

CLINICAL BIOCHEMISTRY

Lecture Notes



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10th Edition



WILEY Blackwell

Clinical Biochemistry

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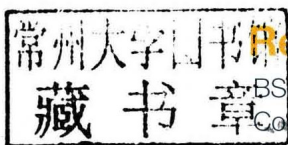
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Clinical Biochemistry

Lecture Notes

Preface

This is the tenth edition of the book originally written by Professor Gordon Whitby, Dr Alistair Smith and Professor Iain Percy-Robb in 1975. It remains an Edinburgh-based book, but both the content and the authorship continue to evolve.

Ever since the first edition this book has been primarily aimed at medical students and junior doctors, but we also believe that it will be of value to specialist registrars, clinical scientists and biomedical scientists pursuing a career in clinical biochemistry and metabolic medicine, and studying for higher qualifications. It has continued to develop in line with changes that have both reshaped the undergraduate curriculum and taken place in medical practice.

Over the course of the book's existence changes in medical education have tended to reduce or abolish courses exclusively covering laboratory medicine disciplines, with their content being integrated into the relevant parts of a systems-based curriculum. This clearly places the laboratory disciplines at the heart of medical teaching in the diagnosis and management of patients, but risks losing the opportunity to take a closer view of the principles behind the use of diagnostic investigations. This book aims to focus on the choice and interpretation of investigations in the diagnosis and management of conditions where biochemical testing plays a key role, with a view to understanding not only their uses but also developing an appreciation of their limitations. This is underpinned by brief summaries of the relevant pathophysiology. There is an emphasis on commonly requested tests and commonly occurring pathology, but less common tests and disorders are also described.

We have reviewed and updated all chapters to ensure that they reflect current clinical practice, the availability of new tests, and where relevant the latest versions of national guidelines, with an emphasis on those published in the UK. Planning this new edition benefited from helpful feedback from a number of sources, including groups of both students and their teachers, commissioned by Wiley, and in response to this we have among other changes increased the numbers of diagrams and tables where these help to summarise useful information. We have also increased the numbers of clinical cases, as these remain a popular feature. Multiple choice questions with an explanation of the answers, and key learning points for each chapter are available as an on-line resource for revision.

Since the last edition, Geoff Beckett, Simon Walker and Peter Ashby have all retired. They were authors since the fourth, fifth and seventh editions, respectively, and have had an enormous effect on the development and success of this book. Their places have been ably taken by Mike Crane and Rebecca Pattenden, who have brought a fresh perspective to many of the topics covered. As ever, we are also indebted to a number of colleagues who read various chapters and provided valuable comment and advice, in particular Catriona Clarke and Jonathan Malo. We remain grateful for the continued interest and support provided by the staff at Wiley towards this title since its first appearance over forty years ago.

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List of abbreviations

α -MSH	α -melanocyte stimulating hormone	CKD	chronic kidney disease
AAT	α_1 -antitrypsin	CNS	central nervous system
ABP	androgen-binding protein	CoA	coenzyme A
A&E	accident and emergency	COC	combined oral contraceptive
ACE	angiotensin-converting enzyme	COHb	carboxyhaemoglobin
ACTH	adrenocorticotrophic hormone	CRH	corticotrophin-releasing hormone
ADH	antidiuretic hormone	CRP	C-reactive protein
AFP	α -foetoprotein	CSF	cerebrospinal fluid
AI	angiotensin I	CT	computed tomography
AIi	angiotensin II	CV	coefficient of variation
AIll	angiotensin III	DDAVP	1-deamino,8-D-arginine vasopressin
AIP	acute intermittent porphyria	DHEA	dehydroepiandrosterone
AIS	androgen insensitivity syndrome	DHEAS	dehydroepiandrosterone sulphate
ALA	aminolaevulinic acid	DHCC	dihydrocholecalciferol
ALP	alkaline phosphatase	DHT	dihydrotestosterone
ALT	alanine aminotransferase	DIT	di-iodotyrosine
AMA	anti-mitochondrial antibodies	DKA	diabetic ketoacidosis
AMH	anti-Mullerian hormone	DPP	4 dipeptidyl peptidase-4
AMP	adenosine 5-monophosphate	DSD	disorder of sexual differentiation
ANP	atrial natriuretic peptide	DVT	deep venous thrombosis
AST	aspartate aminotransferase	ECF	extracellular fluid
ATP	adenosine triphosphate	ECG	electrocardiogram/electrocardiography
AT	Pase adenosine triphosphatase	ED	erectile dysfunction
β -LPH	β -lipotrophin	EDTA	ethylenediamine tetraacetic acid
BChE	butylcholinesterase	eGFR	estimated glomerular filtration rate
BMI	body mass index	EPH	electrophoresis
BMR	basal metabolic rate	EPP	erythropoietic protoporphyria
BNP	B-type natriuretic peptide	ERCP	endoscopic retrograde cholangiopancreatography
CABG	coronary artery bypass grafting	ESR	erythrocyte sedimentation rate
CAH	congenital adrenal hyperplasia	FAD	flavin adenine dinucleotide
cAMP	cyclic adenosine monophosphate	FAI	free androgen index
CBG	cortisol-binding globulin	FBHH	familial benign hypocalciuric hypercalcaemia
CCK	cholecystokinin	FIT	faecal immunochemical test
CCK-PZ	cholecystokinin-pancreozymin	FMN	flavin mononucleotide
CDT	carbohydrate-deficient transferrin	FOB	faecal occult blood
CEA	carcinoembryonic antigen	FPP	free protoporphyrin
CFT	calculated free testosterone	FSH	follicle-stimulating hormone
ChE	cholinesterase		
CK	creatine kinase		

