# Bacterial Physiology and Metabolism

J. R. SOKATCH

## Bacterial Physiology and Metabolism

J. R. SOKATCH

Medical Center, University of Oklahoma Oklahoma City, U.S.A.



ACADEMIC PRESS . LONDON AND NEW YORK

## ACADEMIC PRESS INC. (LONDON) LTD.

24-28 Oval Road London, NWT

U.S. Edition published by ACADEMIC PRESS INC. 111 Fifth Avenue New York, New York 10003

Copyright © 1969 By Academic Press Inc. (London) Ltd.

Third Printing 1970

Fourth Printing 1973

Fifth Printing 1974

All Rights Reserved

No part of this book may be reproduced in any form by photostat, microfilm, or any other means, without written permission from the publishers

Library of Congress Catalog Card Number: 68-24692

SBN: 12-654250-3

PRINTID OFFSET PHOTO-LITHOGRAPHY IN GREAT BRITAIN BY THE WHITEFRIARS PRESS LIMITED, LONDON AND LONBRIDGE

### **PREFACE**

This book was written with two purposes in mind: it was intended for use as a textbook for a graduate course in Microbial Biochemistry and as a reference source for workers in fields allied to Microbiology. The emphasis is placed on research done with bacteria with very little space devoted to principles of biochemistry which the student has covered in earlier courses. By eliminating material already covered in books probably owned by the user, the size and cost of this text becomes more reasonable.

A few words about the philosophy used in writing this book are in order. The first section, Bacterial Physiology, deals with non-metabolic aspects of Microbial Biochemistry (nutrition, growth and chemistry) which are important to understanding the features that set bacteria apart from other living organisms. The second section, Energy Metabolism, deals with ways bacteria attack foods that they find in nature, reduce them to a manageable size, ingest and metabolize them to make chemical energy. The third section, Biosynthetic Metabolism. deals with the formation of bacterial protoplasm beginning with simple materials such as inorganic carbon and nitrogen and leading to the formation of the complex molecules of the bacterial cell. Abbreviations used in the text are those recommended in the Instructions to Authors of the Journal of Biological Chemistry. Each enzyme is listed in the text and index by a common name followed by the name given to it by the Enzyme Commission of the International Union of Biochemistry. (The common name used is not necessarily the same one recommended by the Enzyme Commission.)

Many thanks are due to the large number of authors and publishers who graciously permitted portions of their work to be reproduced here. I feel that these figures add a flavor that would be difficult to obtain otherwise and the figures have been reproduced as they were originally published even though there may be some differences in terminology from the rules outlined in the previous paragraph. Thanks are also due to my Chairman who not only permitted but encouraged me to

work on this book and provided the sort of atmosphere that allowed the writing to be done. I would also like to thank the publishers, both in London where the book was published and in New York for their expert technological assistance. Finally, but most important, I would like to thank my wife who helped with typing, proof reading and indexing and provided the expert and dedicated assistance that she did when it was needed most.

November 1968

JOHN R. SOKATCH

## CONTENTS

Preface	V
Part One	
BACTERIAL PHYSIOLOGY	
Chapter 1. Nutrition.	2
I. Requirements for Growth	3
II. Nutritional Classification of Bacteria	3
A. Classification on the Basis of Energy Source B. Classification on the Basis of Carbon Source for Growth	4
References	8
TVOICE CHICOS	0
Chapter 2. Growth of Bacterial Cultures	10
I. Measurement of Bacterial Growth	10
II. Mathematics of Growth	10
III. Growth of Bacteria in Batch Culture	13
A. Lag Phase	13
B. Exponential Growth Phase C. Stationary Phase of Growth	14
IV. Growth of Bacteria in Continuous Culture	1.7
A. Continuous Culture Techniques  B. Mathematics of Growth in Continuous Cultures	17
C. Applications of Continuous Flow Systems	20
V. Synchronous Growth of Bacterial Cultures	20
A. Methods of Obtaining Synchrony of Cell Division	20
B. The Bacterial Cell Cycle	22
References	23
Chapter 3. Chemical Composition of Bacteria	25
I. Proteins	25
II. Lipids	28
A. Simple Lipids	28
B. Complex Lipids	3:

