# MECHANICAL ENGINEERS: HANDBOOK

Volume

Energy and Power

· 4

MYER KUTZ EDITOR

FOURTH EDITION

WILEY

## Mechanical Engineers' Handbook Fourth Edition

# **Energy and Power**

Edited by Myer Kutz Cover Design: Wiley

Cover Image: © denisovd / Thinkstock

This book is printed on acid-free paper. Copyright © 2014 by John Wiley & Sons, Inc. All rights reserved

Published by John Wiley & Sons, Inc., Hoboken, New Jersey Published simultaneously in Canada

No part of this publication may be reproduced, stored in a retrieval system, or transmitted in any form or by any means, electronic, mechanical, photocopying, recording, scanning, or otherwise, except as permitted under Section 107 or 108 of the 1976 United States Copyright Act, without either the prior written permission of the Publisher, or authorization through payment of the appropriate per-copy fee to the Copyright Clearance Center, 222 Rosewood Drive, Danvers, MA 01923, (978) 750-8400, fax (978) 646-8600, or on the web at www.copyright.com. Requests to the Publisher for permission should be addressed to the Permissions Department, John Wiley & Sons, Inc., 111 River Street, Hoboken, NJ 07030, (201) 748-6011, fax (201) 748-6008, or online at www.wiley.com/go/permissions.

Limit of Liability/Disclaimer of Warranty: While the publisher and author have used their best efforts in preparing this book, they make no representations or warranties with the respect to the accuracy or completeness of the contents of this book and specifically disclaim any implied warranties of merchantability or fitness for a particular purpose. No warranty may be created or extended by sales representatives or written sales materials. The advice and strategies contained herein may not be suitable for your situation. You should consult with a professional where appropriate. Neither the publisher nor the author shall be liable for damages arising herefrom.

For general information about our other products and services, please contact our Customer Care Department within the United States at (800) 762-2974, outside the United States at (317) 572-3993 or fax (317) 572-4002.

Wiley publishes in a variety of print and electronic formats and by print-on-demand. Some material included with standard print versions of this book may not be included in e-books or in print-on-demand. If this book refers to media such as a CD or DVD that is not included in the version you purchased, you may download this material at http://booksupport.wiley.com. For more information about Wiley products, visit www.wiley.com.

#### Library of Congress Cataloging-in-Publication Data:

Mechanical engineers handbook : energy and power / edited by Myer Kutz. – Fourth edition.

1 online resource.

Includes index.

Description based on print version record and CIP data provided by publisher; resource not viewed. ISBN 978-1-118-95636-6 (ePub) – ISBN 978-1-118-95637-3 (Adobe PDF) –

ISBN 978-1-118-11899-3 (4-volume set) – ISBN 978-1-118-11285-4 (cloth: volume 4: acid-free paper) 1. Mechanical engineering–Handbooks, manuals, etc. I. Kutz, Myer, editor of compilation.

TJ151

621-dc23

2014005952

Printed in the United States of America

# Mechanical Engineers' Handbook

To Arthur and Bess, Tony and Mary-Ann, for all the good times

此为试读,需要完整PDF请访问: www.ertongbook.com

## **Preface**

The fourth volume of the fourth edition of the *Mechanical Engineers' Handbook* comprises 32 chapters divided into two parts, the first on energy and the second on power. Part 1 begins with a chapter on thermophysical properties of fluids, then proceeds to cover fundamentals of mechanics of incompressible fluids, thermodynamics (including a chapter on exergy and entropy generation minimization), heat transfer, and temperature and heat flux measurements. Additional heat transfer topics in this volume include heat exchangers, heat pipes, air heating, and electronic equipment cooling. There are chapters on refrigeration and cryogenic engineering. One chapter deals with environmental issues: indoor environmental control. A chapter on thermal systems optimization rounds out this part of this volume.

