ANIONIC PERMEABILITY OF THE LIVER ER MEMBRANE

Ву

Ashutosh Tripathy

A THESIS

Submitted to
Michigan State University
in partial fulfillment of the requirements
for the degree of

DOCTOR OF PHILOSOPHY

Department of Physiology

1993

ABSTRACT

ANIONIC PERMEABILITY OF THE LIVER ER MEMBRANE

Ву

Ashutosh Tripathy

The ionic pathways present in the liver ER membrane are not known in detail. Studies using ER-derived vesicles have shown that they are permeable to Na⁺, K⁺, choline⁺ and Cl⁻ but less permeable to Ca⁺⁺ and Mg⁺⁺. Though highly permeable to K⁺, the liver ER membrane has been postulated to lack an efficient ion conducting structure for K⁺ like the K⁺, Na⁺ channel in the sarcoplasmic reticulum.

InsP₃, an intracellular second messenger can release Ca⁺⁺ from an intracellular store of many types of cells and that store has been postulated to be the ER. An InsP₃-gated Ca⁺⁺ channel has been shown to exist in canine cerebellar microsomes. But, the the identity of the store in liver tissue is unclear. Though the liver rough ER-derived vesicles have been shown to release Ca⁺⁺ when challenged with InsP₃, the InsP₃- binding sites copurify not with the ER marker, rather with the plasma membrane marker.

The present study was undertaken with the aim to look at the anionic, Ca⁺⁺ and K⁺ permeability pathways present in the ER membrane.

Direct current-voltage recording is a straightforward approach to delineate the ionic pathways present in any membrane. But the membrane of an intracellular organelle like the ER is not accessible to direct cellular patch recording. So, we have fused the liver rough ER-derived vesicles with a planar BLM and have made current-voltage measurements across the reconstituted BLM. In our fusion protocols the vesicles could be readily fused with a BLM by swelling them osmotically in chloride containing solutions.

Using the above experimental approach, We have found that the liver rough ER-derived vesicles possess considerable anionic permeability. The permeability to halides and other anions follows the sequence: SCN⁻ > I⁻ > Br⁻ > CI⁻ >> gluconate⁻, suggesting that the chloride channels have low field-strength sites. It can be pharmacologically dissected to Zn⁺⁺-sensitive and DIDS-sensitive types. DIDS blocked the chloride permeability from the cytoplasmic side of the ER. No InsP₃-gated Ca⁺⁺ channels, rynodine-sensitive Ca⁺⁺ channels and K⁺ channel were found in the liver rough ER membrane.

DEDICATION

To my parents, brothers and sisters and my wife Moni and my son Arnav.

ACKNOWLEDGMENTS

I wish to express my sincere appreciation to my major adviser Dr. H. Ti Tien and to Drs. J. Krier, W. D. Atchison, R. A. Meyer, S. D. Mahanti, my guidance committee members for guidance and encouragement during the course of this study. I thank Dr. Dixon Woodbury of Wane State University for his help in making cups for the BLM studies.

I also wish to thank Dr. J. E. Chimoskey, Chairperson and Dr. C. C. Chou, Director of Graduate Studies, Department of Physiology for encouragement and support.

Special thanks go to my laboratory colleague, M. Zviman for help and cooperation and many interesting discussions on topics ranging from BLM to Bible.

And finally, mere words cannot express my gratitude to my wife Moni for her extreme patience, faith and encouragement and my three-year old son Arnav for enduring my constant absence. Arnav missed me a lot and I missed terribly a beautiful part of his growing up.