111.	A. Homopolysaccharides B. Heteropolysaccharides C. Complex Polysaccharides	36 36 37 41
IV.	Nucleic Acids.  A. Composition  B. Structure	44 44 46
	References	48
	Part Two	
	ENERGY METABOLISM	
Chapte	r 4. Oligosaccharide Catabolism	55
	Digestion of Starch and Glycogen	55 55
II.	B. Phosphorylases Digestion of Cellulose	59 60
III.	Digestion of Other Polysaccharides by Bacteria	61
	Cleavage of Galactosides.	62
	Cleavage of Glucosides	63 64
	Other Oligosaccharides.	64
	References	65
Chapte	r 5. Transport of Sugars	67
	Transport Systems A. Transport of Galactosides B. Transport of Galactose C. Transport of Other Sugars by Bacteria.	67 67 69 70
II.	Mechanism of Sugar Transport by Bacteria	70
	References	71
	r 6. Fermentation of Sugars	72
	Methods of Study	72
II.	Embden-Meyerhof Fermentations  A. Balances and Isotope Studies.  B. Enzymes of the Embden-Meyerhof Pathway in Bacteria.  C. Formation of Fermentation Products.	74 74 80 87
III.	Non-Embden-Meyerhof Fermentations	92 92 95
IV.	Mixed Pathways  A. Propionic Fermentation by Propionibacterium: Products and Enzymic Reactions.  B. Gluconate Fermentation by Streptococcus faecalis	101
T.	Miscellaneous Pathways	105
	References	

CONTENTS	ix	
OF CARBOHYDRATES	112	

Chapte	r 7. Aerobic Metabolism of Carbohydrates	112
I,	Hexose Oxidation  A. Oxidation by Way of the Embden-Meyerhof Pathway  B. The Oxidative Pentose Cycle.  C. The Entner-Doudoroff Pathway  D. Direct Hexose Oxidation	112 113 116
II.	Hexuronic Acid Oxidation	119
III.	Pentose Oxidation	123
IV.	Polyol Oxidation	123
V.	Miscellaneous Oxidations	127
	References	129
Chapte	er 8. Oxidation of Organic Acids	131
I.	The Tricarboxylic Acid Cycle	131
II.	Metabolism of Other Organic Acids by Bacteria	
	References	
Chapte	er 9. Electron Transport	145
I.	Enzymes of the Cytochrome System	145
II.	Respiratory Chains of Bacteria	149
III.	Oxidative Phosphorylation  A. Phosphorylative Sites.  B. ATP Yield During Oxidative Phosphorylation	152
	References	156
Chapt	er 10. Oxidation of Hydrocarbons.	157
	Oxidation of Saturated Hydrocarbons  A. Terminal Oxidation.  B. Other Pathways of Hydrocarbon Oxidation	$\frac{157}{158}$
	Oxidation of Aromatic Hydrocarbons	
III.	Miscellaneous Oxidations	
	References	163
Chapt	er 11. Protein and Amino Acid Catabolism	165
I.	Digestion of Proteins and Peptides	165
II.	Transport of Peptides and Amino Acids	166
	Metabolism of Amino Acids.  A. Fermentation of Amino Acids B. Oxidation of Amino Acids.	169 169
	Defenence	100

