## Handbook of Reagents for Organic Synthesis

# Reagents, Auxiliaries, and Catalysts for C–C Bond Formation

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## Handbook of Reagents for Organic Synthesis

## Reagents, Auxiliaries, and Catalysts for C–C Bond Formation

Edited by

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University of Illinois at Urbana-Champaign

and

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University of Illinois at Urbana-Champaign

JOHN WILEY & SONS LTD

Chichester · New York · Weinheim · Brisbane · Toronto · Singapore

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National 01243 779777
International (+44) 1243 779777
e-mail (for orders and customer service enquiries): cs-books@wiley.co.uk
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Other Wiley Editorial Offices

John Wiley & Sons Inc., 605 Third Avenue, New York, NY 10158-0012, USA

Wiley-VCH Verlag GmbH, Pappelallee 3, D-69469 Weinheim, Germany

Jacaranda Wiley Ltd, 33 Park Road, Milton, Queensland 4064, Australia

John Wiley & Sons (Asia) Pte Ltd, 2 Clementi Loop #02-01, Jin Xing Distripark, Singapore 129809

John Wiley & Sons (Canada) Ltd, 22 Worcester Road, Rexdale, Ontario M9W 1L1, Canada

#### Library of Congress Cataloguing-in-Publication Data

Handbook of reagents for organic synthesis.

p. cm. Includes bibliographical references.

Contents: [1] Reagents, auxiliaries, and catalysts for C-C bond formation / edited by Robert M Coates and Scott E Denmark [2] Oxidising and reducing agents / edited by Steven D Burke and Riek L. Danheiser [3] Acidic and basic reagents / edited by Hans J. Reich and James H. Rigby [4] Activating agents and protecting groups / edited by Anthony J Pearson and William R Roush ISBN 0-471-97924-4 (v. 1). ISBN 0-471-97926-0 (v. 2) ISBN 0-471-97925-2 (v. 3) ISBN 0-471-97927-9 (v. 4)

1. Chemical tests and reagents 2 Organic compounds—Synthesis QD77.H37 1999

547'.2 dc 21 98-53088

98-53088 CIP

#### British Library Cataloguing in publication Data

A catalogue record for this book is available from the British Library

ISBN 0 471 97924 4

Typset by Thomson Press (India) Ltd., New Delhi Printed and bound in Great Britian by Antony Rowe, Chippenham, Wilts This book is printed on acid-free paper responsibly manufactured from sustainable forestry, in which at least two trees are planted for each one used in paper production.

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Oxidizing and Reducing Agents
Edited by Steven D. Burke and Rick L. Danheiser
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Activating Agents and Protecting Groups
Edited by Anthony J. Pearson and William R. Roush
ISBN 0471979279

This volume is dedicated to the memory of William G. Dauben, an inspiring mentor and forthright colleague. We acknowledge the authors of the original articles in the *Encyclopedia of Reagents for Organic Synthesis* whose work forms the large body of this new edition. We would like to thank Professor Jeremiah P. Freeman and *Organic Syntheses* for providing the original graphics for the *Organic Syntheses* procedures presented in Section 3C.

We appreciate the assistance of Shirley Pierson and Linda Hirsch for the reformatting and compilation of much of the introductory material.

Robert M. Coates and Scott E. Denmark

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## **Preface**

As stated in its Preface, the major motivation for our undertaking publication of the Encyclopedia of Reagents for Organic Synthesis was "to incorporate into a single work a genuinely authoritative and systematic description of the utility of all reagents used in organic chemistry." By all accounts, this reference compendium has succeeded admirably in attaining this objective. Experts from around the globe contributed many relevant facts that define the various uses characteristic of each reagent. The choice of a masthead format for providing relevant information about each entry, the highlighting of key transformations with illustrative equations, and the incorporation of detailed indexes serve in tandem to facilitate the retrieval of desired information.

Notwithstanding these accomplishments, the editors have since recognized that the large size of this eight-volume work and its cost of purchase have often detered the placement of copies of the *Encyclopedia* in or near laboratories where the need for this type of information is most critical. In an effort to meet this demand in a cost-effective manner, the decision was made to cull from the major work that information having the highest probability for repeated consultation and to incorporate same into a set of handbooks. The latter would also be purchasable on a single unit basis.

