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Physical Chemistry of Solid State Materials

- REMCES VI -

**Proceedings of the 6th International Seminar on
the Physical Chemistry of Solid State Materials,
El Jadida, Morocco, 1993**



Editors:

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K. Benhouja
A. Boukhari
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Physical Chemistry of Solid State Materials

PREFACE :

REMCES (the acronym is formed from the French title of the Moroccan Forum for Solid State Chemistry) meets every two years to discuss the latest advances in the physical chemistry of solid state materials. Numbers of participants have grown from **50 participants** at the first meeting, held in **Rabat** in **1983**, to **300** at the sixth meeting which took place in **EI Jadida** from October 28 to 30, **1993**.

This sixth meeting, entitled International Seminar of Physical Chemistry of Solid State Materials, brought together researchers from Algeria, France, Germany, Japan, Spain, Switzerland and Tunisia, as well as from Morocco itself.

The aims of the seminar were to discuss the progress of scientific research, strengthen ties between industry and universities, and promote cooperation in the field of solid state chemistry at national and international level. The topics presented covered methods of preparing solid state materials, their physical properties and industrial applications.

The coordinating committee



Group photo

ACKNOWLEDGEMENTS

The organizing committee wishes to thank the following :

- **The Minister for National Education** for his interest in this meeting, for his encouragement and support, and for the honour he conferred on us by chairing the opening session of the meeting;
- **The Governor** of the Province of El Jadida for the facilities he generously accorded us ;
- **The Rector of Chouaïb Doukkali University, El Jadida**, for his help and encouragement during the preparation of the meeting ;
- **The Dean of the Faculty of Science of Chouaïb Doukkali University, El Jadida**, for his interest in the meeting and his encouragement throughout the period of preparation.
- **The Municipal Council of the Town of El Jadida**, for contributing materially to the organisation of this meeting.

We thank the following warmly for their support and financial assistance :

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- The Rectorate of Hassan II University, Aïn Chock, Casablanca
- The Rectorate of Cadi Ayad University, Marrakesh
- The Rectorate of Moulay Ismail University, Meknes
- The Faculties of Science at the following Universities :
 - Hassan II University, Aïn Chock, Casablanca
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 - Moulay Ismaïl University, Meknes
- Ecole Supérieure de Technologie, Casablanca
- Centre National de Coordination et de Planification de la Recherche Scientifique et Technique.
- ORMVA Doukkala

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- ERAMEDIC

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CONTENTS

Preface	v
Acknowledgements	vii
Guest Editors and Committees	ix
Light-Converting Molecular Devices: Energy-Transfer Processes in Lanthanide Dinuclear Complexes	
J.-C.G. Bünzli, P. Froidevaux and C. Piguet	1
Electrochemical Oxygen Intercalation: A New Chimie Douce Route to High Oxidation States of Transition Metal Oxides	
M. Pouchard, A. Demourgues, F. Weill, A. Villesuzanne, A. Wattiaux and J.C. Grenier	11
Ceramic Fibers from Organosilicon Precursors	
R. Naslain, R. Pailler, G. Chollon, R. Bodet and H. Hannache	19
Thermal Shock and Fatigue Behaviour of Ceramics	
G. Fantozzi, D. Olagnon and M. Saadaoui	35
Densification of Yttria Ceramics	
J.-F. Baumard, G. Gasgnier, H. Vesteghem, M. Boncoeur and M. Bougoin	47
Dielectric Properties in Nickel-Doped Barium-Strontium Titanate Ceramics	
T. Fukami	55
Structural Relationships and Magnetic Behaviour in R_2BaMO_5 Oxides (R=Rare Earth; M=Co, Ni and Cu)	
R. Saez-Puche and J. Hernandez-Velasco	65
Photoluminescent Metaphosphates Activated by Monovalent Silver	
M. Mesnaoui, C. Parent, B. Tanguy, M. Maazaz, G. Le Flem, 7B. Moine and C. Pedrini	83
Electronic Structure of LaF_9^{6-}, YF_9^{6-} and EuF_9^{6-} Clusters Calculated by the MS-LSD Method	
I. Gérard, H. Chermette and J.C. Krupa	89
Infrared to Visible Upconversion of Tm^{3+} Doped BIGaZYbTZr Glasses	
B. Jacquier, R. Mahiou, J.L. Adam and J. Lucas	101
Vitrification of Phosphates of Nasicon-Type Structure	
A. Eljazouli	105