#### CONTENTS

T 12. METABOLISM OF INORGANIC COMPOUNDS	194
Hydrogen Oxidation.	194
Oxidation of Iron	195
Oxidation of Ammonia to Nitrate	196
Reduction of Sulfate; Sulfate Respiration	203
References	205
er 13. Photosynthetic Energy Metabolism	
The Bacterial Photosynthetic Apparatus	
Photosynthesis and Generation of ATP	210
References	213
Part Three	
BIOSYNTHETIC METABOLISM	
er 14. Autotrophic Carbon Dioxide Fixation	217
Production of Reducing Power  A. In Chemoautotrophs B. In Photoautotrophs	217
A. In Algae B. In Chemolithotrophs C. In Pseudomonas oxalaticus Grown on Formate	218 221 223
References	227
er 15. Carbohydrate Biosynthesis	229
From Acetate	229
Formation of Monosaccharides	231
A. Hexoses and Pentoses	231
B. Formation of Hexosamines	232
B. Formation of Hexosamines C. Formation of Deoxysugars	$\frac{232}{234}$
B. Formation of Hexosamines C. Formation of Deoxysugars. D. Hexuronic Acids	232 234 238
B. Formation of Hexosamines C. Formation of Deoxysugars. D. Hexuronic Acids  Formation of Polysaccharides by Bacteria. A. Homopolysaccharides	232 234 238 240 240
B. Formation of Hexosamines C. Formation of Deoxysugars. D. Hexuronic Acids  Formation of Polysaccharides by Bacteria. A. Homopolysaccharides	232 234 238 240 240
B. Formation of Hexosamines C. Formation of Deoxysugars. D. Hexuronic Acids  Formation of Polysaccharides by Bacteria.	232 234 238 240 240 241 242
	Hydrogen Oxidation Oxidation of Iron Oxidation of Ammonia to Nitrate.  Nitrate Respiration A. Reduction of Nitrate, Nitrite and Nitric Oxide. B. Phosphorylation Coupled to Nitrate Respiration Oxidation of Sulfur Compounds. Reduction of Sulfate; Sulfate Respiration. References  13. Photosynthetic Energy Metabolism. Energy Considerations The Bacterial Photosynthetic Apparatus. Photosynthesis and Generation of ATP. References  Part Three BIOSYNTHETIC METABOLISM  14. Autotrophic Carbon Dioxide Fixation.  Production of Reducing Power A. In Chemoautotrophs B. In Photoautotrophs The Reductive Pentose Cycle A. In Algae B. In Chenolithotrophs C. In Pseudomonas oxalaticus Grown on Formate D. In Photosynthetic Bacteria The Reductive Carboxylic Acid Cycle. References  15. Carbohydrate Biosynthesis. From Acetate. Formation of Monosaccharides

CONTENTS	XI
Chapter 16. Biosynthesis of Amino Acids	251
I. Assimilation of Inorganic Nitrogen.	251
A. Nitrate Assimilation  B. Nitrogen Assimilation; Nitrogen Fixation	251
C. Ammonia Assimilation	
II. Biosynthesis of Amino Acids	
A. The Glutamate Family	254
B. The Aspartate Family	258
C. Cysteine D. Histidine	
E. The Branched Chain Amino Acids.	271
F. Alanine	
G. Glycine and Serine	277
H. The Aromatic Amino Acids	
III. Regulation of Metabolic Processes.  A. Feedback Control	286
B. Action of Allosteric Effectors.	287
C. Regulation in Branched Pathways	
References	292
	20=
Chapter 17. Biosynthesis of Lipids	
I. Biosynthesis of Fatty Acids	
A. Straight-chain Fatty Acids B. Branched-chain Fatty Acids	
C. Cyclopropane Fatty Acids	
D. Mycolic Acids	304
E. Unsaturated Fatty Acids	
II. Complex Lipids	307 307
A. Glycolipids B. Phospholipids	309
References	
*	022
Chapter 18. Biosynthesis of Nucleic Acids	313
I. Formation of Purine and Pyrimidine Nucleotides	
A. Purine Nucleotides	313
B. Pyrimidine Nucleotides	
II. Biosynthesis of DNA	331
B. Enzymic Formation of DNA	
III. Biosynthesis of RNA	341
A. RNA Synthesis Primed by DNA	341
B. RNA Synthesis Primed by RNA	
C. Methylation of RNA D. Addition of CCA to sRNA	

#### CONTENTS

III. Translation of the Genetic Code	354
A. Composition and Function of Ribosomes	354
B. Binding of mRNA and sRNA to Ribosomes	355
C. Nucleic Acid Code for Proteins	356
D. Peptide Bond Formation	357
IV. Regulation of Protein Synthesis.	363
A. Transcriptional Control, Jacob and Monod's Operon Theory	
B. Translational Control	366
References	368
AUTHOR INDEX	371
Subject Index	389
~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	

