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VOLUME 18



KIRK-OTHMER

ENCYCLOPEDIA OF CHEMICAL TECHNOLOGY

THIRD EDITION

VOLUME 18

**PLANT-GROWTH SUBSTANCES
TO
POTASSIUM COMPOUNDS**



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NOTE ON CHEMICAL ABSTRACTS SERVICE REGISTRY NUMBERS AND NOMENCLATURE

Chemical Abstracts Service (CAS) Registry Numbers are unique numerical identifiers assigned to substances recorded in the CAS Registry System. They appear in brackets in the *Chemical Abstracts* (CA) substance and formula indexes following the names of compounds. A single compound may have many synonyms in the chemical literature. A simple compound like phenethylamine can be named β -phenylethylamine or, as in *Chemical Abstracts*, benzenethanamine. The usefulness of the *Encyclopedia* depends on accessibility through the most common correct name of a substance. Because of this diversity in nomenclature careful attention has been given the problem in order to assist the reader as much as possible, especially in locating the systematic CA index name by means of the Registry Number. For this purpose, the reader may refer to the CAS Registry Handbook-Number Section which lists in numerical order the Registry Number with the *Chemical Abstracts* index name and the molecular formula; eg, 458-88-8, Piperidine, 2-propyl-, (S)-, $C_8H_{17}N$; in the *Encyclopedia* this compound would be found under its common name, coniine [458-88-8]. The Registry Number is a valuable link for the reader in retrieving additional published information on substances and also as a point of access for such on-line data bases as Chemline, Medline, and Toxline.

In all cases, the CAS Registry Numbers have been given for title compounds in articles and for all compounds in the index. All specific substances indexed in *Chemical Abstracts* since 1965 are included in the CAS Registry System as are a large number of substances derived from a variety of reference works. The CAS Registry System identifies a substance on the basis of an unambiguous computer-language description of its molecular structure including stereochemical detail. The Registry Number is a machine-checkable number (like a Social Security number) assigned in sequential order to each substance as it enters the registry system. The value of the number lies in the fact that it is a concise and unique means of substance identification, which is

independent of, and therefore bridges, many systems of chemical nomenclature. For polymers, one Registry Number is used for the entire family; eg, polyoxyethylene (20) sorbitan monolaurate has the same number as all of its polyoxyethylene homologues.

Registry numbers for each substance will be provided in the third edition cumulative index and appear as well in the annual indexes (eg, Alkaloids shows the Registry Number of all alkaloids (title compounds) in a table in the article as well, but the intermediates have their Registry Numbers shown only in the index). Articles such as Analytical methods, Batteries and electric cells, Chemurgy, Distillation, Economic evaluation, and Fluid mechanics have no Registry Numbers in the text.

Cross-references are inserted in the index for many common names and for some systematic names. Trademark names appear in the index. Names that are incorrect, misleading or ambiguous are avoided. Formulas are given very frequently in the text to help in identifying compounds. The spelling and form used, even for industrial names, follow American chemical usage, but not always the usage of *Chemical Abstracts* (eg, *coniine* is used instead of *(S)*-2-propylpiperidine, *aniline* instead of *benzenamine*, and *acrylic acid* instead of 2-propenoic acid).

There are variations in representation of rings in different disciplines. The dye industry does not designate aromaticity or double bonds in rings. All double bonds and aromaticity are shown in the *Encyclopedia* as a matter of course. For example, tetralin has an aromatic ring and a saturated ring and its structure appears in the



Encyclopedia with its common name, Registry Number enclosed in brackets, and parenthetical CA index name, ie, tetralin, [119-64-2] (1,2,3,4-tetrahydronaphthalene). With names and structural formulas, and especially with CAS Registry Numbers the aim is to help the reader have a concise means of substance identification.

CONVERSION FACTORS, ABBREVIATIONS, AND UNIT SYMBOLS

SI Units (Adopted 1960)

A new system of measurement, the International System of Units (abbreviated SI), is being implemented throughout the world. This system is a modernized version of the MKSA (meter, kilogram, second, ampere) system, and its details are published and controlled by an international treaty organization (The International Bureau of Weights and Measures) (1).

