# ESSENTIALS OF BIOTECHNOLOGY

Robert P. Ouellette

Paul N. Cheremisinoff

# ESSENTIALS OF BIOTECHNOLOGY

# Robert P. Ouellette

VICE PRESIDENT, CORPORATE DEVELOPMENT, VERSAR, INC., SPRINGFIELD, VA

# Paul N. Cheremisinoff

PROFESSOR OF ENVIRONMENTAL ENGINEERING
NEW JERSEY INSTITUTE OF TECHNOLOGY, NEWARK, NJ



Published in the Western Hemisphere by Technomic Publishing Company, Inc. 351 New Holland Avenue Box 3535 Lancaster, Pennsylvania 17604 U.S.A.

Distributed in the Rest of the World by Technomic Publishing AG

©1985 by Technomic Publishing Company, Inc. All rights reserved

No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, recording, or otherwise, without the prior written permission of the publisher.

Printed in the United States of America
10 9 8 7 6 5 4 3 2 1

Main entry under title: Essentials of Biotechnology

A Technomic Publishing Company book Bibliography: p. Includes index p. 221

Library of Congress Card No. 85-51484 ISBN No. 87762-437-2

# **PREFACE**

The outstanding development of biotechnology during the past decade has been a result of science's ability to begin to decipher the codes of life, reprogram the cells and tune molecular switches adopting hosts capable of growing in large quantities. Biotechnology provides the environment on which new bodies of knowledge can expand and act as catalyst for many disciplines.

New technology, equipment and processes will be born in medicine, biochemistry, food, fuels environmental control as a result of the biological revolution of the last generation. Many of these technologies are complementary and this book views the essentials of biotechnology from a technological viewpoint. Presented is the interaction of several disciplines since all biotechnology is multidisciplinary in nature, challenging the technologist and layman alike.

The reader will find this book useful and informative. Chapters include extensive bibliographies on subject matter presented for further reference and research which will serve as a guide to the more recent literature on biotechnology. Overall this volume should prove useful to both the practitioner as well as the student.

ROBERT P. OUELLETTE PAUL N. CHEREMISINOFF

# **CONTENTS**

# Preface ix

1/What is Biotechnology? 1

Introduction 1

The Logic of Life 1

• The Central Dogma • The Building Blocks • The DNA Code • The Process

Genetic Engineering 10

• Terminology • The Tools • Recombinant DNA

Monumental and Momental Issues 19

• Oncogenes • Exon-Intron Organization • Transposons • Preproproteins

Conclusion 26

References 27

# 2/Drug Delivery (A Systems Approach) 31

Introduction 31

The Systems 31

The Receptor 32

Chemical Species 34

Pertal or Mode of Entry 3

Reservoir 35

Force or Energy Source 36

Rate Controlling Mechanism 36

Control Loop 39

Conclusion 39

References 40

### 3/Brain Peptides (The Molecules of Hope) 43

Introduction 43

Peptide Hormones 43

Possible Uses and Misuses 46

• Somatostatin • Growth Factors • Gonadotropin Releasing Hormone •

Thymosins • Endogeneous Opioids • Bombesin • Angiotensin 11 •

Vasopressin • Substance P • Cholecystokinin (CCK) • The Pro-

opiomelanocortin(POMC) Complex

Non-Mammalian Peptides 51

Conclusion 51

References 52

### 4/Chemicals From Plants 57

Introduction 57

Plant Extracts 57

• Drugs and Chemicals • Hydrocarbons • Proteins

Tissue Culture 60

• The Technology • The Market

Conclusion 62

References 62

### 5/Water 65

Introduction 65

The Structure of Water 65°

The Properties of Water 66

Water as a Solvent 68

Water in Chemical Reactions 68

Interaction of Proteins with Water 69

Interaction of Water with DNA 70

Conclusions 71

References 71

# 6/Modified DNA and Proteins 75

Introduction 75

Glycoproteins 75

The Observations 78

The Hypotheses 79

• Cellular Differentiation • Protection Against Restriction Enzymes •

Replication Blocking • DNA Repair • Gene Control

Phospholipids Methylation 80

Protein Methylation 81

Phosphorylation of Protein 82

• Growth • Hormonal and Neural Growth • Malignancy • Cholesterol • Myosin

Conclusion 84 References 84

#### 93 7/Artificial and Synthetic Proteins

Introduction 93

Enzyme Activity 94

Enzyme Specificity 95

Protein Modifications 96

- Amino Acid Substitution Chain Length Modification Design of an Analogue Protein • Metals Substitution • Genes and Proteins Synthesis Synthetic Enzymes 99
- Enzyme Analogs Polymer Containing Metals [37] Polymer Bond Photosensitizers [37] • Cyclodextrins • Synthetic Greening • Host-Guest Chemistry • Modified Enzymes