# Bacterial Physiology and Metabolism

J. R. SOKATCH

## Part One

## BACTERIAL PHYSIOLOGY

#### I. NUTRITION

## I. Requirements for Growth

Bacteria as well as all other living organisms require certain essential nutrients in the medium or diet in order to be able to grow. Growth is understood to mean balanced growth, that is, a uniform increase in protoplasm, as opposed to an increase in one or a few components. Essential nutrients fall into two classes, those required to supply energy for growth and those required to supply the chemical elements needed for biosynthesis. Of the various forms of energy available, bacteria can use chemical and light energy for growth. Most true bacteria use organic compounds for chemical energy, but many soil bacteria are able to produce useable energy by the oxidation of inorganic chemicals. Quantitatively, the most important elements required for biosynthesis are those found in protein, namely C, H, O, N and S. These elements may suffice in their inorganic forms or may be required in the form of organic growth factors. Many other elements are required for growth, such as Mg, K, PO<sub>4</sub>, Fe, Cu, Co, Mn and Zn, and these are also used as inorganic salts.

### II. Nutritional Classification of Bacteria

In contrast to the relatively uniform nutritional requirements of plants and animals, bacteria exhibit characteristic differences in their requirements for energy and carbon sources. A study of the distribution of nutritional types in nature suggests that this might be response to environment; for example, all autotrophic bacteria are soil and water species. A system of classification of nutritional types of organisms which was proposed by a group of prominent microbiologists (Lwoff et al., 1946) forms the basis of the following discussion. Organisms were classified on the basis of energy requirement and on the basis of carbon source for biosyntheses.

Developments in microbiology since 1946 have resulted in changes in the original system of classification. For example, the terms photolithotroph and photoorganotroph are not used here as the organisms formerly grouped in these categories differ in their method of assimilation of carbon (a biosynthetic process) during photosynthesis, but not in the process of converting light energy into chemical energy. If any distinction is to be made among the phototrophs, it should be between those that cleave water during photosynthesis (green plants) and those that do not (bacteria).

Similarly, the requirement for growth factors is no longer considered distinctive enough to merit a separate category, since such a requirement could occur as the result of a single genetic change. This means that the original classification on the basis of minimum nutritional requirements becomes a classification on the basis of carbon source. The organisms formerly classified as mesotrophs are now grouped together with the heterotrophs under this heading. (See review by Guirard and Snell, 1962, for nutritional requirements of bacteria.)

#### A. CLASSIFICATION ON THE BASIS OF ENERGY SOURCE

## Utilization of light energy

Organisms which use light energy are phototrophs. This type of energy metabolism occurs only among green plants and certain pigmented bacteria. Both plants and bacteria convert light energy into chemical energy in the form of ATP (Arnon et al., 1954; Frenkel, 1954). Most probably this occurs by a process similar to oxidative phosphory-lation. Excited electrons, generated during the photochemical reaction pass along the cytochrome chain with the concomitant formation of ATP (Stanier, 1961; Fig. 13.4).

Bacterial phototrophs are all soil and water species, most of which are classified in the suborder *Rhodobacteriineae*, with the exception of the genus *Rhodomicrobium*, in the order *Hyphomicrobiales*. The taxonomic distinctions among the *Rhodobacteriineae* are based on the color of the photosynthetic pigments and the preferred type of carbon source. There are three families of *Rhodobacteriineae*, the sulfur purple bacteria, family *Thiorhodaceae*, the non-sulfur purple bacteria, family *Athiorhodaceae*, and the sulfur green bacteria, family *Chlorobacteriaceae*. Sulfur purple bacteria grow well with carbon dioxide as the sole carbon source, in which case they use inorganic sulfur compounds such as hydrogen sulfide and thiosulfate in order to reduce carbon dioxide. They can also assimilate organic carbon, however, in which case no separate reducing agent is required. Non-sulfur purple bacteria grow best with organic carbon, although some species can grow with carbon dioxide as the sole carbon source, using either molecular hydrogen or thiosulfate as the

reducing agent. Rhodomicrobium is similar to the Athiorhodaceae because it requires an organic carbon source and has not yet been grown with carbon dioxide as the sole carbon source. It is separated from the other photosynthetic bacteria on morphological grounds, being a stalked bacterium. Some of the Athiorhodaceae are also able to grow aerobically in the dark on organic energy sources. This is true also of some species of algae but not of higher plants. The sulfur green bacteria reduce carbon dioxide with inorganic sulfur compounds. Unlike the other photosynthetic bacteria, Chlorobacteriaceae are not able to grow with organic carbon as the sole carbon source, although they can assimilate organic carbon to a certain extent (Sadler and Stanier, 1960).