SI units are divided into three classes:

BASE UNITS

length	meter [†] (m)
mass [‡]	kilogram (kg)
time	second (s)
electric current	ampere (A)
thermodynamic temperature [§]	kelvin (K)
amount of substance	mole (mol)
luminous intensity	candela (cd)

[†] The spellings “metre” and “litre” are preferred by ASTM; however “-er” are used in the Encyclopedia.

[‡] “Weight” is the commonly used term for “mass.”

[§] Wide use is made of “Celsius temperature” (t) defined by

$$t = T - T_0$$

where T is the thermodynamic temperature, expressed in kelvins, and $T_0 = 273.15$ K by definition. A temperature interval may be expressed in degrees Celsius as well as in kelvins.

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SUPPLEMENTARY UNITS

plane angle	radian (rad)
solid angle	steradian (sr)

DERIVED UNITS AND OTHER ACCEPTABLE UNITS

These units are formed by combining base units, supplementary units, and other derived units (2–4). Those derived units having special names and symbols are marked with an asterisk in the list below:

<i>Quantity</i>	<i>Unit</i>	<i>Symbol</i>	<i>Acceptable equivalent</i>
*absorbed dose	gray	Gy	J/kg
acceleration	meter per second squared	m/s ²	
*activity (of ionizing radiation source)	becquerel	Bq	1/s
area	square kilometer	km ²	
	square hectometer	hm ²	ha (hectare)
	square meter	m ²	
*capacitance	farad	F	C/V
concentration (of amount of substance)	mole per cubic meter	mol/m ³	
*conductance	siemens	S	A/V
current density	ampere per square meter	A/m ²	
density, mass density	kilogram per cubic meter	kg/m ³	g/L; mg/cm ³
dipole moment (quantity)	coulomb meter	C·m	
*electric charge, quantity of electricity	coulomb	C	A·s
electric charge density	coulomb per cubic meter	C/m ³	
electric field strength	volt per meter	V/m	
electric flux density	coulomb per square meter	C/m ²	
*electric potential, potential difference, electromotive force	volt	V	W/A
*electric resistance	ohm	Ω	V/A
*energy, work, quantity of heat	megajoule	MJ	
	kilojoule	kJ	
	joule	J	N·m
	electron volt [†]	eV [†]	
	kilowatt-hour [†]	kW·h [†]	

[†] This non-SI unit is recognized by the CIPM as having to be retained because of practical importance or use in specialized fields (1).

<i>Quantity</i>	<i>Unit</i>	<i>Symbol</i>	<i>Acceptable equivalent</i>
energy density	joule per cubic meter	J/m ³	
*force	kilonewton	kN	
	newton	N	kg·m/s ²
*frequency	megahertz	MHz	
	hertz	Hz	1/s
heat capacity, entropy	joule per kelvin	J/K	
heat capacity (specific), specific entropy	joule per kilogram kelvin	J/(kg·K)	
heat transfer coefficient	watt per square meter kelvin	W/(m ² ·K)	
*illuminance	lux	lx	lm/m ²
*inductance	henry	H	Wb/A
linear density	kilogram per meter	kg/m	
luminance	candela per square meter	cd/m ²	
*luminous flux	lumen	lm	cd·sr
magnetic field strength	ampere per meter	A/m	
*magnetic flux	weber	Wb	V·s
*magnetic flux density	tesla	T	Wb/m ²
molar energy	joule per mole	J/mol	
molar entropy, molar heat capacity	joule per mole kelvin	J/(mol·K)	
moment of force, torque	newton meter	N·m	
momentum	kilogram meter per second	kg·m/s	
permeability	henry per meter	H/m	
permittivity	farad per meter	F/m	
*power, heat flow rate, radiant flux	kilowatt watt	kW W	J/s
power density, heat flux density, irradiance	watt per square meter	W/m ²	
*pressure, stress	megapascal kilopascal pascal	MPa kPa Pa	N/m ²
sound level	decibel	dB	
specific energy	joule per kilogram	J/kg	
specific volume	cubic meter per kilogram	m ³ /kg	
surface tension	newton per meter	N/m	
thermal conductivity	watt per meter kelvin	W/(m·K)	
velocity	meter per second kilometer per hour	m/s km/h	
viscosity, dynamic	pascal second	Pa·s	
	millipascal second	mPa·s	
viscosity, kinematic	square meter per second	m ² /s	

