Chemotaxonomy
of
Flowering
Plants
Volume II

R. Darnley Gibbs



# CHEMOTAXONOMY OF FLOWERING PLANTS

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VOLUME II
FAMILIES



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## JUNE, 1974 NOTES, QUERIES AND ERRATA

Professor G. Ourisson has very kindly pointed out two errors in the figure included in the promotional leaflet for this book. I am very grateful to him since his action has prompted me to recheck virtually all the formulae in the 189 figures of the text. This hasty check has revealed what is to me a shocking number of errors (mostly of a minor nature), as well as many cases in which I am unsure of the formulae. I have therefore prepared the following list, which I hope is substantially complete.

It is but a slight consolation to report the detection of several errors in the reference books available to me, and to note that *chemists* (I am a *botanist*) seem as liable to get plant names wrong as I am to err in my formulae.

- Fig. 146 (p. 686) Opuntiol —CH2OH, not —OH2OH
- Fig. 149 (p. 694) Jacareubin Should have —OH at 1. Maclura-xanthone should have —OH at 1. Have I the correct chain at top? Mangiferin —O missing
- Fig. 150 (p. 698) Embelin Add —OH at 5. Plastoquinone, not plastaquinone
- Fig. 154 (p. 711) Cryptotanshinone Should have 2 —CH<sub>3</sub> groups at top of left-hand ring?
- Fig. 159 (p. 723) Cycloartenol Should have 2 —CH<sub>3</sub> groups (not 1) at bottom left (as in butyrospermol)? Elemolic acid —COOH, not —CH<sub>3</sub>, at α
- Fig. 161 (p. 745) Oleandrigenin, not Oldeandrigenin. *Periplogenin*—CH<sub>3</sub> omitted (compare other genins)
- Fig. 162 (p. 747) Digacetigenin I find a formula somewhat different from mine
- Fig. 166 (p. 766) Sinalbin Is the non-sinapine part of my formula correct?
- Fig. 169 (p. 790) 1,8-Cineole —O— should link to 8 position (not 4)
- Fig. 173 (p. 797) Mansonone-A Should there be a second double bond (common to the rings)?
- Fig. 174 (p. 798) Drimenol —CH<sub>3</sub> missing (compare with other formulae). Cinnamodial Is an —OH group omitted from top of right-hand ring? Farnesiferol-C —CH<sub>3</sub> is missing from top right of left-hand ring?

Fig. 175 (p. 801) Petasin Double bond in ring at left?

Fig. 177 (p. 806) Mexicanin-C Should have =CH<sub>2</sub>, not —CH<sub>3</sub>, in ring at right?

Fig. 180 (p. 812) α-Santonin Is a double bond missing from ring at right? Artemisin Is an —OH missing?

ig. 182 (p. 819) Dodonaea-diterpene Should this be?

c-Camphorene Lower end should be -CH=C(CH<sub>3</sub>)<sub>2</sub> (as top end)

Fig. 183 (p. 843) Bacogenin-A A formula proposed in 1973 transposes the —CH<sub>3</sub> and —CH<sub>2</sub>OH groups (middle level of my fig.) and has a terminal 5-membered heterocyclic ring

Panaxatriol The two —CH<sub>3</sub> groups at right should be on the C below (next to the O)

Fig. 184 (p. 856) Samaderine-B Should be?

Swietenine Should the chain at left be -CH2.CO.OCH3 or -CH(OH).CO.OCH3?

Fig. 185 (p. 857) Arnidiol = CH<sub>2</sub>, not = O, at top? Limonin Should double bond in ring at left be deleted? Entandrophragmin I am unsure of this formula. Quassin Is my formula correct?

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#### **MELANINS**

#### GENERAL

The name 'melanin' has been variously used. It has been employed for any dark-brown or black substance (or mixture) occurring in plant or animal. Because there is abundant evidence that a group of substances so named would be chemically and biosynthetically unnatural, attempts have been made to use the term *melanins* only for those dark-coloured polymeric indole derivatives that involve tyrosine, dihydroxy-phenylalanine (DOPA), and dopamine, and tyrosinase and oxygen.