Intercalation of Organic and Inorganic Anions into Anionic Clays	115
A. De Roy, C. Forano and J-P. Besse	
Phase Transitions in V₂O₅ Intercalation Compounds	129
H. Katzke and W. Depmeier	
Relaxation Study in Y₂O₃ after Polarization Induced by an Electron Beam	137
G. Moya, A. Sami and D. Moya-Siesse	
Precipitation and Formation Mechanism of Type AB Carbonate Apatites Analogous to Dental Enamel	147
A. Barroug, C. Rey and J.C. Trombe	
Removing Cadmium from Phosphate Ores by Pyrometallurgical Process	155
A. Nounah, M. Ferhat and J.L. Lacout	
Optimization of a Hydraulic Calcium Phosphate Cement	163
E. Mejdoubi, J.L. Lacout, J.C. Heughebaert and P. Michaud	
A New Nasicon-Like Phosphate Co_{0.5}Ti₂(PO₄)₃	173
A. El Bouari, A. El Jazouli, J.M. Dance, G. Le Flem and R. Olazcuaga	
The Structure of the High Temperature Forms of Cu_{0.5}^{II}Ti₂(PO₄)₃ and D_{0.5}Cu^I_{0.5}Zr₂(PO₄)₃	177
R. Olazcuaga, G. Le Flem, A. Boireau and J. L. Soubeyroux	
Synthesis and Biocompatibility of New Materials	189
M. Draoui, M. Imbenotte, G. Palavit, L. Montagne and L. Benarafa	
Investigations of the Acid Properties of the Chabasite-Type Silico-aluminophosphate SAPO-34 Synthesized with Morpholine as the Organic Template	195
M. Achache, H. Kessler and A. Azzouz	
Synthesis and Crystal Chemistry of the Rare Earth Phosphate: HoPO₄,xKOH (x≤1)	201
M. Zaki, A. Aamili, M. El Ghazzi, D. Zambon, M. Zahir, A. Sadel and J.C. Cousseins	
Crystal Structures of the Isotypical Tetrametaphosphates: Cd_{0.50}Co_{1.50}P₄O₁₂ and Cd_{0.37}Ni_{1.63}P₄O₁₂	209
A. Alaoui El Belghiti, A. El Marzouki, A. Boukhari and E.M. Holt	
Preparation and Characterisation of the Compounds MFePO₅ (M=Ni, Cu)	215
M. Touaiher and A. El Hajbi	

