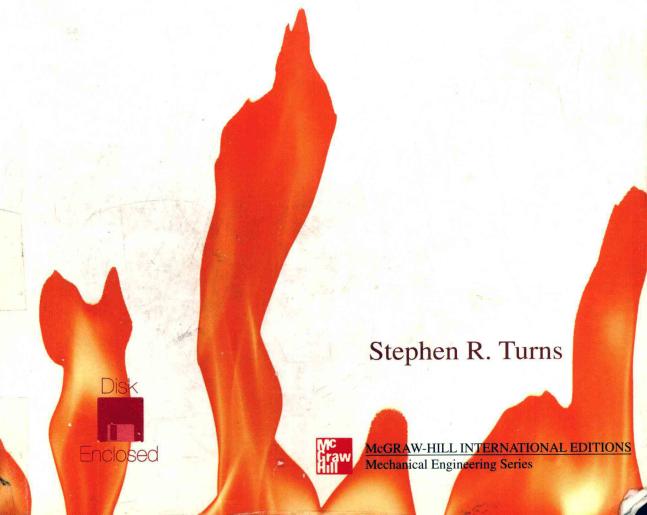
An Introduction to Combustion

Concepts and Applications second edition



AN INTRODUCTION TO COMBUSTION

Concepts and Applications

SECOND EDITION

Stepnen K. 1 urns

Propulsion Engineering Research Center and Department of Mechanical and Nuclear Engineering The Pennsylvania State University

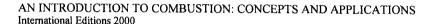


Boston • Burr Ridge, IL • Dubuque, IA • Madison, WI
New York • San Francisco • St. Louis

Bangkok • Bogotá • Caracas • Lisbon • London • Madrid • Mexico City
Milan • New Delhi • Seoul • Singapore • Sydney • Taipei • Toronto

McGraw-Hill Higher Education

A Division of The McGraw-Hill Companies



Exclusive rights by McGraw-Hill Book Co – Singapore, for manufacture and export. This book cannot be re-exported from the country to which it is consigned by McGraw-Hill.

Copyright © 2000, 1996 by The McGraw-Hill Companies, Inc. All rights reserved. Except as permitted under the United States Copyright Act of 1976, no part of this publication may be reproduced or distributed in any form or by any means, or stored in a data base or retrieval system, without the prior written permission of the publisher.

10 9 8 7 6 5 4 20 9 8 7 6 5 4 3 2 CTP SLP

ISBN 0-07-230096-5 (book) ISBN 0-07-230098-1 (disk) ISBN 0-07-235044-X (book bound with disk)

Library of Congress Cataloging-in-Publication Data

Turns, Stephen R.

An introduction to combustion : concepts and applications / Stephen R. Turns. – 2nd ed. p. cm. – (McGraw-Hill series in mechanical engineering)

ISBN 0-07-230096-5

1. Combustion engineering. I. Title. II. Series.

TJ254.5.T88 2000

621.402'3-dc21

99-25421

CIP

www.mhhe.com

When ordering this title, use ISBN 0-07-116910-5

Printed in Singapore

ABOUT THE AUTHOR

Stephen R. Turns received degrees in mechanical engineering from The Pennsylvania State University (B.S., 1970), Wayne State University (M.S., 1974), and the University of Wisconsin at Madison (Ph.D., 1979). He was a research engineer at General Motors Research Laboratories from 1970 to 1975. He joined the Penn State faculty in 1979 and is currently Professor of Mechanical Engineering. Dr. Turns teaches a wide variety of courses in the thermal sciences and has received several awards for teaching excellence at Penn State. He is an active combustion researcher, publishing widely, and is an active member of The Combustion Institute, the American Society of Mechanical Engineers, and the Society of Automotive Engineers.

This Book Is Dedicated to My Wife, Joan, and Our Sons, Matthew and Michael -SRT By contrast, the first fires flickering at a cave mouth are our own discovery, our own triumph, our grasp upon invisible chemical power. Fire contained, in that place of brutal darkness and leaping shadows, the crucible and the chemical retort, steam and industry. It contained the entire human future.