We may cite first of all the brief review by Thomas (1955), who points out that the 'melanins' of higher plants *may* include indole derivatives—though no such melanin had at that time been isolated and analysed—and other non-nitrogenous 'melanins' such as the *phytomelanes* of the *Compositae*, Japanese lac, and many oxidation products of phenols.

The true N-containing melanins, he says, may be variable. Some may contain sulfur, and some (all?) may exist as *melanoproteins*.

The animal (true) melanins are insoluble in hot strong acids, but more or less soluble in alkalis. He gives, as a possible unit of such melanins, the dimer of fig. 144.

In 1958 Thomas has a further article on *melanin* (he uses the singular form). He says that the first real clue as to the nature of melanin came from the work of Bertrand and Bourquelot (1894–6) who noted that in *Rhus* spp., the sources of lac, there seems to exist the system:

while in other plants one may have:

Thomson (1965) says that true melanins are probably produced by the legumes *Vicia faba*, *Cytisus nigricans* and *Sarothamnus scoparius*, and the banana. He gives a melanin unit (fig. 144) that differs slightly from that of Thomas.

Andrews and Pridham (1967) investigated the 'melanins' of some higher plants which have DOPA ( $\beta$ -(3,4-dihydroxy-phenyl)-L-alanine) or dopamine ( $\beta$ -(3,4-dihydroxy-phenyl) ethylamine), and compared them with enzyme-synthesized 'melanins'. They got the following results for nitrogen content, etc.

Astragalus cicer (pod), N 1.2% Baptisia australis (pod), N 1.6%

Thomas (1955)

Thomson (1965)

Fig. 144. Possible melanin units.

Lupinus polyphyllus (pod), N 1·3%; alkali fusion gave catechol, protocatechnic acid and 5,6-dihydroxy-indole.

Vicia angustifolia (pod), N 1.4%; alkali fusion as above

V. faba (fl.), N 2.1%; alkali fusion as above

V. faba (pod), N 1.3%

Musa sp. (epicarp), N 1.5%

Tyrosine-melanin (synthesized with phenolase), N 7.1%

Dopamine-melanin (synthesized with phenolase), N 6.8%

They concluded: 'The melanins from plants which contain DOPA and related compounds have been examined and shown to be largely composed of the catechol-type pigment. Some indole units also appear to be present, however.'

The book Melanins by Nicolaus (1968) contains little more about the

melanins of higher plants.

The co-called *phytomelanes* of the *Compositae*, which blacken fruitwalls, and sometimes other parts of many species, seem not to contain nitrogen, or to have very little of it, and are not true *melanins*. Hegnauer (1964) lists work, largely by Hanausek, on the distribution of these substances in the tribes of the *Compositae*:

Vernonieae absent (5 genera)

Eupatorieae present in fruits of many (19?) genera.

Astereae absent (26 genera)

Inuleae rare; but present in Caesulia axillaris (frt), Sphaeranthus sp., Ammobium sp., Inula helenium (rhiz., rt).

Heliantheae present in spp. of 56 genera

Helenieae present in Jaumeinae, Heleniinae and Tagetinae; but absent from Riddelliinae.

Anthemideae absent from spp. of 10 genera.

Senecioneae rare; absent from spp. of 20 genera; but present in Arnica.

Calenduleae absent from spp. of 3 genera.

Arctoteae absent from spp. of 8 genera:

Cardueae (Cynareae) rare; absent from Brotera, Carlina, Centaurea, Cirsium, Cynara, Galactites, Onopordon, Serratula, Silybum (but see below), and Xeranthemum; but present in Echinops, Carthamus, and Silybum marianum (fruits of some).

Mutisieae rare; present in Perezia (rhiz., rt); but not in fruits of spp. of Dicoma, Gerbera, Leuceria, and Moscharia.

Cichorieae absent from spp. of 24 genera.

Much information on blackening of plants or their extracts is scattered through this book under *cigarette* and *hot-w ster tests*, the *aucubin-type glycosides*, *irritant plants* (*urushiol*, etc.).

