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Handbook of Proton-NMR Spectra and Data

Volume 9

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A. Spectrum Number

The organic compounds for which spectra are listed in this handbook are ordered by molecular complexity, from simple to more complex. Each entry includes a number in parentheses to refer to the index section where reference is made to the spectrum.

A User's Guide

This data handbook has been planned as a part of projected open-ended series, with publication of additional data books scheduled upon completion of further data compilation. The handbook consists of proton-NMR spectra and data for 4,000 organic compounds, divided into five volumes with 800 entries each. The handbook is divided into data sections and index sections.

1. Proton-NMR Spectra and Data Sections

All data entries are in the form:

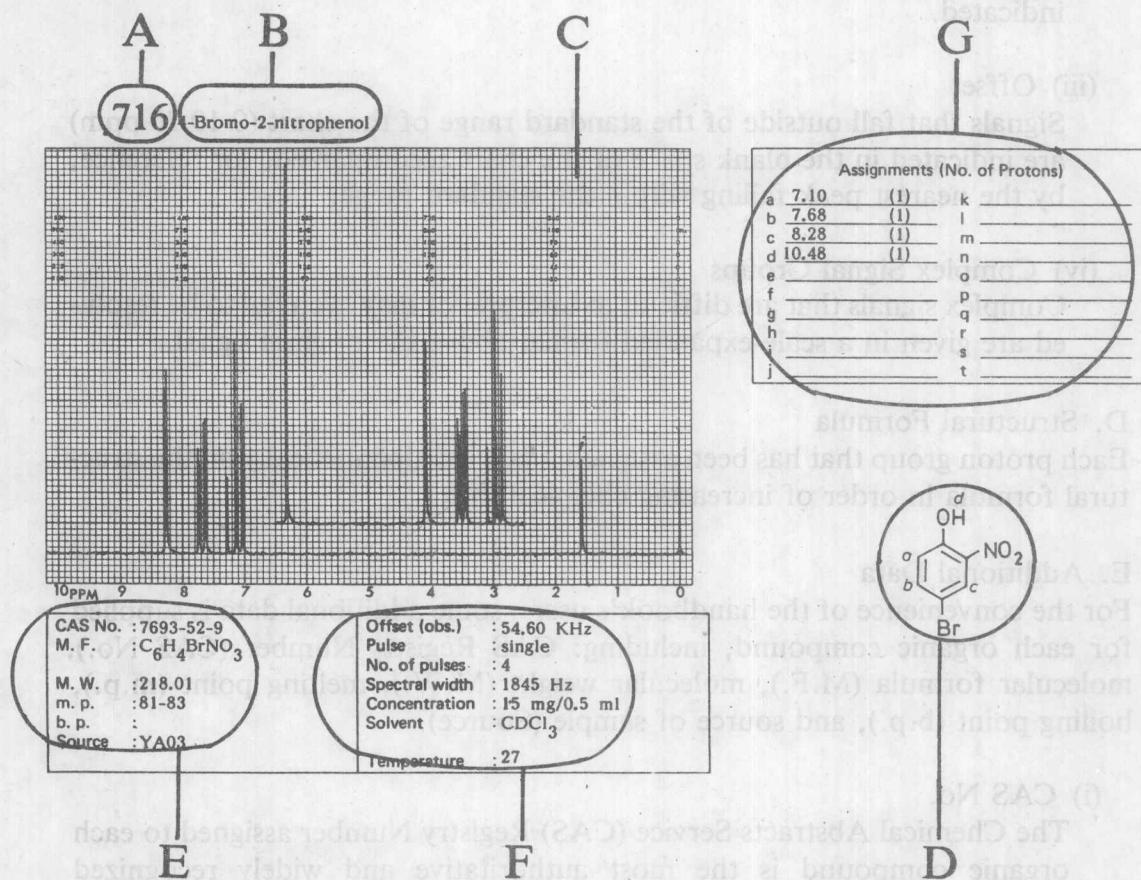


Figure 1. Illustration of Spectral Data

A. Spectrum Number

The organic compounds for which spectral data are provided are arranged in order of molecular complexity, from simpler to more complex. Each entry is assigned a number in sequence which serves as a key for easy indexing and quick reference.

B. Chemical Name

In general, all chemical names are in accord with the IUPAC nomenclature system. There are, however, some compounds that are more frequently referred to by trivial names, such as vanillin. In these cases, the trivial name is used.

C. Spectrum

(i) Units

For easy recognition of peak positions of signals, the smallest unit of the spectrum chart is 0.1 ppm (9 Hz).

(ii) Impurities

Signals resulting from a trace amount of impurities in the sample are indicated by the letter "i" over the signal in question. Signals resulting from impurities in the solvent, however, such as CHCl_3 in CDCl_3 , are not indicated.

(iii) Offset

Signals that fall outside of the standard range of the chart (0-10.20 ppm) are indicated in the blank space of the chart accompanied, for reference, by the nearest peak falling within the standard range.

(iv) Complex Signal Groups

Complex signals that are difficult to interpret or may even fail to be resolved are given in a scale-expanded format above the original signal.

D. Structural Formula

Each proton group that has been assigned is labeled alphabetically on the structural formula in order of increasing chemical shift.

E. Additional Data

For the convenience of the handbook's users, some additional data is supplied for each organic compound, including: CAS Registry Number (CAS No.), molecular formula (M.F.), molecular weight (M.W.), melting point (m.p.), boiling point (b.p.), and source of sample (Source).

(i) CAS No.

The Chemical Abstracts Service (CAS) Registry Number assigned to each organic compound is the most authoritative and widely recognized chemical identification number. As a result, it is extremely helpful as a link

to information available in many other database systems such as CAS-ONLINE and CIS, among others.

(ii) Molecular Formula

Molecular formulas are arranged according to the Hill system, that is, C before H, with all other elements in alphabetical order.

(iii) Molecular Weight

Average molecular weights derived from average atomic weights are given.

(iv) Melting and Boiling Points

Both values are recorded in Celsius, based on the literature or data recorded by sample suppliers. In the case of compounds that decompose, the decomposition point is given, as, for example, 250 dec.

(v) Source of Sample

Sample suppliers are recorded in a four-letter code. Full identification of suppliers is provided in Chart 5.

F. Spectrum Measurement

(i) Spectrometer

The spectrometer used for all measurements is an FX-90Q manufactured by JEOL Ltd. This is a modern Pulse Fourier Transform spectrometer. The 90 MHz spectrometer is the most widely used spectrometer at present.

(ii) Specifications

The specifications for the spectrometer are as follows:

• Magnetic field	21100 gauss
• Observation frequency	89.55-89.65 MHz
• NMR lock	D2 internal lock
• Insert coil	5 mm diameter
• Resolution	$^1\text{H} \leq 0.2$ Hz
• Pulse mode	Single
• Accumulation times	Four (average)
• Data points	16 K

(iii) Measurement Conditions

Spectra measurements are carried out under constant conditions as much as possible. Measurement is carried out, as a rule, at 27°C. In most cases, the solvent used is CDCl_3 ; D_2O , DMSO-d_6 , or polysol, a mixture of CDCl_3 , and DMSO-d_6 which is used as a solvent for compounds that do not dissolve in CDCl_3 . Tetramethylsilane (TMS) is employed as the internal standard except when the solvent is D_2O , in which case 3-(trimethylsilyl)propionic acid- d_4 sodium salt (TSP- d_4) is employed.