New Compounds in the Bi₂O₃-MO-P₂O₅ (M=Cu, Ni) Systems at 800°C	
F. Abraham, M. Ketatni and B. Mernari	223
Polymorphism, Structural and Physical Properties of Pb₃(VO₄)₂ Related Solid Solutions	
A. Mizrahi, M. Drache, P. Conflant, J.P. Wignacourt, A. Lorriaux-Rubbens and F. Wallart	233
Air Jet Micronisation and Reactivity of Micronised Powders	
M. Ouammou, L. Clerc, A. Benhassaine and A. Nadiri	243
Influence of Testing Temperature on the Weibull Modulus of Post-Hipped α-SiC	
F. Osterstock and A. Charif	249
Damage Accumulation in a Unidirectional SiC-MAS.L Composite Material Tested under Uniaxial Tensile Loading	
M. Drissi-Habti and M.Gomina	257
The Wear Characteristics of Ceramic/Ceramic and Ceramic/Metal Pairs	
J. Takadoum, Z. Zsiga and C. Roques-Carmes	267
Synthesis and Characterization of Siliconated Polymers Fluorinated or not, from Hydroxylterminated Liquid Polybutadienes	
B. Ameduri, B. Boutevin, M. Nouiri and M. Talbi	283
Vitrification of the Systems Water-Propanediol 1,2	
A. Nakheli, N. Lotfi and J. Huck	287
Vitrification and Crystallization of the Systems Propylene Carbonate-Toluene	
N. Lofti, A. Nakheli and J. Huck	293
Synthesis Characterization and Protonic Conduction in Mixed Caesium-Ammonium Hydrogenosulphate Cs_{0.95}(NH₄)_{0.05}HSO₄	
H. Khemakhem, T. Mhiri, Z. Fakhfakh and A. Daoud	299
Low Temperature Modulated Phase in the Two Dimensional Compound (C₃H₇NH₃)₂PbCl₄	
Y. Abid, M.Kamoun, A.Daoud and F.Romain	309
Synthesis and Crystal Structure of Bis (3,5-Dicarboxypyrazole)-Bis (Aqua) Copper(II) Dihydrate	
B. Mernari, F. Abraham and M. Lagrenée	317
Electrolytic Codeposition of Gold and Synthetic Diamond Particles	
M. Bendahmane, J. Ginies, M. Tachez, M.-P. Gigandet and J. Picaut	325

Effect of Slightly Solubles Solutes on the Recrystallization of Rolled Silver

F. Salhi, J. Aride, J.Bernardini and G. Moya

333

Observation of the Atomic Structures of (111) $\Sigma 3$ Boundaries in the Intermetallics Ni₃Al (L1₂) and TiAl (L1₀)

C. Derder, R. Bonnet and J.M. Penisson

337

High Temperature XRD and DTA Studies of Sr₃Cu₅O₈, Bi_{1.6}Pb_{0.4}Sr₂CaCu₂O_x, and Sr_{14-y}Ca_yCu₂₄O_z Compositions

H. Faqir, G. Vacquier, P. Conflant and A. Casalot

345

Thermogravimetric Analysis of the Removal of Organic Binders and Plasticizers from an Alumina Matrix

M. El Morabit, J.P. Millet, M. Murat and G. Fantozzi

355

Defects Induced in Silicon by Rapid Thermal Processing

B. Hartiti, H. Amzil, D. Sayah, J.C. Muller and P. Siffert

361

Metakaolinite Behaviour in Aqueous Suspension, the Relationship between Behaviour and Structure

M. Chamoumi, E. Garcia-Diaz, A. Crespy and A. Benhassaine

369

Experimental and Theoretical Approaches to the Internal Stress Field in an Annealed Nickel Base Superalloy

R. Bonnet and A. Ati

375

Vacancy Defects Recovery of Quenched Ni₂Si Intermetallic Compound

A. Jennane, E.H. Sayouti, J. Bernardini, P. Moser and G. Moya

385

Effects of Annealing and Ion Bombardment on Structure and Composition of Ti-Al Alloy Films

M. Fahoume, M. El Khamlchi, C. Baltzinger and C. Burggraf

391

Influence of Thermal Annealing Atmosphere on Electrical Properties of Polycrystalline Silicon Grain Boundaries