#### NAPHTHALENE AND SOME OF ITS DERIVATIVES

#### GENERAL

Again we have a dilemma. Naphthalene itself (fig. 145) is a hydrocarbon and might be discussed with other members of that 'group'. Its derivatives include aldehydes, alcohols, lactones and quinones. We have sections for naphthaquinones (p. 699) and for lactones (p. 670) elsewhere. Some derivatives, such as those of Ulmus, are really sesquiterpenes, and are considered with them (p. 796).

It seems desirable to deal with *naphthalene* itself and yet other of its derivatives here.

According to Ruwet (1966) the cotyledons of Impatiens balsamina have a naphtholglycosidase which hydrolyses 1,2,4-trihydroxy-naphthalene-4-glucoside (fig. 145). The free 1,2,4-trihydroxy-naphthalene is then auto-oxidized to 2-hydroxy-1,4-naphthaquinone (lawsone; fig. 145). Ruvet says that the same or a similar enzyme from leaves of Juglans regia may be responsible for the formation of juglone.

#### List and Occurrence

- 4,5-Dihydroxy-2-methyl-naphthalene (fig. 145): see also diospyrol. Eben. Diospyros mollis (frt)
- 4,5-Dimethoxy-6-hydroxy-2-methyl-naphthalene (Macassar-II; fig. 145)

  Eben. Diospyros celebica (htwd—'macassar ebony')
- 4,5-Dimethoxy-6-hydroxy-2-naphthaldehyde

Eben. Diospyros ebenum (htwd)

Diospyrol (fig. 145) is a dimer of 4,5-dihydroxy-2-methyl-naphthalene. Eben. Diospyros mollis (frt)

Fig. 145. Naphthalene and some derivatives.

Musizin (3-Acetyl-4,5-dihydroxy-2-methyl-naphthalene; fig. 145)

. Rhamn. Maesopsis eminii (htwd-'musizi')

Lili. Dianella laevis

Naphthalene (fig. 145) is said to occur in a lichen and in

Myrt. Eugenia caryophyllata (cloves)

Comp. Saussurea lappa (rt-oil)

Irid. Iris germanica

Gram. Oryza sativa (sdlg, with methyl- and p-dimethyl-naphthalene) 1,2,4-Trihydroxy-naphthalene-4-glucoside (fig. 145) may be the precursor of lawsone (fig. 145, and see above).

Balsamin. Impatiens balsamina (cotyledons)

1,4,5-Trihydroxy-naphthalene-4-glucoside may be the precursor of juglone (fig. 151 and p. 70).

4,5,6-Trimethoxy-2-methyl-naphthalene (Macassar-III)

Eben. Diospyros celebica (htwd)

4,5,6-Trimethoxy-2-naphthaldehyde

Eben. Diospyros ebenum (htwd)

#### **PYRONES**

#### GENERAL

We may distinguish  $\alpha$ -pyrones, of which  $\alpha$ -pyrone (coumalin) itself (fig. 146) may be considered to be the 'parent'; and  $\gamma$ -pyrones, of which  $\gamma$ -pyrone (fig. 147) may be considered to be the parent.

#### I α-PYRONES

#### GENERAL

Only a few simple  $\alpha$ -pyrones are known. These include the kawa-pyrones of Piper. They are lactones and some of them are named as such.

The benzo- $\alpha$ -pyrones include the coumarins, isocoumarins, furocoumarins, etc. They are dealt with in a separate section (p. 440).

Some \alpha-pyrones are phthalides and some are derivatives of naphthalene.

#### List and Occurrence

Aparajitine (fig. 146) is the δ-lactone of 2-methyl-4-hydroxy-pentacosanoic acid.

Legum. Clitoria maritima, ternata

Demethoxy-yangonin

Piper. Piper methysticum

Dihydro-kawain (Marindinin)

Piper. Piper methysticum (st., rt)

Dihydro-methysticin (Pseudo-methysticin)

Piper. Piper methysticum (rt)

Gentiopicroside (Erytaurine; Gentiamarin; Gentiopicrin; Sabbatin; Swertiamarin; fig. 146) has, says Paris (1963), the formula shown. It yields mesogentiogenin and glucose.

Gentian. Chlora (1), Cicendia (1), Erythraea (1), Gentiana (20), Pleurogyna (1), Sabbatia (at least 1), Swertia (3).