G. Assignment

Assignment is carried out in order of increasing chemical shift relative to TMS or TSP-d₄. Related proton groups are identified with letters of the alphabet for easy correlation.

Integration curves have been omitted from the spectra chart because of the problem of poor legibility. The number of assigned protons is recorded instead in parentheses following the chemical shift value.

Determination of the chemical shift value has been carried out by standard methods. In addition, digital output from the spectrometer has been referred to. For those multiplets for which first-order approximation is sufficient, the center of the multiplets has been regarded as the chemical shift. For those which clearly exhibit a second-order signal pattern, when all the signals constituting the multiplet can be positively identified, the first approximation or the center of the multiplet is regarded as the chemical shift.

Spin-spin coupled multiplets with complex overlapping are represented by the appropriate signal range, as, for example, 7.22-8.54.

When assignment is difficult because of similar chemical environments of proximate molecular structures or overlapping signals, the several alternative assignment values are each marked with an asterisk and a number which indicates that other asterisked values of the same number may be interchangeable.

When no appropriate signal is observable throughout the entire spectrum, the proton group is marked u.o. (unobserved). When the appropriate signal clearly overlaps with another signal and the chemical shift value is difficult to determine, the proton group is marked u.d. (undetermined). When the precise chemical shift value cannot be determined but an extremely close approximation is possible, the value is marked with the prefix ca. (circa).

2. Index Sections

Thorough, carefully prepared indexes are crucial tools for effective use of any data collection. Four types of exhaustive indexes accompany the present handbook, designed for quick and convenient cross checking, in a handy format that provides extremely useful combinations of data at a glance and greatly enhances the use of the entire handbook. The indexes and the data they contain are as follows:

Table 1

Index Name	First Column	Second Column	Third Column
Chemical Name Index	Chemical name	Spectrum No.	
Molecular Formula Index	Molecular formula	Chemical name	Spectrum No.
Substructure Index	Substructure	Chemical shift	Spectrum No.
Chemical Shift Index	Chemical shift	Substructure	Spectrum No.

A. Chemical Name Index

All chemical names are in alphabetical order. Numerals, Greek characters,

and codes indicating substitution positions (such as 1-, beta-, N-, etc.) and geometrical, stereochemical, and optical isomeric prefixes (such as cis-, endo-, levo-, etc.) are disregarded in alphabetization. Each name is followed by a spectrum number, as can be seen in the sample page of the Chemical Name Index (Chart 1).

B. Molecular Formula Index

When the desired organic compound cannot be found in the Chemical Name Index or its nomenclature is unclear, it becomes necessary to look for a compound by means of its molecular formula. This is just the case for which the Molecular Formula Index was designed. Molecular formulas are listed following the Hill system, in order of complexity. When several compounds have the same molecular formula, the different compounds are listed in alphabetical order, as can be seen in the sample page of the Molecular Formula Index (Chart 2). The spectrum number is also provided for easy cross reference to the data handbook.

C. Substructure Index

Determining what sorts of proton correspond to the various signals within a spectrum, investigating their chemical environments, and suggesting models for their structure are all very important parts of spectral analysis.

A unique notation system for representing substructure has been developed in the Substructure Index to allow easy correlation of the chemical environments of protons and chemical shift, and at the same time permit fast consideration of a wide variety of model structures.

In the substructure notation of the index, the proton group which is assigned as the TARGET is grouped with its neighbors (e.g. α position) and next nearest neighbors (e.g. β position). The TARGET occupies the head position of the substructure notation, as can be seen in Fig. 2. It is followed by the neighboring atoms, given outside parentheses; the next-nearest atoms are written inside parentheses.

The diagram illustrates a polymer substructure and its neighborhood. On the left, a 'Target' node is shown as a silicon atom (Si) bonded to four oxygen atoms (O). It is also bonded to a methyl group (CH₃) and an ethyl group (CH₂CH₃). The methyl group is further bonded to a hydrogen atom (H). To the right, the 'Notation' section defines the 'Immediate neighbors' as CH/CH₂(CH₂), NH(CH₂), and Si(O,O,O). Below this, the 'Next nearest neighbors' are listed as CH₃, CH₂, NH₂, and O.

Figure 2. Substructure Notation

bn All non-identical substructures relating to a particular TARGET are cited in the index. In order to simplify the substructure notation, single bonds have not been represented. Code for other atomic groups is found in Chart 3-1, as is their order of priority. Chart 3-2 shows a sample page of the Substructure Index. ('CH₂U' indicates that two protons of the methylene group are unequivalent.) For the user's convenience, chemical shift value and spectrum number are also provided for each entry.

D. Chemical Shift Index

Entries on the Chemical Shift Index are arranged in order of increasing delta value (ppm). For easy correlation of chemical shift with substructure, substructure notations are listed for each entry as well. Chart 4 shows a sample page of the Chemical Shift Index.

Chemical Shift Index
Entries on the Chemical Shift Index are arranged in order of increasing delta value (ppm). For easy correlation of chemical shift with substructure, substructure notations are listed for each entry as well. A sample page of the Chemical Shift Index is shown below.



