

THE BIOSYNTHESIS OF
STEROIDS, TERPENES,
AND ACETOGENINS

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and

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Preface

In the last decade, acetate as a metabolic entity has assumed a position of central importance in biological processes, particularly as a prime substrate for the biosynthesis of a wide variety of natural substances. Indeed, the symbol adopted for the Fifth International Congress of Biochemistry, held in Moscow in the summer of 1961, was a somewhat stylized representation of an acetic acid molecule. As a result of this progress, acetate is now recognized as the common progenitor of a host of widely divergent structural types found in nature; e.g., terpenes, steroids, fatty acids, complex oxygen heterocycles, phenolics, and even some alkaloids.

In every area of study there comes a time when hypotheses solidify into theory, when various, at one time independent, understandings merge into a clear and satisfying theoretical unity that generates predictions, the basic tenets of which are safe from experimental destruction. This is the propitious moment for a detailed exposition, since such an exposition may confidently be expected to remain substantially correct thereafter, while providing a useful condensation and organization that exposes the remaining areas of theoretical and experimental uncertainty, thus affording a springboard into future efforts. The study of biosynthesis from acetate seems to be at this stage; although many important questions still remain open, the theoretical outline is generally clear and is unlikely to undergo major revision in the future. We were therefore encouraged to undertake this summary of the field up to the end of 1962 and, in most cases, through 1963.

We should like to acknowledge that the original initiative for this work came from Professor Marshall Gates, who invited us to participate in a general volume on the biogenesis of natural substances. For reasons beyond his control (the life of an editor is not an easy one!), Professor Gates was

obliged to dissolve the enterprise. This work represents the crystallization of our manuscripts with considerable expansion, revision, and updating.

In order to identify ourselves as targets for criticism, one of us (J. B. H.) claims primary responsibility for the first part of the book (Chapters 2 through 5) and the other (J. H. R.) is the principal author of the remainder. J. B. H. would like to record his gratitude to Dr. David Dalton for assistance in literature searching, and J. H. R. is grateful to an anonymous reviewer for a great many very helpful criticisms.

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March 1964

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Contents

Preface	v
<i>one</i> Introduction	1
1-1 Acetogenins	1
1-2 Biological Isoprene Unit	4
1-3 Polymerization of Isopentenyl Pyrophosphate	6
1-4 Cyclization	8
1-5 Bile Acids and Steroids	9
<i>two</i> Principles of Biogenetic Theory	10
2-1 Data	11
2-2 Functional Group Pattern	11
2-3 Economy	12
2-4 Enzyme Control	13
2-5 Nature of the Reactions	13
2-6 Nature of the Starting Materials	14
2-7 Limitations	14
<i>three</i> Construction of the Acetate Hypothesis	16
3-1 Expansion of the Hypothesis	19
3-2 Summary	26

four Statistical Survey of Natural Compounds	27
4-1 Linear Carbon Skeletons	28
4-2 Phenylpropanoid Skeletons	32
4-3 Acylphloroglucinol Derivatives (Cyclization Path A)	41
4-4 Orsellinic Acid Derivatives (Cyclization Path B)	62
4-5 More Complex Cyclizations	75
4-6 Miscellaneous Compounds	95
4-7 Correlation and Commentary	104
4-8 Alkaloids	116
 five Acetogenins: Experimental Verification	 121
5-1 Fatty Acids and Other Linear Carbon Skeletons	121
5-2 Aromatic Compounds	140
5-3 Phenylpropanoid and Flavonoid Compounds	155
5-4 Miscellaneous Compounds	164
5-5 Acetate Hypothesis: Final Summary	170
 six The Isoprene Unit	 173
6-1 Acetate \rightarrow Mevalonate	175
6-2 Mevalonate \rightarrow Isopentenyl Pyrophosphate	190
6-3 Polymerization of Δ^3 -Isopentenyl Pyrophosphate	198
 seven Monoterpenes	 207
7-1 Acyclic Monoterpenes	207
7-2 Non-Head-to-Tail Monoterpenes	211
7-3 Monocyclic Monoterpenes	213
7-4 Bicyclic Monoterpenes	217
7-5 Cyclopentanoid Monoterpenes	221
 eight Sesquiterpenes	 225
8-1 Acyclic Sesquiterpenes	225
8-2 Bisabolene	226
8-3 β -Santalene and Cedrol	227
8-4 Longifolene	229
8-5 Helminthosporal	231

8-6	Santonin and Guaiazulene Skeletons	231
8-7	Vetivazulenes	235
8-8	Eremophilones	236
8-9	Sesquiterpenes with Medium-Sized Rings	236
8-10	Polygodial and Iresin	237
8-11	Trichothecin	238

nine Diterpenes

9-1	Linear Diterpenes	240
9-2	Catic Acid, Sclareol, and Manoöl	241
9-3	Pimaranes and Abietanes	243
9-4	Phyllocladene-Type Skeletons	244
9-5	Stereochemistry	245
9-6	Diterpenes with Rearranged Skeletons	247
9-7	Diterpene Alkaloids	249
9-8	Experimental Biosynthetic Results	253

ten Triterpenes

10-1	Principles of Reaction Mechanisms	257
10-2	Olefin Cyclizations	259
10-3	Triterpene Rearrangements	263
10-4	Theoretical Postulates	264
10-5	Ambrein (Chair-Chair-Unfolded-Unfolded-Chair)	266
10-6	Onocerin (Chair-Chair-Unfolded-Chair-Chair)	266
10-7	Tetracyclic Triterpenes (Chair-Chair-Chair-Boat-Unfolded)	267
10-8	Pentacyclic Triterpenes (Chair-Chair-Chair-Boat-Chair)	268
10-9	Lanosterol (Chair-Boat-Chair-Boat-Unfolded)	273
10-10	Hydroxyhopenone (Chair-Chair-Chair-Chair-Chair)	277
10-11	Alternate Possibilities	277
10-12	Modified Triterpenes	278
10-13	Experimental Evidence for Mechanism of Squalene Cyclization	280
10-14	Experimental Studies of Triterpene Biosynthesis	285

eleven Higher Terpenoids

11-1	Carotenoids	289
11-2	Ubiquinones and Plastoquinones	302

twelve Cholesterol Biogenesis	305
12-1 Origin of Carbon Atoms	305
12-2 Conversion of Lanosterol to Cholesterol	311
12-3 Ergosterol	324
 thirteen Further Transformations of Cholesterol	 326
13-1 Reaction Types	326
13-2 Bile Acids	329
13-3 Corticoids	341
13-4 Androgens	351
13-5 Estrogens	357
13-6 Summary	362
13-7 Steroidal Alkaloids	363
 Appendix	 365
 Index	 387

chapter one

Introduction

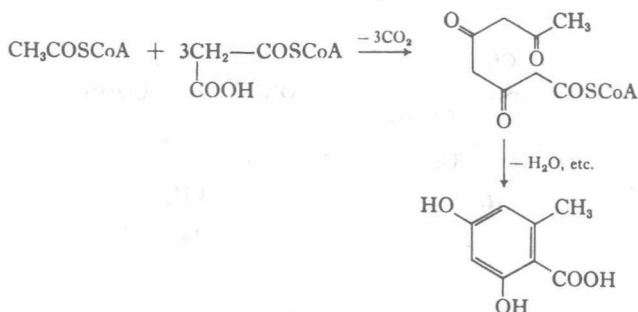
The role of acetate in the biosynthesis of almost all groups of natural products has received extensive theoretical and experimental consideration, especially within the last two decades. As a result the casual reader of this book may easily find himself inundated by the mass of structural correlations that is now apparent and by the large amount of detailed biochemical knowledge that is presently available on many aspects of this question. We propose, therefore, in this brief discussion to give a panoramic view of the role of acetate in the biosynthesis of natural products, and hope that it may serve as a gentle and concise introduction to the subject.

The importance of acetate as an intermediate in a host of biochemical processes has long been known. Acetate results from fatty acid breakdown, is a product of carbohydrate metabolism, and can be produced from certain amino acids. Acetate can also serve as a source of these substances and is thus a molecule that interconnects the three major classes of biological compounds: fats, carbohydrates, and proteins. Acetate can also serve as a source of energy through oxidation to carbon dioxide and water via the familiar Krebs cycle (or tricarboxylic acid cycle).

I-1 ACETOGENINS

The implication of acetate in the biosynthesis of a wide variety of straight-chain and aromatic natural compounds was first deduced on structural grounds; the acetate hypothesis stated that a linear polyketomethylene chain formed from head-to-tail self-condensation of acetate units could cyclize into a remarkable array of complex structures.

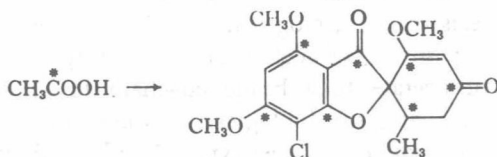
Malonyl coenzyme A is itself formed from acetate by carboxylation of acetyl coenzyme A.



[1]

In the production of other natural products derived from acetate, it is useful to envision the intermediacy of an unreduced polyketo chain that can undergo internal, aldol-type reactions to yield cyclic products. The formation of orsellinic acid [1] points out two of the major features of acetogenins: first, they are derived from the appropriate folding of a chain whose backbone is composed of acetate units linked linearly head to tail; and, second, these substances often carry oxygen at those positions derived from the carboxyl group of the acetate precursor. In this fashion oxygen functions frequently mark the carboxyl carbons of acetate and so serve as an important key to the structural deduction of acetate biogenesis.

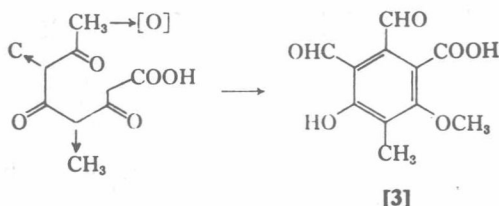
These comments apply to griseofulvin [2], which also demonstrates another aspect of acetogenin biosynthesis: the acquisition of secondary substituents (in this case chlorine). These substituents can be of a wide variety and include halogen, oxygen, and carbon in various forms, e.g., methyl, formyl, carboxyl, or even isoprenoid chains.



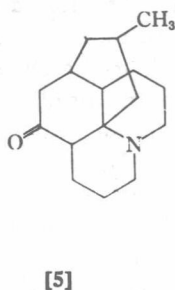
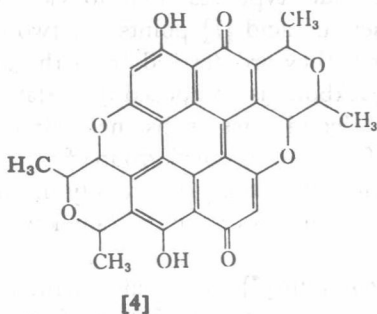
[2]

An additional secondary feature is further oxidation (or reduction) to give products in different over-all oxidation states from those of the

formal polyketo-chain precursor. Cyclopaldic acid [3] demonstrates some of these features. Other modifications and variations of these few basic themes are possible and afford an extraordinarily wide



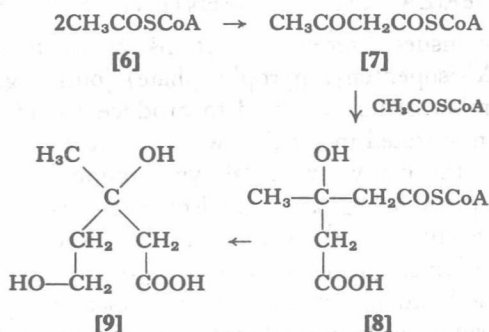
array of substances whose basic carbon skeleton is composed of a linear arrangement of acetate units. For example, such complex and apparently diverse substances as erythroaphin [4] and the alkaloid lycopodine [5] are probably derived directly from acetate.



I-2 BIOLOGICAL ISOPRENE UNIT

For the incorporation of acetate into terpenes and their derivatives, a route is followed that differs at an early stage from that just discussed for the biosynthesis of the acetogenins. Whereas the acetogenins are formed by a linear linking of acetate units, the terpenes are generated by conversion of acetate to a branched-chain intermediate, Δ^3 -isopentenyl pyrophosphate, the biological isoprene unit. The condensation of two moles of acetyl coenzyme A [6] produces acetoacetyl coenzyme A [7], which, upon acquisition of another acetyl residue at the central carbonyl group, forms a branched, six-carbon substance, hydroxymethylglutaryl coenzyme A [8]. A stepwise reduction of the esterified carboxyl carbon of this substance produces mevalonate [9].

The intermediates in the biosynthetic sequence prior to mevalonate are capable of interconversion to many other substances. However, the formation of mevalonate is an essentially irreversible process, and mevalonate once formed has essentially only one biochemical role—the production of isoprenoid substances. Its discovery was, therefore, one of the important breakthroughs in terpene biosynthesis.



The path after mevalonate commences with a series of stepwise phosphorylations ([9] → [10] → [11]) by which the terminal hydroxyl function of the mevalonate is activated as a pyrophosphate ester. The tertiary hydroxyl group is also phosphorylated ([11] → [12]), thus activating the resulting molecule for decarboxylation concerted with loss of phosphate and generation of Δ^3 -isopentenyl pyrophosphate [13].

