Second Edition

THE PICTURE BOOK OF QUANTUM MECHANICS

图解量子力学

第2版

S. Brandt H. D. Dahmen

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OUANTUM MECHANICS

With 486 Illustrations

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Siegmund Brandt Physics Department Siegen University D-57068 Siegen Germany Hans Dieter Dahmen Physics Department Siegen University D-57068 Siegen Germany

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To Renate Brandt and Ute Dahmen

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Foreword to the Second Edition

In the present edition aim and style of the *Picture Book* were left unchanged. It is our aim to explain and exemplify the concepts and results of quantum mechanics by visualization through computer graphics and, in parallel, by the discussion of the relevant physical laws and mathematical formulae. The scope of the book, however, was widened appreciably.

The most important extension is the chapter about spin and magnetic resonance. In three-dimensional quantum mechanics the presentation of wave-packet motion on elliptical and hyperbolic Kepler orbits should help to establish the correspondence and the differences between the classical and the quantum-mechanical description of planetary motion.

Maybe the best known quantum-mechanical concept is that of uncertainty introduced by Heisenberg who formulated the celebrated uncertainty relation $\Delta x \Delta p \geq \hbar/2$ for the product of the uncertainties of a particle in position and in momentum. Also in classical mechanics position and momentum of a particle may be known only up to some uncertainty so that a probability density in phase space (spanned by position x and momentum p) is needed to describe the particle. For many physical situations we present and compare the time development of this classical phase-space probability density with the probability density of quantum mechanics.

An additional tool we use is the *analyzing amplitude* introduced in Appendix C. It allows the definition of a directional distribution which is very helpful in visualizing angular momentum and spin wave functions.

To generate the computer graphics of the first edition of the *Picture Book* we developed an interactive program on quantum mechanics. A modernized version which we call INTERQUANTA (abbreviated IQ) together with an accompanying text has been published by Springer-Verlag in various editions. ¹⁻⁴ It is a pleasure to acknowledge the generous help provided by IBM Germany in the development of IQ. In particular we want to thank Dr. U. Groh for his competent help in the early phase of the work. At various stages of the project we were helped considerably by friends and students in Siegen. We would particularly like to thank Tilo Stroh for his many valuable contributions.

Foreword to the Second Edition

All computer-drawn figures in the present edition were produced using the published version of IQ or recent extensions realized with the help of Sergei Boris, Anli Shundi, and Tilo Stroh. The computer typesetting and the layout of the text were done by Ute Bender and Anli Shundi. We would like to thank them for their excellent work.

Last but not least, we thank Drs. H.-U. Daniel, T. von Foerster, and H. J. Kölsch of Springer-Verlag for their constant interest and support.

Siegen, May 1994

Siegmund Brandt Hans Dieter Dahmen

¹S. Brandt and H. D. Dahmen, *Quantum Mechanics on the Personal Computer*, 3rd ed., Springer-Verlag, Berlin, Heidelberg, New York, 1994.

²S. Brandt and H. D. Dahmen, *Quantum Mechanics on the Macintosh*, Springer-Verlag, New York, Berlin, Heidelberg, 1991.

³S. Brandt and H. D. Dahmen, *Pasocon de Manebu Ryoushi Nikigacu*, Springer-Verlag, Tokyo, 1992.

⁴S. Brandt and H. D. Dahmen, *Quantenmechanik auf dem Personalcomputer*, Springer-Verlag, Berlin, Heidelberg, New York, 1993.

Preface

Students of classical mechanics can rely on a wealth of experience from everyday life to help them understand and apply mechanical concepts. Even though a stone is not a mass point, the experience of throwing stones certainly helps them to understand and analyze the trajectory of a mass point in a gravitational field. Moreover, students can solve many mechanical problems on the basis of Newton's laws and, in doing so, gain additional experience. When studying wave optics, they find that their knowledge of water waves, as well as experiments in a ripple tank, are very helpful in forming an intuition about the typical wave phenomena of interference and diffraction.