Part 2 opens with a chapter on combustion. This part also includes chapters on conventional energy sources—gaseous and liquid fuels and coal (one chapter on properties of coals, lignite, and peat and a second chapter on clean power generation from coal)—and alternative energy sources—biofuels, solar, geothermal and fuel cells. There are, in addition, chapters on cogeneration and hydrogen energy. There are six chapters on power machinery: one on fans, blowers, compressors, and pumps; one each on gas, wind, and steam turbines; one on internal combustion engines and one on fluid power.

Two chapters—on cryogenic engineering and steam turbines—replace the old versions of the chapters on these important topics. To provide greater emphasis on sustainability than in earlier editions, I have included four chapters—on clean power generation from coal, wind power generation, cogeneration, and hydrogen energy—from my book, *Environmentally Conscious Alternative Energy Production* (chapters updated as contributors found necessary) and one chapter on biofuels from *Environmentally Conscious Transportation*. I have also included three chapters—on temperature, heat flux, and solar energy measurements—from my *Handbook of Measurement in Science and Engineering* and one on mechanics of incompressible fluids from the current edition of *Eshbach's Handbook of Engineering Fundamentals*, which I edited. Inclusion of these chapters enriches this handbook. All told, more than half the chapters in this volume contain material new to this handbook.

## Vision for the Fourth Edition

Basic engineering disciplines are not static, no matter how old and well established they are. The field of mechanical engineering is no exception. Movement within this broadly based discipline is multidimensional. Even the classic subjects, on which the discipline was founded, such as mechanics of materials and heat transfer, keep evolving. Mechanical engineers continue to be heavily involved with disciplines allied to mechanical engineering, such as industrial and manufacturing engineering, which are also constantly evolving. Advances in other major disciplines, such as electrical and electronics engineering, have significant impact on the work of mechanical engineers. New subject areas, such as neural networks, suddenly become all the rage.

In response to this exciting, dynamic atmosphere, the Mechanical Engineers' Handbook expanded dramatically, from one to four volumes for the third edition, published in November 2005. It not only incorporated updates and revisions to chapters in the second edition, published seven years earlier, but also added 24 chapters on entirely new subjects, with updates and revisions to chapters in the Handbook of Materials Selection, published in 2002, as well as to chapters in Instrumentation and Control, edited by Chester Nachtigal and published in 1990, but never updated by him.

The fourth edition retains the four-volume format, but there are several additional major changes. The second part of Volume I is now devoted entirely to topics in engineering mechanics, with the addition of five practical chapters on measurements from the Handbook of Measurement in Science and Engineering, published in 2013, and a chapter from the fifth edition of Eshbach's Handbook of Engineering Fundamentals, published in 2009. Chapters on mechanical design have been moved from Volume I to Volumes II and III. They have been augmented with four chapters (updated as needed) from Environmentally Conscious Mechanical Design, published in 2007. These chapters, together with five chapters (updated as needed, three from Environmentally Conscious Manufacturing, published in 2007, and two from Environmentally Conscious Materials Handling, published in 2009) in the beefed-up manufacturing section of Volume III, give the handbook greater and practical emphasis on the vital issue of sustainability.

Prefaces to the handbook's individual volumes provide further details on chapter additions, updates and replacements. The four volumes of the fourth edition are arranged as follows:

Volume 1: Materials and Engineering Mechanics—27 chapters

Part 1. Materials-15 chapters

Part 2. Engineering Mechanics—12 chapters

Volume 2: Design, Instrumentation and Controls—25 chapters

Part 1. Mechanical Design—14 chapters

Part 2. Instrumentation, Systems, Controls, and MEMS -11 chapters

Volume 3: Manufacturing and Management—28 chapters

Part 1. Manufacturing—16 chapters

Part 2. Management, Finance, Quality, Law, and Research—12 chapters

Volume 4: Energy and Power - 35 chapters

Part 1: Energy—16 chapters Part 2: Power—19 chapters

The mechanical engineering literature is extensive and has been so for a considerable period of time. Many textbooks, reference works, and manuals as well as a substantial number of journals exist. Numerous commercial publishers and professional societies, particularly in the United States and Europe, distribute these materials. The literature grows continuously, as applied mechanical engineering research finds new ways of designing, controlling, measuring, making, and maintaining things, as well as monitoring and evaluating technologies, infrastructures, and systems.