Hyptolide is like massoilactone.

Lab. Hyptis pectinata

Kawain (fig. 146)

Piper. Piper methysticum (rt)

Massoilactone (fig. 146)

Laur. Cryptocarya (Massoia) aromatica (bk-oil)

4-Methoxy-paracotoin

Laur. Aniba fragrans

Methysticin (fig. 146)

Piper. Piper methysticum (rt)

Mevalolactone (fig. 146) may be placed here. The biologically active form is R-(-)-mevalolactone.

Opuntiol (fig. 146)

Cact. Opuntia elatior

Paracotoin (fig. 146) is said to occur in Bolivian 'coto-bark' and 'Paracoto-bark'.

H<sub>3</sub>CO  
HOH<sub>2</sub>O O O H<sub>11</sub>C<sub>5</sub>O O H<sub>41</sub>C<sub>2</sub>O O O  

$$\leftarrow$$
 Pyrone (Coumalin) Opuntiol Massoilactone Aparajitine (Coumalin) Paracotoin lactone -coumalin OCH<sub>3</sub>  $\leftarrow$  OCH<sub>3</sub>

Fig. 146. Some α-pyrones.

Parasorbic acid

Ros. Sorbus aucuparia (unripe frt)

6-Phenyl-coumalin (fig. 146) is said to occur in 'true' coto-bark. Yangonin

Piper. Piper methysticum (rt)

### II y-PYRONES

#### GENERAL

We may recognize three groups of  $\gamma$ -pyrones:

- 1. Simple γ-Pyrones.
- 2. Chromones, derivatives of benzo-γ-pyrone (fig. 148). See also flavonoids.
- 3. Xanthones, derivatives of dibenzo-\gamma-pyrone (xanthone; fig. 149).

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- 6-phenyl-Y-pyrone

Fig. 147. Simple γ-pyrones.

## II.1 Simple γ-Pyrones

#### GENERAL

Only a few simple  $\gamma$ -pyrones seem to have been recorded from higher plants. One of them—chelidonic acid—is, however, very widely distributed.

#### List and Occurrence

Chelidonic acid (fig. 147) was found in more than half of the more than 1100 species examined by Ramstad (1953). Kwasniewsky (1953) also lists many plants as containing the acid. We list only a few examples to show how widely spread it is.

Papaver. Chelidonium majus, Stylophorum diphyllum (rt, bound to an alkaloid)

Berberid. Berberis vulgaris

Campanul. Lobelia inflata and other spp.

Rubi. Uragoga (Cephaelis)

Lili. Asparagus (plt, frt), Colchicum, Convallaria (lvs), Gloriosa (lvs), Polygonatum (2), Schoenocaulon (sd), Veratrum (2)

Comanic acid (fig. 147)

Occurrence?

2,3-Dihydro-3-methyl-6-phenyl-γ-pyrone (fig. 147)

Myrt. Myrtus bullata (ess. oil)

Maltol (Laricinic acid; fig. 147) occurs in Larix and in

Papaver. Corydalis ochotensis

Meconic acid (fig. 147): belongs here? I find two formulae for it.

Papaver. Papaver dubium, rhoeas, somniferum

Pyromeconic acid (fig. 147)

Comp. Erigeron annuus (lvs, fl.)

# II.2 Benzo-γ-pyrones (chromones)

#### GENERAL

Many of the benzo-y-pyrones have other rings too. Thus some furo-chromones are known. Other substances which might be included here, such as deguelin, elliptone, and rotenone, are essentially isoflavanone derivatives and are treated with them (p. 609). Some of the other groups of flavonoids are also chromone or near-chromone derivatives.

The occurrence of a unique group of benzo- $\gamma$ -pyrones in Ptaeroxylon and Cedrelopsis is of interest. These genera are removed from the Meliaceae by some taxonomists and placed in a little family Ptaeroxylaceae (q.v.). I know of no benzo- $\gamma$ -pyrone in other members of the Meliaceae.

