

ENCYCLOPEDIA OF CHEMICAL TECHNOLOGY

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VOLUME 15

WAXES

to

ZYMOSTEROL

INDEX to VOLUMES 1-15

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ABBREVIATIONS AND SYMBOLS

A.	Ångström unit(s)	A.S.M.E.	American Society of
A	anion; as, HA		Mechanical Engineers
abs.	absolute	A.S.T.M.	American Society for
a.c.	alternating current		Testing Materials
ac-	alicyclic; as, <i>ac</i> -derivatives of tetrahydronaphthalene	atm.	atmosphere(s), atmospheric
A.C.S.	American Chemical Society	at. no.	atomic number
addn.	addition	at. wt.	atomic weight
A.G.A.	American Gas Association	av.	average
A.I.Ch.E.	American Institute of Chemical Engineers	b. (as, b ₁₁)	boiling (at 11 mm.)
A.I.M.E.	American Institute of Mining and Metallurgical Engineers	<i>B</i>	base; as, <i>B.2HCl</i>
alc.	alcohol, alcoholic	bbl.	barrel(s)
alk.	alkaline (not alkali)	Bé.	Baumé
Alk	alkyl	b.p.	boiling point
amp.	ampere(s)	B.t.u.	British thermal unit(s)
amp.-hr.	ampere-hour(s)	bu.	bushel(s)
amt.	amount (noun)	C.	centigrade
anhyd.	anhydrous	C-	denoting attachment to carbon; as, <i>C</i> -alkyl derivatives of aniline
A.P.I.	American Petroleum Institute	cal.	calorie(s)
app.	apparatus	calcd.	calculated
approx.	approximate (adj.), approximately	c.f.m.	cubic foot (feet) per minute
aq.	aqueous	cg.	centigram(s)
Ar	aryl	c.g.s.	centimeter-gram-second
ar-	aromatic; as, <i>ar</i> -derivatives of tetrahydronaphthalene	chem.	chemical
as-	asymmetric; as, <i>as-m</i> -xylidine	<i>C.I.</i>	<i>Colour Index</i> no.
ASA	American Standards Association	cks.	centistokes
A.S.M.	American Society for Metals	c.l.	car lots
		cm.	centimeter(s)
		coeff.	coefficient
		com.	commercial
		compd.	compound (noun)
		compn.	composition
		concd.	concentrated
		concn.	concentration
		cond.	conductivity
		const.	constant
		cor.	corrected

c.p.	chemically pure	ff.	following (pages)
cps.	centipoise	fl.oz.	fluid ounce(s)
crit.	critical	f.o.b.	free on board
cryst.	crystalline	f.p.	freezing point
crystd.	crystallized	ft.	foot (feet)
crystn.	crystallization	ft.-lb.	foot-pound(s)
cu.	cubic	g.	gram(s)
d (as, d_4^{20})	density (conveniently, specific gravity)	gal.	gallon(s)
d	differential operator	g.p.d.	grams per denier
d-	<i>dextro</i> -, dextrorotatory	g.p.m.	gallons per minute
d-	denoting configurational relationship, as to <i>dex</i> - <i>tro</i> -glyceraldehyde	hp.	horsepower
		hr.	hour(s)
		hyd.	hydrated, hydrous
		i.	insoluble
		i-	inactive; as, <i>i</i> -methio- nine
d.c.	direct current		
dec., decomp.	decompose(s)	i.b.p.	initial boiling point
decompn.	decomposition	I.C.C.	Interstate Commerce Commission
deriv.	derivative		
detd.	determined	I.D.	inner diameter
detn.	determination	in.	inch(es)
diam.	diameter	insol.	insoluble
dielec.	dielectric (adj.)	I.P.T.	Institute of Petroleum Technologists
dil.	dilute		
dist.	distilled	I.U.	International Unit(s)
distn.	distillation	I.U.C.,	International Union of
DL-, dl-	racemic	I.U.P.A.C.	Chemistry, Interna- tional Union of Pure and Applied Chem- istry
dm.	decimeter		
e	electron	j.	joule
ed.	edition, editor	K.	Kelvin
elec.	electric, electrical	K	dissociation constant
elev.	elevated	Kev	kilo electron volt
e.m.f.	electromotive force	kg.	kilogram(s)
eng.	engineering	kg.-cal.	kilogram-calorie(s)
eq.	equation	kv.	kilovolt(s)
equil.	equilibrium	kv.-amp.	kilovolt-ampere(s)
equiv.	equivalent	kw.	kilowatt(s)
esp.	especially	kw.-hr.	kilowatt-hour(s)
estd.	estimated	l.	liter(s)
estn.	estimation	l-	<i>levo</i> -, levorotatory
e.s.u.	electrostatic unit(s)	L-	denoting configurational relationship, as to <i>levo</i> -glyceraldehyde
e.u.	entropy unit(s)		
e.v.	electron volt(s)	lb.	pound(s)
expt.	experiment	LC ₅₀	concentration lethal to 50% of animals tested
exptl.	experimental		
ext.	extract		
extd.	extracted		
extn.	extraction		
F.	Fahrenheit		
Fedl.	Federal		