N. M'Gafad, H. Amzil, B.M. Semega and D. Sayah

399

Thermal Non Destructive Detection of Defect in Multilayer Wall

S. Belattar, A. Elouahlouli and B. Duthoit

405

Structural Approach of Glasses Belonging to the Li₂O-Li₂WO₄-P₂O₅ Ternary System

A. Nadiri, A. Yacoubi, L. Bih, M. Haddad and A. Levasseur

413

ESR Study of Paramagnetic Centers in Irradiated Phosphate Glasses

M.E. Archidi, M. Haddad, A. Nadiri and R. Berger

421

Physico-Chemical Properties of an Hybrid Glass

J.C. Bureau, A. Bakkali, Z. Sassi, C.Mai, J.F. Cornu and F. Babonneau

427

Contribution of Raman and Luminescence Spectroscopies to the Investigation of Phase Demixion in the System $\text{Bi}_x\text{Ln}_{(1-x)}\text{VO}_4$	
A. Lorriaux-Rubbens, J. Corset, J. Ghamri and H. Baussart	433
Correlations between Mn(III) Optical Spectroscopy and Structural and Magnetic Properties in $\text{Ti}_2\text{MnF}_5\cdot\text{H}_2\text{O}$ Single Crystals	
P. Nuñez, F. Rodriguez and M.C. Marco de Lucas	447
Luminescence Properties of Heavily Doped $\text{CaF}_2 : \text{Mn}^{2+}$	
A. Aamili, R. Mahiou, D. Avignant, S. Chou, J.C. Cousseins, M. Zahir and A. Sadel	453
Dielectric Properties of Carbon Black-Epoxy Resin Composites in the Frequency Range of 100Hz-15MHz	
M.E. Achour, L. Salome, K. Benaboud, F. Carmona and J.L. Miane	461
Ferroelastic Domain Study in Crystals with Formula K_3TiOF_5, $\text{K}_3\text{MO}_2\text{F}_4$ and $\text{K}_3\text{M}'\text{O}_3\text{F}_3$ ($\text{M} = \text{Nb}, \text{Ta} ; \text{M}' = \text{Mo}, \text{W}$)	
M. Fouad, J.P. Chaminade, J. Ravez and A. Sadel	469
Anionic Conductivity in Fluorapatites: Correlation between Structure and Electrical Properties	
A. Laghzizil, A. Bouhaouss, M. Ferhat, P. Barboux, R. Morineau and J. Livage	479
Conductimetric Determination of the Isotherms 25°C of Ternary Systems $\text{H}_2\text{O}-\text{M}_2\text{HPO}_4-(\text{NH}_4)_2\text{HPO}_4$ ($\text{M} = \text{Na}, \text{K}$)	
M. Nadifiyine, R. Lazrak, B. Tanouti, A. Mokhlisse, J. Guion, R. Tenu, J. Berthet and J.J. Counioux	489
Conductivity and Viscosity Studies of Lithium Ion Conductive Electrolytes Gelled with Poly(Methylmethacrylate)	
M. Rezrazi, M. Mullet and O. Bohnke	495
Dependence of the Stored Elastic Energy at Copact Cubic/Cubic Heterointerfaces on the Lattice Misfit	
A. Bouzaher and R. Bonnet	501
$\text{Ag}_2(\text{UO}_2)_2\text{V}_2\text{O}_8$: A New Compound with the Carnotite Structure. Synthesis, Structure and Properties	
F. Abraham, C. Dion, N. Tancret and M. Saadi	511
Theoretical Study of Electro-Optic Effect in the Tetragonal Phase of BaTiO_3 and PbTiO_3	
D. Khatib, B. Jannot and L. Lifsal	521
Effects of Molecular Hydrogen on Electrical Properties of Multicrystalline Silicon Grain Boundaries	
N. M'Gafad, H. Amzil and D. Sayah	531

Thermally Stimulated Current Study of Molecular Motions in Pentachlorotoluene

A. Bennis, N. Hitmi, A. Lamure and C. Lacabanne

537

Crystallographic and Magnetic Study of Mixed Pyrophosphates $Zn_xM_{2-x}P_2O_7$ ($0 \leq x \leq 2$) ($M=Mn, Co, Ni, Cu$)