Figure 2. Superimposed Molecule

Chart 1. The Chemical Name Index

name	No.	name	No.
3-Methyl-2-nitrobenzoyl chloride	4822	3-Phenyl-1,3-butanediol	5466
Methyl 4-nitrophenylacetate	5108	1-Phenyl-1,3-butanedione	5360
2-Methyl-2-nonanol	5577	1-Phenyl-2-butanone	5402
(E)-2-Methyl-4-nonen-2-ol	5554	1-Phenyl-1-butyne	5341
2-Methyl-2-(3-oxobutyl)cyclopentane-1,3-dione	5468	cis-2-Phenylcyclopropane-1-carboxylic acid	5361
4-Methyl-3-(3-oxobutyl)-4-pentenal	5503	trans-2-Phenylcyclopropane-1-carboxylic acid	5362
Methyl (2-oxocyclohexane)acetate	5225	2-Phenyl(2- ² H ₁)-1,3-dioxane	5379
Methyl 8-oxo-1,4-dioxaspiro[4.5]decane-7-carboxylate	5472	(1-Phenylethyl)oxamohydrazide	5439
2-Methyl-2-pentenal dimethyl acetal	5009	Phenylglyoxylic acid	4834
4-Methyl-trans-perhydroquinolizine	5544	4-Phenyl-3-oxolen-2-one	5319
1-Methyl-3-phenylpropylamine	5482	2-Phenylperhydro-1,2-oxazine-3,5-dione	5337
1-Methyl-5-phenyl-1H-tetrazole	4880	2-Phenylperhydro-1,2-oxazine-3,5-dione 5-oxime	5354
5-Methyl-1-phenyl-1H-tetrazole	4881	N-Phenylpiperazine	5448
Methyl 3-(phenylthio)-3-(² H ₁)propenoate	5329	3-Phenylpropionaldehyde	5134
2-(2-Methyl-2-propenyl)cyclohexanone	5497	2-Phenylpropionamide	5167
N'-Methylpyridine-2,3-dicarboximide	4832	DL-2-Phenylpropionic acid	5142
6-Methyl-2-(2-pyridyl)-1,3-thiazin-4-one	5311	2-Phenylsuccinic acid	5369
2-[2-(Methylthio)phenyl]ethylammonium chloride	5215	3-(Phenylsulfonyl)propionic acid	5159
Methyl para-tolyl sulfone	4924	3-Phenylthiazolidine-2,4-dione	5068
Methyl para-tolyl sulfoxide	4921	(2-Phenyl)thioacetamide	4910
3-Methyl-1-para-tolyltriazene	4944	3-(Phenylthio)propionic acid	5144
Methyl 3-(trimethylsilylmethyl)-3-butenoate	5253	4-Phenyl-1,2,4-triazolidine-3,5-dione	4863
Naphthalene	5300	1-Phenylvinyl acetate	5363
2,3-Naphthalenediol	5316	Phenyl vinyl sulfone	4888
2,6-Naphthalenediol	5317	Phenyl vinyl sulfoxide	4884
2,7-Naphthalenediol	5318	exo-2-Pinen-4-ol	5498
1-Naphthalenesulfonyl chloride	5293	Potassium 1,4-dioxo-1,4-dihydronaphthalene-2-sulfonate	5280
Naphthionic acid	5338	Potassium hydrogen phthalate	4810
1-Naphthylamine	5331	Potassium 3-indoleacetate	5307
3,3',3"-Nitrilotripropionamide	5247	2-Propionoxybenzoic acid	5370
3'-Nitroacetophenone oxime	4878	Propyl 2-furoate	4925
4-Nitrobenzyl chloroformate	4823	N-Propylheptanamide	5568
4-Nitro-1-naphthoic acid	5591	N-Propylhexanamide	5267
2-(3-Nitrophenoxy)-5-nitropyridine	5593	N-Propylisovaleramide	5017
2-(2'-Nitrophenoxy)pyridine	5598	N-Propyl-4-methylpentanamide	5268
5-Nitro-2-phenoxypyridine	5599	4-Propylphenol	5188
2-Nitrophenylacetic acid	4860	2-Propylpyridine	4935
2-(4-Nitrophenyl)-1,3-dithiolane	5104	Propyl 3,4,5-trihydroxybenzoate	5421
4-Nitroquinoline 1-oxide	5051	N-Propylvaleramide	5018
Nitropthalic acid	4814	1-Propynylbenzene	5072
4-Nitro-2-vinylaniline	4876	2-Propynyl 3,5-dinitrobenzoate	5284
Nonanenitrile	5241	2-Propynyl 2-oxanyl ether	4961
(Z)-6-Nonenol	5249	N ¹ -Pyridin-N ² -(ethoxycarbonyl)acetamidin-N ¹ -ide	5440
Nonylamine	5276	4-(1-Pyrrolidinyl)-3-penten-2-one	5227
Nonyl chloroformate	5541	4-(1-Pyrrolidinyl)pyridine	5182
2-Nonyne	5230	Quinazoline	4828
3-Nonyne	5231	2,4-Quinolinediol	5067
4-Nonyne	5232	2-Quinolinol	5064
4-Nonyn-3-one	5221	8-Quinolyl methanesulfonate	5339
2,2,3,3,4,4,5,5-Octafluoro-2'-hydroxy-4'-nitrovaleranilide	5587	Sodium 4-acetamidozenesulfonate	4873
Octanal	5008	Sodium (2-carbamoylphenoxy)acetate	5074
Octanedioic acid	4992	Sodium 5,5-diethyl-4,6-dioxo-3,4,5,6-tetrahydropyrimidin-2-olate	4943
1,3-Octanediol	5026	Sodium 4-dimethylaminophenyldiazosulfonate	4920
Octanethiol	5027	Sodium 1-naphthol-5-sulfonate	5298
(E)-2-Octene	4998	Sodium 2-naphthol-6-sulfonate	5299
2-Octyne	4970	Sodium 1-nonanesulfonate	5269
3-Octyne	4971	Sodium octanoate	4995
1-Octyn-3-ol	4981	Spiro[3.3]heptane-2,6-dicarboxylic acid	5202
2-Octyn-1-ol	4982	Spiro[4.4]nona-2,7-diene-1,6-dione	5077
7-Oxabicyclo[2.2.1]heptane-2,3-dicarboxylic acid	4928	Terephthalaldehydic acid	4835
1-Oxabisp[4.5]dec-3-en-2-one	5195	Terephthalamide	4877
(3-Oxa-5-thiahexyl)benzene	5461	Tetraethylgermane	5037
4-Oxatricyclo[5.2.1.0 ^{2,6}]dec-8-en-3-one	5141	Tetraethyl orthocarbonate	5275
3-(3-Oxobutyl)cyclohexanone	5504	5,6,7,8-Tetrahydro-4H-cyclohepta[c]furan-6-one	5143
7-Oxo-1,3,5-cycloheptatrienyl thiocyanate	4811	1,2,3,4-Tetrahydroisoquinoline	5166
1,1,4,7,7-Pentamethylidihylenetriamine	5278	1,2,3,4-Tetrahydro-1-naphthol	5403
2,5,8,11,14-Pentaoxapentadecane	5581	5,6,7,8-Tetrahydro-1-naphthylamine	5427
N-Pentylbutyramide	5265	1,2,4,5-Tetrakis(bromomethyl)benzene	5347
N-Pentylisobutyramide	5266	4,4',5,5'-Tetrakis(methylthio)-2,2'-bi-1,3-dithiolylidene	5423
N-Pentylisovaleramide	5566	2,3,4,5-Tetramethylaniline	5483
N-Pentylpropionamide	5016	Tetramethyl-para-benzoquinone	5410
N-Pentylvaleramide	5567	1,1,3,3-Tetramethylbutylamine	5031
2-(1-Perhydroazocinyl)ethylguanidine sulfate	5572	1,1,3,3-Tetramethyl-1,3-divinyldisilazane	5035
2-Phenoxyethyl acetate	5416	1,1,3,3-Tetramethyl-1,3-divinyldisiloxane	5025
Phenylacetamide	4901	Tetramethylene diacetate	4993
D-Phenylalanine	5168		

