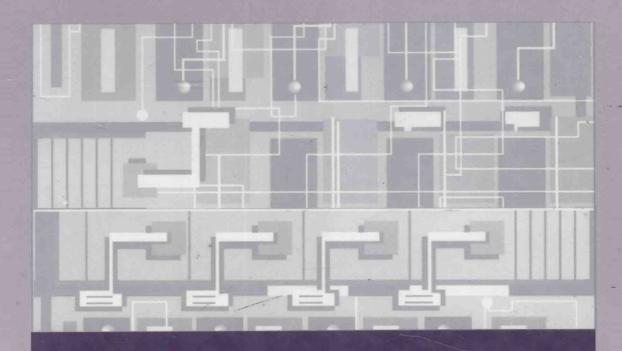
Functional and Smart Materials

Structural Evolution and Structure Analysis



Z. L. Wang Z. C. Kang

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Z. L. Wang and Z. C. Kang

Georgia Institute of Technology Atlanta, Georgia

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Functional and Smart Materials Structural Evolution and Structure Analysis

To all of our family members

Foreword

At the end of this century, the technological importance of oxides are growing extremely fast. Most of the information is transported by optical fibers because light can carry more information than conventional electromagnetic waves. This implies new microlasers and new microamplifiers where polyoxides, such as lithium niobate crystals, are required. Research on these crystals is experiencing a superfast development and many new discoveries have reached the industrial stage of large scale production. As sparked by the extraordinary discovery of high temperature superconductors in 1986, growth and characterization of oxides are a forefront research field in materials science. Less spectacular, but very important, also are the progresses made during the last thirty years in the field of industrial ceramics, for instance the dramatic improvements obtained with stabilized zirconia. Progresses are made daily in the field of "Research and Development" with oxides presenting some special physical property and functionality. The largest domain of interest is presently a possible coupling between at least two different kinds of properties (i.e., the smart structure). These progresses have been possible because of the fundamental understanding of their structure and microstructure. In the book by Z. L. Wang and Z. C. Kang one can find a very interesting concentration of basic physical properties of the most important polyoxides, related to their structures (and microstructures) and evolution behavior. The approach of Z. L. Wang and Z. C. Kang is very interesting and rather new: they have classified oxides through their structures and their physical properties. Rock, salt, rutile, fluorite, perovskite and many other related (or mixed) structure types are comprehensively described in the first four chapters with an emphasis on the connections among different structure systems. The fifth chapter is about the important process known as "Soft Chemistry" or "Chimie Douce". The second part, Chapters 6 through 8, are devoted to the optimal techniques and technologies used for study of these compounds and their physical properties. This book is unique because it focuses specifically on the intrinsic connections among several crystal structure systems and their evolution behavior. It relates the basic principles for molecular and structural design of functional materials to the fundamental structure modules. These materials are described from the mixed-valence and stoichiometry points of view to understand their structural transformation and the evolution of different materials systems.

viii FOREWORD The mixed valences of transition and rare earth metals have been shown as a fundamental for oxides with specific functionalities. There are numerous books describing the properties, preparations, electronic and crystal structures of transition, rare earth metals and their oxides. This book fills a gap in that field, not only because it focuses on the role played by the evolution of crystal structures in functional materials, but also gives the solution of structure determination through advanced techniques such as spectroscopy and transmission electron microscopy. Because this specific approach has been followed I am confident this book will be a basic reference in the domain of oxides which are to be the basis of functional and smart materials.

C. Boulesteix *Pr. Univ. Aix-Marseille 3, France*

Preface

Functional materials, a new emerging materials system, have attracted the interests of many scientists, since they are synthesized to perform specific functionality. Functional materials include but are not limited to smart materials, and they cover a large range of materials with important applications in modern and future technologies. To be unique, this book is not a compiled list of various functional materials, rather it is on the intrinsic connection and evolution behavior among and in different structure systems which are frequently observed in oxide functional materials. Each structural system is described from the basic modules that are the building blocks for constructing all of the related structures. The structural evolution is linked with mixed valences of rare earth and transitional metal elements, and this is shown to be the fundamental principle for producing new materials with unique functionality. The book aims to explore the fundamental structural mosaics that likely lead to some new routes for synthesizing new functional materials. The book is also unique in the way that it integrates structural evolution with structure analysis using transmission electron microscopy and associate techniques.

We have written this book for advanced graduate students and scientists who are interested in studying and developing functional materials. The intended readers are materials scientists, solid state chemists (material chemists), electron microscopists, mineralogists, and solid state physicists (material physicists). The book also fulfills the need as a text book for advanced research and education in oxide functional materials and transmission electron microscopy.

This book was written based on our research experience on the subject. Chapters 2–4 were primarily written by ZCK. The Introduction section, Chapters 6–8 and all of the Appendixes were written by ZLW. Chapters 1 and 5 were co-authored by ZCK and ZLW. ZLW was responsible for organizing and editing the entire manuscript, and he was also cited as liason during the publication process.