Loren Eiseley The Unexpected Universe

PREFACE TO THE SECOND EDITION

This second edition retains the same primary objectives as the original text: first, to present basic combustion concepts using relatively simple and easy-to-understand analyses; and second, to introduce a wide variety of practical applications that motivate or relate to the various theoretical concepts. The overarching goal is to provide a textbook that is useful for both formal undergraduate study in mechanical engineering and related fields, and informal study by practicing engineers.

The author, as well as many of his colleagues around the world, also have found the book to be useful in a first course in combustion at the graduate level. In this usage, however, the book alone did not suffice: more detailed treatments and advanced topics needed to be developed by the instructor to supplement the text. Nevertheless, many have reported success in using the book in this manner. The second edition specifically addresses this need for additional topics and greater depth in some areas. Chapter 7 now contains a section dealing with multicomponent diffusion, including thermal diffusion. The development of the one-dimensional energy conservation equation is expanded in Chapters 7 and 8 to include multicomponent and thermal diffusion in a form consistent with that used in the various flame codes developed at Sandia National Laboratories, Livermore, CA. This provides a good link for those instructors who use such codes in conjunction with CHEMKIN software in their courses. In the same spirit, Chapter 9 now includes a section on counterflow diffusion flames. None of these additions detract in any way from the ability to use the book at a lower level. The more advanced topics are arranged so that they can be skipped with no loss in continuity. Furthermore, these additions, in general, are not particularly lengthy—thus, the overall length of the text is not greatly increased, and the book retains its original comfortable, compact feel.

Changes to the basic text include a brief discussion of the molecular structure of fuels in an appendix to Chapter 2. This appendix provides information useful for appreciating many of the thermochemical concepts of Chapter 2, as well as providing background information useful for understanding some of the chemical-kinetic concepts developed in Chapters 4 and 5. Chapter 4 now includes a section on partial equilibrium and a discusion of characteristic chemical time scales, while Chapter 5 has been updated to include the latest advances in methane kinetics (GRI-Mech). Reaction pathway diagrams for CH₄-air combustion are also included to give a much clearer—and holistic—picture of methane combustion kinetics. Chapter 6 sees the addition of a well-stirred-reactor example that employs detailed kinetics, providing yet another link to CHEMKIN software. In Chapter 8, the discussion of premixed

flame structure is expanded to give a much clearer, and detailed, view of this important subject. A chapter on detonations (Chapter 16) also has been added to meet the needs of those desiring to include this topic in their courses. Logically, this material could be included following Chapter 2 or Chapter 8. New problems have been added to many chapters and several additional examples are included. Those problems requiring, or benefiting from, the use of a computer for their solution are indicated. The computer software has been updated to be compatible with Windows.

The author hopes that this new edition will continue to serve well those who desire to use the book at its most basic level and that the additional topics presented in this edition also will make the book more useful at a somewhat more advanced level.

Stephen R. Turns University Park, PA

PREFACE TO THE FIRST EDITION

High interest in combustion and combustion applications exists among many engineering students. Although undergraduate, senior-level courses in combustion and combustion-related areas are offered at many institutions, finding an appropriate textbook for such courses is difficult, at best. The need for an introductory text on combustion, specifically structured for an undergraduate readership, has served as the motivation for writing this book. The offering of an introductory course at Penn State and the development of an introductory textbook were conceived jointly, and this book is the result of those developments.

Although the primary audience is intended to be senior-level students in mechanical and related engineering majors, others may find the text useful as a bridge between the basic undergraduate thermal sciences and advanced treatments of combustion. Many examples and problems are presented to aid in understanding and to relate to practical applications. Thus, it is hoped that both first-year graduate students and practicing engineers can benefit from the material presented here.

In its organization, the text provides flexibility. The 15 chapters provide much more material than can be covered in a single-semester course; this overkill makes it easy for an instructor to tailor a course to a particular theme or set of topics, while allowing the theme to evolve or change from one course offering to another. For example, a one-semester course providing a general overview could cover Chapters 1–6, 15, 8, 9, and 14; while a course with some emphasis on spark-ignition engines could cover Chapters 1–6, 8, 11, 12, 15, and 9.

Located in Chapters 1–3 are topics considered essential for an undergraduate course. Chapter 1 defines combustion and the types of flames, and introduces the effects and control of combustion-generated air pollution, which is treated in greater detail in Chapter 15.