In quantum mechanics, however, beginners are without any intuition. Because quantum-mechanical phenomena happen on an atomic or a subatomic scale, we have no experience of them in daily life. The experiments in atomic physics involve more or less complicated apparatus and are by no means simple to interpret. Even if students are able to take Schrödinger's equation for granted, as many students do Newton's laws, it is not easy for them to acquire experience in quantum mechanics through the solution of problems. Only very few problems can be treated without a computer. Moreover, when solutions in closed form are known, their complicated structure and the special mathematical functions, which students are usually encountering for the first time, constitute severe obstacles to developing a heuristic comprehension. The most difficult hurdle, however, is the formulation of a problem in quantummechanical language, for the concepts are completely different from those of classical mechanics. In fact, the concepts and equations of quantum mechanics in Schrödinger's formulation are much closer to those of optics than to those of mechanics. Moreover, the quantities that we are interested in - such as transition probabilities, cross sections, and so on - usually have nothing to do with mechanical concepts such as the position, momentum, or trajectory of a particle. Nevertheless, actual insight into a process is a prerequisite for understanding its quantum-mechanical description and interpreting basic properties in quantum mechanics like position, linear and angular momentum, as well as cross sections, lifetimes, and so on.

Preface

Actually, students must develop an intuition of how the concepts of classical mechanics are altered and supplemented by the arguments of optics in order to acquire a roughly correct picture of quantum mechanics. In particular, the time evolution of microscopic physical systems has to be studied to establish how it corresponds to classical mechanics. Here computers and computer graphics offer incredible help, for they produce a large number of examples which are very detailed and which can be looked at in any phase of their time development. For instance, the study of wave packets in motion, which is practically impossible without the help of a computer, reveals the limited validity of intuition drawn from classical mechanics and gives us insight into phenomena like the tunnel effect and resonances, which, because of the importance of interference, can be understood only through optical analogies. A variety of systems in different situations can be simulated on the computer and made accessible by different types of computer graphics.

Some of the topics covered are

- scattering of wave packets and stationary waves in one dimension,
- the tunnel effect,
- · decay of metastable states,
- bound states in various potentials,
- · energy bands,
- distinguishable and indistinguishable particles,
- angular momentum,
- three-dimensional scattering,
- cross sections and scattering amplitudes,
- eigenstates in three-dimensional potentials, for example, in the hydrogen atom, partial waves and resonances,
- · motion of wave packets in three dimensions,
- spin and magnetic resonance.

As conceptual tools bridging between classical and quantum concepts serve

- the phase-space probability density of statistical mechanics,
- the Wigner phase-space distribution,

 the absolute square of the analyzing amplitude as probability or probability density.

The graphical aids comprise

- time evolutions of wave functions for one-dimensional problems,
- parameter dependences for studying, for example, the scattering over a range of energies,
- three-dimensional surface plots for presenting two-particle wave functions or functions of two variables,
- polar (antenna) diagrams in two and three dimensions,
- plots of contour lines or contour surfaces, i.e., constant function values, in two and three dimensions.
- ripple-tank pictures to illustrate three-dimensional scattering.

Whenever possible, how particles of a system would behave according to classical mechanics has been indicated by their positions or trajectories. In passing, the special functions typical for quantum mechanics, such as Legendre, Hermite, and Laguerre polynomials, spherical harmonics, and spherical Bessel functions, are also shown in sets of pictures.

The text presents the principal ideas of wave mechanics. The introductory Chapter 1 lays the groundwork by discussing the particle aspect of light, using the fundamental experimental findings of the photoelectric and Compton effects and the wave aspect of particles as it is demonstrated by the diffraction of electrons. The theoretical ideas abstracted from these experiments are introduced in Chapter 2 by studying the behavior of wave packets of light as they propagate through space and as they are reflected or refracted by glass plates. The photon is introduced as a wave packet of light containing a quantum of energy.

To indicate how material particles are analogous to the photon, Chapter 3 introduces them as wave packets of de Broglie waves. The ability of de Broglie waves to describe the mechanics of a particle is explained through a detailed discussion of group velocity, Heisenberg's uncertainty principle, and Born's probability interpretation. The Schrödinger equation is found to be the equation of motion.