FT3	free tri-iodothyronine	IFG	impaired fasting glucose
FT4	free thyroxine	Ig	immunoglobulin
GAD	glutamic acid decarboxylase	IGF	insulin-like growth factor
Gal-1-PUT	galactose-1-phosphate uridylyl-transferase	IGFBP	insulin-like growth factor-binding protein
GDM	gestational diabetes mellitus	IGT	impaired glucose tolerance
GFR	glomerular filtration rate	IM	intramuscular
GGT	γ -glutamyltransferase	INR	international normalised ratio
GH	growth hormone	IV	intravenous
GHD	growth hormone deficiency	LCAT	lecithin cholesterol acyltransferase
GHRH	growth hormone-releasing hormone	LDH	lactate dehydrogenase
GI	gastrointestinal	LDL	low-density lipoprotein
GIP	glucose-dependent insulinotropic peptide/ gastric inhibitory polypeptide	LH	luteinising hormone
GLP-1	glucagon-like polypeptide-1	Lp(a)	lipoprotein (a)
GnRH	gonadotrophin-releasing hormone	LSD	lysergic acid diethylamide
GP	general practitioner	MCAD	medium chain acyl-CoA dehydrogenase
GSH	glucocorticoid-suppressible hyperaldosteronism	MCV	mean cell volume
GTT	glucose tolerance test	MDRD	Modification of Diet in Renal Disease
Hb	haemoglobin	MEGX	monoethylglycinexylidide
HC	hereditary coproporphria	MEN	multiple endocrine neoplasia
HCC	hydroxycholecalciferol	MGUS	monoclonal gammopathy of unknown significance
hCG	human chorionic gonadotrophin	MI	myocardial infarction
HDL	high-density lipoprotein	MIT	mono-iodotyrosine
HGPRT	hypoxanthine-guanine phosphoribosyltransferase	MODY	maturity onset diabetes of the young
HHS	hyperosmolar hyperglycaemic state	MOM	multiples of the median
5-HIAA	5-hydroxyindoleacetic acid	MRCP	magnetic resonance cholangiopancreatography
HIV	human immunodeficiency virus	MRI	magnetic resonance imaging
HLA	human leucocyte antigen	MTC	medullary thyroid cancer
HMG-CoA	β -hydroxy- β -methylglutaryl-coenzyme A	MUST	Malnutrition Universal Screening Tool
HMMA	4-hydroxy-3-methoxymandelic acid	NABQI	<i>N</i> -acetyl- <i>p</i> -benzoquinoneimine
HNF	hepatic nuclear factor	NAC	<i>N</i> -acetylcysteine
HPA	hypothalamic-pituitary-adrenal	NAD	nicotinamide-adenine dinucleotide
HPLC	high-performance liquid chromatography	NADP	NAD phosphate
HRT	hormone replacement therapy	NAFLD	nonalcoholic fatty liver disease
hsCRP	highly sensitive C-reactive protein	NASH	nonalcoholic steatohepatitis
5-HT	5-hydroxytryptamine	NICE	National Institute for Health and Clinical Excellence
5-HTP	5-hydroxytryptophan	NIPT	noninvasive prenatal testing
IBS	irritable bowel syndrome	NSAID	nonsteroidal anti-inflammatory agent
ICF	intracellular fluid	NTD	neural tube defect
ICU	intensive care unit	NTI	nonthyroidal illness
IDL	intermediate-density lipoprotein	OCP	oral contraceptive pill
IFCC	International Federation for Clinical Chemistry	OGTT	oral glucose tolerance test