The thermochemistry needed for a study of combustion is presented in Chapter 2. This chapter emphasizes the importance of chemical equilibrium to combustion. Software provided with this book provides students with a simple means of calculating complex equilibria for combustion gases; this software can be put to good use in many interesting and pedagogically helpful projects. Chapter 3 introduces mass transfer. The approach taken here, and throughout the book, is to simplify theoretical developments by treating all mass transfer within the context of simple binary systems. Except for a brief mention in Chapter 7, the treatment of multicomponent diffusion is left to more advanced texts. Such an approach allows students with no previous exposure to mass transfer to gain an appreciation of the subject without getting



bogged down in its inherent complexities. Chapter 3 uses both the classical Stefan problem and simple droplet evaporation to illustrate mass-transfer theory.

Onward to the subject of chemistry, Chapters 4 and 5 deal with chemical kinetics by presenting basic concepts (Chapter 4) and discussing chemical mechanisms of importance to combustion and combustion-generated air pollution (Chapter 5). In addition to showing the unavoidable complexity of hydrocarbon combustion chemistry, simple single- and multistep kinetics are presented that can be used to incorporate chemical kinetic effects in simple analyses or models, recognizing, of course, the pitfalls of simplified kinetics.

The interrelation of chemical kinetics and thermodynamic modeling is the subject of Chapter 6. Here, models of constant-pressure and constant-volume reactors, and well-stirred and plug-flow reactors, are developed. These simple models allow a student to grasp clearly how chemical kinetics fits into the bigger picture. This chapter also offers many opportunities for projects involving reactor analysis and/or design. Both the usefulness and uniqueness of this chapter make it a lot of fun.

Having completed our study of thermochemistry, molecular transport, and chemical kinetics, we devote Chapter 7 to the development of the simplified conservation equations for reacting systems used in subsequent chapters. The conserved-scalar concept is introduced here. This chapter is intended to provide a background from which more rigorous developments can be followed. For an undergraduate course, this chapter is clearly optional, and is probably best skipped; however, for an introductory graduate-level course, the chapter may be quite useful.

Elementary treatments of flames are presented in Chapters 8–13. Laminar premixed flames are discussed in Chapter 8, and laminar nonpremixed flames in Chapters 9 and 10; turbulent flames are dealt with in Chapter 12 (premixed) and Chapter 13 (nonpremixed). Topics treated include flame propagation, ignition and quenching, and flame stabilization. Simplified analyses are presented wherever possible, and practical applications emphasized. In all cases, rigorous mathematical development is eschewed in favor of developing the most basic understanding. This approach has the shortcoming of not being able to deal with some phenomena at all, and others, incompletely at best. Usually in these areas, warnings are given and references cited to help the reader who seeks a more complete understanding. Because the wealth of material in these chapters, one can conveniently choose to cover only laminar flames (Chapters 8, 9, and 10) or to focus only on premixed flames (Chapters 8, 11, and 12) or nonpremixed flames (Chapters 9, 10, and 13). Particular emphases on specific applications might suggest which topics to cover.

Linking droplet vaporization theory to practical devices is the subject of the second half of Chapter 10, where a model of a one-dimensional vaporization-controlled combustor is developed. The primary purposes of this section are to reinforce previous concepts of equilibrium and evaporation, help develop students' powers of analysis, and provide ideas and concepts that can be used in applications-oriented projects. Design projects can easily be fitted into the framework of Chapter 10. Depending on course objectives, this section of Chapter 10 can be treated as optional.

In Chapter 14, burning of solids is introduced, using carbon combustion as the archetypical system. Again, simplified analyses are presented to illuminate heterogeneous combustion concepts and to introduce the ideas of diffusionally and kinetically controlled combustion. This chapter also acquaints the student with coal combustion and its applications.

Omitting a treatment of combustion-generated pollutants would be unthinkable in a modern book on combustion. Chapter 15 focuses on this topic. This chapter introduces the reader to the quantification of emissions and discusses the mechanisms of pollutant formation and their control. This chapter emphasizes applications and should be of particular interest to the intended readers of this book. The placement of this chapter does not suggest its relative importance. Depending on course objectives, the material here could be covered following Chapters 1–6.

Now, in summary, this book attempts to present an introduction to combustion at a level easily comprehended by students nearing the completion of an undergraduate study in mechanical engineering and related fields. Through the use of examples and homework problems, students can develop confidence in their understanding and go on to apply this to various projects and "real world" problems. It is hoped that this text will fit the needs of instructors, and others, who desire simplified and appropriately structured materials for an introductory study of the fascinating field of combustion.