Conclusion 106 References 106

# 8/Enzymes 113

Introduction 113

Enzymes Classification 113

Enzyme Activity 119

• Enzyme Kinetics • Effect of Temperature and pH

Enzyme Immobilization 125

The Market 127

Enzyme Uses 127

• Detergents • Monitoring and Diagnostic • Enzyme Reagents and Indicators • DNA Probes • Enzyme Electrodes • Enzyme Thermistors • Enzyme Immunoassays • Monoclonal Antibodies • Food Industry • Chemicals • Pharmaceutical • Medical Applications • Fibrinolytic Agents • Anti-inflammatory Agents • Collagen • Other Applications • Future Prospects

Enzymes in Research 151

Survey of Enzymes Used in Industry 152

Perspective 152

References 163

### 9/Enzymes in Hostile Environments 167

Introduction 167

Water and Salinity Stress 167

Temperature 168

• Properties at High Temperatures • Properties at Low Temperatures Light 172

Xenochemicals 174

pH Effect 175

### viii Contents

Oxygen Toxicity 176 Trace Metals 177 Pressure 179 Conclusion 180 References 181

# 10/Transducers in Biotechnology 185

Introduction 185

Measurement System 185

- Accuracy Sensitivity Specificity Output Information Independence from Interferences Reproducibility Stability Response Time
- Sterilization Lack of Toxicity/Contamination Environment Size The State of the Art 189
- Glass Electrodes Membrane Electrodes Enzyme Electrodes Interpretation 195

Conclusions 196

• Miniaturization • Electrical Output • On-Board Computing • Multiple Parameter Measurements • Dynamic Measurements • Optical Sensing References 198

# 11/Chemical Field Effect Transistors (ChemFETs) 201 Introduction 201

ChemFets 201

- Semiconductors Enzymes Membranes Un Mariage a Trois Applications 212
- Biomedical Applications Industrial Applications Environmental Applications Defense Applications

New Directions 216

References 217

Index 221

# What is Biotechnology?

### INTRODUCTION

Diotechnology is exploding like a Nova, bursting on the everyday scene, challenging the technologist and the layman alike. The field is ancient and new. It is as old as thinking man; and as new as the latest discovery.

It is not possible to review, with scientific rigor, all the findings, advances and discoveries of recent years. We will limit ourselves to a broad sketch on a canvas. We hope that major lines will emerge, guiding our review of what biotechnology has achieved and what it promises to accomplish.

There are no limits to scientific inquiry, still nagging ethical questions persist. We will stay far from such crags and concentrate on the easier topic of science and technology. We display in the following pages the logic of life, the confusing terminology in use today, the prerequisite tools for biotechnological achievements, and some of the emerging concepts.

### THE LOGIC OF LIFE

All life is reducible, at the molecular level, to the sequences in DNA (deoxyribonucleic acid). A precise mathematical logic insures that the DNA message is translated into RNA (ribonucleic acid) and that this last code is eventually transcribed into proteins. Proteins serve many functions but none is more important than their role as enzymes catalyzing the entire chemistry of life (Figure 1-1).

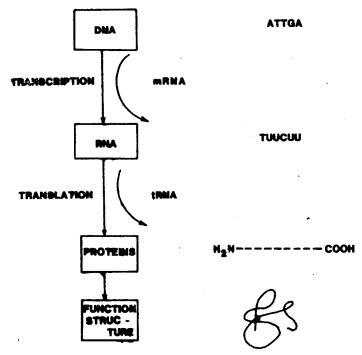


Figure 1-1. The logic of life.

# The Central Dogma

A central dogma on the sense and direction of the cellular machinery from DNA to RNA to protein has been proposed. Like all dogma, it sins in its absolute character.