PAPP-A	pregnancy-associated plasma protein A	SGLT	sodium-glucose cotransporter
PBG	porphobilinogen	SUR	sulphonylurea receptor
PCI	percutaneous coronary intervention	T3	tri-iodothyronine
PCOS	polycystic ovarian syndrome	T4	thyroxine
PCSK9	proprotein convertase subtilisin/kexin type 9	TBG	thyroxine-binding globulin
PCT	porphyria cutanea tarda	TDM	therapeutic drug monitoring
PE	pulmonary embolism	TDP	thiamin diphosphate
PEG	percutaneous endoscopic gastrostomy	TGN	6-thioguanine nucleotide
PEM	protein-energy malnutrition	THR	thyroid hormone resistance
PIIINP	pro-collagen type III	TIBC	total iron-binding capacity
PKU	phenylketonuria	TNF	tumour necrosis factor
PLP	pyridoxal 5'-phosphate	TPMT	thiopurine S-methyltransferase
POCT	point of care testing	TPN	total parenteral nutrition
POP	progestogen-only pill	TPOAb	thyroid peroxidase antibody
PRPP	5-phosphoribosyl-1-pyrophosphate	TPP	thiamin pyrophosphate
PSA	prostate-specific antigen	TRAb	thyrotrophin receptor antibody
PT	prothrombin time	TRH	thyrotrophin-releasing hormone
PTH	parathyroid hormone	TSH	thyroid-stimulating hormone
PTHrP	PTH-related protein	TSI	thyroid-stimulating immunoglobulin
RBP	retinol-binding protein	tTG	tissue transglutaminase
RDA	recommended dietary allowance	U&Es	urea and electrolytes
RF	rheumatoid factor	UFC	urinary free cortisol
RMI	risk of malignancy index	UV	ultraviolet
ROC	receiver operating characteristic	VIP	vasoactive intestinal peptide
SAAG	serum-ascites albumin gradient	VLDL	very low density lipoprotein
SAH	subarachnoid haemorrhage	VMA	vanillylmandelic acid
SD	standard deviation	VP	variegate porphyria
SHBG	sex hormone-binding globulin	WHO	World Health Organization
SIADH	inappropriate secretion of ADH	XO	xanthine oxidase
		ZPP	zinc protoporphyrin

About the companion website

This book is accompanied by a companion website:



www.lecturenoteseries.com/clinicalbiochemistry

The website includes:

- Interactive multiple-choice questions
- Key revision points for each chapter

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Requesting and interpreting tests

Learning objectives

To understand:

- ✓ how sample handling, analytical and biological factors can affect test results in health and disease and how these relate to the concept of a test reference range;
- ✓ the concepts of accuracy, precision, test sensitivity, test specificity in the quantitative assessment of test performance.

Introduction

Biochemical tests are crucial to modern medicine. Most biochemical tests are carried out on blood using plasma or serum, but urine, cerebrospinal fluid (CSF), faeces, kidney stones, pleural fluid, etc. are sometimes required. Plasma is obtained by collecting blood into an anticoagulant and separating the fluid, plasma phase from the blood cells by centrifugation. Serum is the corresponding fluid phase when blood is allowed to clot. For many (but not all) biochemical tests on blood, it makes little difference whether plasma or serum is used.

There are many hundreds of tests available in clinical biochemistry but a core of common tests makes up the majority of tests requested. These core tests are typically available from most clinical laboratories throughout the 24-h period. Tests are sometimes brought together in profiles, especially when a group of tests provides better understanding of a problem than a single test (e.g. the liver function test profile). More specialist tests may be restricted to larger laboratories or specialist centres offering a national or regional service.