M. Bettach, K. Benkhouja, A. Sadel, M. Zahir and M. Drillon

543

Magnetic Properties of some Copper and Nickel Tellurates and Tellurites

S. Bendaoud, J. Aride, M. Taibi, M. Belaiche, A. Boukhari
and M. Drillon

553

Synthesis, Crystal Structure and Magnetic Properties of a Dinickel (II) Complex with 3,6-Diformyl-Pyridazine Dioxime

B. Mernari, F. Abraham and M. Lagrenée

563

Polynuclear Ferromagnetic Ni(II) Complexes with Pseudohalide as Bridging Ligands

J. Ribas, M. Monfort, C. Diaz, A. Escuer, R. Vicente, M. Salah El Fallah,
X. Solans and M. Font-Bardia

573

Magneto-Structural Correlations in 1-D μ -(1,3-Azido)Nickel(II) Complexes

A. Escuer, R. Vincente, J. Ribas, M. Salah El Fallah, X. Solans
and M. Font-Bardia

581

Stability of a Spanish Sepiolite in Neutral and Basic Media

S. Martinez-Ramirez, F. Puertas and M.T. Blanco-Varela

587

Study of the Reaction at the Cr/Si Interface

A. Bouabellou, R. Halimi and N. Benouattas

593

Model of Formation of Thin Insulating Films on Silicon

J.C. Bureau, K. Chafik, Z. Sassi, B. Balland and A. Glachant

601

The Magnesium Oxide- Mono Aluminium Phosphate Reaction: Mechanism and Concentration Dependence

I. Borges, J.P. Wignacourt, J.C. Boivin, A. Nonat, A. Lorriaux-Rubbens,
F. Wallart and J.M. Canini

609

An Alternative Process for the Valorization of Natural Phosphates: Natural Microposphates and Reactivity

M. Ouammou, A. Banhassaine and A. Nadiri

621

Adherence Measurement of Duplex Systems onto Cold Rolled and Zinc Coated Steels

M. Bouzirri, A. Roche, L. Brochard and L. Pege

627

A Coprecipitation Process Using Heterocyclic Diacids for the Preparation of Conducting Oxides

- S. Nadir, J.P. Wignacourt, M. Drache, P. Conflant, M. Lagrenée
and B. Mernari 633

Behaviour of the Alloys F17Ti and AISI 304 in an Atmosphere Containing Hydrogen Sulphide at High Temperature

- A. Nyassi, A. Bendriss and J.P. Larpin 639

Gettering of Metallic Impurities by Rapid Thermal Processing

- B. Hartiti, H. Amzil, D. Sayah, J.C. Muller and P. Siffert 649

Author Index 657

Keyword Index 661

LIGHT-CONVERTING MOLECULAR DEVICES: ENERGY-TRANSFER PROCESSES IN LANTHANIDE DINUCLEAR COMPLEXES

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Keywords: Lanthanide, Rare Earth, Luminescence, Energy Transfer, Probe, Light Conversion, Calixarene, Dinuclear Complexes, Self-Assembly, Triple Helical Complexes, Aromatic Polyamines

Abstract

The design of efficient light-converting molecular devices containing a lanthanide ion is discussed. In order to unravel the energy transfer processes taking place in such a device, two series of model dinuclear complexes are studied : coordination compounds with calixarenes and self-assembled triple helical complexes based on aromatic polyamines.

Introduction

The 4f-block elements form an extended series of trivalent cations with the same formal charge and with similar chemical properties, mostly due to very weak ligand-field effects. The $4f^n5s^25p^6$ electronic configuration is indeed special in the sense that 4f electrons are shielded by the 5s and 5p subshells, resulting in atom-like absorption and emission spectra. The latter are easy to identify and make luminescent probes based on Eu(III) and Tb(III) especially efficient (Fig. 1). The development of luminescent chemical probes and sensors is the subject of intensive research, both in natural and medical sciences [1][2]. Furthermore, trivalent lanthanide ions may be incorporated into supramolecular complexes acting as molecular photonic devices [3]. In these nanometric machines, the various steps of the overall process involving absorption of light, transfer and re-emission at another wavelength are performed separately by distinct parts of the complex molecule. Such supramolecular constructions are termed light-conversion molecular devices.