Chapters 4 through 8 are devoted to the one-dimensional quantum-mechanical systems. Study of the scattering of a particle by a potential helps us understand how it moves under the influence of a force and how the probability interpretation operates to explain the simultaneous effects of transmission and reflection. We study the tunnel effect of a particle and the excitation and decay

of a metastable state. A careful transition to a stationary bound state is carried out. Quasi-classical motion of wave packets confined to the potential range is also examined.

Chapters 7 and 8 cover two-particle systems. Coupled harmonic oscillators are used to illustrate the concept of indistinguishable particles. The striking differences between systems composed of different particles, systems of identical bosons, and systems of identical fermions obeying the Pauli principle are demonstrated.

Three-dimensional quantum mechanics is the subject of Chapters 9 through 14. We begin with a detailed study of angular momentum and discuss methods of solving the Schrödinger equation. The scattering of plane waves is investigated by introducing partial-wave decomposition and the concepts of differential cross sections, scattering amplitudes, and phase shifts. Resonance scattering, which is the subject of many fields of physics research, is studied in detail in Chapter 13. Bound states in three dimensions are dealt with in Chapter 12. The hydrogen atom and the motion of wave packets under the action of a harmonic force as well as the Kepler motion on elliptical orbits are among the topics covered. Chapter 14 is devoted to Coulomb scattering in terms of stationary wave functions as well as wave-packet motion on hyperbolical orbits.

Spin is treated in Chapter 15. After the introduction of spin states and operators the Pauli equation is used for the description of the precession of a magnetic moment in a homogeneous magnetic field. The discussion of Rabi's magnetic resonance concludes this chapter.

The last chapter is devoted to results obtained through experiments in atomic, molecular, solid-state, nuclear, and particle physics. They can be qualitatively understood with the help of the pictures and the discussion in the body of the book. Thus examples for

- typical scattering phenomena,
- spectra of bound states and their classifications with the help of models,
- · resonance phenomena in total cross sections,
- phase-shift analyses of scattering and Regge classification of resonances,
- radioactivity as decay of metastable states,
- magnetic resonance phenomena,

taken from the fields of atomic and subatomic physics, are presented. Comparing these experimental results with the computer-drawn pictures of the book and their interpretation gives the reader a glimpse of the vast fields of science that can be understood only on the basis of quantum mechanics.

In Appendix A the simplest aspects of the structure of quantum mechanics are discussed and the matrix formulation in an infinite-dimensional vector space is juxtaposed to the more conventional formulation in terms of wave functions and differential operators. Appendix B gives a short account of two-level systems which is helpful for the discussion of spin. In Appendix C we introduce the analyzing amplitude using as examples the free particle and the harmonic oscillator. Appendix D discusses Wigner's phase-space distribution. Appendixes E through G give short accounts of the gamma, Bessel, and Airy functions, as well as the Poisson distribution.

There are more than a hundred problems at the ends of the chapters. Many are designed to help students extract the physics from the pictures. Others will give them practice in handling the theoretical concepts. On the endpapers of the book are a list of frequently used symbols, a short list of physical constants, and a brief table converting SI units to particle-physics units. The constants and units will make numerical calculations easier.

We are particularly grateful to Professor Eugen Merzbacher for his kind interest in our project and for many valuable suggestions he gave before the publication of the first edition and which helped to improve the book.