In dealing with the large number of routine test requests, the modern clinical biochemistry laboratory depends heavily on automated instrumentation linked to a laboratory computing system. Test results are assigned to electronic patient files that allow maintenance of a cumulative patient record. Increasingly, test requests can be electronically booked at the ward, clinic or in General Practice via a terminal linked to the main laboratory computer. Equally, the test results can be displayed on computer screens at distant locations, removing the need to issue printed reports.

In this first chapter, we set out some of the principles of requesting tests and of the interpretation of results. The effects of analytical errors and of physiological factors, as well as of disease, on test results are stressed. Biochemical testing in differential diagnosis and in screening is discussed.

Collection of specimens

Test requests require unambiguous identification of the patient (patient's name, sex, date of birth and, increasingly, a unique patient identification number), together with the location, the name of the

requesting doctor and the date and time of sampling. Each test request must specify the analyses requested and provide details of the nature of the specimen itself and relevant clinical diagnostic information. This may be through a traditional request form and labelled specimen or be provided electronically in which case only the sample itself need be sent to the laboratory with its own unique identifier (typically a bar code which links it to the electronic request).

Clinical laboratories have multiple procedures at every step of sample processing to avoid errors. Regrettably, errors do occur and these arise at different stages between the sample being taken and the result being received:

- Pre-analytical. These arise prior to the actual test measurement and can happen at the clinical or laboratory end. Most errors fall into this category (see Table 1.1).
- Analytical. Laboratory based analytical errors are rare but may occur, e.g. reagent contamination, pipetting errors related to small sample volumes, computing errors.
- Post-analytical. These are increasingly rare because of electronic download of results from the analyser but include, for example, transcription errors when entering results from another laboratory into the

computer manually; results misheard when these are telephoned to the clinician.

Despite the scale of requesting of biochemical tests, errors are fortunately very rare. However, occasional blunders do arise and, if very unexpected results are obtained, it is important that the requesting doctor contacts the laboratory immediately to check whether the results are indeed correct or whether some problem may have arisen. Occasionally this reveals that more than one problem has occurred, for example two samples were labelled with each other's details on the ward, so querying the results can have wider benefits.

The use of clinical biochemistry tests

Biochemical tests are most often *discretionary*, meaning that the test is requested for defined diagnostic purposes. The justification for discretionary testing is well summarised by Asher (1954):

- 1 Why do I request this test?
- 2 What will I look for in the result?
- 3 If I find what I am looking for, will it affect my diagnosis?

Table 1.1 Some more common causes of pre-analytical errors arising from use of the laboratory.

Error	Consequence
Crossover of addressograph labels between patients	This can lead to two patients each with the other's set of results. Where the patient is assigned a completely wrong set of results, it is important to investigate the problem in case there is a second patient with a corresponding wrong set of results.
Timing error	There are many examples where timing is important but not considered. Sending in a blood sample too early after the administration of a drug can lead to misleadingly high values in therapeutic monitoring. Interpretation of some tests (e.g. cortisol) is critically dependent on the time of day when the blood was sampled.
Sample collection tube error	For some tests the nature of the collection tube is critical, which is why the Biochemistry Laboratory specifies this detail. For example, using a plasma tube with lithium–heparin as the anti-coagulant is not appropriate for measurement of a therapeutic lithium level. Electrophoresis requires a serum sample rather than plasma so that fibrinogen does not interfere with the detection of any monoclonal bands. Topping up a biochemistry tube with a haematology (potassium ethylenediamine tetraacetic acid [EDTA]) sample will lead to high potassium and low calcium values in the biochemistry sample.
Sample taken from close to the site of an intravenous (IV) infusion	The blood sample will be diluted so that all the tests will be correspondingly low with the exception of those tests that might reflect the composition of the infusion fluid itself. For example, using normal saline as the infusing fluid would lead to a lowering of all test results, but with sodium and chloride results that are likely to be raised.

- 4 How will this investigation affect my management of the patient?
- 5 Will this investigation ultimately benefit the patient?

The main reasons for this type of testing are summarised in Table 1.2. Tests may also be used to help evaluate the future risk of disease (e.g. total cholesterol and HDL-cholesterol levels contribute to assessment of an individual's risk of cardiovascular disease), or in disease prognosis (e.g. biochemical tests to assess prognosis in acute pancreatitis or liver failure), or to screen for a disease, without there being any specific indication of its presence in the individual (e.g. maternal screening for foetal neural tube defects).