Stephen R. Turns University Park, PA

ACKNOWLEDGMENTS

Many people contributed their support, time, and psychic energy to either the first or second edition of this book. First, I would like to thank the many reviewers who contributed along the way; in particular, Jim Driscoll, The University of Michigan; Norm Laurendeau, Purdue University; John Lloyd, Michigan State University; Michel Louge, Cornell University; Jon Van Gerpen, Iowa State University; and Carl Wikstrom, University of Arkansas, all provided many useful comments on early drafts of the first edition. Also, the detailed comments of Steve Goebel and Alan Feitelberg, who used various versions of the book in teaching a course at Rensselaer Polytechnic Institute. were extremely valuable. They also generously shared some interesting end-ofchapter problems, which have been incorporated into the second edition. My friend and colleague, Chuck Merkle, continually provided moral support and served as a sounding board for ideas on both content and pedagogy in the first edition. Many students at Penn State contributed in various ways, and I want to acknowledge the particular contributions of Jeff Brown, Jongguen Lee, and Don Michael. Sankaran Venkateswaran deserves special thanks for providing the turbulent jet-flame model calculations, as does Dave Crandall for his assistance with the software for the second edition. A major debt of thanks is owed to Donn Mueller, who painstakingly solved all of the first-edition end-of-chapter problems. Thanks are also due to my dear friend Kathy Wendland for her editorial help with the second edition. I would also like to thank the Gas Research Institute for their support of my research activities through the years, as it was these activities that provided the initial inspiration and impetus to write this book. Invaluable to my efforts throughout was the unflagging support of my family. They tolerated amazingly well the time spent writing on weekends and holiday breaks—time that I could have spent with them. The last acknowledgment is reserved for Cheryl Adams. How either edition could have been completed without her is difficult to conceive. She typed and word-processed the many drafts and created many of the line drawings that appear in the text. Cheryl's skill and dependability contributed greatly to making the writing of this book an enjoyable experience.

CONTENTS

•	
Preface to the Second Edition xi Preface to the First Edition xiii Acknowledgments xvii Introduction 1 Motivation to Study Combustion 1 A Definition of Combustion 6 Combustion Modes and Flame Types 6 Approach to Our Study 8	Flue- (or Exhaust-) Gas Recirculation 58 Summary 66 Nomenclature 66 References 68 Review Questions 69 Problems 70 Appendix 2A — Some Fuel Chemistry 78 3 Introduction to Mass Transfer 83
References 8	Overview 83
2 Combustion and Thermochemistry 9	Rudiments of Mass Transfer 83 Mass Transfer Rate Laws 84 Species Conservation 90
Overview 9 Review of Property Relations 9 Extensive and Intensive Properties 9 Equation of State 10 Calorific Equations of State 11 Ideal-Gas Mixtures 13 Latent Heat of Vaporization 15 First Law of Thermodynamics 16	Some Applications of Mass Transfer 92 The Stefan Problem 92 Liquid-Vapor Interface Boundary Conditions 94 Droplet Evaporation 98 Summary 105 Nomenclature 105 References 107
First Law—Fixed Mass 16 First Law—Control Volume 17	Review Questions 107
Reactant and Product Mixtures 18	Problems 108
Stoichiometry 18	4 Chemical Kineties 111
Absolute (or Standardized) Enthalpy and	
Enthalpy of Formation 24	Overview 111
Enthalpy of Combustion and Heating	Global versus Elementary Reactions 112
Values 27	Elementary Reaction Rates 113
Adiabatic Flame Temperatures 32	Bimolecular Reactions and Collision Theory 113
Chemical Equilibrium 36	Other Elementary Reactions 118
Second-Law Considerations 36	Rates of Reaction for Multistep
Gibbs Function 38	Mechanisms 119
Complex Systems 44	Net Production Rates 119
Equilibrium Products of Combustion 45	Compact Notation 121
Full Equilibrium 45	Relation between Rate Coefficients and
Water-Gas Equilibrium 47 Pressure Effects 51	Equilibrium Constants 122 Steady-State Approximation 125
	The Mechanism for Unimolecular Reactions 126
Some Applications 52 Recuperation and Regeneration 52	Chain and Chain-Branching Reactions 127
Recuperation and Regeneration 32	