The ultimate result of these deliberations is the publication of the *Handbook of Reagents for Organic Synthesis* consisting of the following four volumes: Reagents, Auxiliaries, and Catalysts for C-C Bond Formation
edited by Robert M. Coates and Scott E. Denmark

Oxidizing and Reducing Agents
edited by Steven D. Burke and Rick L. Danheiser

Acidic and Basic Reagents
edited by Hans J. Reich and James H. Rigby

Activating Agents and Protecting Groups

edited by Anthony J. Pearson and William R. Roush

Each of the volumes contains a selected compilation of those entries from the original *Encyclopedia* that bear on the specific topic. Ample listings can be found to functionally related reagents contained in the original work. For the sake of current awareness, references to recent reviews and monographs have been included, as have relevant new procedures from *Organic Syntheses*.

The end product of this effort by eight of the original editors of the *Encyclopedia* is an affordable, enlightening set of books that should find their way into the laboratories of all practicing synthetic chemists. Every attempt has been made to be of the broadest synthetic relevance and our expectation is that our colleagues will share this opinion.

Leo A. Paquette Columbus, Ohio USA

## Introduction

Every practicing organic chemist recognizes the critical importance of selecting the most appropriate reagent and reaction conditions for executing a given chemical transformation. While a working knowledge of the most common reagents employed in organic synthesis is essential for every synthetic chemist, even the most dedicated practitioner can hardly claim fluency with the overwhelming array of inorganic and organic reagents now at his or her disposal. Moreover, the development of new and more selective reagents has accelerated exponentially and still constitutes one of the most vigorous areas of organic chemistry research. In recognition of this challenge, the Editors of the Encyclopedia of Reagents for Organic Synthesis (EROS) have selected approximately 500 of the most important and useful reagents employed in organic synthesis. In keeping with our goal to provide a concise, desktop reference work, we culled the most important and commonly used reagents from the over 3000 entries in the original EROS. These selections comprise the Handbook of Reagents for Organic Synthesis and cover the full range of chemical types and transformations. The Handbook is divided into four volumes that contain reagents of similar type and/or function to assist the user in rapidly locating a reagent of choice: Reagents, Auxiliaries, and Catalysts for C-C Bond Formation; Oxidizing and Reducing Agents; Acidic and Basic Reagents; and Activation Agents and Protection Groups.

This volume entitled Reagents, Auxiliaries, and Catalysts for C-C Bond Formation contains the largest and most diverse group of reagents by virtue of the broad scope of chemical reactions included in this general category. Two major types of reagents selected from the parent EROS are essential carbon-nucleophiles and carbon-electrophiles widely used to form C-C bonds in synthesis. The C-nucleophile group includes numerous organometallic reagents, carbanions, enolates, ylides, and their precursors. Familiar alkylating, acylating, and cyclopropanating reagents together with Michael acceptors and other electron-deficient olefins comprise a fundamental group of C-electrophiles. An indispensable subset of reagents for cycloadditions includes acetylenes, allenes, dienes, dienophiles, and

ketenes. However, many other reagents extensively represented in this volume are critical participants in C-C bond forming reactions, which are not themselves incorporated into the products, such as catalysts, chiral auxiliaries, and selected heteroatom electrophiles. Particular attention was paid to the selection of the most important chiral catalysts and auxiliaries in recognition of the growing importance of enantioselective and diastereoselective transformations. Numerous transition-metal catalysts and stoichiometric metalloid reagents vital for C-C coupling reactions and cyclopropanations were deemed appropriate. While efforts were made to minimize duplication, a number of entries found in this volume have multiple applications and therefore appear in other volumes, e.g. 1,4-benzoquinone (see also Oxidizing and Reducing Agents), cerium trichloride (see also Acidic and Basic Reagents), chromium(II) chloride (see also Oxidizing and Reducing Agents), diazomethane (see also Activating Agents and Protecting Groups) and thexylborane (see also Oxidizing and Reducing Agents). On the other hand, some obvious redox reagents such as nickel(II) chloride, palladium(II) acetate and chloride, and manganese(III) acetate were chosen exclusively for this volume on account of their primary function in C-C bond formation. Although chlorotrimethylsilane, tri-n-butylchlorostannane, (R)- and (S)-menthyl p-toluenesulfinates, ptoluenesulfonyl azide, and p-toluenesulfonylhydrazide are not themselves used for C-C bond formation, their crucial role in the synthesis of C-C bond forming reagents provided a compelling rationale for their inclusion in this volume. In contrast, reagents such as the isomeric butyllithiums and pyrrolidine, while obviously closely associated with the generation of C-C bond forming reagents, are assigned exclusively to Acidic and Basic Reagents in recognition of their primary function as bases.