Screening may take several forms:

- In well-population screening a spectrum of tests is carried out on individuals from an apparently healthy population in an attempt to detect pre-symptomatic or early disease. It is easy to miss significant abnormalities in the large amount of data provided by the laboratory, even when the abnormalities are highlighted in some way. For these and other reasons, the value of well-population screening has been called into question and certainly should only be initiated under certain specific circumstances (Table 1.3).

Table 1.2 Test selection for the purposes of discretionary testing.

Category	Example
To confirm a diagnosis	Serum free T4 and thyroid-stimulating hormone (TSH) in suspected hyperthyroidism
To aid differential diagnosis	To distinguish between different forms of jaundice
To refine a diagnosis	Use of adrenocorticotrophic hormone (ACTH) to localise Cushing's syndrome
To assess the severity of disease	Serum creatinine or urea in renal disease
To monitor progress	Plasma glucose and serum K ⁺ to follow treatment of patients with diabetic ketoacidosis (DKA)
To detect complications or side effects	Alanine aminotransferase (ALT) measurements in patients treated with hepatotoxic drugs
To monitor therapy	Serum drug concentrations in patients treated with anti-epileptic drugs

Table 1.3 Requirements for well-population screening.

- The disease is common or life-threatening
- The tests are sensitive and specific
- The tests are readily applied and acceptable to the population to be screened
- Clinical, laboratory and other facilities are available for follow-up
- Economics of screening have been clarified and the implications accepted

Table 1.4 Examples of tests used in case-finding programmes.

Programmes to detect diseases in	Chemical investigations
Neonates	
Phenylketonuria (PKU)	Serum phenylalanine
Hypothyroidism	Serum TSH
Adolescents and young adults	
Substance abuse	Drug screen
Pregnancy	
Diabetes mellitus in the mother	Plasma glucose
Open neural tube defect (NTD) in the foetus	Maternal serum α -foetoprotein
Industry	
Industrial exposure to lead	Blood lead
Industrial exposure to pesticides	Serum cholinesterase activity
Elderly	
Malnutrition	Serum vitamin D levels
Thyroid dysfunction	Serum TSH and thyroxine

- In case-finding screening programmes appropriate tests are carried out on a population sample known to be at high risk of a particular disease. These are inherently more selective and yield a higher proportion of useful results (Table 1.4).

Point of care testing (POCT)

These are tests conducted close to the patient, for example in the emergency department, an out-patient clinic, or a general practitioner's surgery.

Table 1.5 Examples of POCT that are in common use.

Common POCT in blood	Common POCT in urine
Blood gases	Glucose
Glucose	Ketones
Urea and creatinine	Red cells/haemoglobin
Na, K and Ca	Bilirubin
Bilirubin	Protein
Alcohol	hCG

The instrumentation used is typically small and fits on a desk or may even be handheld. This approach can be helpful where there is a need to obtain a result quickly (e.g. blood gas results in the emergency department in a breathless patient), or where a result can be used to make a real-time clinical management decision (e.g. whether to adjust someone's statin dose on the basis of a cholesterol result). A further attraction is the immediate feedback of clinical information to the patient. POCT can be used to monitor illness by the individual patient and help identify if a change in treatment is needed (e.g. blood glucose monitoring in a diabetic patient). There is also an increasing number of urine test sticks that are sold for home use (e.g. pregnancy and ovulation testing by measuring human chorionic gonadotrophin (hCG) and luteinising hormone (LH), respectively). Table 1.5 shows examples of POCT tests in common use.

The introduction of POCT methodology requires attention to cost, ease of use, staff training, quality, health and safety as well as need. The advantages and disadvantages of POCT are summarised in Table 1.6.

Interpretation of clinical biochemistry tests

Most reports issued by clinical biochemistry laboratories contain numerical measures of concentration or activity, expressed in the appropriate units. Typically, the result is interpreted in relation to a reference range (see Chapter 1: Reference ranges) for the analyte in question. Results within and outside the reference range may be subject to variation caused by a number of factors. These include analytical variation, normal biological variation, and the influence of pathological processes.

Table 1.6 Advantages and disadvantages of point-of-care testing (POCT).