Chemical Time Scales 133
Partial Equilibrium 138
Summary 140
Nomenclature 140
References 141
Questions and Problems 143

5 Some Important Chemical Mechanisms 148

Overview 148 The H₂-O₂ System 148 Carbon Monoxide Oxidation 152 Oxidation of Higher Paraffins 153 General Scheme 153 Global and Quasi-global Mechanism 156 Methane Combustion 158 Complex Mechanism 158 High-Temperature Reaction Pathway Analysis 158 Low-Temperature Reaction Pathway Analysis 166 Oxides of Nitrogen Formation 168 Summary 171 References 172 Ouestions and Problems 174

6 Coupling Chemical and Thermal Analyses of Reacting Systems 178

Overview 178 Constant-Pressure, Fixed-Mass Reactor Application of Conservation Laws 179 Reactor Model Summary 182 Constant-Volume, Fixed-Mass Reactor 182 Application of Conservation Laws 182 Reactor Model Summary 183 Well-Stirred Reactor 189 Application of Conservation Laws 190 Reactor Model Summary 192 Plug-Flow Reactor 200 Assumptions 200 Application of Conservation Laws 201 Applications to Combustion System Modeling 205 Summary 206 Nomenclature 206

References 208
Problems and Projects 209
Appendix 6A—Some Useful Relationships among Mass Fractions, Mole
Fractions, Molar Concentrations, and Mixture
Molecular Weights 214

7 Simplified ConservationEquations for Reacting Flows 215

Overview 215 Overall Mass Conservation (Continuity) Species Mass Conservation (Species Continuity) 218 Multicomponent Diffusion General Formulations 222 Calculation of Multicomponent Diffusion Coefficients 223 Simplified Approach 226 Momentum Conservation 229 One-Dimensional Forms 229 Two-Dimensional Forms 230 Energy Conservation 234 General One-Dimensional Form 234 Shvab-Zeldovich Forms 236 Useful Form for Flame Calculations 240 The Concept of a Conserved Scalar 241 Definition of Mixture Fraction 241 Conservation of Mixture Fraction 242 Conserved Scalar Energy Equation 246 Summary 247 Nomenclature 248 References 249 Review Questions 250 Problems 251

8 Laminar Premixed Flames 253

Overview 253
Physical Description 254
Definition 254
Principal Characteristics 254
Typical Laboratory Flames 256
Simplified Analysis 261
Assumptions 262
Conservation Laws 262
Solution 264
Detailed Analysis 269

Governing Equations 269	Structure of CH ₄ -Air Flame 350
Boundary Conditions 270	Summary 354
Structure of CH ₄ -Air Flame 271	Nomenclature 354
Factors Influencing Flame Velocity and	References 356
Thickness 274	Review Questions 359
Temperature 274	
Pressure 277	Problems 359
Equivalence Ratio 278	
Fuel Type 278	na Duamlet Evenementian and
Flame Speed Correlations for Selected	10 Droplet Evaporation and
Fuels 278	Burning 362
Quenching, Flammability, and Ignition 283	0 : 000
Quenching by a Cold Wall 284	Overview 362
Flammability Limits 289	Some Applications 362
Ignition 291	Diesel Engines 363
Flame Stabilization 294	Gas-Turbine Engines 365
Summary 298	Liquid-Rocket Engines 367
Nomenclature 299	Simple Model of Droplet Evaporation 367
References 300	Assumptions 370
	Gas-Phase Analysis 372
Review Questions 302	Droplet Lifetimes 375
Problems 302	Simple Model of Droplet Burning 378
• Laminar Diffusion Flames 305	Assumptions 379
4 Lammar Direction Frames 303	Problem Statement 380
Overview 305	Mass Conservation 381
Nonreacting Constant-Density Laminar	Species Conservation 381
Jet 306	Energy Conservation 383
	Summary and Solution 389
Physical Description 306	Burning Rate Constant and Droplet
Assumptions 307 Conservation Laws 308	Lifetimes 391
	Extension to Convective Environments 395
Boundary Conditions 308	Additional Factors 398
Solution 309	One-Dimensional Vaporization-Controlled
Jet Flame Physical Description 314	Combustion 399
Simplified Theoretical Descriptions 317	
Primary Assumptions 318	Physical Model 399 Assumptions 401
Basic Conservation Equations 318	Mathematical Problem Statement 401
Additional Relations 319	Analysis 402
Conserved Scalar Approach 320	Model Summary 410
Various Solutions 327	
Flame Lengths for Circular-Port and Slot	Summary 410
Burners 331	Nomenclature 413
Roper's Correlations 331	References 415
Flowrate and Geometry Effects 336	Problems 418
Factors Affecting Stoichiometry 336	Projects 420
Soot Formation and Destruction 343	Appendix 10A—Sir Harry R. Ricardo's
Counterflow Flames 347	Description of Combustion in Diesel
Mathematical Description 348	Engines 421