To familiarize the user with the spectrum of reagents contained in this volume and to organize the pertinent reference material, we have subdivided the 203 featured reagents into 22 classes based on their chemical structures. Clearly many reagents are multi-functional and could be logically assigned to several different classes. In many

cases the class assignment was based upon the functionality that would normally be retained in the product of their common synthetic applications. Some examples are ethyl bromozincacetate (Class 9: "Enolates, Homoenolates, and Dicarbonyl Compounds"), lithium bis[dimethyl(phenyl)silyllcuprate (Class 18: "Silicon and Tin Reagents"), and trifluoromethyltrimethylsilane (Class 11: "Halo Compounds"). All but one (nitromethane) of the designated "C1 Reagents" in Class 3 are carbonyl compounds commonly utilized as electrophilic reagents to introduce oxygen functionality in C-C bond forming reactions (e.g. formaldehyde and N,N-dimethylformamide). On the other hand, the common electrophilic C<sub>1</sub> reagents used to introduce nitrogen functionality are found in Class 12: "Imines, Iminium ions, and Amide Acetals" (e.g. formaldehyde-dimethlyamine and dimethylformamide acetal). classifications should afford conceptual and organizational guidance for the users. It is worth while to remind readers that although headings designate a single compound, often related reagents (e.g. other esters, enantiomers of chiral reagents) are also covered. The reagent articles are arranged alphabetically in the volume, as they are in the original EROS. In addition, the Table of Contents also shows the class number to aid readers in finding reagents of similar structure and function.

- 1. Acetylene and Allenes
- 2. Aluminum and Boron Reagents
- 3. C<sub>1</sub> Reagents
- 4. Chiral Auxiliaries and Reagents
- 5. Copper Reagents
- 6. Cyano, Isocyano, and Isocyanato Reagents
- 7. Diazo, Hydrazido, and Azido Reagents
- 8. Dienes, Dienophiles, and Michael Acceptors
- 9. Enolates, Homoenolates, and Dicarbonyl Compounds
- 10. Epoxides
- 11. Halo Compounds
- 12. Imines, Iminium Ions, and Amide Acetals
- 13. Ketenes, Ketene Acetals, and Ortho Esters
- 14. Lithium and Magnesium Reagents
- 15. Nickel Reagents
- 16. Palladium Reagents
- 17. Phosphorus Reagents
- 18. Silicon and Tin Reagents

- 19. Sulfur Reagents
- 20. Titanium Reagents
- 21. Transition Metal and Lanthanide Reagents
- 22. Zinc Reagents

Each of the reagent entries appears almost verbatim from the original work (EROS). However, to make this new work as current as possible and to provide additional, useful information, the section Editors have compiled lists of recent reviews (1992- and 1998) in the general areas of each Volume. In addition, a collection of recent procedures from Organic Syntheses (Vol. 70-75) that feature preparations, reactions or applications pertinent to the content of this volume is provided in the front material. The reagent classes have been used to good advantage for the organization of this additional material that appears after the Table of Contents. The recent reviews and selected Organic Syntheses procedures are sorted and presented under the appropriate rubric of the 22 reagent classes. The Editors have taken a liberal view in the selection of recently published reviews, chapters, and monographs concerning important C-C bond forming methods related to each reagent class, whether or not the specific reagents appear in this volume. Furthermore, the Editors have culled out only the most relevant chemical transformations from multistep Organic Syntheses procedures. Finally, while the bulk of the entries are directly reproduced from EROS, they now all contain a Related Reagents section that guides the reader to reagents of similar structure or function that can be found either in the other volumes or in the original work.