To design a good europium (or terbium) luminescent sensor, a unique combination of features must be realized : (i) protection of the included ion from quenching due to the interaction with solvent molecules and/or high-energy vibrations from the ligating groups, (ii) presence of multiple absorbing groups suitable for energy transfer (antenna effect [2]), (iii) high thermodynamic stability and kinetic inertness, and (iv) non-toxicity of the sensor if it is to be used in *in vivo* applications.

The yield of the energy transfer between the donor and the acceptor is related to the donor-acceptor distance. Determination of this distance allows one to realize a fluorescent mapping of large molecules such as proteins, leading to useful structural information, or to calculate the distance between ion sites [1]. In order to unravel the intimate mechanism of energy-transfer processes both between an organic chromophore and a Ln(III) ion and between two lanthanide ions, we have undertaken the study of several classes of heterodinuclear complexes. This paper presents some of the results obtained these last three years on dinuclear complexes with calixarenes and planar multidentate nitrogen-containing ligands analogous to terpyridine.

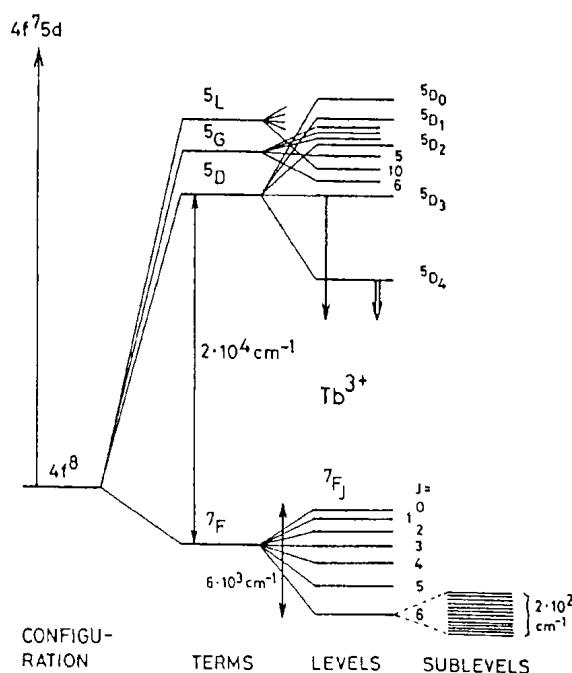
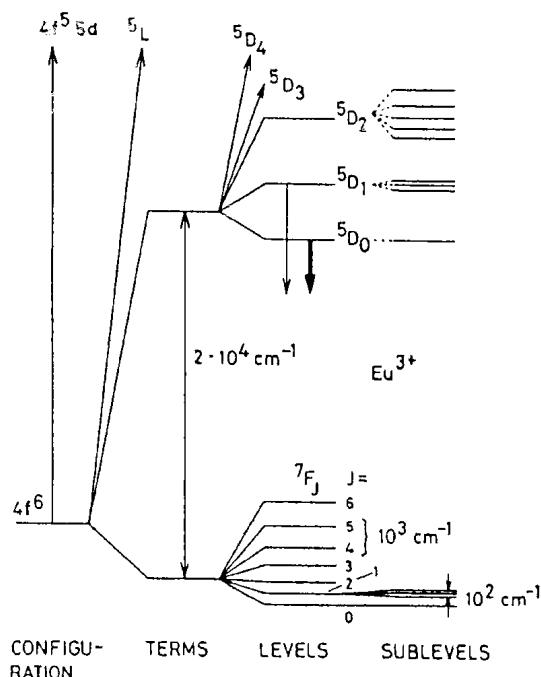


Figure 1 : Partial energy level diagrams for Eu^{III} and Tb^{III} .