l.c.l.	less than car lots	N.O.I.B.N.	not otherwise indexed
LD ₅₀	dose lethal to 50% of animals tested		by name
ln	logarithm (natural)	<i>o</i> -	ortho; as, <i>o</i> -xylene
log	logarithm (common)	<i>O</i> -	denoting attachment to oxygen; as, <i>O</i> -acetyl- hydroxylamine
m.	meter(s)		
<i>m</i> -	meta; as, <i>m</i> -xylene	O.D.	outer diameter
M	metal	oz.	ounce(s)
<i>M</i>	molar (as applied to concn.; not molal, which is written out)	p., pp.	page, pages
		<i>p</i> -	para; as, <i>p</i> -xylene
ma.	milliampere(s)	pos.	positive (adj.)
manuf.	manufacture	powd.	powdered
manufd.	manufactured	p.p.m.	parts per million
manufg.	manufacturing	ppt.	precipitate
max.	maximum	pptd.	precipitated
M.C.A.	Manufacturing Chem- ists' Association	pptn.	precipitation
m.c.f.	million cubic feet	prepd.	prepared
m.e., meq.	milliequivalent(s)	prepn.	preparation
mech.	mechanical	Pr. no.	Foreign Prototype no. (for dyes)
M.e.v.	million electron volts	p.s.i.(g.), (a.)	pound(s) per square inch (gage), (absolute)
mg.	milligram(s)		
m.g.d.	million gallons per day	pt.	point
min.	minimum; minute(s)	pts.	parts
misc.	miscellaneous	quad. pt.	quadruple point
mixt.	mixture	qual.	qualitative
ml.	milliliter(s)	quant.	quantitative
M.L.D.	minimum lethal dose	<i>q.v.</i>	"which see"
mm.	millimeter(s)	R	univalent hydrocarbon radical (or hydrogen)
mM	millimole(s)		
mol.	molecule, molecular	R.	Rankine
m.p.	melting point	ref.	reference
m.p.h.	miles per hour	resp.	respectively
M.R.	molar refraction	r.h.	relative humidity
mv.	millivolt(s)	<i>R.I.</i>	<i>Ring Index</i> no.
mμ	millimicron(s)	r.p.m.	revolutions per minute
<i>n</i> (as, <i>n</i> _D ²⁰)	index of refraction (for 20°C., and sodium light)	r.p.s.	revolutions per second
		s.	soluble
<i>n</i> -	normal; as, <i>n</i> -butyl	<i>s</i> -	symmetric(al); as, <i>s</i> - <i>m</i> - xylydine
<i>N</i>	normal (as applied to concn.)	<i>S</i> -	denoting attachment to sulfur; as, <i>S</i> -methyl- cysteine
<i>N</i> -	denoting attachment to nitrogen; as, <i>N</i> -meth- ylaniline	S.A.E.	Society of Automotive Engineers
neg.	negative (adj.)	satd.	saturated
n.o.	number	satn.	saturation