Chart 2. The Molecular Formula Index

name	No.	name	No.
C ₁₀ H ₁₃ K ₃ N ₂ O ₈ Tripotassium hydrogen ethylenediaminetetraacetate	5426	C ₁₀ H ₁₄ O ₄ 3,4,5-Trimethoxybenzyl alcohol	5469
C ₁₀ H ₁₃ N 5,6,7,8-Tetrahydro-1-naphthylamine	5427	4,7,7-Trimethyl-3-oxo-2-oxabicyclo[2.2.1]heptane-1-carboxylic acid	5470
C ₁₀ H ₁₃ NO N-Benzylpropionamide	5428	C ₁₀ H ₁₄ O ₄ Pd Bis(2,4-pentanedionato-O,O')palladium(II)	5471
3-(Dimethylamino)acetophenone	5429	C ₁₀ H ₁₄ O ₅ Methyl 8-oxo-1,4-dioxaspiro[4.5]decane-7-carboxylate	5472
4-(Dimethylamino)acetophenone	5430	C ₁₀ H ₁₄ S ₂ 4,5-Bis(mercaptomethyl)-ortho-xylene	5473
C ₁₀ H ₁₃ NO ₂ 3-Hydroxy-3-phenylbutyramide	5431	C ₁₀ H ₁₅ Br 2-Bromotricyclo[3.3.1.1 ^{3,7}]decane	5474
Methyl 2-dimethylaminobenzoate	5432	C ₁₀ H ₁₅ Cl 1-Chlorotricyclo[3.3.1.1 ^{3,7}]decane	5475
2,3,4,5-Tetramethyl-1-nitrobenzene	5433	C ₁₀ H ₁₅ D (¹ H ₁)Tricyclo[3.3.1.1 ^{3,7}]decane	5476
C ₁₀ H ₁₃ NO ₂ S N-Allyl-para-toluenesulfonamide	5434	C ₁₀ H ₁₅ IO ₅ 3-O-Acetyl-5-deoxy-5-iodo-1,2-O-isopropylidene-alpha-D-xylofuranose	5477
C ₁₀ H ₁₃ NO ₃ 3,4-Dimethoxyacetanilide	5435	C ₁₀ H ₁₅ N 1,1-Dimethyl-2-phenylethylamine	5478
L-Tyrosine methyl ester hydrochloride	5436	2-(1-Ethylpropyl)pyridine	5479
C ₁₀ H ₁₃ NO ₃ S 4-(Benzenesulfonyl)morpholine	5437	4-(1-Ethylpropyl)pyridine	5480
C ₁₀ H ₁₃ NO ₄ Methyl 4-amino-2,5-dimethoxybenzoate	5438	N-Isopropylbenzylamine	5481
C ₁₀ H ₁₃ N ₃ O ₂ (1-Phenylethyl)oxamohydrazide	5439	1-Methyl-3-phenylpropylamine	5482
N ¹ -Pyridinio-N ² -(ethoxycarbonyl)acetamidin-N ¹ -ide	5440	2,3,4,5-Tetramethylaniline	5483
C ₁₀ H ₁₃ N ₅ O ₅ Guanosine	5441	C ₁₀ H ₁₅ NO 1-(3-Dimethylaminophenyl)ethanol	5484
C ₁₀ H ₁₄ meta-Cymene	5442	C ₁₀ H ₁₅ NO ₂ 3-(1-Hydroxy-3-azapentyl)phenol hydrochloride	5485
meta-Diethylbenzene	5443	C ₁₀ H ₁₅ NO ₂ S 4-Dimethylaminobenzyl methyl sulfone	5486
ortho-Diethylbenzene	5444	C ₁₀ H ₁₅ NO ₃ Ethyl 3-oxo-2-(2-pyrrolidinylidene)butyrate	5487
para-Diethylbenzene	5445	C ₁₀ H ₁₅ N ₂ Na ₃ O ₇ Trisodium N'-(2-hydroxyethyl)ethylenediamine-N,N,N'-triacetate dihydrate	5488
1-Ethyl-2,4-dimethylbenzene	5446	C ₁₀ H ₁₅ N ₃ O 6-Methyl-4-(1-methylpropylidenehydrazino)-2(1H)-pyridone	5489
C ₁₀ H ₁₄ CaO ₆ Calcium bis(3-methyl-2-oxobutyrate)	5447	C ₁₀ H ₁₅ P Diethylphenylphosphine	5490
C ₁₀ H ₁₄ N ₂ N-Phenylpiperazine	5448	C ₁₀ H ₁₆ (+)-3-Carene	5491
C ₁₀ H ₁₄ N ₂ O N,N-Diethyl-4-nitrosoaniline	5449	C ₁₀ H ₁₆ CIN Benzyltrimethylammonium chloride	5492
4,6,7,8,9,10-Hexahydro-2-methylpyrimidino[1,2-a]azepin-4-one	5450	C ₁₀ H ₁₆ N ₂ N,N-Diethyl-para-phenylenediamine sulfate	5493
C ₁₀ H ₁₄ N ₂ O ₂ N,N-Diethyl-2-nitroaniline	5451	N,N,N',N'-Tetramethyl-para-phenylenediammonium dichloride	5494
C ₁₀ H ₁₄ N ₅ O ₇ P Adenosine 5'-monophosphate	5452	C ₁₀ H ₁₆ N ₂ O ₄ N,N'-Diallyl-L-tartaramide	5495
C ₁₀ H ₁₄ O 2-tert-Butylphenol	5453	C ₁₀ H ₁₆ O Bicyclo[4.4.0]decan-3-one	5496
4-sec-Butylphenol	5454	2-(2-Methyl-2-propenyl)cyclohexanone	5497
2-(1-Hydroxyethyl)-1,4-dimethylbenzene	5455	exo-2-Pinen-4-ol	5498
1,4(8)-para-Menthadien-3-one	5456	Tricyclo[3.3.1.1 ^{3,7}]decan-1-ol	5499
1,8-para-Menthadien-3-one	5457	Tricyclo[3.3.1.1 ^{3,7}]decan-2-ol	5500
7-Methylenebicyclo[3.3.1]nonan-3-one	5458	C ₁₀ H ₁₆ O ₂ 4-tert-Butylcyclohexane-1,3-dione	5501
3,4,5-Trimethylanisole	5459	(Z)-3-Hexenyl crotonate	5502
2,6,6-Trimethyl-2,4-cycloheptadienone	5460	4-Methyl-3-(3-oxobutyl)-4-pentenal	5503
C ₁₀ H ₁₄ OS (3-Oxa-5-thiahexyl)benzene	5461	3-(3-Oxobutyl)cyclohexanone	5504
C ₁₀ H ₁₄ O ₂ 3-Butoxyphenol	5462		
4-Butoxyphenol	5463		
tert-Butylhydroquinone	5464		
1-Methoxy-4-methylbicyclo[2.2.2]oct-5-en-2-one	5465		
3-Phenyl-1,3-butenediol	5466		
C ₁₀ H ₁₄ O ₂ S Isopropyl para-tolyl sulfone	5467		
C ₁₀ H ₁₄ O ₃ 2-Methyl-2-(3-oxobutyl)cyclopentane-1,3-dione	5468		