Advantages	Disadvantages
Rapid results on acutely ill patients	More expensive than centralised tests
Allows more frequent monitoring	Wide staff training may be needed
Immediate patient feedback	Nontrained users may have access, with potential for errors
Available 24 h if required	Calibration and quality control may be less robust
	Health and Safety may be less well monitored
	Results less often integrated into patient electronic record

Sources of variation in test results

Analytical sources of variation

Analytical results are subject to error, no matter how good the laboratory and no matter how skilled the analyst. The words “accuracy” and “precision” have carefully defined meanings in this context.

An *accurate* method will, on average, yield results close to the true value of what is being measured. It has no systematic bias. Lack of accuracy means that results will always tend to be either high or low.

A *precise* method yields results that are close to one another (but not necessarily close to the true value) on repeated analysis. If multiple measurements are made on one specimen, the spread of results will be small for a precise method and large for an imprecise one. Lack of precision means that results may be scattered, and unpredictably high or low.

A ‘dartboard’ analogy is often used to illustrate the different meanings of the terms accuracy and precision, and this is illustrated in Figure 1.1.

The standard deviation (SD) is the usual measure of scatter around a mean value. If the spread of results is wide, the SD is large, whereas if the spread is narrow, the SD is small. For data that have a Gaussian distribution, as is nearly always the case for analytical errors, the shape of the curve (Figure 1.2) is completely defined by the mean and the SD, and these characteristics are such that:

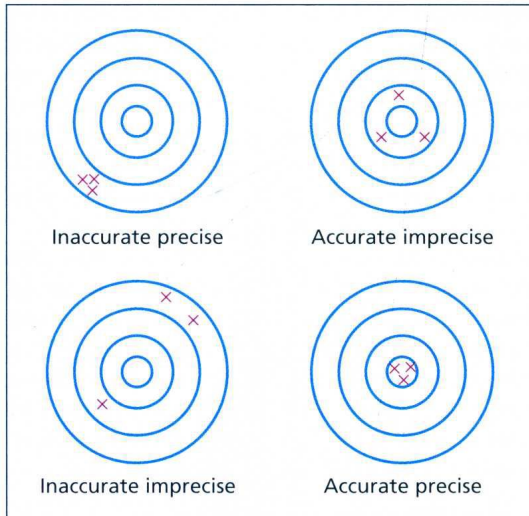


Figure 1.1 The 'dartboard' analogy can be used to illustrate accuracy and precision.

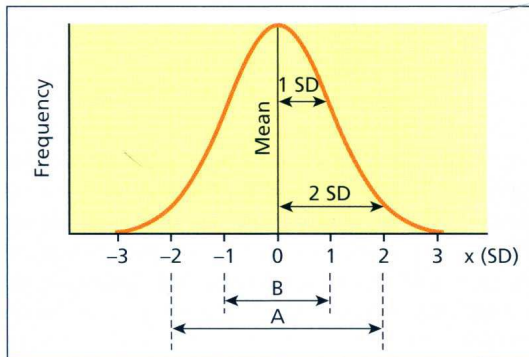


Figure 1.2 Diagram of a Gaussian (normal or symmetrical) distribution curve. The span (A) of the curve, the distance between the mean ± 2 SD, includes about 95% of the 'population'. The narrower span (B), the distance between the mean ± 1 SD, includes about 67% of the 'population'.

- About 67% of results lie in the range mean ± 1 SD.
- About 95% of results lie in the range mean ± 2 SD.
- Over 99% of results lie in the range mean ± 3 SD.

Blunders are grossly inaccurate results that bear no constant or predictable relationship to the true value. They arise, for instance, from mislabelling of specimens at the time of collection, or transcription errors when preparing or issuing reports (see Table 1.1).

If different results for the same test are obtained on two or more occasions on the same patient, then an

important question that arises is whether that difference is due to analytical imprecision or to a true change in the patient's clinical condition. Statistically, if the results differ by more than 2.8 times the analytical SD then there is a chance of over 95% that a genuine change in concentration of the substance has occurred.

Biological causes of variation

Test results also show biological variation in both health and disease. The concentrations of all analytes in blood vary with time due to diverse physiological factors *within* the individual. There are also differences *between* individuals.