ABBREVIATIONS AND SYMBOLS

S.C.F.	standard cubic foot (feet)	t.s.i.	tons per square inch
Sch.	Schultz no. (for dyes)	Twad.	Twaddell
sec.	second(s)	u.v.	ultraviolet
sec-	secondary; as, <i>sec</i> -butyl	v.	volt(s)
S.F.s.	Saybolt Furol second(s)	var.	variety
sl.s.	slightly soluble	vic-	vicinal; as, <i>vic-m</i> -xyli- dine
sol.	soluble	vol.	volume(s) (not volatile)
soln.	solution	v.s.	very soluble
soly.	solubility	w.	watt(s)
sp.	specific	wt.	weight
sp., spp.	species	X.U. (10^{-10} mm.)	X-unit
spec.	specification	yd.	yard(s)
sp.gr.	specific gravity	yr.	year(s)
sq.	square	$[\alpha]_D^{20}$	optical rotation (for 20°C. and sodium light)
S.T.P.	standard temperature and pressure	γ	microgram(s)
subl.	sublime(s), subliming	∂	differential operator (partial)
S.U.s.	Saybolt Universal second(s)	Δ	finite difference
<i>sym</i> -	symmetric(al); as, <i>sym</i> - <i>m</i> -xylidine	η	viscosity
T.A.P.P.I.	Technical Association of the Pulp and Paper Industry	λ	wave length
tech.	technical	μ	micron(s)
temp.	temperature	Ω	ohm(s)
<i>tert</i> -	tertiary; as, <i>tert</i> -butyl	<	less than
theoret.	theoretical	>	more than
t.p.h.	tons per hour	$\sim$	cycle(s)
		$\approx$	approximately equal to

Other letter symbols may be found in "Standard System of Nomenclature for Chemical Engineering Unit Operations" adopted by the American Institute of Chemical Engineers.

SHIPPING REGULATIONS

Complete information for the U.S. is given in "Tariff No. 9 Publishing Interstate Commerce Commission Regulations for Transportation of Explosives and Other Dangerous Articles by Land and Water in Rail Freight Service and by Motor Vehicle (Highway) and Water Including Specifications for Shipping Containers," with supplements, issued by H. A. Campbell, Agent, 30 Vesey Street, New York 7, N.Y. (1954). The following terms for labeling explosives and other dangerous articles have been used in the Encyclopedia:

- Red label (for inflammable liquids)
- Yellow label (for inflammable solids and oxidizing materials)
- White label (for acids and corrosive liquids)
- Red label (for inflammable compressed gases)
- Green label (for noninflammable compressed gases)
- N.O.I.B.N. (not otherwise indexed by name)

In the text of the Encyclopedia the preferred terms "flammable" and "nonflammable" are used in place of "inflammable" and "noninflammable," respectively.

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PERIODICAL ABBREVIATIONS

The abbreviations used are, for the most part, those given in the "List of Periodicals Abstracted by Chemical Abstracts" (Vol. 45, No. 24, Pt. 2 (1951), also published separately). See also *Literature (survey)*, especially the sections on "Reviews, yearbooks, and monographs" and "Periodicals," Vol. 8, pp. 437-40.

- Am. Soc. Testing Materials, Proc.*
Anal. Chem. (superseding *Ind. Eng. Chem., Anal. Ed.*)
Angew. Chem. (superseding *Die Chemie; Z. angew. Chem.*)
Ann. Chem., Justus Liebig's
Arch. Biochem. and Biophys. (superseding *Arch. Biochem.*)
Arch. Ind. Hyg. and Occupational Med. (superseding *J. Ind. Hyg. Toxicol.*)
Biochem. J. (London)
Biochem. Z.
Biochim. et Biophys. Acta
BIOS Repts.
Bull. Chem. Soc. Japan
Bull. soc. chim. or Bull. soc. chim. France C.A.
Can. J. Research
Chem. Ber. (superseding *Ber.*)
Chem. Eng. (superseding *Chem. & Met. Eng.*)
Chem. Eng. News (superseding *News Ed. (Am. Chem. Soc.); Ind. Eng. Chem., News Ed.*)
Chem. Eng. Progress (superseding *Trans. Am. Inst. Chem. Engrs.*)
Chem. Eng. Science
Chemische Industrie
Chemistry & Industry (formerly part of *J. Soc. Chem. Ind.*)
Chem. Revs.
Chem. Tech. (Berlin) (superseding *Chem. Fabrik*)
Chem. Week (superseding *Chem. Inds. Week*)
Chem. Zentr.
Chem.-Ztg.
- Chimica e industria (Italy) or Chimica e industria (Milan)*
Chimie & industrie
CIOS Repts.
- Compt. rend.*
- FIAT Repts.*
Fortschr. chem. Forsch.
Gazz. chim. ital.
Helv. Chim. Acta
Ind. Chemist
- American Society for Testing Materials, Proceedings
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 Gazzetta chimica italiana
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 Industrial Chemist and Chemical Manufacturer, The

Ind. Eng. Chem. (superseding *J. Ind. Eng. Chem.*)