The Editors of the *Handbook* are pleased with the content and organization of each of the four volumes in this set and we hope that your chemical enterprises will benefit from the efforts of the original researchers, authors and editors who have contributed to the creation of this valuable new resource.

Robert M. Coates
University of Illinois at Urbana-Champaign

Scott E. Denmark
University of Illinois at Urbana-Champaign

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#### **CLASS 1** Acetylenes and Allenes

#### A. Reagents

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Bis(trimethylsilyl)acetylene
Ethoxyacetylene
Lithium Acetylide
1-Methyl-1-(trimethylsilyl)allene
Trimethylsilylacetylene

#### B. Reviews (1992–1998)

- (1) Gleiter, R. and Kratz, D. Conjugated Enediynes-An Old Topic in a Different Light. Angew. Chem., Int. Ed. Engl. 1993, 32, 842-45.
- (2) Hopf, H. and Witulski, B. Cyanoalkynes: Magic Wands for the Preparation of Novel Aromatic Compounds. *Pure Appl. Chem.* 1993, 65, 47-56.
- (3) Kanematsu, K. Molecular Design and Syntheses of Biologically Active Compounds Via Intramolecular Allene Cycloaddition Reaction Strategy. Rev. Heteroatom. Chem. MYU K.K.; Tokyo; 1993, Vol. 9.
- (4) Zimmer, R. Alkoxyallenes-Building Blocks in Organic Synthesis. Synthesis 1993, 165-78.
- (5) Magnus, P. A General Strategy Using  $\eta^2 CO_2(CO)_6$  Acetylene Complexes For the Synthesis of the Enediyne Antitumor Agents Esperamicin, Calicheamicin, Dynemicin, and Neocarzinostatin. *Tetrahedron* 1994, 50, 1397-418.
- (6) Masuda, T. and Tachimori, H. Design, Synthesis, and Properties of Substituted Polyacetylenes. J. Macromol. Sci. Chem. 1994, 31, 1675-90.
- (7) Nicolaou, K. The Magic of Enediyne Chemistry. Chem. Britain 1994, 30, 33-6.
- (8) Casson, S. and Kocienski, P. The Hydrometallation, Carbometallation, and Metallometallation of Heteroalkynes. *Contemp. Org. Synth.* 1995, 2, 19-34.
- (9) Stang, P. J. and Diederich, F. Modern Acetylene Chemistry; VCH: Weinheim, 1995.

#### C. Organic Syntheses Procedures (Vols. 70-75)

Authors	Citation	Title
Peter J. Stang and Tsugio Kitamura	Org. Synth. 1992, 70, 215.	ALKYNYL(PHENYL)IODONIUM TOSYLATES: PREPARATION AND STEREOSPECIFIC COUPLING WITH VINYLCOPPER REAGENTS. FORMATION OF CONJUGATED ENYNES. 1-HEXYNYL(PHENYL)-IODONIUM TOSYLATE AND $(E)$ -5-PHENYLDODEC-5-EN-7-YNE
C <sub>4</sub> H <sub>9</sub> C≡CSi(	$CH_3)_3$ + PhiO $\frac{BF_3 \cdot Et_2O, CHCt_3}{0^{\circ}C \rightarrow 25^{\circ}C}$	$\frac{\text{aq. NaOTs}}{25^{\circ}\text{C}}  \text{C}_4\text{H}_9\text{C} \equiv \text{C} - \text{I}^{+} - \text{C}_6\text{H}_5 - \text{OTs}$
C.H.Rr —	Ag CuBr-SMe <sub>2</sub> C <sub>6</sub> H <sub>5</sub> C≡CH	$ \frac{C_6H_5C \equiv C-I^4-C_6H_5 \text{ OTs}}{C_6H_5} =  \begin{array}{c} C_6H_5 & H \\ C=C & C \end{array} $
		C₄H₀

#### CLASS 2 Aluminum and Boron Reagents

#### A. Reagents

Allenylboronic Acid B-Allyldiisopinocampheylborane 9-Borabicyclo[3.3.1]nonane Dimer Di-n-butylboryl Trifluoromethanesulfonate Diisopinylcampheylboron Trifluoromethanesulfonate Diisopropyl 2-Crotyl-1,3,2-dioxaborolane-4,5-dicarboxylate Thexylborane Triethyaluminum