The following may be important causes of within-individual variation:

- 1 **Diet:** Variations in diet can affect the results of many tests, including serum triglyceride, the response to glucose tolerance tests and urinary calcium excretion.
- 2 **Time of day:** Several plasma constituents show diurnal variation (variation with the time of day), or a sleep/wake cycle. Examples include iron, adrenocorticotrophic hormone (ACTH) and cortisol concentrations.
- 3 **Posture:** Proteins and all protein-bound constituents of plasma show modest differences in concentration between blood collected from upright individuals and blood from recumbent individuals. Examples include serum calcium, cholesterol, cortisol and total thyroxine concentrations.
- 4 **Muscular exercise:** Recent exercise, especially if vigorous or unaccustomed, may increase serum creatine kinase (CK) activity and blood lactate, and lower blood pyruvate.
- 5 **Menstrual cycle:** Several substances show variation with the phase of the cycle. Examples include serum iron, and the serum concentrations of the pituitary gonadotrophins, ovarian steroids and their metabolites, as well as the amounts of these hormones and their metabolites excreted in the urine.
- 6 **Drugs:** These can have marked effects on chemical results. Attention should be drawn particularly to the many effects of oestrogen-containing oral contraceptives on serum constituents (Chapter 10: Steroid contraceptives).

Even after allowing for known physiological factors that may affect plasma constituents and for analytical imprecision, there is still considerable residual individual



CASE 1.1

A 52-year-old man taking a statin drug to reduce his cholesterol level attended for a routine follow-up appointment. He was well, and had recently started training for a half-marathon as part of his determination to get fitter and reduce his risk of future cardiovascular problems. Statins sometimes cause muscle side effects, so among his other blood tests a creatine kinase (CK) was checked and it was very high. Should his statin be stopped?

Comments: Muscular exertion, especially if unaccustomed or severe, can give rise to high CK results. He was asked to refrain from training for a few days, and on repeat a CK level was normal.

variation (Table 1.7). The magnitude of this variation depends on the analyte, but it may be large and must be taken into account when interpreting successive values from a patient.

Differences between individuals can affect the concentrations of analytes in the blood. The following are the main examples:

- 1 **Age:** Examples include serum phosphate and alkaline phosphatase (ALP) activity, and serum and urinary concentrations of the gonadotrophins and sex hormones.
- 2 **Sex:** Examples include serum creatinine, iron and urate concentrations, and serum and urinary concentrations of the sex hormones.
- 3 **Race:** Racial differences have been described for serum cholesterol and protein. It may be difficult to distinguish racial from environmental factors, such as diet.

Reference ranges

When looking at results, we need to compare each result with a set of results from a particular defined (or reference) population. This reference range is

determined, in practice, by measuring a set of reference values from a sample of that population, usually of healthy individuals. The nature of the reference population should be given whenever reference ranges are quoted, although a healthy population is usually assumed. Even age-matched and sex-matched reference ranges are often difficult to obtain, since fairly large numbers of individuals are needed.

When results of analyses for a reference population are analysed, they are invariably found to cluster around a central value, with a distribution that may be symmetrical (often Gaussian, Figure 1.3a) or asymmetrical (often log-Gaussian, Figure 1.3b). However, reference ranges can be calculated from these data without making any assumptions about the distribution of the data, using nonparametric methods.

Because of geographical, racial and other biological sources of variation between individuals, as well as differences in analytical methods, each laboratory should ideally define and publish its own reference ranges. By convention, these encompass the central 95% of the results obtained for each analysis from the reference population.

Analytical factors can affect the reference ranges for individual laboratories. If an inaccurate method is used, the reference range will reflect the method bias. If an imprecise method is used, the reference range will be widened, that is, the observed span of results (reflected in the SD) will be greater. In statistical terms, the observed variance (i.e. the square of the SD) of the population results will equal the sum of the true or biological variance of the population plus the analytical variance of the method.

How do results vary in disease?

Biochemical test results do not exist in isolation, and when laboratory tests are requested, the clinician will often have made a list of differential diagnoses based on the patient's history, symptoms and signs, and may have a provisional diagnosis that is the likeliest possibility from this list

Table 1.7 Residual individual variation of some serum constituents (expressed as the approximated day-to-day, within-individual coefficient of variation). CV = coefficient of variation.