Quantity	Unit	Symbol	Acceptable equivalent
	square millimeter		
	per second	mm ² /s	
volume	cubic meter	m ³	
	cubic decimeter	dm ³	L(liter) (5)
	cubic centimeter	cm ³	mL
wave number	1 per meter	m ⁻¹	
	1 per centimeter	cm ⁻¹	

In addition, there are 16 prefixes used to indicate order of magnitude, as follows:

Multiplication

factor	Prefix	Symbol	Note
10 ¹⁸	exa	E	
10 ¹⁵	peta	P	
10 ¹²	tera	T	
10 ⁹	giga	G	
10 ⁶	mega	M	
10 ³	kilo	k	
10 ²	hecto	h ^a	^a Although hecto, deka, deci, and centi are SI prefixes, their use should be avoided except for SI unit-multiples for area and volume and nontechnical use of centimeter, as for body and clothing measurement.
10	deka	da ^a	
10 ⁻¹	deci	d ^a	
10 ⁻²	centi	c ^a	
10 ⁻³	milli	m	
10 ⁻⁶	micro	μ	
10 ⁻⁹	nano	n	
10 ⁻¹²	pico	p	
10 ⁻¹⁵	femto	f	
10 ⁻¹⁸	atto	a	

For a complete description of SI and its use the reader is referred to ASTM E 380 (4) and the article Units and Conversion Factors which will appear in a later volume of the *Encyclopedia*.

A representative list of conversion factors from non-SI to SI units is presented herewith. Factors are given to four significant figures. Exact relationships are followed by a dagger. A more complete list is given in ASTM E 380-79(4) and ANSI Z210.1-1976 (6).

Conversion Factors to SI Units

To convert from	To	Multiply by
acre	square meter (m ²)	4.047 × 10 ³
angstrom	meter (m)	1.0 × 10 ^{-10†}
are	square meter (m ²)	1.0 × 10 ^{2†}
astronomical unit	meter (m)	1.496 × 10 ¹¹
atmosphere	pascal (Pa)	1.013 × 10 ⁵
bar	pascal (Pa)	1.0 × 10 ^{5†}
barn	square meter (m ²)	1.0 × 10 ^{-28†}

† Exact.

<i>To convert from</i>	<i>To</i>	<i>Multiply by</i>
barrel (42 U.S. liquid gallons)	cubic meter (m ³)	0.1590
Bohr magneton (μ_B)	J/T	9.274×10^{-24}
Btu (International Table)	joule (J)	1.055×10^3
Btu (mean)	joule (J)	1.056×10^3
Btu (thermochemical)	joule (J)	1.054×10^3
bushel	cubic meter (m ³)	3.524×10^{-2}
calorie (International Table)	joule (J)	4.187
calorie (mean)	joule (J)	4.190
calorie (thermochemical)	joule (J)	4.184 [†]
centipoise	pascal second (Pa·s)	$1.0 \times 10^{-3}\dagger$
centistoke	square millimeter per second (mm ² /s)	1.0 [†]
cfm (cubic foot per minute)	cubic meter per second (m ³ /s)	4.72×10^{-4}
cubic inch	cubic meter (m ³)	1.639×10^{-5}
cubic foot	cubic meter (m ³)	2.832×10^{-2}
cubic yard	cubic meter (m ³)	0.7646
curie	becquerel (Bq)	$3.70 \times 10^{10}\dagger$
debye	coulomb-meter (C·m)	3.336×10^{-30}
degree (angle)	radian (rad)	1.745×10^{-2}
denier (international)	kilogram per meter (kg/m)	1.111×10^{-7}
dram (apothecaries')	tex [†]	0.1111
dram (avoirdupois)	kilogram (kg)	3.888×10^{-3}
dram (U.S. fluid)	kilogram (kg)	1.772×10^{-3}
dyne	cubic meter (m ³)	3.697×10^{-6}
dyne/cm	newton (N)	$1.0 \times 10^{-5}\dagger$
electron volt	newton per meter (N/m)	$1.0 \times 10^{-3}\dagger$
erg	joule (J)	1.602×10^{-19}
fathom	joule (J)	$1.0 \times 10^{-7}\dagger$
fluid ounce (U.S.)	meter (m)	1.829
foot	cubic meter (m ³)	2.957×10^{-5}
footcandle	meter (m)	0.3048 [†]
furlong	lux (lx)	10.76
gal	meter (m)	2.012×10^{-2}
gallon (U.S. dry)	meter per second squared (m/s ²)	$1.0 \times 10^{-2}\dagger$
gallon (U.S. liquid)	cubic meter (m ³)	4.405×10^{-3}
gallon per minute (gpm)	cubic meter (m ³)	3.785×10^{-3}
gauss	cubic meter per second (m ³ /s)	6.308×10^{-5}
gilbert	cubic meter per hour (m ³ /h)	0.2271
gill (U.S.)	tesla (T)	1.0×10^{-4}
grad	ampere (A)	0.7958
grain	cubic meter (m ³)	1.183×10^{-4}
gram force per denier	radian	1.571×10^{-2}
	kilogram (kg)	6.480×10^{-5}
	newton per tex (N/tex)	8.826×10^{-2}

