiny particles of itself. These are of three kinds, named after the first three letters of the Greek alphabet: alpha, beta, gamma. The gamma rays have the strongest effect.

The reason why elements emit radioactive rays is that they are constantly decaying, and radioactivity is the energy they liberate little by little. There are other elements that are radioactive, besides uranium and radium. Some radiation from space is added to the radiation from the elements in the earth; fortunately, the air mass, 250 miles high, that surrounds our earth protects us against this radiation. Only a very small fraction of it reaches us.

We are, then, constantly being exposed to radioactive radiation coming from the earth and from space. It is so

2 Radioactivity

weak, however, that it does not hurt us. Stronger sources of radiation, as for instance X-ray machines and exposed radium, have harmful effects if one is exposed to them for some time. These rays are invisible. How can we tell that they are there, or how strong they are?

Thanks to the German physicist Hans Geiger, who died in 1945 as a victim of X-rays, we have an instrument which makes that possible.

Explanation of words in italic

WORDS

abrasion [ə'breiʒn] sore, painful skin where the surface has been rubbed off

alpha ['alfə] see GREEK ALPHABET in phrases

apparatus [apə'reitəs] equipment: scientific instruments and machines

beta ['bi:tə] see GREEK ALPHABET in phrases

cell [sel] (here) a small piece of living matter: see tissue

decay [di'kei] to rot away, die

demonstrate ['demonstreit] to show, prove

disposal [di'spouzl] (see phrases below)

element ['eliment] something that cannot be split into simpler things but is part of other matter: for example, water contains the elements hydrogen and oxygen

emit [i'mit] to send out

expose [ik'spouz] to uncover; be exposed to, to have no protection against

fraction ['fraksn] (see Exercise 6, Language)

gamma ['gamə] see GREEK ALPHABET in phrases

growth [groud] a diseased mass (of cells) growing in the body humanity [hju:'maniti] people: the human race; humanitarian [hju:mani'teəriən] someone who loves and works for

humanity

incur [in'kə:] to bring something bad upon oneself

involve [in'volv]: be involved in, (here) to be associated with marrow ['marou] the fat that is inside the bones of the body

material [mo'tioriol] (see phrases below)

mineral [imineral] a metal, etc., that comes from the ground: