GREEN CHEMISTRY AND CHEMICAL ENGINEERING

Fotis Rigas • Paul Amyotte

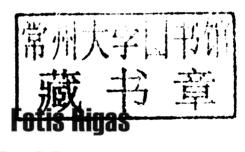
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Hydrogen Safety



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Hydrogen Safety

GREEN CHEMISTRY AND CHEMICAL ENGINEERING

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Series Preface

The subjects and disciplines of chemistry and chemical engineering have encountered a new landmark in the way of thinking about developing and designing chemical products and processes. This revolutionary philosophy, termed green chemistry and chemical engineering, focuses on the designs of products and processes that are conducive to reducing or eliminating the use and/or generation of hazardous substances. In dealing with hazardous or potentially hazardous substances, there may be some overlaps and interrelationships between environmental chemistry and green chemistry. Whereas environmental chemistry is the chemistry of the natural environment and the pollutant chemicals in nature, green chemistry proactively aims to reduce and prevent pollution at its very source. In essence, the philosophies of green chemistry and chemical engineering tend to focus more on industrial application and practice rather than academic principles and phenomenological science. However, as both a chemistry and chemical engineering philosophy, green chemistry and chemical engineering derive from and build on organic chemistry, inorganic chemistry, polymer chemistry, fuel chemistry, biochemistry, analytical chemistry, physical chemistry, environmental chemistry, thermodynamics, chemical reaction engineering, transport phenomena, chemical process design, separation technology, automatic process control, and more. In short, green chemistry and chemical engineering are the rigorous use of chemistry and chemical engineering for pollution prevention and environmental protection.

The Pollution Prevention Act of 1990 in the United States established a national policy to prevent or reduce pollution at its source whenever feasible. And adhering to the spirit of this policy, the Environmental Protection Agency (EPA) launched its Green Chemistry Program in order to promote innovative chemical technologies that reduce or eliminate the use or generation of hazardous substances in the design, manufacture, and use of chemical products. Global efforts in green chemistry and chemical engineering have recently gained a substantial amount of support from the international communities of science, engineering, academia, industry, and government in all phases and aspects.

Some of the successful examples and key technological developments include the use of supercritical carbon dioxide as a green solvent in separation technologies; application of supercritical water oxidation for destruction of harmful substances; process integration with carbon dioxide sequestration steps; solvent-free synthesis of chemicals and polymeric materials; exploitation of biologically degradable materials; use of aqueous hydrogen peroxide for efficient oxidation; development of hydrogen proton exchange membrane (PEM) fuel cells for a variety of power generation needs; advanced biofuel

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productions; devulcanization of spent tire rubber; avoidance of the use of chemicals and processes causing generation of volatile organic compounds (VOCs); replacement of traditional petrochemical processes by microorganism-based bioengineering processes; replacement of chlorofluorocarbons (CFCs) with nonhazardous alternatives; advances in design of energy-efficient processes; use of clean, alternative, and renewable energy sources in manufacturing; and much more. This list, even though it is only a partial compilation, is undoubtedly growing exponentially.

This book series on Green Chemistry and Chemical Engineering by CRC Press/Taylor & Francis is designed to meet the new challenges of the twenty-first century in the chemistry and chemical engineering disciplines by publishing books and monographs based on cutting-edge research and development to the effect of reducing adverse impacts on the environment by chemical enterprise. In achieving this, the series will detail the development of alternative sustainable technologies that will minimize the hazard and maximize the efficiency of any chemical choice. The series aims at delivering the readers in academia and industry with an authoritative information source in the field of green chemistry and chemical engineering. The publisher and its series editor are fully aware of the rapidly evolving nature of the subject and its long-lasting impact on the quality of human life in both the present and future. As such, the team is committed to making this series the most comprehensive and accurate literary source in the field of green chemistry and chemical engineering.

Sunggyu Lee

Preface

This book on the safe handling and use of hydrogen is intended as much to provoke thought and dialogue as to provide practical guidance. We of course hope that readers will find the information we present to be helpful in their endeavors to bring the envisaged hydrogen economy into reality. But we also hope that important areas of safety research, often viewed as being nontechnological, will be welcomed into the discussion of hydrogen safety along with the more familiar technological topics.

The book is organized to first address questions associated with the hazards of hydrogen and the ensuing risk in its use within industry and by the public. What are the properties of hydrogen that can render it a hazardous substance? How have these hazards historically resulted in undesired incidents? How might these hazards arise in the storage of hydrogen and with its use in vehicular transportation?

We then turn to the issues of inherently safer design and, in accordance with the previous comment on so-called nontechnological topics, safety management systems and safety culture. The European Commission (EC) Network of Excellence for Hydrogen Safety, *HySafe*, is singled out for separate coverage, as are various case studies associated with hydrogen and constructional materials. We conclude with a brief look at future research requirements and current legal requirements for hydrogen safety.

Our approach, then, has been to attempt a balanced view of hydrogen safety. Such a perspective comes not only from our respective areas of expertise, but also a collective belief that the safety of any material or activity is most effectively addressed through a combination of the physical sciences and engineering principles together with the management and social sciences. Non-hydrogen-related industries have undergone painful experiences in this regard; there should be no need for industrial applications involving hydrogen to experience the same difficulties.