† Exact.

† See footnote on p. xiv.

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<i>To convert from</i>	<i>To</i>	<i>Multiply by</i>
hectare	square meter (m ²)	$1.0 \times 10^{4\dagger}$
horsepower (550 ft-lbf/s)	watt (W)	7.457×10^2
horsepower (boiler)	watt (W)	9.810×10^3
horsepower (electric)	watt (W)	$7.46 \times 10^{2\dagger}$
hundredweight (long)	kilogram (kg)	50.80
hundredweight (short)	kilogram (kg)	45.36
inch	meter (m)	$2.54 \times 10^{-2\dagger}$
inch of mercury (32°F)	pascal (Pa)	3.386×10^3
inch of water (39.2°F)	pascal (Pa)	2.491×10^2
kilogram force	newton (N)	9.807
kilowatt hour	megajoule (MJ)	$3.6^\dagger$
kip	newton (N)	4.48×10^3
knot (international)	meter per second (m/s)	0.5144
lambert	candela per square meter (cd/m ²)	3.183×10^3
league (British nautical)	meter (m)	5.559×10^3
league (statute)	meter (m)	4.828×10^3
light year	meter (m)	9.461×10^{15}
liter (for fluids only)	cubic meter (m ³)	$1.0 \times 10^{-3\dagger}$
maxwell	weber (Wb)	$1.0 \times 10^{-8\dagger}$
micron	meter (m)	$1.0 \times 10^{-6\dagger}$
mil	meter (m)	$2.54 \times 10^{-5\dagger}$
mile (statute)	meter (m)	1.609×10^3
mile (U.S. nautical)	meter (m)	$1.852 \times 10^3\dagger$
mile per hour	meter per second (m/s)	0.4470
millibar	pascal (Pa)	1.0×10^2
millimeter of mercury (0°C)	pascal (Pa)	$1.333 \times 10^{2\dagger}$
minute (angular)	radian	2.909×10^{-4}
myriagram	kilogram (kg)	10
myriameter	kilometer (km)	10
oersted	ampere per meter (A/m)	79.58
ounce (avoirdupois)	kilogram (kg)	2.835×10^{-2}
ounce (troy)	kilogram (kg)	3.110×10^{-2}
ounce (U.S. fluid)	cubic meter (m ³)	2.957×10^{-5}
ounce-force	newton (N)	0.2780
peck (U.S.)	cubic meter (m ³)	8.810×10^{-3}
pennyweight	kilogram (kg)	1.555×10^{-3}
pint (U.S. dry)	cubic meter (m ³)	5.506×10^{-4}
pint (U.S. liquid)	cubic meter (m ³)	4.732×10^{-4}
poise (absolute viscosity)	pascal second (Pa-s)	$0.10^\dagger$
pound (avoirdupois)	kilogram (kg)	0.4536
pound (troy)	kilogram (kg)	0.3732
poundal	newton (N)	0.1383
pound-force	newton (N)	4.448
pound per square inch (psi)	pascal (Pa)	6.895×10^3
quart (U.S. dry)	cubic meter (m ³)	1.101×10^{-3}

[†] Exact.