J. Agr. Food Chem.

J. Am. Chem. Soc.

J. Am. Med. Assoc.

J. Am. Pharm. Assoc.

J. Appl. Chem. (U.S.S.R.) (see also *Zhur. Priklad. Khim.*)

J. Appl. Phys. (superseding *Physics*)

J. Assoc. Offic. Agr. Chemists

J. Biol. Chem.

J. Chem. Phys.

J. Chem. Soc.

J. Colloid Sci.

J. Electrochem. Soc. (superseding *Trans. Electrochem. Soc.*; *Trans. Am. Electrochem. Soc.*)

J. Gen. Chem. (U.S.S.R.) (see also *Zhur. Obshchei Khim.*)

J. Indian Chem. Soc.

J. Inst. Metals

J. makromol. Chem. (superseding *J. prakt. Chem.*)

J. Org. Chem.

J. Phys. Chem. (superseding *J. Phys. & Colloid Chem.*)

J. Polymer Sci. (superseding *J. Polymer Research*)

J. Research Natl. Bur. Standards (superseding *Bur. Standards J. Research*)

J. Sci. Food Agr.

J. Soc. Chem. Ind. or J. Soc. Chem. Ind. (London) (formerly containing *Chemistry & Industry*)

J. Soc. Chem. Ind., Japan
Kolloid-Z.

Mfg. Chemist

Monatsh. Chem.

Nature

Nucleonics

Office Tech. Services (OTS) Repts. (superseding *Office Publication Board Repts.*)

Oil, Paint Drug Repr.

Phys. Rev.

Rec. trav. chim.

Research (London)

Revs. Mod. Phys.

Science

Trans. Am. Inst. Mining Met. Engrs.

Trans. Am. Soc. Metals (superseding

Trans. Am. Soc. Steel Treating)

Trans. Inst. Chem. Engrs. (London)

Z. anorg. u. allgem. Chem. (superseding *Z. anorg. Chem.*)

Z. Elektrochem.

Zhur. Obshchei Khim.

Zhur. Priklad. Khim.

Z. physik. Chem.

Industrial and Engineering Chemistry

Journal of Agricultural and Food Chemistry
Journal of the American Chemical Society, The
Journal of the American Medical Association, The
Journal of the American Pharmaceutical Association
Journal of Applied Chemistry (U.S.S.R.)

Journal of Applied Physics

Journal of the Association of Official Agricultural Chemists

Journal of Biological Chemistry, The

Journal of Chemical Physics, The

Journal of the Chemical Society (London)

Journal of Colloid Science

Journal of the Electrochemical Society

Journal of General Chemistry (U.S.S.R.)

Journal of the Indian Chemical Society

Journal of the Institute of Metals and Metallurgical Abstracts

Journal für makromolekulare Chemie

Journal of Organic Chemistry, The

Journal of Physical Chemistry, The

Journal of Polymer Science

Journal of Research of the National Bureau of Standards

Journal of the Science of Food and Agriculture

Journal of the Society of Chemical Industry (London)

Journal of the Society of Chemical Industry, Japan
Kolloid-Zeitschrift

Manufacturing Chemist and Pharmaceutical and Fine Chemical Trade Journal Incorporating Manufacturing Perfumer

Monatshefte für Chemie und verwandte Teile anderer Wissenschaften

Nature

Nucleonics

Office of Technical Services Reports

Oil, Paint and Drug Reporter

Physical Review, The

Recueil des travaux chimiques des Pays-Bas

Research, A Journal of Science and Its Applications

Reviews of Modern Physics

Science

Transactions of the American Institute of Mining and Metallurgical Engineers

Transactions of the American Society for Metals

Transactions of the Institution of Chemical Engineers (London)

Zeitschrift für anorganische und allgemeine Chemie

Zeitschrift für Elektrochemie und angewandte physikalische Chemie

Zhurnal Obshchei Khimii (Journal of General Chemistry (U.S.S.R.))

Zhurnal Prikladnoi Khimii (Journal of Applied Chemistry (U.S.S.R.))