The central dogma of molecular biology states that information transfer never proceeds from protein to nucleic acid. This part of the dogma has never been refuted. The DNA to RNA pathway part of the original formulation had to be revised with the observation that retroviruses transcribe RNA to DNA. The common use of reverse transcriptase in genetic engineering experiments is ample confirmation of the reality of the reverse pathway.

A more modern version of the dogma would be as follows:



Indicating that RNA to DNA information transfers while rare, can occur.

# The Building Blocks

Today's engineers might not assemble DNA the way nature did it, but given the basic building blocks, the logic of the assembly is easily recognized. The basic blocks are three: the bases, the sugar moitie, and the phosphate groups.

### The Bases

The bases are of five kinds: 2 purines and 3 pyrimidines (Figure 1-2). The two purines, adenine and guanine, are used both in RNA and DNA; while the pyrimidine thymine is found only in DNA, and uracil appears in RNA. These bases are linked together to form a polynucleotide. The phosphate group of one nucleotide is linked to the deoxyribose group of another, and so forth.

PURINES

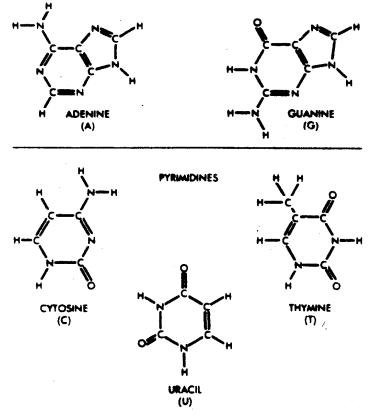


Figure 1-2. The five bases.

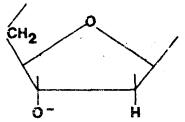


Figure 1-3. The deoxyribose.

### The Sugar

The sugar backbone is a deoxyribose (Figure 1-3).

### The Phosphodiester Bond

The phosphate group serves as a linkage (Figure 1-4). The linking is from the 5' carbon of a deoxyribose molecule to the 3' carbon of a nucleotide (Figure 1-5).

### The Double Helix

The double helix of DNA is held together by hydrogen bonds between purines and pyrimidines according to an exact pairing system where A pairs with T and G with C (Figure 1-6).

The double helix structure of DNA was demonstrated in 1953 by Watson and Crick, building on the work of earlier crystallographers [1].

The double helix can now be seen as made up of sugar-phosphate backbones linking the bases. The two strands of the helix are connected by hydrogen bonds (Figure 1-7). The two strands of the DNA molecule have complementary shapes. The strands are twisted in a right-handed double helix. The DNA helix is supercoiled in a compact form. Supercoiling is controlled by enzymes: topoisomerases and gyrases.

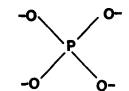


Figure 1-4. The phosphate group.

Figure 1-5. Phosphodiester bonds.

. Figure 1-6. Base pairing with hydrogen bonds.

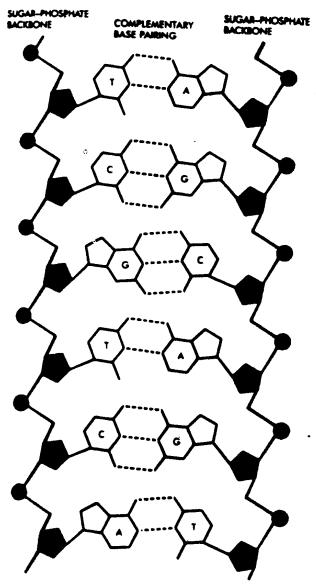


Figure 1-7. The DNA chain.

#1.

# The DNA Code

We now know how the basic information is stored, transmitted from generation to generation, modified by external agents and noise, and decoded. We know how cells use this information to build structures and condition the functioning of cells, tissues, organs, and organisms.

The basic message is made up of a sequence of four bases: Adenine

,,,

(A), Thymine (T) or Uracil (U), Guanine (G), Cytocine (C). The bases are complementary, insuring geometry into the message.

The genetic code is a language based on four nucleic acid bases, through which the 64 triplets specify the 20 amino acids and operators. The DNA code is a coma free, generally non-overlapping degenerate triplet code (Table 1-1).