#### Triethylborane

Trimethylaluminum

#### B. Reviews (1992-1998)

- (1) Lohray, B. B. and Bhushan, V. Oxazaborolidines and Dioxaborolidines in Enantioselective Catalysis, Angew. Chem., Int. Ed. Engl. 1992, 31, 729-30.
- (2) Deloux, L. and Srebnik, M. Asymmetric Boron-Catalyzed Reactions. Chem. Rev. 1993, 93, 763-84.
- (3) Imamoto, T. Synthesis and Reactions of New Phosphine-Boranes. Pure Appl. Chem. 1993, 65, 655-60.
- (4) Brown, H. C. and Ramachandran, P. V. Recent Advances in the Boron Route to Asymmetric Synthesis. Pure Appl. Chem. 1994, 66, 201-12.
- (5) Kabalka, G. W. Current Topics in the Chemistry of Boron; Royal Society of Chemistry: Cambridge, U.K., 1994.
- (6) Suzuki, A. New Synthetic Transformations Via Organoboron Compounds. Pure Appl. Chem. 1994, 66, 213-22.
- (7) Brown, H. C. and Ramachandra, P. V. In Advances in Asymmetic Synthesis; JAI Press: Greenwich 1995; Vol. 1.
- (8) Matteson, D. S. Stereodirected Synthesis with Organoboranes; Springer: Berlin, 1995.
- (9) Suzuki, A. Haloboration of 1-Alkynes and Its Synthetic Application. Rev. Heteroatom Chem. 1997, 17, 271-314.
- (10) Yamamoto, H. Organoaluminum Compounds; In Organometallics in Synthesis A Manual Wiley: New York, 1994; Chapt. 7.

#### C. Organic Syntheses Procedures (Vols. 70-75)

Authors

Citation

Title

John A. Soderquist and Alvin Negron

Org. Synth. 1992, 70, 169.

9-BORABICYCLO[3.3.1]NONANE DIMER

Tatsuo Ishiyama, Norio Miyaura, and Akira Suzuki

Org. Synth. 1992, 71, 89.

PALLADIUM(0)-CATALYZED REACTION OF 9-ALKYL9-BORABICYCLO[3.3.1]-NONANE WITH 1-BROMO-1-PHENYLTHIOETHENE: 4-(3-CYCLOHEXENYL)-2-PHENYLTHIO-1-BUTENE

T. Ooi, K. Maruoka, and H. Yamamoto

Org. Synth. 1993, 72, 95.

REARRANGEMENT OF trans-STILBENE OXIDE TO DIPHENYLACETALDEHYDE WITH CATALYTIC METHYLALUMINUM BIS(4-BROMO-2,6-DI-tert-BUTYLPHENOXIDE)

George W. Kabalka, John T. Maddox, Timothy Shoup, and Karla R. Bowers

Org. Synth. 1995, 73, 116.

A SIMPLE AND CONVENIENT METHOD FOR THE OXIDATION OF ORGANOBORANES USING SODIUM PERBORATE: (+)-ISOPINOCAMPHEOL

Shoji Hara and Akira Suzuki

Org. Synth. 1997, 75, 129

SYNTHESIS OF 4-(2-BROMO-2-PROPENYL)-4-METHYL-γ-BUTYROLACTONE BY THE REACTION OF ETHYL LEVULINATE WITH (2-BROMOALLYL) DIISOPROPOXYBORANE PREPARED BY HALOBORATION OF ALLENE

$$H_2C=C=CH_2 \xrightarrow{1) BBr_3} \xrightarrow{Br} B(OP_{T-i})_2 \xrightarrow{O} CO_2Et \xrightarrow{OH} OH$$

#### CLASS 3 C<sub>1</sub> Reagents

#### A. Reagents

Carbon Monoxide
Diethyl Carbonate
Dimethoxycarbenium Tetrafluoroborate
N,N-Dimethylformamide
Formaldehyde
Methyl Cyanoformate
Methyl Formate
Nitromethane