Most professional-level mechanical engineering publications tend to be specialized, directed to the specific needs of particular groups of practitioners. Overall, however, the mechanical engineering audience is broad and multidisciplinary. Practitioners work in a variety of organizations, including institutions of higher learning, design, manufacturing, and consulting firms, as well as federal, state, and local government agencies. A rationale for a general mechanical engineering handbook is that every practitioner, researcher, and bureaucrat cannot be an expert on every topic, especially in so broad and multidisciplinary a field, and may need an authoritative professional summary of a subject with which he or she is not intimately familiar.

Starting with the first edition, published in 1986, my intention has always been that the Mechanical Engineers' Handbook stand at the intersection of textbooks, research papers, and design manuals. For example, I want the handbook to help young engineers move from the college classroom to the professional office and laboratory where they may have to deal with issues and problems in areas they have not studied extensively in school.

With this fourth edition, I have continued to produce a practical reference for the mechanical engineer who is seeking to answer a question, solve a problem, reduce a cost, or improve a system or facility. The handbook is not a research monograph. Its chapters offer design techniques, illustrate successful applications, or provide guidelines to improving performance, life expectancy, effectiveness, or usefulness of parts, assemblies, and systems. The purpose is to show readers what options are available in a particular situation and which option they might choose to solve problems at hand.

The aim of this handbook is to serve as a source of practical advice to readers. I hope that the handbook will be the first information resource a practicing engineer consults when faced with a new problem or opportunity—even before turning to other print sources, even officially sanctioned ones, or to sites on the Internet. In each chapter, the reader should feel that he or she is in the hands of an experienced consultant who is providing sensible advice that can lead to beneficial action and results.

Can a single handbook, even spread out over four volumes, cover this broad, interdisciplinary field? I have designed the Mechanical Engineers' Handbook as if it were serving as a core for an Internet-based information source. Many chapters in the handbook point readers to information sources on the Web dealing with the subjects addressed. Furthermore, where appropriate, enough analytical techniques and data are provided to allow the reader to employ a preliminary approach to solving problems.

The contributors have written, to the extent their backgrounds and capabilities make possible, in a style that reflects practical discussion informed by real-world experience. I would like readers to feel that they are in the presence of experienced teachers and consultants who know about the multiplicity of technical issues that impinge on any topic within mechanical engineering. At the same time, the level is such that students and recent graduates can find the handbook as accessible as experienced engineers.

## **Contributors**

Andrew Alleyne University of Illinois, Urbana–Champaign Urbana, Illinois

Avram Bar-Cohen University of Maryland College Park, Maryland

Prabir Basu Dalhousie University Halifax, Nova Scotia

Adrian Bejan Duke University Durham, North Carolina

Peter D. Blair National Academy of Sciences Washington, DC

James W. Butler Dalhousie University Halifax, Nova Scotia

Jerald A. Caton Texas A&M University College Station, Texas

Peter R. N. Childs Imperial College London, England

Carroll Cone Toledo, Ohio

T. E. Diller Virginia Polytechnic Institute and State University Blacksburg, Virginia

Eric G. Eddings University of Utah Salt Lake City, Utah D. Y. Goswami University of South Florida Tampa, Florida

Cesar Granda Texas A&M University College Station, Texas

Mark Holtzapple Texas A&M University College Station, Texas

Wade W. Huebsch West Virginia University Morgantown, West Virginia

James G. Keppeler Progress Materials, Inc. St. Petersburg, Florida

Allan Kraus Beachwood, Ohio

Peter E. Liley Purdue University West Lafayette, Indiana

Hongbin Ma University of Missouri Columbia, Missouri

Keith Marchildon Queen's University Kingston, Ontario, Canada

Matthew M. Mench University of Tennessee Knoxville, Tennessee and Oak Ridge National Lab Oak Ridge, Tennessee