<i>To convert from</i>	<i>To</i>	<i>Multiply by</i>
quart (U.S. liquid)	cubic meter (m ³)	9.464×10^{-4}
quintal	kilogram (kg)	$1.0 \times 10^{2\dagger}$
rad	gray (Gy)	$1.0 \times 10^{-2\dagger}$
rod	meter (m)	5.029
roentgen	coulomb per kilogram (C/kg)	2.58×10^{-4}
second (angle)	radian (rad)	4.848×10^{-6}
section	square meter (m ²)	2.590×10^6
slug	kilogram (kg)	14.59
spherical candle power	lumen (lm)	12.57
square inch	square meter (m ²)	6.452×10^{-4}
square foot	square meter (m ²)	9.290×10^{-2}
square mile	square meter (m ²)	2.590×10^6
square yard	square meter (m ²)	0.8361
stere	cubic meter (m ³)	1.0 [†]
stokes (kinematic viscosity)	square meter per second (m ² /s)	$1.0 \times 10^{-4\dagger}$
tex	kilogram per meter (kg/m)	$1.0 \times 10^{-6\dagger}$
ton (long, 2240 pounds)	kilogram (kg)	1.016×10^3
ton (metric)	kilogram (kg)	$1.0 \times 10^{3\dagger}$
ton (short, 2000 pounds)	kilogram (kg)	9.072×10^2
torr	pascal (Pa)	1.333×10^2
unit pole	weber (Wb)	1.257×10^{-7}
yard	meter (m)	0.9144 [†]

Abbreviations and Unit Symbols

Following is a list of commonly used abbreviations and unit symbols appropriate for use in the *Encyclopedia*. In general they agree with those listed in *American National Standard Abbreviations for Use on Drawings and in Text (ANSI Y1.1)* (6) and *American National Standard Letter Symbols for Units in Science and Technology (ANSI Y10)* (6). Also included is a list of acronyms for a number of private and government organizations as well as common industrial solvents, polymers, and other chemicals.

Rules for Writing Unit Symbols (4):

1. Unit symbols should be printed in upright letters (roman) regardless of the type style used in the surrounding text.
2. Unit symbols are unaltered in the plural.
3. Unit symbols are not followed by a period except when used as the end of a sentence.
4. Letter unit symbols are generally written in lower-case (eg, cd for candela) unless the unit name has been derived from a proper name, in which case the first letter of the symbol is capitalized (W, Pa). Prefix and unit symbols retain their prescribed form regardless of the surrounding typography.
5. In the complete expression for a quantity, a space should be left between the numerical value and the unit symbol. For example, write 2.37 lm, *not* 2.37lm, and 35 mm, *not* 35mm. When the quantity is used in an adjectival sense, a hyphen is often used, for example, 35-mm film. *Exception:* No space is left between the numerical value and the symbols for degree, minute, and second of plane angle, and degree Celsius.

xx FACTORS, ABBREVIATIONS, AND SYMBOLS

6. No space is used between the prefix and unit symbols (eg, kg).
7. Symbols, not abbreviations, should be used for units. For example, use "A," not "amp," for ampere.
8. When multiplying unit symbols, use a raised dot:

N·m for newton meter

In the case of W·h, the dot may be omitted, thus:

Wh

An exception to this practice is made for computer printouts, automatic typewriter work, etc, where the raised dot is not possible, and a dot on the line may be used.