### B. Reviews (1992-1998)

Paraformaldehyde

- (1) Sonoda, N. Selenium-Assisted Carbonylation With Carbon Monoxide. Pure Appl. Chem. 1993, 65, 699-706.
- (2) Tsuda, T. Utilization of Carbon-Dioxide in Organic Synthesis and Polymer Synthesis by the Transition-Metal Catalyzed Carbon-Dioxide Fixation into Unsaturated Hydrocarbons. Gazz. Chim. Ital. 1995, 125, 101-10.
- (3) Ono, Y. Dimethyl Carbonate for Environmentally Benign Reactions. Pure Appl. Chem. 1996, 68, 367-76.
- (4) Ryan, T. A. Phosgene and Related Compounds; Elsevier Science: New York, 1996.
- (5) Ryu, I. and Sonoda, N. Free-Radical Carbonylations: Then and Now. Angew. Chem., Int. Ed. Engl. 1996, 35, 1050-66.

#### C. Organic Syntheses Procedures (Vols. 70-75)

Authors
Citation
Title

Simon R. Crabtree, Lewis N. Mander, Org. Synth. 1992, 70, 256.
and S. Paul Sethi
Synthesis Of β-KETO ESTERS BY C-ACYLATION OF PREFORMED ENOLATES WITH METHYL CYANOFORMATE: PREPARATION OF METHYL (1α, 4aβ, 8aα)-2-OXO-

Li, NH<sub>3</sub> (liq.)

H

NCCO<sub>2</sub> Me

X. Wang, S. O. deSilva, J. N. Reed, R. Billadeau, E. J. Griffen, A. Chan, and V. Snieckus

Org. Synth. 1993, 72, 163.

7-METHOXYPHTHALIDE

**DECAHYDRO-1-NAPHTHOATE** 

#### **CLASS 4** Chiral Auxiliaries and Reagents

#### A. Reagents

(S)-1-Amino-2-methoxymethylpyrrolidine

(S)-4-Benzyl-2-oxazolidinone

(R)- and (S)-2,2'-Bis(diphenylphosphino)-1,1'-binaphthyl

(R)-2-t-Butyl-6-methyl-4H-1,3-dioxin-4-one

(R,R)-2-t-Butyl-5-methyl-1,3-dioxolan-4-one

(R)-(+)-Butyl 2-(p-Tolylsulfinyl)acetate

10,2-Camphorsultam

Chloro(cyclopentadienyl)-

bis[3-O-(1,2;5,6-di-O-isopropylidene-α-p-glucofuransoyl)]titanium

10-Dicyclohexylsulfonamidoisoborneol

(S)-(+)-2,5-Dihydro-2-isopropyl-3,6-dimethylpyrazine

 $(R^*,R^*)$  –  $\alpha$ -2,6-Diisopropylbenzyloxy-5-oxo-1,3,2-dioxaborolane-4-acetic Acid

(S)-N,N-Dimethyl-N'-(1-t-butoxy-3-methyl-2-butyl) formamidine

2,2'-(Dimethylmethylene)bis(4-t-butyl-2-oxazoline)

trans-2,5-Dimethylpyrrolidine

(R,R)-1,2 Diphenyl-1,2-diaminoethane, N,N'-Bis[3,5-bis(trifluoromethyl)benzenesulfonamide]

