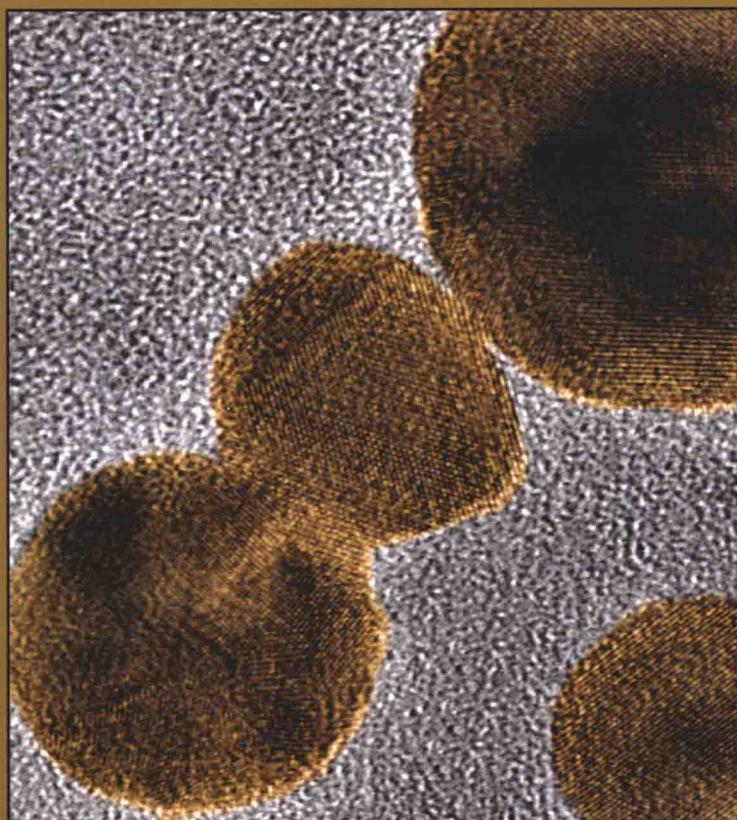


Advances in Materials Science and Engineering

# Semiconductor Nanocrystals and Metal Nanoparticles

Physical Properties  
and Device Applications



Edited by

Tupei Chen • Yang Liu



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# Preface

A material particle or cluster having a size in the range of one to several hundreds of nanometers is often referred to as “nanocrystal” or “nanoparticle.” *Nanocrystal* is considered as crystalline clusters in either a single crystalline or a polycrystalline arrangement, while *nanoparticle* can represent both crystalline and noncrystalline clusters. Semiconductor nanocrystals with small dimensions are also described as quantum dots. Semiconductor nanocrystals and metal nanoparticles exhibit fascinating behaviors and unique physical properties as a result of quantum size effect. For example, for a semiconductor nanocrystal whose size is smaller than twice the size of its exciton Bohr radius, the excitons are squeezed, leading to quantum confinement. Quantum confinement will modify the electronic and optical properties of the semiconductor nanocrystal, for example, the bandgap of the nanocrystal increases with the decrease of the nanocrystal size. A direct experimental observation of the quantum confinement effect is that different sized quantum dots emit different color light. A good example of the fascinating behaviors of metal nanoparticles is the localized surface plasmon resonance (LSPR) in noble metal (for example, silver and gold) nanoparticles where the conduction electrons oscillate coherently under irradiation by light in the visible and near-infrared regions of the electromagnetic spectrum. The wavelength of excitation of the LSPR shows strong dependence on the size, shape, and dielectric environment of the metal nanoparticles.

Semiconductor nanocrystals and metal nanoparticles are the building blocks of the next generation of electronic, optoelectronic, and photonic devices. For instance, semiconductor quantum dots are particularly significant for optoelectronic/photonic device applications such as photovoltaic devices, light-emitting devices, and photodetector devices due to their tunable absorption spectrum and high extinction coefficient. In electronic applications, semiconductor nanocrystals or metal nanoparticles with sizes smaller than ~5 nm can be used to realize single-electron or few-electron devices (e.g., transistors and memory devices) as such tiny nanostructures exhibit the Coulomb blockade effect at room temperature.

This book examines in detail the physical properties and device applications of semiconductor nanocrystals and metal nanoparticles. It begins by giving a review on the synthesis and characterization of various semiconductor nanocrystals and metal nanoparticles and goes on to discuss in detail their electronic, optical, and electrical properties. Based on the knowledge of the physical properties, the book illustrates some exciting applications in nanoelectronic devices (e.g., memristors, single-electron devices), optoelectronic devices (e.g., UV detectors, quantum dot lasers, solar cells), and other applications (e.g., gas sensors, metallic nanopastes for power electronics packaging).

This book contains 13 chapters. It covers a rapidly developing, interdisciplinary field, and it is also a compilation of some research efforts in the field. It is intended as a reference book for senior undergraduate and graduate students and researchers in microelectronics/nanoelectronics, optoelectronics, photonics, nanoscience and nanotechnology, physics, and materials science.