Harold E. Miller G.E. Energy Schenectady, New York

#### xiv Contributors

David Mody Queen's University Kingston, Ontario, Canada

Tariq Muneer Edinburgh Napier University Edinburgh, Scotland

Todd S. Nemec GE Energy Schenectady, New York

Dennis L. O'Neal Texas A&M University College Station, Texas

Egemen Ol Ogretim West Virginia University Morgantown, West Virginia

Joseph W. Palen Eugene, Oregon

William W. Peng California State University Fresno, California

G. P. Peterson Georgia Institute of Technology Atlanta, GA

Reinhard Radermacher University of Maryland College Park, Maryland Richard J. Reed North American Manufacturing Company Cleveland, Ohio

Aaron Smith Heat Transfer Research, Inc. Navasota, TX

Jelena Srebric University of Maryland College Park, MD

S. S. Srinivasan Florida Polytechnic University Lakeland, FL

E. K. Stefanakos University of South Florida Tampa, Florida

Abhay A. Wative Intel Corp. Chandler, Arizona

Yieng Wei Tham Edinburgh Napier University Edinburgh, Scotland

J. G. Weisend II European Spallation Source Lund, Sweden

Feng-Yuan Zhang University of Tennessee Space Institute Tullahoma, Tennesee

## **Contents**

Preface

Vision for the Fourth Edition

хi

Con	tributors xiii
PA	RT 1 ENERGY
1.	Thermophysical Properties of Fluids 3 Peter E. Liley
2.	Mechanics of Incompressible Fluids 41 Egemen Ol Ogretim and Wade W. Huebsch
3.	Thermodynamics Fundamentals 135 Adrian Bejan
4.	Exergy Analysis, Entropy Generation Minimization, and the Constructal Law Adrian Bejan
5.	Heat Transfer Fundamentals 183 G. P. Peterson
6.	Temperature Measurement 247 Peter R. N. Childs
7.	Heat Flux Measurement 285 T. E. Diller
8.	Furnaces 313 Carroll Cone
9.	Heat Exchangers, Vaporizers, and Condensers 375  Joseph W. Palen
10.	Heat Pipes 415 Hongbin Ma
11.	Air Heating 443 Richard J. Reed
12.	Cooling Electric Equipment 451 Allan Kraus, Avram Bar-Cohen, and Abhay A. Wative
13.	Refrigeration 499 Dennis L. O'Neal
14.	Cryogenic Engineering 543  J. G. Weisend II
15.	Indoor Environmental Control 577  Jelena Srebric
16.	Thermal Systems Optimization 599 Vikrant C. Aute

#### PART 2 POWER

17.	Combustion 635 Eric G. Eddings
18.	Gaseous Fuels 673 Richard J. Reed
19.	Liquid Fossil Fuels from Petroleum 683 Richard J. Reed
20.	Coals, Lignite, and Peat 703  James G. Keppeler
21.	Clean Power Generation from Coal  James W. Butler and Prabir Basu  719
22.	Biofuels for Transportation 767 Aaron Smith, Cesar Granda, and Mark Holtzapple
23.	Solar Energy Measurements 805 Tariq Muneer and Yieng Wei Tham
24.	Geothermal Resources and Technology: Introduction 839 <i>Peter D. Blair</i>
25.	Pumps, Fans, Blowers, and Compressors 857 Keith Marchildon and David Mody
26.	Gas Turbines 901 Harold E. Miller and Todd S. Nemec
27.	Wind Power Generation 955 Todd S. Nemec
28.	Cogeneration 963 Jerald A. Caton
29.	Hydrogen Energy 991 E. K. Stefanakos, D. Y. Goswami, and S. S. Srinivasan
30.	Steam Turbines 1029 William W. Peng
31.	Fuel Cells 1055  Matthew M. Mench and Fens-Yuan Zhano