We would like to express gratitude to our collaborators related to the research described in this book. Thanks to Professor L. Eyring, a pioneer in the field, for your advice and encouragement. Thanks also go to Professor C. Boulesteix, Dr. D.M. Kroeger, Dr. Jiming Zhang and Professor R.L. Whetten for collaborative research in the past few years. We are also grateful to those who kindly permit us to use their data for illustrating the text, and each of them is acknowledged in the corresponding figure caption.

X **PREFACE**

Finally, our heartfelt gratitude goes to our wives, children, and parents, for their constant encouragment, support and understanding. Their support and help are indispensable for finishing this book.

> Zhong Lin Wang* Zhen Chuan Kang† School of Materials Science and Engineering

^{*} e-mail: Zhong-Wang@mse.gatech.edu † Currently at: Department of Chemistry, Arizona State University

Symbols and Definitions

Listed below are the symbols frequently used in this book. All quantities are defined in SI units except that Angstrom (Å) is used occasionally for convenience.

_	
R	Lattice displacement vector
I an	Eattlee displacement vector

 E_c Lattice energy A Madelung constant Cation radius r_M Anion radius r_X

Coordination number p_i electrostatic bond strength S_i Charge of the jth anion

d(MX)Bond length

 Δ_0 Energy gap between t and e orbitals in octahedral coordination Δ_{t} Energy gap between t and e orbitals in tetrahedral coordination

 ψ_i Molecular orbit ϕ_i Atom wave function S_{12} Overlap integral

Valence delocalization coefficient α_v

 ΔG Free enthalpy of reaction ΔH Transition enthalpy ΔS Transition entropy Magnetic susceptibility χ

M Magnetization HMagnetic field

 M_r Remnant magnetization H_c , E_c Coercive force/field P_s T_c Spontaneous polarization

Ferromagnetic transition temperature (or Curie temperature)

Superparamagnetic freezing temperature

Density of states Bohr magneton

xx	w_b	Band gap
AA	U_b	Transfer energy
SYMBOLS AND		Radius of anions
DEFINITIONS	r_X	Radius of cation
	r_M	Electronegativity
	Xx	Chemical potential
	μ	Hardness of atom X
	$\eta_{\mathbf{X}}$ h	Planck's constant
	ħ	$=h/2\pi$
	c	Speed of light in vacuum Rest mass of electron
	m_0	Mass of electron with relativistic correction
	m_e	
	e	Absolute charge of electron
	U_0	Accelerating voltage of electron microscope
	λ	Electron wavelength in free space
	p	Momentum of incident electron
	\mathbf{K}_0, \mathbf{K}	Electron wave-vectors
	θ	Electron scattering semiangle
	f_{α}^{e}	Electron scattering factor of α th atom
	f_{α}^{x}	X-ray scattering factor of ath atom
	FT	Fourier transform from real space to reciprocal space
	FT ⁻¹	Inverse Fourier transform
	r	=(x, y, z) real space vector
	b	=(x, y) real space vector
	g (or h)	Reciprocal lattice vector
	u , τ	Reciprocal space vector
	$V(\mathbf{r})$	Electrostatic potential distribution in crystal
	$V_{\kappa}(\mathbf{r})$	Electrostatic potential of κth atom
	$\rho_{\kappa}(\mathbf{r})$	Electron density distribution of κth atom
	S	Scattering vector, $\mathbf{s} = \mathbf{u}/2$
	S	$=(\sin \theta)/\lambda$
	Z	Atomic number
	V_{σ}	Fourier coefficient of crystal potential
	$V_{g} \ V_{lpha}(\mathbf{g})$	Fourier transform of ath atom in unit cell
	$\exp(-W_{\alpha})$	Debye-Waller factor of αth atom
	Ω	Volume of unit cell
	\mathbf{r}_{α}	$= \mathbf{r}(\alpha)$, position of α th atom within unit cell
	\mathbf{R}_n	Position vector of <i>n</i> th unit cell
	a, b, c	Base vectors of unit cell
	a*, b*, c*	Base vectors of reciprocal lattice vector
	\mathbf{a}_s , \mathbf{b}_s , \mathbf{c}_s	Base vectors of superstructure unit cell
	$\mathbf{a}_s^*, \mathbf{b}_s^*, \mathbf{c}_s^*$	Base vectors of reciprocal lattice vectors for the superstructure
	θ_{α}	Bragg angle
	d	Interplanar distance
	$egin{array}{c} heta_g \ d_g \ arpropto \end{array}$	Convolution calculation
	$T_{ m obj}$	Transfer function of objective lens
	$A_{ m obj}$	Shape function of objective aperture
	C_s	Spherical aberration coefficient of objective lens
	$\mathcal{C}_{\mathcal{S}}$	ophorical abeliation coefficient of objective tens