Siegmund Brandt Hans Dieter Dahmen

Frequently Used Symbols

a	Bohr radius	Н	Hamiltonian
$a(x_0, p_0, x_S, p_S)$	analyzing amplitude	j	probability current density
A_1, A_{11}, \ldots	amplitude factors in regions	$j_{\ell}(ho)$	spherical Bessel function
B_1, B_{11}, \ldots	I, II, of space	k	wave number
В	magnetic-induction field	l	angular-momentum
c	speed of light		quantum number
c	correlation coefficient	L_n^{α}	Laguerre polynomials
$d_{mm'}^{\ell}$	Wigner function	L	angular momentum
$D_{mm'}^{\ell}$	Wigner function	Ĺ	angular-momentum operator
$\mathrm{d}\sigma/\mathrm{d}\Omega$	differential scattering	m	quantum number
	cross section		of z component
<i>e</i>	elementary charge	17	of angular momentum
E_{c}	complex electric field strength	m, M	mass
E	energy	M(x,t)	amplitude function
E_{n}	energy eigenvalue	M	magnetization
$E_{ m kin}$	kinetic energy	n	refractive index
E	electric field strength	n	principal quantum number
f(k)	spectral function with respect to	$n_{\ell}(ho)$	spherical Neumann function
	wave number	n	unit vector
f(p)	spectral function with respect to	p	momentum
6/ 0)	momentum	\hat{p}	momentum operator
$f(\vartheta)$	scattering amplitude	$\langle p \rangle$	momentum expectation value
f_t	partial scattering amplitude	p	momentum vector
$f_{\ell m}(\varTheta, \Phi)$	directional distribution	p	vector operator of momentum
g	gyromagnetic factor	P_{ℓ}	Legendre polynomial
$h = h/(2\pi)$	Planck's constant	P_{ℓ}^{m}	associated Legendre function
		r	relative coordinate
$h_{\ell}^{(+)}(ho)$	spherical Hankel function of the first kind	τ	radial distance
$h_t^{(-)}(\rho)$		r	position vector
ι ι (<i>p</i>)	spherical Hankel function of the second kind	- R	center-of-mass coordinate
			Senser of mass cool dinate