1089

Index 1123

32. Fluid Power Systems

Andrew Alleyne

# PART 1

# **ENERGY**



## CHAPTER 1

### THERMOPHYSICAL PROPERTIES OF FLUIDS

Peter E. Liley Purdue University West Lafayette, Indiana

- Table 1 Conversion Factors
- Table 2 Phase Transition Data for Elements
- Table 3 Phase Transition Data for Compounds
- Table 4 Thermodynamic Properties of Liquid and Saturated Vapor Air
- Table 5 Ideal Gas Thermophysical Properties of Air
- Table 6 Thermophysical Properties U.S. Standard Atmosphere
- Table 7 Thermophysical Properties of Condensed and Saturated Vapor Carbon Dioxide from 200 K to Critical Point
- Table 8 Thermophysical Properties of Gaseous Carbon Dioxide at 1 Bar Pressure
- Figure 1 Enthalpy-Log Pressure Diagram for Carbon Dioxide
- Table 9 Thermodynamic Properties of Saturated Mercury
- Figure 2 Enthalpy-Log Pressure Diagram for Mercury
- Table 10 Thermodynamic Properties of Saturated Methane
- Table 11 Thermophysical Properties of Methane at Atmospheric Pressure
- Table 12 Thermophysical Properties of Saturated Refrigerant 22
- Table 13 Thermophysical Properties of Refrigerant 22 at Atmospheric Pressure
- Figure 3 Enthalpy-log Pressure Diagram for Refrigerant 22
- Table 14 Thermodynamic Properties of Saturated Refrigerant 134a
- Table 15 Thermophysical Properties of Refrigerant 134a
- Figure 4 Compressibility Factor of Refrigerant 134a
- Figure 5 Enthalpy-Log Pressure Diagram for Refrigerant 134a
- Table 16 Thermodynamic Properties of Saturated Sodium
- Table 17 Thermodynamic Properties of Ice/Water
- Table 18 Thermodynamic Properties of Saturated Steam/Water
- Table 19 Thermophysical Properties of Miscellaneous Substances at Atmospheric Pressure
- Table 20 Physical Properties of Numbered Refrigerants
- Table 21 Specific Heat (kJ/kg · K) at Constant Pressure of Saturated Liquids
- Table 22 Ratio of Principal Specific Heats,  $c_p/c_v$ , for Liquids and Gases at Atmospheric Pressure
- Table 23 Surface Tension (N/m) of Liquids
- Table 24 Thermal Conductivity (W/m · K) of Saturated Liquids
- Table 25 Viscosity (10<sup>-4</sup> Pa·s) of Saturated Liquids
- Table 26 Thermochemical Properties at 1.013 Bars, 298.15 K
- Table 27 Ideal Gas Sensible Enthalpies (kJ/kg · mol) of Common Products of Combustion
- Figure 6 Pscyhometric Chart

#### 4 Thermophysical Properties of Fluids

In this chapter, information is usually presented in the System International des Unités, called in English the International System of Units and abbreviated SI. Various tables of conversion factors from other unit systems into the SI system and vice versa are available. The following table is intended to enable rapid conversion to be made with moderate, that is, five significant figure, accuracy, usually acceptable in most engineering calculations. The references listed should be consulted for more exact conversions and definitions.