Zeitschrift für physikalische Chemie

W *continued*

WAXES

The term wax is now generally applied to all waxlike materials found naturally, as well as synthetic materials of a waxy nature, since their uses are usually based on their physical characteristics. These characteristics include: resistance to water and water vapor, tensile strength, ductility, gloss, hardness, ability to emulsify, solvent retention, moldability, and melting range. Originally, wax denoted beeswax, which is probably the oldest wax known, and later only products with characteristics similar to beeswax. Today, in addition to the broadening of the classification on the basis of physical properties, a chemical basis is sometimes recognized. Thus waxes can be distinguished from fats (*q.v.*) not only by their greater hardness and brittleness and lesser greasiness, but also by their composition. Lipide waxes contain esters, acids, and alcohols of higher molecular weight than those of fats (which are glycerides), and in addition higher saturated hydrocarbons. Wool grease (*q.v.*) is classed chemically as a wax, and sperm oil (see p. 9) as a *liquid wax*, whereas Japan wax (see p. 8) and bayberry wax are fats.

The major classes of waxes are vegetable, animal (and insect), petroleum (see Vol. 10, p. 211), and the naturally occurring mineral waxes, and synthetic waxes. The origins of the more important of the naturally occurring waxes are shown in Table I. Wax products are used for a wide variety of applications in many industries; the largest uses, however, are in paper coating, candles, polishes, electrical insulation and coating, carbon paper, textiles, and leather. (Modern sealing wax is not a wax, and contains no wax. It consists chiefly of shellac and Venice turpentine, an oleoresin. Medieval sealing wax contained beeswax.)

PROPERTIES

As with any naturally occurring product, the physical and chemical properties of a natural wax vary over rather wide limits, and are further accentuated by differences in grades and processing (5). Table II lists the principal physical and chemical properties of the commercial natural waxes (13).

Composition. The main constituents of vegetable and animal waxes are high-molecular-weight acids, alcohols, esters, and saturated hydrocarbons, the esters usually

TABLE I. Commercial Natural Waxes.

Wax	Source	Locality of origin	Original appearance	Commercial grades
<u>Animal waxes</u>				
Beeswax	<i>Apis mellifica</i> , Ghedda, <i>Melipona</i>	World-wide	Light yellow to dark greenish brown	Crude; yellow re- fined; bleached white
Chinese (insect wax)	<i>Coccus ceriferus</i>	Western China	White, opaque	Crude, refined
Shellac	<i>Carteria lacca</i>	India	Yellow	Crude
Spermaceti	<i>Physeter</i> <i>Macrocephalus</i>	—	White, crystalline	Technical; U.S.P.
<u>Mineral waxes</u>				
Ceresin	Purified ozocerite from lignite	—	Yellow, white	Sold according to melting point range
Montan	Lignite	Germany (Saxony, Thuringia), California	Dark brown, brownish black	Crude; refined; distilled
Ozocerite	Lignite	Chiefly Galicia, Tcheleken Islands; also Utah, Austria, Egypt, Serbia	Yellow, dark brown, dark amber	Crude; natural yellow; white; purified grade called ceresin
<u>Vegetable waxes</u>				
Candelilla	<i>Pedilanthus pavonis</i>	Mexico and South- west Texas	Dark brown to greenish	Technical; refined; bleached
Carnauba	<i>Copernicia cerifera</i>	Northeastern Brazil	Sulfur yellow to dark brown- black	No. 1 (No. 1 Yel- low) No. 2 (No. 2 Yel- low) No. 3 (Light Fatty Grey or Cauhype) No. 4 (No. 4 North Country or Fatty Grey) No. 5 (Chalky) Crude; refined; bleached Refined
Douglas fir bark	<i>Pseudotsuga taxifolia</i>	Oregon, Washing- ton	Yellow-brown	
Japan	<i>Rhus surcedanea</i> , <i>Rhus vernicifera</i> , <i>Rhus sylvestris</i>	Western Japan, China	Greenish, pale yel- low or light brown	Kitagumi, Chichusan
Ouricury	<i>Syageus cororata</i>	Brazil	Dark brown, black	Crude; refined; bleached
Palm	<i>Ceroxylon andicolum</i>	Columbian Andes	White	Lump; powder; crude
Sugar cane	<i>Saccharum officinarum</i>	Cuba, Puerto Rico, Louisiana	Dark brown	Crude, refined (various grades)

TABLE II. Chemical and Physical Properties of Commercial Natural Waxes.