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# 1 Synthesis and Characterization of Nanocrystals and Nanoparticles

*Yang Liu*

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## 1.1 INTRODUCTION

Nanocrystals and nanoparticles are described as clusters having diameters in the range of one to several hundreds of nanometers. A “nanocrystal” is considered as a cluster of crystals, whereas a “nanoparticle” can consist of both crystalline and noncrystalline clusters. The synthesis of nanocrystals/nanoparticles has attracted extensive studies in the past several decades. These materials show size-dependent physical properties that are not shown by their bulk counterparts. For example, a Si nanocrystal shows photoluminescence (PL) and electroluminescence (EL), which cannot be achieved in bulk silicon due to its indirect bandgap. Therefore, precise control during the synthesis of nanocrystals/nanoparticles is of key importance for their applications.

Many techniques are used to synthesize nanocrystals/nanoparticles. Chemical vapor deposition (CVD), physical vapor deposition, ion implantation, sputtering, laser ablation, and spray pyrolysis are all gas-based techniques and have been used successfully for the synthesis of nanocrystals/nanoparticles. They provide the possibility of controlling the nanocrystal/nanoparticle size by adjusting the synthesis parameters such as temperature, pressure, power density, gas flow rate, and so on. Solution-based techniques are of intensive research interest, as they are highly effective in synthesizing nanocrystals/nanoparticles with good size control. Besides, they also have the advantages of low synthesis temperature, flexibility, and low cost. A large variety of methods have been developed to synthesize nanocrystals/nanoparticles, such as reduction, thermal decomposition, and processes with hydrolysis and alcoholysis. At the same time, various techniques have been utilized to characterize the structural and physical properties of nanocrystals/nanoparticles, such as transmission electron microscopy (TEM), scanning electron microscopy (SEM), atomic force microscopy (AFM), X-ray photoemission spectroscopy (XPS), X-ray diffraction (XRD), Raman scattering, and so on. In this chapter, we review the methods for synthesizing nanocrystals/nanoparticles. We also discuss the characterization of nanocrystals/nanoparticles using various techniques.



## 1.2 SEMICONDUCTOR NANOCRYSTALS

### 1.2.1 COMPOUND SEMICONDUCTOR NANOCRYSTALS

Compound semiconductor nanocrystals such as CdSe, CdS, GaAs, InAs, and CuCl have been widely investigated. In this section, we discuss the synthesis of compound semiconductor nanocrystals as well as their characterization using various techniques.

Synthesis of II–VI (A2B6) nanocrystals/nanoparticles such as CdS and CdSe is attractive because these materials have potential applications in photosensitive, photovoltaic, optical, and electronic devices. The shape, concentration, and size of the nanocrystals, as well as their size distribution, strongly depends on the synthesis process. CdSe nanoparticles showing quantum size effects was reported in 1993 [1]. Research on nanocrystalline cadmium selenide (CdSe) has been going on for years due to its excellent properties such as light emission, low bandgap ( $E_g = 1.75$  eV), low toxicity, facile surface modification, and thermostability, with applications in light-emitting diodes, solar cells, hydrogen-producing catalysts, and biological imaging [2–15].

Chemical methods are simple, flexible, effective, and low cost to synthesize isolated CdSe nanocrystals, and include solvothermal methods and hydrothermal methods. The solvothermal methods always utilize TOP (tri-*n*-octylphosphine)/TOPO (tri-*n*-octylphosphine oxide), TOP/TOPO/HPA (hexylphosphonic), TOP/TOPO/HDA (hexadecylam), or other reagents to synthesize CdSe nanocrystals [16–23].

The TOP/TOPO method was reported in [16,20] and is illustrated in Figure 1.1. In [16], the synthesis of ZnS-capped CdSe nanocrystals was carried out by using TOP/TOPO with a size of 2.7–3.0 nm. In [20], CdSe nanocrystals were synthesized with a diameter of ~3 nm with TOP/TOPO. The nanocrystals were dried under  $N_2$  flow and dissolved in toluene. Finally, the nanocrystal solution was filtered using a 0.45 mm syringe filter to remove other contaminations from the synthesis process.

The TOP/TOPO/HPA method was used in [17], where  $Cd(CH_3)_2$  in heptane and Se/TBP (tri-*n*-butylphosphine) were mixed with TBP and the solution was injected into HPA (hexylphosphonic acid) in TOPO. The size of SeCd nanocrystals could be controlled in the range 5–20 nm by varying the injection volume. The TOP/TOPO/HDA method was reported in [19,21]. In [19],  $Cd(CH_3)_2$  and Se/TOP were injected into TOPO and HDA. The size of prepared CdSe nanocrystals was in the range 4.5–5.0 nm. In [21], CdO/SA (stock solution of stearic acid with CdO) was injected into

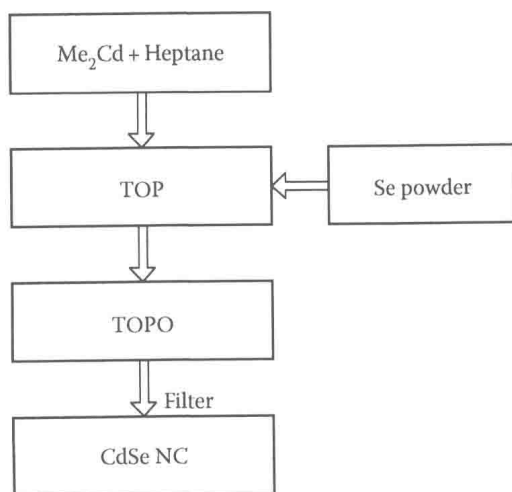


FIGURE 1.1 Process of the TOP/TOPO method.