### The Process

The conversion of the DNA code into a protein molecule is a two-step process, involving numerous enzymes.

# **DNA** Duplication

DNA can direct its own replication. The first step in DNA replication is the separation of the DNA double helix. Each strand then serves as a

TABLE 1-1. The DNA Code.

First Position (5' end)		Third Position (3' end)			
	U	С	A	·G	
U	PHE	SER	TYR	CYS	U
	PHE	SER	TYR	CYS	C
	LEU	SER	Stop	Stop	Α
	LEU	SER	Stop	TRP	G
c ,	LEU	PRO	HIS	ARG	U
	LEU	PRO	HIS	-ARG	С
	LEU	PRO	GLN	ARG	Α
	LEU	PRO	GLN	ARG	G
A	ILE	THR	ASN	SER	U
	ILE	THR	ASN	SER	C
	ILE	THR	LYS	ARG	Α
	MET	THR	LYS	ARG	G
G	VAL	ALA	ASP	GLY	U
	VAL	ALA	ASP	GLY	С
	VAL	ALA	GLU	GLY	A
	VAL	ALA	GLU	GLY	G

template for the production of two daughters DNA molecules. The separation-replication process goes hand-in-hand.

# Transcription

The conversion of DNA to RNA is accomplished by an RNA polymerase. This enzyme binds to the DNA at a correct starting point, then moves rapidly down the DNA chain, adding the correct nucleotides to the RNA chain until the task is completed.

It takes several steps to transform the primary DNA into mature messenger RNA. Certain nucleotides in the chain are modified and others are excerpted by a splicing mechanism, yielding a mRNA chain shorter than the original primary DNA [2].

### Translation

The synthesis of protein is accomplished by intracellular particles called ribosomes. The ribosome is a complex apparatus made up of 52 different proteins and three different RNA molecules. It is a quite efficient machine, adding up some 15 amino acids, to the growing chain, per second [3].

The instructions for coding protein are delivered to the ribosome by the messenger RNA (mRNA). Each amino acid is transported to the ribosome synthesis sites by transfer RNA's (tRNA). The tRNA carries an anticodon which matches, in a complementary fashion, the codon of the messenger RNA.

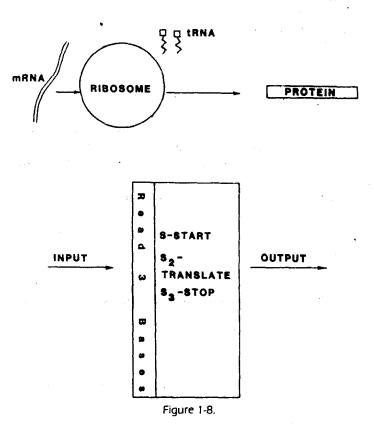
The ribosome has two binding sites, one for positioning the arriving tRNA carrying the selected amino acid, and the other for holding the growing polypeptide chain.

The process we have described above has been compared to a language, to a computer and more generally, to a finite state machine.

The skeleton of a finite state machine can be uncovered by reducing most modern devices for their bare bone essential elements. A mouse-trap, a household appliance, a computer, is a finite state machine [4].

At the molecular level, the ribosome and the various transfer RNA's operate as a finite state machine. The input are the four RNA bases (A,U,G,C); the output is a protein made up of infinite rearrangement and ordering of the basic 20 amino acids. The machine must be able to recognize a start signal (AUG) and a stop signal (UAA, UAG, UGA). The machine reads three bases at a time and can perform either of three functions: start the processing, halt, or translate the input into an output (Figure 1-8).





### The Protein

Twenty amino acids are commonly occurring in protein.

Glycine	GLY	Lysine	LYS
Alanine	ALA	Arginine	ARG
Valine	VAL	Asparagine	ASN
Isoleucine	ILE	Glutamine	GLN
Leucine	LEU	Cysteine	CYS
Serine	SER ·	Methionine	MET
Threonine	THR	Tryptophan	TRP
Proline	PRO <sub>、</sub>	Phenylalanine	PHE
Aspartic acid	ASP	Tyrosine	TYR
Glutamic acid	GLU	Histidine	HIS

Proteins are the basis of all living phenomena as structural support, catalyst of equisite selectivity, storage media, transport systems, hormone messengers, etc.