#### List and Occurrence

Ammiol (2-Oxymethyl-5,8-dimethoxy-furo-4',5',6,7-chromone)

Umbell. Ammi visnaga (sd)

Angustifolionol (fig. 148)

Myrt. Backhousia angustifolia (ess. oil)

Chellol (fig. 148) is the aglycone of khellinin (chellol-2-glucoside). Does it occur free?

Eugenin (fig. 148)

Myrt. Eugenia aromatica, caryophyllata ('cloves')

Eugenitin (6-Methyl-eugenin)

Myrt. Eugenia caryophyllata

Heteropeucenin (fig. 148)

Meli. Ptaeroxylon obliquum (htwd)

Heteropeucenin-dimethyl ether

Meli. Ptaeroxylum obliquum

Heteropeucenin-7-methyl ether

Meli. Ptaeroxylon obliquum

Fig. 148. Some benzo-y-pyrones (chromones).

Isoeugenitin

Myrt. Eugenia caryophyllata

Isoeugenitol

Myrt. Eugenia caryophyllata

Karenin

Meli. Ptaeroxylon obliquum (htwd)

Khellin (Kellin; Visammin)

Umbell. Ammi visnaga (sd)

Khellinin (Chellol-2-glucoside)

Umbell. Ammi visnaga (sd)

Khellinol (5-Norkhellin)

Umbell. Ammi visnaga (sd)

Peucenin (fig. 148)

Meli. Ptaeroxylon obliquum (htwd)

Umbell. Peucedanum ostruthium (rhiz.)

Ptaerochromenol

Meli. Ptaeroxylon obliquum (htwd)

Ptaerocyclin

Meli. Ptaeroxylon obliquum (htwd)

Ptaeroglycol

Meli. Ptaeroxylon obliquum (htwd)

Ptaeroxylin (Deoxy-karenin; fig. 148)

Meli. Cedrelopsis grevei, Ptaeroxylon obliquum (htwd)

Ptaeroxylinol

Meli. Ptaeroxylon obliquum (htwd)

Ptaeroxylone

Meli. Ptaeroxylon obliquum

Sorbifolin

Rut. Spathelia sorbifolia (rt)

# II.3 Dibenzo-γ-Pyrones (Xanthones)

#### GENERAL

These plant constituents are derivatives of dibenzo- $\gamma$ -pyrone (xanthone; fig. 149). Karrer (1958) says that only a few xanthones have been found in plants but I have records of about 70, many of them from a paper by Gottlieb (1968). A useful review is that by Roberts (1961). Mostly they are free, but some at least occur also as glycosides. They are found in fungi, in lichens, and in at least one fern. In the higher plants they may occur in all parts. All are yellow to red-yellow in colour.

Xanthone itself may not occur in plants, at least I have no record of it. The naturally occurring xanthones are hydroxy-, methoxy-, and other derivatives of xanthone.

We have records of them from the following dicotyledonous families: Anacardiaceae (a few), Flacourtiaceae, Gentianaceae (several), Guttiferae (many: see discussion under that family), Hippocrateaceae, Leguminosae, Moraceae, Polygalaceae and Sapotaceae. From the monocotyledons we have: Liliaceae and Iridaceae.

#### List and Occurrence

Alvaxanthone

Mor. Maclura pomifera

Bellidifolin (1,5,8-Trihydroxy-3-methoxy-xanthone; fig. 149)

Gentian. Gentiana bellidifolia

Celebixanthone (3,4,8-Trihydroxy-2-methoxy-1-(3-methyl-2-butenyl)-xanthone)

Gutt. Cratoxylon celebicum

Corymbiferin (4,5-Di-O-methyl-corymbin)

Gentian. Gentiana bellidifolia (rt), corymbifera (as glycoside)

Decussatin (8-Hydroxy-1,3,7-trimethoxy-xanthone)

Gentian. Swertia decussata

Dehydrocyclo-guanandin

Gutt. Calophyllum brasiliense

6-Dehydroxy-jacareubin

Gutt. Calophyllum brasiliense, scriblitifolium (htwd); Kielmeyera ferruginea (bk), speciosa

Demethyl-bellidifolin (Demethyl-swertianol)