Serum constituent	CV (%)	Serum constituent	CV (%)
Sodium	1	ALT activity	25
Calcium	1–2	AST activity	25
Potassium	5	Iron	25
Urea	10		

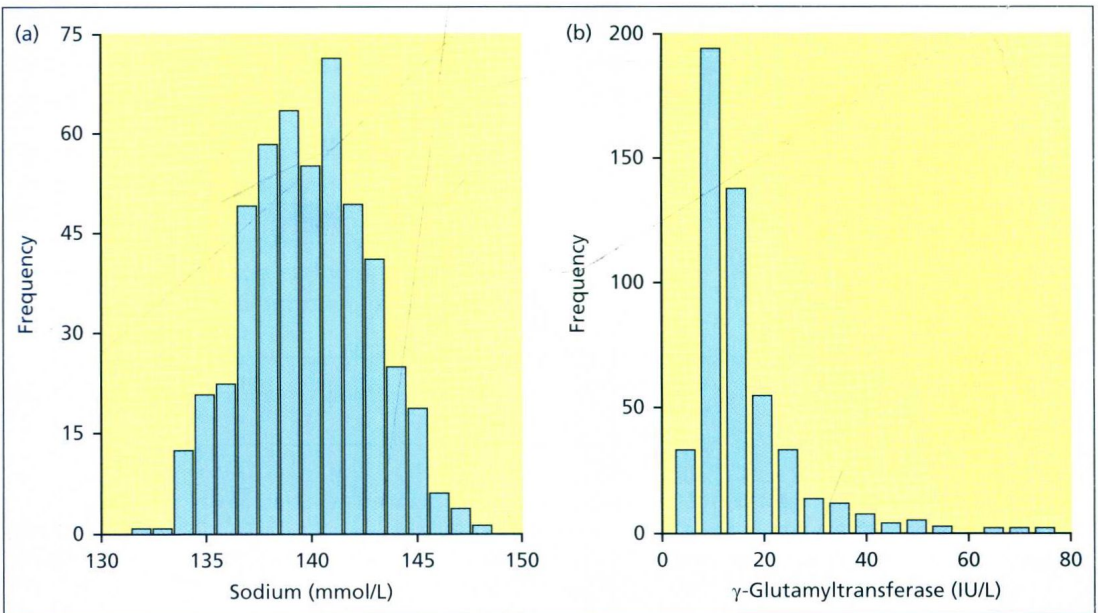


Figure 1.3 Histograms showing the relative frequency with which results with the values indicated were obtained when serum Na^+ and γ -glutamyltransferase (GGT) activities were measured in a reference population of healthy adult women. (a) The sodium data are symmetrically distributed about the mean whereas (b) the GGT data show a log-Gaussian distribution.

For example, in a patient with severe abdominal pain, tenderness and rigidity, there may be several diagnoses to consider including acute pancreatitis, perforated peptic ulcer and acute cholecystitis. In all three conditions, the serum amylase activity may be raised above the upper reference value for healthy adults. We need to know how the serum amylase activity might vary in the clinically likely diagnoses. It would be useful to know, for instance, that very high serum amylase activities are more likely to be associated with one of these diagnostic possibilities (pancreatitis), than with the other two.

To summarise, to interpret results on patients adequately, we need to know:

- the reference range for healthy individuals of the appropriate age range and sex;
- the values to be expected for patients with the disease, or diseases, under consideration;
- the prevalence of the disease, or diseases, in the population to which the patient belongs.

The assessment of diagnostic tests – sensitivity and specificity

In evaluating and interpreting a test, it is necessary to know how it behaves in health and disease. Central to

this understanding are the terms “sensitivity” and “specificity.” These define how well a test performs in the diagnosis of a disease, but in order to calculate them it is necessary to know whether the disease is present or not by some method (this could be some other definitive test, or may be a diagnosis made later once the clinical course has made this more obvious).

- Test sensitivity refers to how effective the test is in detecting individuals who have the disease in question. It is expressed as the percentage of true positives in all the individuals who have disease (all the individuals with disease will encompass the true positives (TP) and false negatives (FN)). So:

$$\text{Sensitivity} = \text{TP} / (\text{TP} + \text{FN}) \times 100\%.$$

- Test specificity is a measure of how good the test is at providing a negative result in the absence of disease. It is expressed as the percentage of true negatives in all those without the disease (all the individuals without disease will encompass the true negatives (TN) and the false positives (FP)). So:

$$\text{Specificity} = \text{TN} / (\text{TN} + \text{FP}) \times 100\%.$$