Table 1 Conversion Factors

```
Density: 1 \text{ kg/m}^3 = 0.06243 \text{ lb}_m/\text{ft}^3 = 0.01002 \text{ lb}_m/\text{U.K.} gallon = 8.3454 \times 10^{-3} \text{ lb}_m/\text{U.S.} gallon = 1.9403 \times 10^{-3} \text{ lb}_m/\text{U.S.}
    \times 10^{-3} slug/ft<sup>3</sup> = 10^{-3} g/cm<sup>3</sup>
Energy: 1 \text{ kJ} = 737.56 \text{ ft} \cdot 10_{\text{f}} = 239.01 \text{ cal}_{\text{th}} = 0.94783 \text{ Btu} = 3.7251 \times 10^{-4} \text{ hp h} = 2.7778 \times 10^{-4} \text{ kWh}
Specific energy: 1 \text{ kJ/kg} = 334.54 \text{ ft} \cdot 10_f/10_m = 0.4299 \text{ Btu/lb}_m = 0.2388 \text{ cal/g}
Specific energy per degree: 1 \text{ kJ/kg} \cdot \text{K} = 0.23901 \text{ Btu}_{tb}/1b \cdot \text{°F} = 0.23901 \text{ cal}_{tb}/g \cdot \text{°C}
Mass: 1 \text{ kg} = 2.20462 \text{ lb}_m = 0.06852 \text{ slug} = 1.1023 \times 10^{-3} \text{ U.S. ton} = 10^{-3} \text{ tonne} = 9.8421 \times 10^{-4} \text{ U.K.}
    ton
Pressure: 1 bar = 10^5 \text{ N/m}^2 = 10^5 \text{ Pa} = 750.06 \text{ mm} Hg at 0^{\circ}\text{C} = 401.47 \text{ in}. H<sub>2</sub>O at 32^{\circ}\text{F} = 29.530 \text{ in}. Hg
    at 0^{\circ}C = 14.504 lb/in.<sup>2</sup> = 14.504 psia = 1.01972 kg/cm<sup>2</sup> = 0.98692 atm = 0.1 MPa
Temperature: T(K) = T(^{\circ}C) + 273.15 = [T(^{\circ}F) + 459.69]/1.8 = T(^{\circ}R)/1.8
Temperature difference: \Delta T(K) = \Delta T(^{\circ}C) = \Delta T(^{\circ}F)/1.8 = \Delta T(^{\circ}R)/1.8
Thermal conductivity: 1 \text{ W/m} \cdot \text{K} = 0.8604 \text{ kcal/m} \cdot \text{h} \cdot {}^{\circ}\text{C} = 0.5782
    Btu/ft · h · °F = 0.01 W/cm · K = 2.390 \times 10^{-3} cal/cm · s · °C
Thermal diffusivity: 1 \text{ m}^2/\text{s} = 38,750 \text{ ft}^2/\text{h} = 3600 \text{ m}^2/\text{h} = 10.764 \text{ ft}^2/\text{s}
Viscosity, dynamic: 1 \text{ N} \cdot \text{s/m}^2 = 1 \text{ Pa} \cdot \text{s} = 10^7 \ \mu\text{P} = 2419.1 \text{ lb}_m/\text{ft} \cdot \text{h} = 10^3 \text{ cP} = 75.188
    slug/ft \cdot h = 10 P = 0.6720 lb_m/ft \cdot s = 0.02089 lb_f \cdot s/ft^2
Viscosity, kinematic (see thermal diffusivity)
```

Source: E. Lange, L. F. Sokol, and V. Antoine, Information on the Metric System and Related Fields, 6th ed., G. C. Marshall Space Flight Center, AL (exhaustive bibliography); B. N. Taylor, The International System of Units, NBS S.P. 330, Washington, D.C., 2001; E. A. Mechtly, The International System of Units, Physical Constants and Conversion Factors, NASA S.P. 9012, 1973, numerous revisions periodically appear: see, for example, Pure Appl. Chem., 51, 1–41 (1979) and later issues.

Table 2 P	hase Tra	nsition	Data	for E	lementsa.
-----------	----------	---------	------	-------	-----------

Name	Symbol	Formula Weight	$T_m\left(\mathbb{K}\right)$	$\Delta h_{\mathrm{fus}}  (\mathrm{kJ/kg})$	$T_b(K)$	$T_c(K)$
Actinium	Ac	227.028	1323	63	3,475	
Aluminum	Al	26.9815	933.5	398	2,750	7,850
Antimony	Sb	121.75	903.9	163	1,905	5,700
Argon	Ar	39,948	83	30	87.2	151
Arsenic	As	74.9216	885	_		2,100
Barium	Ba	137.33	1,002	55.8	4	4,450
Beryllium	Ве	9.01218	1,560	1,355	2,750	6,200
Bismuth	Bi	208.980	544.6	54.0	1,838	4,450
Boron	В	10.81	2,320	1,933	4,000	3,300
Bromine	Br	159.808	266	66.0	332	584
Cadmium	Cd	112.41	594	55.1	1,040	2,690
Calcium	Ca	40.08	1,112	213.1	1,763	4,300
Carbon	C	12.011	3,810	_	4,275	7,200
Cerium	Ce	140.12	1,072	390		9,750
Cesium	Cs	132.905	301.8	16.4	951	2,015
Chlorine	$\text{Cl}_2$	70.906	172	180.7	239	417