$R_{n\ell}$ radial eigenfunction ϑ , Θ polar angle s spin quantum number ϑ scattering angle $S = (S_1, S_2, S_3)$ spin-vector operator λ wavelength S_ℓ scattering-matrix element μ_0 vacuum permeability t time μ reduced mass T oscillation period μ magnetic moment T transmission coefficient ρ probability density T kinetic energy ρ^{cl} classical phase-space probability T transition-matrix elements density density U voltage σ_0 width of ground state of harmonic oscillator U voltage σ_0 width of ground state of harmonic oscillator V potential (energy) σ_1 σ_2 pull matrices V potential (energy) σ_1 partial cross section V potential (energy) σ_2 width in wave number V potential (energy) σ_2 width in momentum V V V	$R(r), R_{\ell}(k)$	(r) radial wave function	$\eta_\ell(\mathbf{r})$	scattered partial wave
$S = (S_1, S_2, S_3)$ spin-vector operator λ wavelength S_ℓ scattering-matrix element μ_0 vacuum permeability t time μ reduced mass T oscillation period μ magnetic moment T transmission coefficient ρ probability density T kinetic energy $\rho^{\rm cl}$ classical phase-space probability T transmission-matrix elements T densition-matrix elements T voltage T width of ground state of harmonic oscillator T phase velocity T phase velocity T potential (energy) T width in wave number T effective potential T potential (energy) T width in momentum T with in momentum T stationary wave function T stationary wave function T stationary wave function T stationary wave function T station number T and T magnetic susceptibility T and T momentum uncertainty T wave number uncertainty T state vector T angular frequency solid angle T momentum uncertainty T solid angle T nabla (or del) operator	$R_{n\ell}$	radial eigenfunction	ϑ, ⊖	polar angle
S_ℓ scattering-matrix element μ_0 vacuum permeability t time μ reduced mass T oscillation period μ magnetic moment T transmission coefficient ρ probability density T kinetic energy ρ^c classical phase-space probability T transition-matrix elements σ width of ground state U voltage σ_0 width of ground state v_0 group velocity σ_1 σ_2 σ_3 v_0 phase velocity σ_1 σ_2 width in matrices V potential (energy) σ_k width in wave number V potential (energy) σ_k width in momentum W wigner distribution σ_2 width in momentum W Wigner distribution σ_5 width in position W wigner distribution σ_5 stationary wave function W wigner distribution σ_5 stationary wave function W W W W	<i>s</i>	spin quantum number	v	scattering angle
t time μ reduced mass T oscillation period μ magnetic moment T transmission coefficient ρ probability density T kinetic energy ρ^{c1} classical phase-space probability T transition-matrix elements ρ classical phase-space probability density T transition-matrix elements ρ width of ground state of harmonic oscillator ρ phase velocity ρ phase velocity ρ potential (energy) ρ width in wave number ρ potential (energy) ρ width in wave number ρ width in momentum ρ width in momentum ρ width in momentum ρ width in momentum ρ width in position ρ width in position ρ stationary wave function ρ stationary wave function ρ stationary wave function ρ state vector of stationary state ρ and ρ scattering phase shift ρ scattering phase shift ρ width in energy ρ state vector of stationary state ρ scattering phase shift ρ wave-number uncertainty ρ state vector ρ	$S = (S_1, S_2)$, S ₃) spin-vector operator	λ	wavelength
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T transmission coefficient ρ probability density T kinetic energy ρ^{cl} classical phase-space probability T, T_R transition-matrix elements density U voltage σ_0 width of ground state v_0 group velocity σ_1 , σ_2 , σ_3 Pauli matrices v_0 phase velocity σ_1 , σ_2 , σ_3 Pauli matrices v_0 potential (energy) σ_k width in wave number v_0 effective potential σ_ℓ partial cross section v_0 wight in momentum v_0 average energy density σ_p width in momentum v_0 wight in momentum v_0 wight in position v_0 total cross section v_0 stationary wave function v_0 stationary wave function v_0 position v_0 stationary wave function v_0 position v_0 state vector of stationary state v_0 position expectation value v_0 state vector of stationary state v_0 fine-structure constant v_0 magnetic susceptibility v_0 fine-structure constant v_0 state vector of station v_0 state vector of station v_0 state vector of the properties of the state vector v_0 state vec	t	time	μ	reduced mass
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The strategy of the contents of the state o	T	transmission coefficient	ρ	probability density
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δ_{ℓ} scattering phase shift $\psi(x,t)$ time-dependent wave function Δk wave-number uncertainty $\psi_{\mathbf{p}}(\mathbf{r},t)$ harmonic wave function Δp momentum uncertainty ψ state vector Δx position uncertainty ω angular frequency ε_0 vacuum permittivity Ω solid angle ∇ nabla (or del) operator Ω	\boldsymbol{Z}	atomic number	-	•
scattering phase shift $\psi_p(\mathbf{r},t)$ harmonic wave function Δp momentum uncertainty ψ state vector Δz position uncertainty ω angular frequency ε_0 vacuum permittivity Ω solid angle ∇ nabla (or del) operator ∇^2	α	fine-structure constant	,-	•
Δp momentum uncertainty ψ state vector Δx position uncertainty ω angular frequency ε_0 vacuum permittivity Ω solid angle ∇ nabla (or del) operator η_1, η_{-1} spin $\frac{1}{2}$ base states ∇^2	δ_{ℓ}	scattering phase shift	• • •	_
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ϵ_0 vacuum permittivity ∇ nabla (or del) operator η_1, η_{-1} spin $\frac{1}{2}$ base states ∇^2	Δx	position uncertainty	_	•
η_1, η_{-1} spin $\frac{1}{2}$ base states ∇^2	ϵ_0	vacuum permittivity		•
$\eta_{\mathbf{k}}(\mathbf{r})$ scattered wave	η_1, η_{-1}	spin $\frac{1}{2}$ base states		•
	$\eta_{\mathbf{k}}(\mathbf{r})$	scattered wave	v -	Laplace operator