11 Introduction to Turbulent Flows 423 Overview 423 Definition of Turbulence 424 Length Scales in Turbulent Flows 42

Definition of Turbulence 424
Length Scales in Turbulent Flows 427
Four Length Scales 427
Turbulence Reynolds Numbers 430
Analyzing Turbulent Flows 434
Reynolds Averaging and Turbulent Stresses 435
The Closure Problem 437
Axisymmetric Turbulent Jet 441
Beyond the Simplest Model 444
Summary 445

Summary 445 Nomenclature 446 References 447

Questions and Problems 449

12 Turbulent Premixed Flames 450

Overview 450
Some Applications 450
Spark-Ignition Engines 450
Gas-Turbine Engines 451
Industrial Gas Burners 452
Definition of Turbulent Flame Speed 454
Structure of Turbulent Premixed Flames 456
Experimental Observations 456
Three Flame Regimes 457
Wrinkled Laminar-Flame Regime 462

Wrinkled Laminar-Flame Regime 462 Distributed-Reaction Regime 466

Flamelets-in-Eddies Regime 468

Flame Stabilization 470

Bypass Ports 470 Burner Tiles 471 Bluff Bodies 471

Swirl or Jet-Induced Recirculating Flows 473

Summary 474
Nomenclature 475
References 476
Problems 478

13 Turbulent Nonpremixed Flames 481

Overview 481 Jet Flames 484 General Observations 484
Simplified Analysis 489
Flame Length 495
Flame Radiation 501
Liftoff and Blowout 504
Other Configurations 509
Summary 512
Nomenclature 513
References 514
Review Questions 517
Problems 517

14 Burning of Solids 519

Overview 519
Coal-Fired Boilers 520
Heterogeneous Reactions 520
Burning of Carbon 522
Overview 523
One-Film Model 524
Two-Film Model 536
Particle Burning Times 542
Coal Combustion 544
Other Solids 545
Summary 545
Nomenclature 546
References 547
Questions and Problems 548

15 Pollutant Emissions 550

Overview 550 Effects of Pollutants 551 Quantification of Emissions 553 Emission Indices 553 Corrected Concentrations 555 Various Specific Emission Measures 558 Emissions from Premixed Combustion 559 Oxides of Nitrogen 559 Carbon Monoxide 567 Unburned Hydrocarbons 568 Catalytic Aftertreatment 569 Particulate Matter 571 Emissions from Nonpremixed Combustion 572 Oxides of Nitrogen 573 Unburned Hydrocarbons and Carbon Monoxide 584

Particulate Matter 586
Oxides of Sulfur 586
Summary 587
Nomenclature 588
References 589
Ouestions and Problems 594

16 Detonations 598

Overview 598
Physical Description 598
Definition 598
Principal Characteristics 599
One-Dimensional Analysis 600
Assumptions 600
Conservation Laws 601
Combined Relations 602
Detonation Velocities 609
Structure of Detonation Waves 613
Summary 617
Nomenclature 618
References 619
Problems 620

Appendix A Selected Thermodynamic Properties of Gases Comprising C-H-O-N System 621

Appendix B Fuel Properties 648

Appendix C Selected Properties of Air, Nitrogen, and Oxygen 653

Appendix D Binary Diffusion Coefficients and Methodology for their Estimation 656

Appendix E Generalized Newton's Method for the Solution of Nonlinear Equations 659

Appendix F Computer Codes for Equilibrium Products of Hydrocarbon—Air Combustion 662

Index 665