Table 2 (Continued)

Name	Symbol	Formula Weight	$T_m(\mathbf{K})$	$\Delta h_{\rm fus}  ({\rm kJ/kg})$	$T_b(\mathbf{K})$	$T_c(\mathbf{K})$
Chromium	Cr	51.996	2,133	325.6	2,950	5,500
Cobalt	Co	58.9332	1,766	274.7	3,185	6,300
Copper	Cu	63.546	1,357	206.8	2,845	8,280
Dysprosium	Dy	162.50	1,670	68.1	2,855	6,925
Erbium	Er	167.26	1,795	119.1	3,135	7,250
Europium	Eu	151.96	1.092	60.6	1,850	4,350
Fluorine	$F_2$	37.997	53.5	13.4	85.0	144
Gadolinium	Gd	157.25	1,585	63.8	3,540	8,670
Gallium	Ga	69.72	303	80.1	2,500	7,125
Germanium	Ge	72.59	1,211	508.9	3,110	8,900
Gold	Au	196.967	1,337	62.8	3,130	7,250
Hafnium	Hf	178.49	2,485	134.8	4,885	10,400
Helium	He	4.00260	3.5	2.1	4.22	5
Holmium	Но	164.930	1,744	73.8	2,968	7,575
Hydrogen	H <sub>2</sub>	2.0159	14.0	_	20.4	
Indium	In	114.82	430	28.5	2,346	6,150
Iodine	I,	253.809	387	125.0	457	785
Iridium	Ir	192.22	2,718	13.7	4,740	7,800
Iron	Fe	55.847	1,811	247.3	3,136	8,500
Krypton	Kr	83.80	115.8	19.6	119.8	209.
Lanthanum	La	138.906	1,194	44.6	3,715	10,500
Lead	Pb	207.2	601	23.2	2,025	5,500
Lithium	Li	6.941	454	432.2	1,607	3,700
Lutetium	Lu	174.967	1,937	106.6	3,668	
Magnesium	Mg	24.305	922	368.4	1,364	3,850
Manganese	Mn	54.9380	1,518	219.3	2,334	4,325
Mercury	Hg	200.59	234.6	11.4	630	1,720
Molybdenum	Mo	95.94	2,892	290.0	4,900	1,450
Neodymium	Nd	144.24	1,290	49.6	3,341	7,900
Neon	Ne	20.179	24.5	16.4	27.1	44.
Neptunium	Np	237.048	910		4,160	12,000
Nickel	Ni	58.70	1,728	297.6	3,190	8,000
Niobium	Nb	92.9064	2,740	283.7	5,020	12,500
Nitrogen	N <sub>2</sub>	28.013	63.2	25.7	77.3	126.
Osmium	Os	190.2	3,310	150.0	5,300	12,700
Oxygen	O <sub>2</sub>	31.9988	54.4	13.8	90.2	154.
Palladium	Pd	106.4	1,826	165.0	3.240	7,700
Phosphorus	P	30.9738	317	103.0	553	995
Platinum	Pt	195.09	2,045	101	4,100	10,700
Plutonium	Pu	244	913	11.7	3,505	10,500
Potassium	K	39.0983	336.4	60.1	1,032	2,210
Praseodymium	Pr	140.908	1,205	49	3,785	8,900
Promethium	Pm	145	1,353	92	2,730	0,900
Protactinium	Pa	231	1,500	64.8	4,300	
		226.025	973	04.6	1,900	_
Radium	Ra	220.023	202	12.3	211	377
Radon	Rn					18,900
Rhenium	Re	186.207	3,453	177.8	5,920	
Rhodium	Rh	102.906	2,236	209.4	3,980	7,000
Rubidium	Rb	85.4678	312.6	26.4	964	2,070
Ruthenium	Ru	101.07	2,525	256.3	4,430	9,600
Samarium	Sm	150.4	1,345	57.3	2,064	5,050
Scandium	Sc	44.9559	1,813	313.6	3,550	6,410

(continued)