## 2. Utilization of chemical energy

Organisms which use chemical energy for growth are chemotrophs. Chemotrophs are divided into chemolithotrophs, those which use inorganic energy sources, and chemoorganotrophs, those which use organic energy sources.

Bacteria which use chemolithotrophic energy metabolism are soil and water species in the order *Pseudomonadales*, suborder *Pseudomonadineae*. Some of the non-sulfur purple bacteria are also able to grow in the dark on inorganic energy sources such as hydrogen gas and thiosulfate (van Niel, 1944). This ability does not occur widely outside of bacteria, but strains of *Scenedesmus* and other blue-green algae can be obtained which will grow in the dark with hydrogen as the energy source (Gaffron, 1940).

Chemolithotrophic bacteria whose physiology has been studied with pure cultures can be classified into four groups on the basis of energy source used. It is possible that other physiological groups might be discovered since there are many species of bacteria in the Pseudomonadineae which have not been obtained in pure culture. Organisms which oxidize nitrogen compounds are classified in the family Nitrobacteriaceae, and these constitute the first physiological group. Nitrosomonas, which oxidizes ammonia to nitrite, and Nitrobacter, which oxidizes nitrite to nitrate, are the best studied examples of this family. These organisms are strict chemolithotrophs, since they will not grow with organic carbon energy sources. Nitrosomonas and Nitrobacter are important in maintaining soil fertility because they effect an oxidation of ammonia to nitrate, the preferred nitrogen source for green plants.

Hydrogenomonas (family Methanomonadaceae) represents the second group of chemolithotrophs, the hydrogen oxidizers. The ability to oxidize hydrogen occurs frequently among bacteria and occasionally among lower plants such as blue-green algae. Hydrogenomonas is also

able to grow well with organic energy sources in contrast to Nitro-somonas and Nitrobacter.

Sulfur oxidizers comprise the third group of chemolithotrophs, most of which are classified in the family *Thiobacteriaceae*. They are able to grow on inorganic sulfur compounds such as hydrogen sulfide, elemental sulfur, thiosulfate and thiocyanate, and produce tetrathionate and sulfate as end products of the oxidation of these compounds. The species of *Thiobacillus* are the best known sulfur oxidizers. At least one species of *Thiobacillus* can grow on organic media (Santer *et al.*, 1959), but this appears to be an exceptional case.

The fourth group, the iron oxidizers, is represented by Ferrobacillus. Ferrobacillus (family Siderocapsaceae) obtains its energy by the oxidation of ferrous iron. This organism grows best in acid media, 3.5 being the optimum pH for growth (Leathen et al., 1956).

Chemoorganotrophs satisfy their energy requirement by the oxidation or fermentation (anaerobic metabolism) of organic compounds. Chemoorganotrophy is the most common type of energy metabolism among bacteria and almost the only kind found in the animal kingdom. Chemoorganotrophy occurs in the plant kingdom among the non-photosynthetic groups such as yeast and fungi. Some species of algae are facultative chemoorganotrophs, being able to grow on organic carbon sources in the dark (Danforth, 1962).

Taxonomically, chemoorganotrophs are almost all those bacteria that have not been mentioned to this point. Rickettsia are able to oxidize a limited number of organic substrates (Moulder, 1962) and should possibly be considered as chemoorganotrophs, although they have not been grown on lifeless media as yet.

The compounds which chemoorganotrophs use for energy range from simple materials such as formate and oxalate to complex hydrocarbons such as camphor. It seems possible that any organic compound which can be oxidized to produce energy is subject to attack by chemoorganotrophs.

## 3. Energy supplied by metabolism of the host cell

Organisms which obtain energy for biosynthetic reactions from the metabolism of host cells are paratrophs. This category includes bacterial, plant and animal viruses. Although rickettsia are known to have some oxidative ability, it is possible that they may obtain part of their growth energy as paratrophs. Many of the reactions involved in supplying energy to paratrophs are those of the host cell and, from that point of view, will be covered in this book. The subject of viral replication as a separate topic will not, however, be treated here.