D. C C. alice disc. lens
Defocus of objective lens
$= eU_0[1 + eU_0/2m_0c^2]$, energy of incident electron
Fourier coefficient of the modified potential U
Velocity of incident electron $(1 + (1 + 1)^2)^{1/2}$
= $(1-(v/c_0)^2)^{1/2}$, relativistic correction factor
= $(2\gamma m_0 e/h^2)V(\mathbf{r})$, modified crystal potential
Electron wave function
Electron wave function excluding $\exp(2\pi i \mathbf{K} \cdot \mathbf{r})$ factor,
$\Phi(\mathbf{r}) = \Psi(\mathbf{r}) \exp(-2\pi_i \mathbf{K} \cdot \mathbf{r})$
ith branch Bloch wave
Wave vector of ith Bloch wave
Superposition coefficients of Bloch waves
Eigenvector of <i>i</i> th Bloch waves
Eigenvalue of ith Bloch waves
Two-beam extinction distance
Thickness of crystal slab
Amplitude of g reflection
Static displacement vector of atoms in imperfect crystal
Burgers vector of dislocations
Direction of dislocation line
Projected crystal potential along z direction
Scherzer defocus
Defocus due to chromatic aberration
Scherzer resolution
Coherence function
$=\pi e \gamma / \lambda E = 1/\hbar v.$
Thickness of crystal slice
Propagation function of slice with thickness Δz
Phase grating function of slice with thickness Δz
Transformation matrix
Vector potential of magnetic field
Magnetic field
$= \mathbf{R}(n)$, position of <i>n</i> th unit cell
Temperature
Volume of crystal
Unit vector along z axis
Position of ath atom in unit cell
X-ray absorption coefficient
Integrated x-ray line intensity
Fluorescence yield
Number of A element per unit volume
Ionization cross section of the inner shell
Fraction of the K, L, or M line intensity measured by the detector
Detector efficiency
Absorption factor
K factor for x-ray microanalysis
Concentration of impurity X
Fraction of impurity X in B atom sites

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SYMBOLS AND DEFINITIONS

xxii	E	Electric field
SYMBOLS AND	D	Displacement vector
DEFINITIONS	ϑ_E	Characteristic angle of inelastic scattering
DEI II VIII IOI VO	Λ	Mean-free path length of inelastic electron scattering
	ε , (ω, \mathbf{q})	Dielectric function of solid
	$d^2P_v/dz\ d\omega$	Differential excitation probability of valence states
	ρ	Free charge density function
	ω_p	Resonance frequency of the plasmon
	σ_I	Integrated ionization cross section
	β	Collection semiangle of the EELS spectrometer
	Δ	Energy width of the integration window

SIGN CONVENTIONS

Free-space plane wave $\exp \left[2\pi i \mathbf{K} \cdot \mathbf{r} - i\omega t\right]$

Fourier transforms

Real space to reciprocal space $F(\mathbf{u}) = \int d\mathbf{r} \exp[-2\pi i \mathbf{u} \cdot \mathbf{r}] f(\mathbf{r}) \equiv \mathbf{F} \mathbf{T}[f(\mathbf{r})]$ Reciprocal space $f(\mathbf{r}) = \int d\mathbf{u} \exp[2\pi i \mathbf{u} \cdot \mathbf{r}] F(\mathbf{u}) \equiv \mathbf{F} \mathbf{T}^{-1}[F(\mathbf{u})]$

where the limits of integration are $(-\infty, \infty)$ unless otherwise specified.

ACRONYMS

ALCHEMI	Atom location by channeling-enhanced microanalysis
ADF	Annular dark field
BF	Bright field
bcc	Body-centered cubic
BZ	Brillouin zone
CBED	Convergent beam electron diffraction
c.n.	Coordination number
CSL	Coincident site lattice
CVD	Chemical vapor deposition
DF	Dark field
DOS	Density of states
DTA	Differential thermal analysis
EDS	Energy dispersive x-ray spectroscopy
EELS	Electron energy loss spectroscopy
ELNES	Energy loss near-edge structure
fcc	Face centered cubic
FWHM	Full width at half-maximum
GB	Grain boundary
HAADF	High-angle annular dark field
hcp	Hexagonal close packing
HOLZ	High-order Laue zone
HOMO	Highest occupied molecular orbital

HRTEM High-resolution transmission electron microscopy **LACBED**

Large-angle convergent beam electron diffraction

LFSE Ligand field stabilization energy Longitudinal magnetic recording **LMR**

Metal organic chemical vapor deposition MOCVD

Molecular beam epitaxy **MBE** NCS Nanocrystal superlattices Partial charge model **PCM**

Perpendicular magnetic recording **PMR**

PMN Pb(Mg,Nb)O₃ **PZT** Pb(Zr,Ti)O₃

Reflection electron microscopy **REM**

Reflection high-energy electron diffraction RHEED

RTRoom temperature SAD Selecting area diffraction

SC Soft chemistry

STEM Scanning transmission electron microscopy

Transmission electron microscopy TEM

TDS Thermal diffuse scattering

WPOA Weak phase object approximation Phase object approximation POA

Zero-order Laue zone **ZOLZ** 1-D One dimensional 2-D Two dimensional 3-D Three dimensional

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SYMBOLS AND **DEFINITIONS**

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