Chart 3-1. The Code Table of Substructure Notation

Priority	Code	Substructure	Priority	Code	Substructure
1	CH3	-CH ₃	26	=N=	=N=
2	CH2	-CH ₂ -	27	N[*]	aromatic nitrogen
3	CH2U	-CH ₂ - a]	28	NO2	-NO ₂
4	CH	-CH<	29	# N	≡N
5	C	>C<	30	OH	-OH
6	CH2=	=CH ₂	31	O	-O-
7	CH=	=CH-	32	=O	=O
8	CH=[E]	=CH- b]	33	SH	-SH
9	CH=[Z]	=CH- c]	34	S	(HARA)-S-
10	C=	=C<	35	=S	(HDO)=S<
11	=C=	=C=	36	SO2	-SO ₂
12	# CH	≡CH	37	P	-P<
13	C#	≡C-	38	PH=	=PH
14	AH	d]	39	P=	-P= or >P=
15	A	e]	40	F	-F
16	TRL	f]	41	Cl	-Cl
17	CHO	-CHO	42	Br	-Br
18	C=O	>C=O	43	I	-I
19	=C=O	=C=O	44	As	>As-
20	NH3 ⁺	ammonium cation	45	As=	=As-
21	NH2	-NH ₂	46	Si	>Si<
22	NH	>NH	47	+	cation
23	N	>N-	48	-	anion
24	=NH	=NH	49	•	radical
25	N=	-N=	50	/	chelation
			51	g]	other atoms

a) Unequivalent methylene

b) Entgegen

c) Zusammen

d) Aromatic carbon with proton

e) Aromatic carbon without proton

f) -C(OH)-C(O)- in troponoid

g) Other element symbols are listed in alphabetical order.

Chart 3-2. The Substructure Index

substructure	chemical shift	No.	substructure	chemical shift	No.
CH2/CH(CH3,CH3),NH(C=O)	3.09	5014	CH2/A(AH,AH),C=O(OH)	3.64	4852
CH2/CH(CH3,CH3),NH(C=O)	3.09	5561	CH2/A(AH,AH),C=O(O)	3.74	5108
CH2/CH(CH3,CH3),NH(C=O)	3.09	5562	CH2/A(AH,A),C=O(C=O)	4.58	5046
CH2/CH(A,OH),NH(CH2)	ca. 3.32	5485	CH2/A(AH,A),C=O(N)	3.50	5099
CH2/CH(CH,NH2),OH	3.29	5184	CH2/A(AH,A),C=O(OH)	3.80	4824
CH2/CH(CH2,CH),O(CH3)	4.22	5535	CH2/A(AH,A),C=O(OH)	4.02	4860
CH2/CH(CH,O),O(C)	3.40- 4.40	4987	CH2/A(AH,AH),NH2	3.69	4950
CH2/CH(CH3,CH3),S(S)	2.60	5028	CH2/A(AH,AH),NH2	3.80	4939
CH2/CH(CH2U,CH),S(CH2)	1.50- 2.01	5181	CH2/A(AH,AH),NH2	3.81	4933
CH2/CH(CH2,O),Br	3.29- 3.81	5509	CH2/A(AH,AH),NH2	3.84	4938
CH2/CH(CH,I),I	3.07- 3.40	5477	CH2/A(AH,AH),NH2	3.90	4949
CH2/C(CH3,CH3,CH3),C(CH3,CH3,NH2)	1.44	5031	CH2/A(AH,A),NH2	3.82	4937
CH2/C(CH3,CH3,OH),CH=(E)(CH=(E))	ca. 2.15	5554	CH2/A(AH,AH),NH(CH3)	3.75	4934
CH2/C(CH3,CH3,OH),CH=(E)(CH=(E))	2.17	5552	CH2/A(AH,AH),NH(CH2)	3.80	5208
CH2/C(CH3,CH3,NH2),A(AH,AH)	2.66	5478	CH2/A(AH,AH),NH(CH2)	3.80	5219
CH2/C(CH3,CH3,CH=),C=O(C=)	2.67	5460	CH2/A(AH,AH),NH(CH2)	3.82	4952
CH2/C(CH3,CH2,CH2),C=O(NH)	2.43	4965	CH2/A(AH,AH),NH(CH)	3.79	5481
CH2/C(CH3,CH2,CH2),C=O(OH)	2.30	4991	CH2/A(AH,AH),NH(C=O)	4.43	5428
CH2/C(CH3,CH2,CH=),C=O(C)	2.03	5465	CH2/A(AH,A),NH(CH2)	4.00	5166
CH2/C(F,F,F),O(C=O)	4.73	5063	CH2/A(AH,AH),N(CH3,CH3,CH3,+)	5.06	5492
CH2/C(CH3,CH3,A),Cl	3.65	5425	CH2/A(AH,AH),OH	4.48	4922
CH2/CH=(CH2=),A(AH,A)	3.37	5400	CH2/A(AH,AH),OH	4.52	4889
CH2/CH=(CH=),A(AH,A)	3.40	5071	CH2/A(AH,AH),OH	4.61	5469
CH2/CH=(CH2=),NH(A)	3.79	5165	CH2/A(AH,AH),OH	4.76	5151
CH2/CH=(CH2=),NH(C=O)	3.74	5495	CH2/A(AH,A),OH	4.52	5194
CH2/CH=(CH2=),NH(SO2)	3.60	5434	CH2/A(AH,A),OH	4.53	5198
CH2/CH=(CH2=),O(A)	4.53	5056	CH2/A(AH,A),OH	4.69	5199
CH2/CH=(CH2=),O(A)	4.61	5332	CH2/A(AH,A),OH	4.80	5197