Coauthor Fotis Rigas would like to acknowledge and thank his family, and especially his wife, Betty, for their (almost) uncomplaining and enduring patience during his long-drawn-out academic activities. He also wishes to express his gratitude to Ms. Allison Shatkin, senior editor in Materials Science and Chemical Engineering of CRC Press/Taylor & Francis, for entrusting him and coauthor Paul Amyotte with the task of writing this book.

Coauthor Paul Amyotte would like to acknowledge and thank his family for their loving support in all his professional undertakings. He is especially grateful to his wife, Peggy. He also extends gratitude to Dalhousie University, and in particular Dr. Joshua Leon, dean of Engineering, for his recent appointment as C. D. Howe Chair in Engineering.

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List of Acronyms

ADNR (inland Regulation for the Carriage of Dangerous Substances on

waterways) the Rhine (EU)

ADR (road) Accord Européen Relatif au Transport International

des Marchandises Dangereuses par Route - European Agreement Concerning the International Carriage of

Dangerous Goods by Road

AIChE American Institute of Chemical Engineers
ALARP As Low As Reasonably Practicable Principle

ATEX Appareils destinés à être utilisés en ATmosphères

EXplosives (EU Directive)

BAMs Bulk Amorphous Materials

BLEVE Boiling Liquid Expanding Vapor Explosion CCPA Canadian Chemical Producers' Association

CCPS Center for Chemical Process Safety
CEI Dow Chemical Exposure Index

CFCs Chlorofluorocarbons

CFD Computational Fluid Dynamics

CL Checklist

CNG Compressed Natural Gas

COD Code of Practice

CPU Central Processing Unit
CSB Chemical Safety Board (USA)

CSChE Canadian Society for Chemical Engineering

CVCE Confined Vapor Cloud Explosion
DDT Deflagration-to-Detonation Transition
DGR IATA Dangerous Goods Regulations

DNR Department of Naval Research

DOE United States Department of Energy
EIGA European Industrial Gases Association

EPR European Pressure Reactor

ETA Event Tree Analysis

FDS Fire Dynamics Simulator Code F&EI Dow Fire and Explosion Index

FMCSA Federal Motor Carrier Safety Administration

FMEA Failure Modes and Effects Analysis

FTA Fault Tree Analysis

FVM Finite Volume Method

GH₂ Compressed Hydrogen Gas

HAZOP Hazard and Operability Analysis

HIAD European Hydrogen Incident and Accident Database

HRAM Hydrogen Risk Assessment Method

HSE Health and Safety Executive

HTHA High-Temperature Hydrogen Attack
IATA International Air Transport Association
IChemE Institution of Chemical Engineers (UK)

IEA International Energy Agency
IGC Industrial Gases Council

IMO (sea) International Maritime Organization

ISD Inherently Safer Design

ISO International Standards Organization

ITER International Thermonuclear Experimental Reactor

LBLOCA Large Break Loss of Coolant Accident

LDL Lower Detonability Limit LFL Low Flammability Limit

LH₂ Liquid Hydrogen

LLNL Lawrence Livermore National Laboratory

LNG Liquefied Natural Gas LPG Liquefied Petroleum Gas

MH Metal Hydride

MHIDAS Major Hazard Incident Database Service

MIACC Major Industrial Accidents Council of Canada)

MIC Methyl Isocyanate
MIL-STD 882 Military Standard 882
MOC Management of Change
MOFs Metal-Organic Frameworks

MVFRI Motor Vehicle Fire Research Institute

NASA National Aeronautics and Space Administration

NBP Normal Boiling Point

NFPA National Fire Protection Association

NHTSA National Highway Traffic Safety Administration NIST National Institute of Standards and Technology

NRA National Railway Authority

NTP Normal Temperature and Pressure

NWC Naval Weapons Center OHA Operating Hazard Analysis

OHSAS Occupational Health and Safety Assessment Series

OSHA U.S. Occupational Safety and Health Administration

PAR Passive Auto-Catalytic Recombiners
PED European Pressure Equipment Directive

PHA Preliminary Hazard Analysis

PRD Pressure Relief Devices
PSM Process Safety Management
QRA Quantitative Risk Assessment

RID (rail) Règlement Concernant le Transport International

Ferroviaire des Marchandises Dangereuses - Regulation

Concerning the Transport of Dangerous Goods by

International Railway

RRR Relative Risk Ranking

SHHSV Stationary High-pressure Hydrogen Storage Vessel

SRB Solid Rocket Boosters

SST Shear Stress Transport Models
STP Standard Temperature and Pressure

SUV Suburban Utility Vehicle SwRI Southwest Research Institute

TNT Trinitrotoluene

TPED Transportable Pressure Equipment Directive

TRD Thermal Relief Device
UFL Upper Flammability Limit
UDL Upper Detonability Limit

UN ECE United Nations' Economic Commission for Europe

UVCE Unconfined Vapor Cloud Explosion

WI What-If Analysis

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