1R,2S-Ephedrine

(R)- and (S)-Ethyl 3-Hydroxybutyrate

(S)- and (R)-Ethyl lactate

3-Hydroxyisoborneol

(S)-(+)- and (R)-(-)-Mandelic Acid

p-(S)- and (R)-Menthyl Toluenesulfinate

(S)-2-Methoxymethylpyrrolidine

(1R,2S) - N-Methylephedrine

α-Methyltoluene-2,α-sultam

(R)-Pantolactone

(2R,4R)-2,4-Pentanediol

(-)-8-Phenylmenthol

1,1,2-Triphenyl-1,2-ethanediol

#### B. Reviews (1992-1998)

- (1) Blaser, H.-U. The Chiral Pool as a Source of Enantioselective Catalysts and Auxiliaries. Chem. Rev. 1992, 92, 935-52.
- (2) Braun, M. Recent Developments in Stereoselective Aldol Reactions; Advances in Carbanion Chemistry; JAI Press: Greenwich, CT, 1992, Vol. 1.
- (3) Chelucci, G. Chiral Ligands Based on the Pyridine Framework: Synthesis and Application in Asymmetric Catalysis. Gazz. Chim. Ital. 1992, 122, 89-98.
- (4) Faber, K. and Riva, S. Enzyme-Catalyzed Irreversible Acyl Transfer. Synthesis 1992, 895-910.
- (5) Halgas, J. Biocatalysts in Organic Synthesis (Studies in Organic Chemistry); Elsevier: Amsterdam, 1992.
- (6) Hayashi, T.; Kubo, A. and Ozawa, F. Catalytic Asymmetric Arylation of Olefins. Pure Appl. Chem. 1992, 64, 421-7.
- (7) Inoue, Y. Asymmetric Photochemical Reactions in Solution. Chem. Rev. 1992, 92, 741-70.
- (8) Knochel, P. Asymmetric Deprotonations as an Efficient Enantioselective Preparation of Functionalized Secondary Alcohols. Angew. Chem., Int. Ed. Eng. 1992, 31, 1459-61.
- (9) Poppe, L. and Novak, L. Selective Biocatalysis. A Synthetic Approach; VCH: Weinheim, 1992.
- (10) Rosini, C.; Franzini, L.; Raffaelli, A. and Salvadori, P. Synthesis and Applications of Binaphthylic  $C_2$ -Symmetry Derivatives as Chiral Auxiliaries in Enantioselective Reactions. Synthesis 1992, 503–17.
- (11) Santaniello, E.; Ferraboschi, P.; Grisenti, P. and Manzoocchi, A. The Biocatalytic Approach to the Preparation of Enantiomerically Pure Chiral Building Blocks. Chem. Rev. 1992, 92, 1071-140.
- (12) Sawamura, M. and Ito, Y. Catalytic Asymmetric Synthesis by Means of Secondary Interaction Between Chiral Ligands and Substrates. Chem. Rev. 1992, 92, 857-781.