CH2/CH=(CH2=),O(C=O)	4.75	5366	CH2/A(AH,TPL),OH	4.60	5092
CH2/CH=(CH2=),O(C=O)	4.87	5103	CH2/A(A,A),OH	4.61	4941
CH2/CH=(CH2=),S(A)	3.51	5422	CH2/A(AH,AH),O(CH2)	4.57	5192
CH2/CH=(CH2=),S(A)	3.55	5162	CH2/A(AH,AH),O(CH2)	4.59	5164
CH2/CH=(E)(CH=(E)),C=(CH3,CH2=)	2.70	5520	CH2/A(AH,AH),O(C=O)	5.10	4904
CH2/CH=(E)(CH=(E)),OH	4.15	4960	CH2/A(AH,AH),O(C=O)	5.40	4823
CH2/CH=(E)(C=(E)),OH	4.16	5530	CH2/A(AH,A),O(C=O)	5.20	5368
CH2/CH=(E)(CH=(E)),O(A)	4.46	5325	CH2/A(AH,AH),SH	3.72	4930
CH2/CH=(E)(CH=(E)),O(C=O)	4.53	5532	CH2/A(AH,A),SH	3.81	5473
CH2/CH=(Z)(CH=(Z)),O(C=O)	4.68	4962	CH2/A(AH,AH),S(C=)	4.59	4919
CH2/C-(CH2,N=),A(AH,A)	3.86	5100	CH2/A(AH,AH),SO2(CH3)	4.14	5486
CH2/C-(NH2,-S),A(AH,AH)	4.11	4910	CH2/A(AH,AH),SO2(Cl)	4.84	4898
CH2/C-(CH2,CH2=),C=O(O)	3.00	5253	CH2/A(AH,A),SO2(Cl)	4.96	5163
CH2/C-(CH2,N=),C=O(N)	3.59	5354	CH2/A(AH,AH),P=(O,O,=O)	3.18	5216
CH2/C-(CH2,N=),C=O(N)	3.61	5328	CH2/A(AH,AH),Br	4.50	4895
CH2/C-(CH=,A),C=O(OH)	3.53	5336	CH2/A(AH,A),Br	4.60	5347
CH2/C-(CH=,A),C=O(O)	3.67	5307	CH2/A(AH,AH),D	2.29	5117
CH2/C-(CH2,N=),O(N)	5.00	5354	CH2/C=O(CH2),C=O(CH)	3.40	5501
CH2/C-(CH2,N=),O(N)	5.10	5328	CH2/C=O(CH2),C=O(N)	3.80	5305
CH2/C-(CH=,A),O(C=O)	5.23	5319	CH2/C=O(CH2),C=O(N)	3.80	5337
CH2/C-(CH=,O),S(S)	3.71	5364	CH2/C=O(CH2),C=O(N)	3.81	5304
CH2/C-(CH=,C=),Br	4.46	5324	CH2/C=O(CH2),C=O(O)	3.62	5226
CH2/C-(CH=,C=),Br	4.53	5343	CH2/C=O(C),C=O(C=)	3.50	5506
CH2/C-(CH=,C=),Br	4.55	5324	CH2/C=O(O),C=O(O)	6.47	4808
CH2/C-(CH=,C=),Br	4.60	5343	CH2/C=O(NH),NH(C=O)	3.31	5239
CH2/C-(CH=,C=O),Br	4.46	5324	CH2/C=O(OH),NH(C=O)	3.87	5394
CH2/C-(CH=,C=O),Br	4.53	5343	CH2/C=O(OH),NH(C=O)	3.88	5125
CH2/C-(CH=,C=O),Br	4.55	5324	CH2/C=O(O),NH(C=O)	3.92	5387
CH2/C-(CH2,CH2=),Si(CH3,CH3,CH3)	1.63	5253	CH2/C=O(OH),N(CH2,CH2)	4.53	4955
CH2/C-(E)(CH=(E),C=O),C=O(O)	3.98	5204	CH2/C=O(O),N(CH2,CH2)	3.63	5426
CH2/C-(E)(CH3,CH=)(E),OH	4.00	5530	CH2/C=O(O),N(CH2,CH2)	3.16	5488
CH2/C#(#N),A(AH,AH)	3.61	4874	CH2/C=O(CH3),N*(AH,AH,+)	3.38	5393
CH2/C#(#N),A(AH,AH)	3.68	5102	CH2/C=O(CH3),N*(AH,AH,+)	3.63	5426
CH2/C#(#N),A(AH,AH)	3.71	5096	CH2/C=O(O),OH,	5.94	4913
CH2/C#(#N),A(AH,AH)	3.72	4805	CH2/C=O(CH2),O(N)	4.40	5323
CH2/C#(#N),A(AH,AH)	3.80	5308	CH2/C=O(CH2),O(N)	4.56	5337
CH2/C#(#N),A(AH,A)	3.68	5101	CH2/C=O(CH2),O(N)	4.57	5305
CH2/C#(#N),A(AH,A)	3.81	4803	CH2/C=O(A),O(C=O)	4.69	5304
CH2/C#(#N),A(A,A)	4.02	4804	CH2/C=O(OH),O(A)	5.43	5323
CH2/C#(#N),C=O(N)	3.39	5113	CH2/C=O(OH),O(A)	4.61	5152
CH2/C#(C#),OH	4.25	4982	CH2/C=O(OH),O(A)	4.66	5417
CH2/C#(CH#),O(CH)	ca. 4.27	4961	CH2/C=O(OH),O(A)	4.71	5158
CH2/C#(CH#),O(C=O)	5.09	5284	CH2/C=O(O),O(A)	4.75	4826
CH2/C#(CH#),Br	3.93	4964	CH2/C=O(O),O(A)	4.62	5074
CH2/A(AH,AH),C=O(CH2)	3.69	5402	CH2/C=O(N),S(C=)	4.62	5378
CH2/A(AH,AH),C=O(NH2)	3.58	4901	CH2/C=O(N),S(C=)	3.19	5333
CH2/A(AH,AH),C=O(OH)	3.63	4851	CH2/C=O(N),S(C=O)	4.22	5334
				4.13	5068