Basic Equations

$$\psi_{\mathbf{p}}(\mathbf{r},t) = \frac{1}{(2\pi\hbar)^{3/2}} \exp\left(-\frac{\mathrm{i}}{\hbar}Et\right) \exp\left(\frac{\mathrm{i}}{\hbar}\mathbf{p}\cdot\mathbf{r}\right)$$

$$i\hbar \frac{\partial}{\partial t}\psi(\mathbf{r},t) = \left[-\frac{\hbar^2}{2M}\nabla^2 + V(\mathbf{r})\right]\psi(\mathbf{r},t)$$

$$\left[-\frac{\hbar^2}{2M}\nabla^2 + V(\mathbf{r})\right]\varphi_E(\mathbf{r}) = E\varphi_E(\mathbf{r})$$

$$\hat{\mathbf{p}} = (\hat{p}_x, \hat{p}_y, \hat{p}_z) = \frac{\hbar}{i} \left(\frac{\partial}{\partial x}, \frac{\partial}{\partial y}, \frac{\partial}{\partial z} \right) = \frac{\hbar}{i} \nabla$$

angular-momentum operators

$$\hat{\mathbf{L}} = \mathbf{r} \times \hat{\mathbf{p}} = \frac{\hbar}{:} \, \mathbf{r} \times \boldsymbol{\nabla}$$

radial Schrödinger equation for spherically symmetric potential

$$-\frac{\hbar^2}{2M}\left[\frac{1}{r}\frac{\mathrm{d}^2}{\mathrm{d}r^2}r-\frac{\ell(\ell+1)}{r^2}-\frac{2M}{\hbar^2}V(r)\right]R_\ell(k,r)=ER_\ell(k,r)$$

stationary scattering wave

$$\varphi_{\mathbf{k}}^{(+)}(\mathbf{r}) \xrightarrow{kr \gg 1} e^{i\mathbf{k}\cdot\mathbf{r}} + f(\vartheta) \frac{e^{ikr}}{r}$$

$$f(\vartheta) = \frac{1}{k} \sum_{\ell=0}^{\infty} (2\ell + 1) f_{\ell}(k) P_{\ell}(\cos \vartheta)$$

$$\frac{\mathrm{d}\sigma}{\mathrm{d}\Omega} = |f(\vartheta)|^2 , \quad \sigma_{\ell} = \frac{4\pi}{k^2} (2\ell + 1) |f_{\ell}(k)|^2 , \quad \sigma_{\text{tot}} = \sum_{k=0}^{\infty} \sigma_{\ell}$$

Physical Constants

Planck's constant
$$h = 4.136 \cdot 10^{-15} \,\text{eV} \cdot \text{s} = 6.626 \cdot 10^{-34} \,\text{J} \cdot \text{s}$$

$$h = h/(2\pi) = 6.582 \cdot 10^{-16} \,\text{eV} \cdot \text{s}$$

= 1.055 \cdot 10^{-34} \,\text{J} \cdot \\\ \text{s}

speed of light
$$c = 2.998 \cdot 10^8 \, \text{ms}^{-1}$$

elementary charge
$$e = 1.602 \cdot 10^{-19} \,\mathrm{C}$$

fine-structure constant
$$\alpha = \frac{e^2}{4\pi\varepsilon_0\hbar c} = \frac{1}{137.036}$$

electron mass
$$m_e = 0.5110 \,\text{MeV}/c^2 = 9.110 \cdot 10^{-31} \,\text{kg}$$

proton mass
$$m_p = 938.3 \,\text{MeV}/c^2 = 1.673 \cdot 10^{-27} \,\text{kg}$$

neutron mass
$$m_n = 939.6 \,\mathrm{MeV}/c^2 = 1.675 \cdot 10^{-27} \,\mathrm{kg}$$

Conversion Factors

mass
$$1 \text{ kg} = 5.609 \cdot 10^{35} \text{ eV}/c^2$$
, $1 \text{ eV}/c^2 = 1.783 \cdot 10^{-36} \text{ kg}$

energy
$$1 J = 6.241 \cdot 10^{18} \,\text{eV}$$
, $1 \,\text{eV} = 1.602 \cdot 10^{-19} \,\text{J}$

momentum
$$1 \text{ kg} \cdot \text{ms}^{-1} = 1.871 \cdot 10^{27} \text{ eV/}c$$
, $1 \text{ eV/}c = 5.345 \cdot 10^{-28} \text{ kg} \cdot \text{ms}^{-1}$

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