- (13) Whitesell, J. K. Cyclohexyl-based Chiral Auxiliaries. Chem. Rev. 1992, 92, 953-64.
- (14) Williams, R. M. Asymmetric Syntheses of α-Amino Acids. Aldrichimica Acta 1992, 24, 11-13 & 15-25.
- (15) Banfi, L. and Guanti, G. Asymmetrized 2-Methyl-1,3-propanediol and its Equivalents: Preparation and Synthetic Applications. Synthesis 1993, 1029-56.
- (16) Brunner, H. and Zettlmeier, W. Handbook of Enantioselective Catalysis; VCH: Weinheim, 1993, Vols. 1 & 2.
- (17) Danieli, B.; Lesma, G.; Passarella, D. and Riva, S. Chiral Synthons Via Enzyme-Mediated Asymmetrization of Meso-Compounds. Advances in the Use of Synthons in Organic Chemistry; JAI Press: Greenwich, 1993, Vol. 1.
- (18) Fuji, K. Asymmetric Creation of Quaternary Carbon Centers. Chem. Rev. 1993, 93, 2037-66.
- (19) Kim, B. H. and Curran, D. P. Asymmetric Thermal Reactions with Oppolzer's Camphor Sultam. Tetrahedron 1993, 49, 293-318.
- (20) Kunz, H. and Rück, K. Carbohydrates as Chiral Auxiliaries in Stereoselective Synthesis. Angew. Chem., Int. Ed. Engl. 1993, 32, 336-58.
- (21) Ojima, I. Catalytic Asymmetric Synthesis; VCH: New York, 1993.
- (22) Pfaltz, A. Chiral Semicorrins and Related Nitrogen Heterocycles as Ligands in Asymmetric Catalysis. Acc. Chem. Res. 1993, 26, 339-45.
- (23) Sakai, K. and Suemune, H. Application of Chiral Cyclic Diols to Asymmetric Synthesis. Tetrahedron: Asymmetry 1993, 4, 2109-18.
- (24) Winterfeldt, E. Enantiomerically Pure Cyclopentadienes. Chem. Rev. 1993, 93, 827-43.
- (25) Bach, T. Catalytic Enantioselective C-C Coupling Allyl Transfer and Mukaiyama Aldol Reaction. Angew. Chem., Int. Ed. Engl. 1994, 33, 417-19.
- (26) Brunner, H. Natural Products by Enantioselective Catalysis With Transition Metal Compounds. Pure Appl. Chem. 1994, 66, 2033-6.
- (27) Cervinka, O. Enantioselective Reactions in Organic Chemistry; Horwood: London, 1994.
- (28) Duthaler, R. O. Recent Developments in the Stereoselective Synthesis of α-Amino Acids. Tetrahedron 1994, 50, 1539-650.
- (29) Franklin, A. S. and Paterson, I. Recent Developments in Asymmetric Aldol Methodology. Contemp. Org. Synth. 1994, 1, 317-38.
- (30) Gant, T. G. and Meyers, A. I. The Chemistry of 2-Oxazolines (1985-present). Tetrahedron 1994, 50, 2297-360.
- (31) Harada, T. and Oku, A. Enantiodifferentiating Transformation of Prochiral Polyols by Using Menthone as Chiral Template. Synlett 1994, 95-104.
- (32) Jones, J. B. Probing the Specificity of Synthetically Useful Enzymes. Aldrichimica Acta 1994, 26, 105-112.
- (33) Nógrádi, M. Stereoselective Synthesis: A Practical Approach, 2nd Ed.; VCH: Weinheim, 1994.
- (34) Noyori, R. Asymmetric Catalysis in Organic Synthesis; Wiley: New York, 1994.
- (35) Atkinson, R. S. An Introduction to Stereoselective Synthesis; Wiley: New York, 1995.
- (36) Kunz, H. Stereoselective Syntheses Using Carbohydrates as Chiral Auxiliaries. Pure Appl. Chem. 1995, 67, 1627-36.
- (37) Waldmann, H. Amino Acid Esters: Versatile Chiral Auxiliary Groups for the Asymmetric Synthesis of Nitrogen Heterocycles. Synlett 1995, 133-41.
- (38) Williams, R. M. Asymmetric Synthesis of α-Amino Acids. Advances in Asymmetric Synthesis; JAI Press: Greenwich, 1995, Vol. 1.
- (39) Ager, D. J. and East, M. B. Asymmetric Synthetic Methodology; CRC Press: Boca Raton, 1996.
- (40) De Lucchi, O. High Symmetry Chiral Auxiliaries Containing Heteroatoms. Pure Appl. Chem. 1996, 68, 945-50.
- (41) Enders, D. and Klatt, M. Asymmetric Synthesis with (S)-2-Methoxymethyl Pyrrolidine (SMP); A Pioneer Auxiliary. Synthesis 1996, 1403-18.
- (42) Gawley, R. E. and Aube, J. Principles of Asymmetric Synthesis; Elsevier: Amsterdam, 1996.
- (43) Klabunovskii, E. I. Catalytic Asymmetric Synthesis of β-Hydroxyacids and Their Esters. Russ. Chem. Rev. 1996, 65, 329-44.
- (44) Pfaltz, A. Design of Chiral Ligands for Asymmetric Catalysis: From C2-Symmetric Semicorrins and Bisoxazolines to Non-Symmetric Phosphinooxazolines. Acta Chem. Scand. 1996, 50, 189-94.
- (45) Studer, A. Amino Acids and Their Derivatives as Stoichiometric Auxilaries in Asymmetric Synthesis. Synthesis 1996, 793-815.
- (46) Abiko, A. Isoxazolidine-Based Chiral Auxiliaries for Asymmetric Syntheses. Rev. Heteroatom Chem. 1997, 17, 51-72.
- (47) Ager, D. J.; Prakash, I. and Schaad, D. R. Chiral Oxazolidinones in Asymmetric Synthesis. Aldrichimica Acta 1997, 30, 3-12.
- (48) Bennani, Y. L. and Hanessian, S. *Trans*-1,2-Diaminocyclohexane Derivatives as Chiral Reagents, Scaffolds, and Ligands for Catalysis: Applications in Asymmetric Synthesis and Molecular Recognition. *Chem. Rev.* 1997, 97, 3161-95.

Avoid Skin Contact with All Reagents