Chart 4. The Chemical Shift Index

chemical shift	substructure	No.	chemical shift	substructure	No.
1.50- 2.50	CH2/CH(CH2,C=O),CH=(C=)	4976	1.65	CH2/CH3,CH2(C=O)	5532
1.50- 2.50	CH/CH2(CH2),CH2(CH=),C=O(NH)	4976	1.65	CH2/CH3,CH2(C=O)	5533
1.50- 2.90	CH2/CH2(C=O),CH(C,C=O)	5501	ca. 1.65	CH2/CH2(CH3),CH2(A)	5206
1.50- 2.90	CH2/CH2(CH),C=O(CH2)	5501	ca. 1.65	CH2/CH2(CH2),CH(CH2,OH)	5579
1.50- 2.90	CH/CH2(CH2),C(CH3,CH3,CH3),C=O(CH2)	5501	ca. 1.65	CH2/CH2(CH2),CH(C=O,NH)	4981
1.51	CH3/CH(A,C=O)	5142	ca. 1.65	CH2/CH2(CH2),CH(C=O,NH)	4974
1.51	CH3/CH(A,C=O)	5167	1.65	CH2U/CH(CH,CH=),CH(CH,CH=)	5141
1.51	NH2/CH2(A)	4949	1.65	CH2U/CH(CH,A),CH(CH,C=O)	5361
ca. 1.52	CH2/CH2(CH3),CH2(C=O)	5260	1.65	OH/(CH3,CH3,CH2)	5554
1.52	CH/CH3,CH3,CH2(CH=(E))	5520	1.65- 2.30	CH2/CH2(C=O),C(CH3,C=,C=O)	5409
1.52- 2.00	CH2/CH2(CH2),CH(CH2,NH)	5257	1.66	CH3/CH(C=O,O)	5091
1.53	CH3/C(CH3,CH3,O)	5539	1.66	CH3/C-(CH2,CH=)	4976
1.53	CH3/C(CH3,O,O)	5477	1.66	CH2/CH3,CH2(CH=)	5399
1.53	CH2/CH3,CH2(NH)	5017	ca. 1.66	CH2/CH2(CH2),CH2(NH)	5265
1.53	CH2/CH3,CH2(NH)	5568	1.67	CH2/CH3,CH2(O)	5421
1.53	CH2/CH2(CH2),CH2(CH)	4984	1.67	OH/CH(CH,CH)	5500
1.53	CH2/CH2(CH2),CH(CH2U,O)	4984	1.67- 1.97	CH2/CH2(C=O),CH(CH3,C=O)	4990
1.53	CH2/CH(CH,CH=),CH(CH,CH=)	5105	1.68	CH3/C=(CH3,CH=)	5526
ca. 1.53	CH/CH3,CH3,CH(CH3,CH2)	5270	1.68	CH2/CH(CH2,CH2),CH(CH2,CH2)	5475
ca. 1.53	CH/CH3,CH2(CH2),CH(CH3,CH3)	5270	1.68	OH/CH(CH,CH=)	5498
1.53	NH2/CH2(A)	4938	1.68- 2.18	CH/CH3,CH3,CH(CH2,CH)	5553
1.53- 2.00	CH2/CH2(CH2),CH2(CH2)	5450	1.68- 2.18	CH/CH3,CH2(CH2),CH2(CH)	5553
1.53- 2.00	CH2/CH2(CH2),CH2(C=)	5450	1.68- 2.18	CH/CH2(CH2),CH(CH3,CH3),CH(CH2,OH)	5553
1.53- 2.00	CH2/CH2(CH2),CH2(N)	5450	1.69	CH3/C=(CH3,CH=)	5236
1.54	CH3/CH(C=O,NH)	5384	1.69	CH3/C=(CH3,CH=)	5530
1.54	CH3/C(C=O,O,O)	5255	1.69	CH3/C=(E)(CH2,CH=(E))	5036
1.54	CH2/CH2(CH2),CH2(C=O)	4995	1.69	CH2/CH3,P=(O,O,O)	5582
ca. 1.54	CH/CH3,CH3,CH2(CH=(E))	5519	1.69	CH2/CH2(CH2),CH2(S)	5578
1.55	CH3/CH(A,NH)	5439	ca. 1.69	CH2/CH2(OH),CH(CH2,OH)	5007
1.55	CH3/CH(C=O,NH2)	5001	1.70	CH3/C=(CH3,CH=)	5527
1.55	CH3/CH(C=O,NH2)	5003	1.70	CH3/C=(CH3,CH=)	5497
1.55	CH3/CH(C=O,NH2)	5245	1.70	CH3/C=(CH2,CH2=)	5272
1.55	CH2/CH3,CH(CH3,A)	5454	1.70	CH2/CH3,CH2(C=O)	5254
1.55	CH2/CH2(CH2),CH2(SH)	5027	1.70	CH2/CH3,CH(C=O)	5505
ca. 1.55	CH/CH3,CH3,C=O(NH)	5262	1.70	CH2/CH3,C(C,CH=,OH)	5195
1.56	CH3/CH(C=O,NH2)	5002	ca. 1.70	CH2/CH2(CH2),CH2(CH2)	5195
ca. 1.56	CH2/CH2(CH=),C(CH3,CH=,OH)	5526	ca. 1.70	CH2/CH2(CH2),CH2(C)	5195
1.57	CH3/CH(CH3,N)	5395	1.70	CH2/CH2(CH2),CH2(O)	4993
1.58	CH3/C(CH2U,A,OH)	5431	1.70	CH2/CH2(OH),CH(CH2,OH)	5026
ca. 1.58	CH2/CH2(CH2),CH2(C#)	5241	ca. 1.70	CH2/CH2(CH2),C(CH2,CH=,O)	5195
1.58	CH2/CH2(C=O),CH(CH2,C)	5220	1.70	CH2/CH(CH2,CH2),CH(CH2,CH2)	5511
1.58- 1.92	CH2/CH2(CH2),CH(C=O,NH)	5246	1.70	CH2U/CH(CH2,S),CH(CH2U,O)	4983
1.59	CH3/CH2(N)	5127	ca. 1.70	CH2U/CH(CH,A),CH(CH,C=O)	5362
1.59	CH3/C(CH2,A,OH)	5466	ca. 1.70	CH/CH2U(CH),CH(CH2U,A),C=O(OH)	5362
1.60	CH3/C=(CH3,CH=)	5526	1.70	OH/CH(CH2U,A)	5133
1.60	CH3/C=(CH2,CH=)	5491	1.70- 2.07	CH2/CH2(CH),CH2(C=O)	5529
1.60	CH3/C=(CH,CH=)	5009	1.70- 2.07	CH2/CH2(CH2),CH(CH2,O)	5529
1.60	CH2/CH3,CH2(A)	5188	1.71	CH2/CH3,CH(CH2,A)	5479
ca. 1.60	CH2/CH3,CH2(C=O)	5551	1.72	CH3/CH(C=O,O)	5090
ca. 1.60	CH2/CH3,CH2(NH)	5267	1.72	CH3/C=(CH2,CH2=)	5520
ca. 1.60	CH2/CH2(CH2),CH2(C=O)	5551	1.72	CH3/C=O(CH2U)	5386
ca. 1.60	CH/CH3,CH3,CH2(CH2)	4969	ca. 1.72	CH2/CH2(CH3),CH2(O)	5462
1.60	CH/CH3,CH2(CH3),CH2(CH)	5021	1.72	CH2/CH2(CH=),CH(CH2,C=O)	4958
1.60	NH/CH2(CH2),CH2(CH2)	5448	1.72	OH/CH(CH,C#)	4980
1.60	OH/CH(CH2,CH)	5524	1.72	SH/CH2(A)	4930
1.60- 1.90	CH2/CH2(CH3),CH(C=O,NH)	5001	1.72- 2.10	CH/CH3,CH2(CH3),CH(C=O,NH)	5245
1.60- 2.00	CH2/CH2(CH2),CH(C=O,OH)	5010	1.73	CH3/C(CH3,C#,N=)	4956
1.60- 2.66	CH2/CH2(C),C(CH3,C,C=O)	5470	1.73	CH3/C=(CH,CH=)	5498
1.60- 2.66	CH2/CH2(C),C(C,C=O,O)	5470	1.73	CH2/CH2(CH3),CH2(O)	5463
1.61	CH3/C=(CH3,CH=)	5527	1.73	CH2/CH2(CH2),CH2(O)	5541
1.61	CH2/CH2(CH3),CH2(NH)	5012	1.73	CH2/CH2(CH2),CH2(SO2)	5269
1.62	CH3/C=(CH3,CH=)	5007	1.74	OH/CH2(CH=(E))	5530
1.62	CH3/C=(CH3,CH=)	5236	1.74	OH/CH2(C=(E))	5530
1.62	CH3/C=(CH3,CH=)	5480	1.75	CH3/C=(CH3,CH=)	5519
1.62	CH2/CH3,CH(CH2,A)	5008	1.75	CH3/C=(CH3,C=)	4940
1.62	CH2/CH2(CH2),CH2(CHO)	5561	1.75	CH3/C=(CH,CH2=)	5457
ca. 1.62	CH2/CH2(CH2),CH2(C=O)	5559	ca. 1.75	CH/CH3,CH3,CH2(NH)	5260
1.62	CH2/CH2(CH2),CH2(NH)	4953	1.76	CH2/CH3,CH2(A)	4935
1.63	CH3/C(CH3,CH3,O)	5032	1.76	CH2/CH3,CH2(O)	4925
ca. 1.63	CH2/CH2(CH2),NH(CH2)	5032	1.76	CH2/CH3,C(O,O,O)	5274
1.63	CH2/C=(CH2,CH2=),Si(CH3,CH3,CH3)	5253	1.76	CH/CH3,CH3,CH2(NH)	5562
1.63	NH2/CH2(A)	4937	1.76	CH3/CH=(E)(CH=(E))	5325
1.64	CH2/CH2(A),CH(CH3,NH2)	5482	1.77	CH2/CH2(CH2),CH2(Cl)	5256
ca. 1.64	CH2/CH2(C=O),CH(CH2,C=)	5503	1.77	OH/CH(CH3,A)	5455
1.65	CH3/C=(E)(CH=(E))	4998	1.77	CH3/C#(C#)	4969
1.65	CH3/C=(CH,CH2=)	5503	1.78	CH2/CH2(CH2),CH2(C=)	5547
1.65	CH2/CH3,CH2(C=O)	5273	1.78		