Wax	Melting range, °C.	Saponification number	Acid number	Iodine number	Sp. gr.	Acetyl number	Main constituents
Beeswax	62-70	86-96	17-21	8-11	0.955-0.975	15	Myricyl palmitate, cerotic acid, hydrocarbons
Candelilla	65-69	46-65	15-16	14-37	0.969-0.993	—	Hydrocarbons, melissic acid, myricyl alcohol, resins
Carnauba	83-91	73-86	1-8	8-13	0.990-0.999	55	See Table IV
Ceresin	64-77	0	0	0	0.88-0.92	—	Aliphatic hydrocarbons
Chinese	65-80	78-93	0.2-13	1.4-2	0.926-0.970	—	Ceryl cerotate
Japan	50-56	207-237	20	4-15	0.975-0.990	27-31	Glycerides of mono- and dibasic acids
Montan	Crude 76-92 Distd. 72-77 Refd. 77-84 79-84	Crude 58 Distd. 75-89 Refd. 70-80 62-85	Crude 25 Distd. 73-85 Refd. 15-20 3-24	Crude 16 Distd. 10-15 7-15	Crude 1.0 0.99-1.06	—	Esters of wax acids, free wax acids, free wax alcohols, resins
Ouricury						—	Myricyl cerotate, hentriacontane, free wax acids, resins
Ozocerite	58-100	0	0	0	0.85-0.95	0	Aliphatic hydrocarbons
Shellac	74-78	100-126	12.5-16.0	1.2	0.97-0.98	—	Esters of wax acids, free wax alcohols
Spermaceti	41-49	121-135	0.5	2.6-3.8	0.905-0.960	2.6	Cetyl palmitate
Sugar cane (refined wax)	76-79	65-77	23-28	5-10	0.997	30-40	Alkyl and steryl esters of fatty acids
Wool wax	31-42	82-140	0.2-40	15-47	0.924-0.960	—	Esters of cholesterol and lanosterol

being predominant. Table III lists some of the common wax acids and alcohols (13); these are all straight-chain compounds (see also *Fatty acids*). Some of the other components of natural waxes are resins, lactones, and sterols. As an example, the approxi-

TABLE III. Some Common Wax Acids and Alcohols.

Name	Formula	M.p., °C.	Occurrence
<u>Acids</u>			
Cerotic acid	$C_{25}H_{51}COOH$	87.7	Free in beeswax, montan, carnauba; as ester in Chinese wax and carnauba
Lauric acid	$C_{11}H_{23}COOH$	44.2	As laurin in Japan wax
Montanic acid	$C_{27}H_{55}COOH$	90	Free in montan wax
Melissic acid	$C_{29}H_{59}COOH$	93.6	Free in beeswax and montan
Palmitic acid	$C_{15}H_{31}COOH$	63.1	As palmatin in Japan wax; cetyl palmitate in spermaceti; myricyl palmitate in beeswax
<u>Alcohols</u>			
Cetyl alcohol (see Vol. 1, p. 320)	$C_{16}H_{33}OH$	49	As cetyl palmitate in spermaceti
Ceryl alcohol	$C_{26}H_{53}OH$	79	As ceryl cerotate in Chinese wax
n-Octadecyl alcohol (stearyl alcohol see Vol. 6, p. 255)	$C_{18}H_{37}OH$	59	In spermaceti
Montanyl alcohol	$C_{28}H_{57}OH$	—	In montan wax
Myricyl alcohol	$C_{30}H_{61}OH$	85	As esters in beeswax, carnauba

mate composition of carnauba wax is given in Table IV (the composition is that of Warth (13) modified to conform to Findley and Brown's functional group analysis and fractionation by molecular distillation of unhydrolyzed carnauba wax (2a)).

TESTING

Testing of waxes consists of the determination of physical and chemical properties by well-recognized test procedures, the specific details of which have not been standardized in many cases for the natural waxes and vary throughout the industry. The tests include: Physical constants: melting point, refractive index, specific gravity, and hardness. Chemical constants: saponification number, acid number, unsaponifiables, hydrocarbon content, iodine number, and acetyl number (1,4,13). (See also Vol. 6, pp. 151-56.)

Other test procedures consist of actual formulation and performance tests, such as emulsifying a wax in a no-rub floor polish formula, and evaluating the product.

METHODS OF PROCESSING

The most important physical operations used in the processing of natural wax are: separation of wax from fiber by heating, shredding, or maceration; selective and non-selective solvent extraction; filtration with and without filter aid; destructive and vacuum distillation; and bleaching by adsorption. Chemical processing (including that used for the manufacture of synthetic waxes) consists of: chemical bleaching; esterification; chlorination; oxidation; condensation; and hydrogenation.