9. When dividing unit symbols use one of the following forms:

$$\text{m/s or m}\cdot\text{s}^{-1} \text{ or } \frac{\text{m}}{\text{s}}$$

In no case should more than one slash be used in the same expression unless parentheses are inserted to avoid ambiguity. For example, write:

$$\text{J}/(\text{mol}\cdot\text{K}) \text{ or } \text{J}\cdot\text{mol}^{-1} \cdot \text{K}^{-1} \text{ or } (\text{J}/\text{mol})/\text{K}$$

but *not*

$$\text{J/mol/K}$$

10. Do not mix symbols and unit names in the same expression. Write:

$$\text{joules per kilogram or J/kg or J}\cdot\text{kg}^{-1}$$

but *not*

$$\text{joules/kilogram nor joules/kg nor joules}\cdot\text{kg}^{-1}$$

ABBREVIATIONS AND UNITS

A	ampere	AIChE	American Institute of
A	anion (eg, HA); mass number		Chemical Engineers
a	atto (prefix for 10 ⁻¹⁸)	AIME	American Institute of
AATCC	American Association of		Mining, Metallurgical,
	Textile Chemists and		and Petroleum Engineers
	Colorists	AIP	American Institute of
ABS	acrylonitrile-butadiene-		Physics
	styrene	alc	alcohol(ic)
abs	absolute	Alk	alkyl
ac	alternating current, <i>n</i> .	alk	alkaline (not alkali)
a-c	alternating current, <i>adj</i> .	amt	amount
ac-	alicyclic	amu	atomic mass unit
acac	acetylacetonate	ANSI	American National
ACGIH	American Conference of		Standards Institute
	Governmental Industrial	AO	atomic orbital
	Hygienists	AOAC	Association of Official
ACS	American Chemical Society		Analytical Chemists
AGA	American Gas Association	AOCS	American Oil Chemist's
Ah	ampere hour		Society

APHA	American Public Health Association	cl	carload
API	American Petroleum Institute	cm	centimeter
aq	aqueous	cmil	circular mil
Ar	aryl	cmpd	compound
ar-	aromatic	CNS	central nervous system
as-	asymmetric(al)	CoA	coenzyme A
ASH-		COD	chemical oxygen demand
RAE	American Society of Heating, Refrigerating, and Air Conditioning Engineers	coml	commercial(ly)
ASM	American Society for Metals	cp	chemically pure
ASME	American Society of Mechanical Engineers	cph	close-packed hexagonal
ASTM	American Society for Testing and Materials	CPSC	Consumer Product Safety Commission
at no.	atomic number	cryst	crystalline
at wt	atomic weight	cub	cubic
av(g)	average	D-	denoting configurational relationship
AWS	American Welding Society	d	differential operator
^b	bonding orbital	d-	dextro-, dextrorotatory
bbl	barrel	da	deka (prefix for 10 ¹)
bcc	body-centered cubic	dB	decibel
Bé	Baumé	dc	direct current, <i>n.</i>
BET	Brunauer-Emmett-Teller (adsorption equation)	d-c	direct current, <i>adj.</i>
bid	twice daily	dec	decompose
Boc	<i>t</i> -butyloxycarbonyl	detd	determined
BOD	biochemical (biological) oxygen demand	detn	determination
bp	boiling point	Di	didymium, a mixture of all lanthanons
Bq	becquerel	dia	diameter
C	coulomb	dil	dilute
°C	degree Celsius	DIN	Deutsche Industrie Normen
C-	denoting attachment to carbon	dl-, DL-	racemic
c	centi (prefix for 10 ⁻²)	DMA	dimethylacetamide
c	critical	DMF	dimethylformamide
ca	circa (approximately)	DMG	dimethyl glyoxime
cd	candela; current density; circular dichroism	DMSO	dimethyl sulfoxide
CFR	Code of Federal Regulations	DOD	Department of Defense
cgs	centimeter-gram-second	DOE	Department of Energy
CI	Color Index	DOT	Department of Transportation
cis-	isomer in which substituted groups are on same side of double bond between C atoms	dp	dew point
		DP	degree of polymerization
		DPH	diamond pyramid hardness
		dstl(d)	distill(ed)
		dta	differential thermal analysis
		(E)-	entgegen; opposed
		ε	dielectric constant (unitless number)
		e	electron