Chart 5. Code Table of Sample Suppliers

Code	Name	Appliation	Code	Name	Appliation
AB01	Abe, Nobuo	Akita University	NA01	Nakasuga, Noriyuki	Nagoya University
AB02	Abe, Yoshio	Keio University	NA02	Nakagawa, Shigeki	Nippon University
AD01	Adachi, Kazuo	Osaka Institute of Technology	NA03	Hoshino, Masamatsu	Saitama University
AL		Aldrich Chemical Co.	NA04	Nagata, Masanori	Toyama Medical and Pharmaceutical University
AN01	Ando, Masayoshi	Tohoku University	NI01	Nishimura, Jun	Kyoto University of Industrial Art and Textile
AR01	Arai, Mannosuke	Tohoku University	NI02	Nishiyama, Shigeru	Keio University
AS01	Asao, Toyonobu	Tohoku University	NO01	Nonaka, Tsutomu	Tokyo Institute of Technology
AS02	Asami, Masatoshi	Yokohama National University	NO02	Nomura, Keiichi	Toyama Medical and Pharmaceutical University
FU01		Kao Corporation	NU01		Numazu College of Technology
FU02	Fujise, Yutaka	Tohoku University	OG01	Ogino, Hiroshi	Tohoku University
FU03		Asahi Chemical Industry Co., Ltd.	OJ01	Ojima, Juro	Toyama University
GO01	Goromaru, Tsuyoshi	The University of Tokushima	OO01	Osawa, Keisuke	Tohoku Pharmaceutical College
HA01	Hamada, Yoshiki	Meijo University	OO02	Onuma, Hiroshi	Akita University
HA02	Hamada, Keinosuke	Nagasaki University	OO03	Omura, Satoshi	Kitasato University
HA03	Harada, Nobuyuki	Tokoku University	OO04		BASF Japan Co., Ltd.
HA04	Hagiwara, Hisahiro	Tohoku University	OZ01	Ozawa, Fumiuki	Tokyo Institute of Technology
HA05		Toyo Jozo Co., Ltd.	SA01	Sato, Gen	Sophia University
HI01	Hirose, Yoshiyuki	Science University of Tokyo	SA02		Fuji Photo Film Co., Ltd.
HI02	Hirota, Hiroshi	The University of Tokyo	SA03	Sato, Masaru	Saitama University
HO01	Horiike, Michio	Kochi University	SA04		Konishiroku Photo Ind. Co., Ltd.
HO02		Nippon Kayaku Co., Ltd.	SA05	Minabe, Masahiro	Utsunomiya University
HO03	Iino, Masashi	Tohoku University	SE01	Senda, Yasuhisa	Yamagata University
IK01	Ikeda, Masazumi	Hokko Chemical Industry Co., Ltd.	SI01	Shibata, Hisao	Shinshu University
IK02	Ikariya, Takao	Osaka University	SI02		Asahi Chemical industry Co., Ltd.
IS01	Ishida, Shin-ichiro	The University of Tokyo	SU01	Suzuki, Minoru	Hokkaido University
IS02	Ishikawa, Yukihiko	Kanazawa University	SU02	Suzuki, Toshio	Akita University
IS03		Tottori University	SU03	Suzuki, Kanji	Tokyo Institute of Technology
IS04	Ishiyama, Junichi	National Food Research Institute	SU04		Sumitomo Chemical Co., Ltd.
IT01	Ito, Kenji	Tohoku University	SU05		Nippon Soda Co., Ltd.
IT02	Ito, Takashi	Toyohashi University of Technology	TA01	Takatori, Masayuki	Fukushima University
KA		Yokohama National University	TA02	Tabei, Katsumi	Tokyo College of Pharmacy
KA01	Kan, Takayuki	Kanto Chemical Co., Inc.	TA03		Zeria Pharmaceutical Co., Ltd.
KA02	Kasahara, Akira	Hiroshima University	TA04	Tanaka, Kuniyoshi	Kinki University
KA03	Kakei, Shoichi	Yamagata University	TA05	Takeda, Kei	Toyama Medical and Pharmaceutical University
KA04	Kanno, Shunroku	Shinshu University	TA06		Asahi Chemical Industry Co., Ltd.
KA05	Karube, Akio	Yamagata University	TO		Tokyo Kasei Kogyo Co., Ltd.
KA06		Akita National College of Technology	TS01		Sumitomo Bakelite Co., Ltd.
KA07	Kato, Norimoto	Konishiroku Photo Ind. Co., Ltd.	TS02	Tsunetsugu, Josuke	Saitama University
KA08	Kawashima, Takayuki	Tohoku University	TS03		Asahi Chemical Industry Co., Ltd.
KA09	Katayama, Hajime	The University of Tokyo	UE01	Uehara, Tadao	Tohoku University
KI01	Kimura, Yoshiharu	Nigata College of Pharmacy	YA01	Yasuda, Seiichi	Nagoya University
KI02	Kido, Hideo	Kyoto University of Industrial Art and Textile	YA02		Central Glass Co., Ltd.
KI03	Kitahara, Haruo	Tohoku University	YA03	Yamato, Masatoshi	Okayama University
KO01	Kosugi, Hiroshi	Hirosaki University	YA04	Yamamoto, Yutaka	Tohoku Pharmaceutical College
KO02	Komiya, Sanshiro	Tohoku University	YA05	Yasunami, Masafumi	Tohoku University
KO03	Koga, Gen	Tokyo University of Agriculture and Technology	YA06	Yamada, Yoichi	Utsunomiya University
KU01	Kusuyama, Yoshiaki	Ibaraki University	YA07	Yashima, Shigetaka	Hokkaido University
KU02	Kurosawa, Kazu	Wakayama University	YO01		Maruzen Petrochemical Co., Ltd.
MA01	Matoba, Katsuhide	Kumamoto University	YO02		The Green Cross Corporation
MA02	Machida, Katsunosuke	Toyama Medical and Pharmaceutical University			
MA03	Maruyama, Masao	Kyoto University			
MA04	Matsuda, Sumu	Miyagi University of Education			
MA05	Maruyama, Kazuhiro	Hirosaki University			
ME01	Meguro, Hiroshi	Kyoto University			
MO01	Mochida, Kunio	Tohoku University			
MO02		Gakushuin University			
MO03	Morita, Yasuo	Fujisawa Pharmaceutical Co., Ltd.			
		Tohoku Pharmaceutical College			

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