Gentian. Gentiana bellidifolia, Swertia tosaensis

6-Deoxy-jacarubin

Gutt. Calophyllum inophyllum (htwd), scriblitifolium (htwd); Kielmeyera speciosa

Deoxy-morellin

Gutt. Garcinia morella

Dihydro-isomorellin

Gutt. Garcinia morella

1,3-Dihydroxy-5-methoxy-xanthone

Gutt. Calophyllum brasiliense
1,7-Dihydroxy-8-methoxy-xanthone

Gutt. Kielmeyera excelsa, ferruginea (bk), petiolaris

5,6-Dihydroxy-7-methoxy-xanthone

Gutt. Kielmeyera corymbosa 6,7-Dihydroxy-8-methoxy-xanthone

Gutt. Kielmeyera speciosa

1,5-Dihydroxy-xanthone (fig. 149)

Gutt. Mammea americana, Mesua ferrea

4,5-Dimethoxy-bellidin (4,7-Di-O-methyl-bellidin)

Gentian. Gentiana bellidifolia (rt)

4,7-Dimethoxy-bellidin

Gentian. Gentiana bellidifolia (rt)

2-(3,3-Dimethylallyl)-1,3,5,6-tetrahydroxy-xanthone *Gutt. Calophyllum inophyllum* (htwd)

2-(3,3-Dimethylallyl)-1,3,5-trihydroxy-xanthone

Gutt. Calophyllum scriblitifolium (htwd)

2-(3,3-Dimethylallyl)-1,3,7-trihydroxy-xanthone Gutt. Calophyllum scriblitifolium (htwd)

Euxanthone (Purrenone; 1,7-Dihydroxy-xanthone) seems to be more widely spread than most xanthones.

Anacardi. Mangifera indica

Gutt. Calophyllum sclerophyllum, Kielmeyera excelsa, Mammea americana, Mesua ferrea, Platonia insignis, Symphonia globulifera Gambogic acid

Gutt. Garcinia hanburyi, morella

Gentioside (Gentiin; Isogentisin-3-primeveroside?)

Gentian. Gentiana lutea

Gentisin (Gentianin (1); 1,7-Dihydroxy-3-methoxy-xanthone; fig. 149)

Gutt. Calophyllum brasiliense

Gentian. Gentiana lutea, Swertia japonica

Globuxanthone

Gutt. Symphonia globulifera

Guanandin

Gutt. Calophyllum brasiliense

1-Hydroxy-3,7-dimethoxy-xanthone

Gutt. Calophyllum brasiliense

1-Hydroxy-7,8-dimethoxy-xanthone

Gutt. Kielmeyera petiolaris

5-Hydroxy-1,3-dimethoxy-xanthone

Gutt. Kielmėyera coriacea, corymbosa, ferruginea (bk), speciosa

5-Hydroxy-6,7-dimethoxy-xanthone

Gutt. Kielmeyera coriacea, corymbosa, ferruginea (bk), rupestris (wd), speciosa

6-Hydroxy-5,7-dimethoxy-xanthone

Gutt. Kielmeyera speciosa

6-Hydroxy-7,8-dimethoxy-xanthone

Gutt. Kielmeyera rupestris (wd), speciosa

1-Hydroxy-7-methoxy-xanthone

Gutt. Kielmeyera corymbosa, excelsa; Mesua ferrea

7-Hydroxy-8-methoxy-xanthone

Gutt. Kielmeyera excelsa, speciosa

5-Hydroxy-6,7-methylenedioxy-xanthone

Gutt. Kielmeyera corymbosa, speciosa

3-Hydroxy-1,5,6-trimethoxy-xanthone

Gutt. Kielmeyera rupestris

5-Hydroxy-xanthone (fig. 149)

Gutt. Calophyllum brasiliense, Mammea americana

7-Hydroxy-xanthone

Gutt. Kielmeyera excelsa, speciosa; Mammea americana

Isobellidifolin (1,3,8-Trihydroxy-5-methoxy-xanthone)

Gentian. Gentiana bellidifolia

Isogentisin (1,3-Dihydroxy-7-methoxy-xanthone)

Gentian. Gentiana lutea

Isoguanandin

Gutt. Calophyllum brasiliense

Isomorellic acid

Gutt. Garcinia morella