TABLE OF CONTENTS

		Page
LIST OF TA	BLES	viii
LIST OF FIG	GURES	ix
INTRODUC	TION	1
	Ca ⁺⁺ Conductance of the Rough ER Membrane Chloride Conductance of the Rough ER Membrane	
EXPERIME	NTAL DESIGN	11
	Isolation of Rough ER Vesicles from Rat Liver Isolation Procedure. Protein Estimation Assay for Marker Enzyme. Planar Bilayer Setup Electrophysiological Setup for Measurement Fusion of ER Vesicles with the BLM A Typical Protocol for a BLM Reconstitution Experiment BLM Forming Solution Preconditioning the Hole Bilayer Formation, Fusion and Recording Buffers and Current - Voltage Measurements	12 13 14 16 19 19
RESULTS		25
	Assay of Marker Enzyme for ER	25 29 33

Channels	44
Permeability Ratios of Halides and Other Anions	44
Inhibition of Chloride Conductance by Zn ⁺⁺ and	- 4
DIDS	54
Studies at the Level of Single Channel	57
DISCUSSION	65
Fusion of ER Vesicles with a BLM	65
In Search of InsP3-gated and Rynodine-Sensitive	
Ca++ Channels in Liver Rough ER Membrane	66
CI- to K+ Permeability Ratios of Liver Rough ER	
Chloride Channels	66
Anionic Permeability Ratios	68
No K+ "Channel" in the ER Membrane	
Inhibition of Chloride Conductance by DIDS and Zinc	
Inhibition by DIDS	
Sidedness of DIDS Block	
Inhibition by Zinc	71
Single Channel Studies	72
Minimal Conductance Induced by ER Vesicle	
Fusion with a BLM	72
Minimal Conductance of a DIDS-Sensitive	
Channel	72
Minimal Conductance of a Zinc-Sensitive	
Channel	73
Physiological Significance of ER Chloride Channels	74
SUMMARY	75
BIBLIOGRAPHY	/ [

LIST OF TABLES

Table	Pag	je
1.	Anionic Permeability Ratios of ER Chloride Channels. Reversal potentials in this study were determined under near bi-ionic conditions in which the <i>cis</i> chamber contained KCl and the <i>trans</i> chamber the test anion. Permeability ratios were calculated using the Goldman-Hodgkin-Katz equation. Anion diameters for halides are taken fron Hille (87).	2
2.	Anionic Permeability Ratios for Different Chloride Channels. Reversal Reversal potentials in this study were determined under near bi-ionic conditions in which the <i>cis</i> chamber contained KCl and the <i>trans</i> chamber the test anion. Permeability ratios were calculated using the Goldman-Hodgkin-Katz equation. The listed permeability ratios for lobster neuron; mouse Gly-R, Gaba-R; CFTR and SR Cl-channels are from literature. n.d not determined.	3
3.	Inhibition of Macroscopic Chloride Conductance by Zn++. Numbers in parentheses in the left column indicate KCI concentration in mM across the BLM ($cis/trans$). The values in the second column indicate the conductance blocked by Zn++. Columns 3 and 4 describe a possible breakdown to unit channel conductance. "Mean unit conductance" taken from the experiments in which both cis and $trans$ KCI concentration was 363 mM are: 0.283 \pm 0.011 pS or 0.589 \pm	
	0.039 pS (mean ± s.d.)56	3
4.	Inhibition of Macroscopic Chloride Conductance by DIDS. Numbers in the left column indicate the initial conductance. The values in the second column indicate the conductance blocked by DIDS. Columns 3 and 4 describe possible breakdown to unit channel conductance. "Mean unit conductance" are: 0.566 ± 0.021 pS or 0.190 ± 0.005 pS	۵
	(mean + s d)	J

LIST OF FIGURES

Figure		Page
1.	Diagrammatic representation of the BLM setup; R - Feedback resistor; v - Command voltage; i - Measured current; R ₂ R ₃ - Voltage dividing resistors for command voltage; OS - Oscilloscope; NR - Neurocorder; VCR - Video recorder; WD - Window discriminator; PCA - Patch clamp amplifier; LPF - 8 pole low pass Bessel filter; GSCR - Gould strip chart recorder.	15
2.	Schematic representation of fusion of vesicles with a planar BLM. A - scheme of a BLM. B - initial stage. C - prefusion state. D - osmotic swelling of the vesicle and fusion. E - final stage: reconstituted BLM.	17
3A.	Time course of reduction of cyt-c by liver smooth ER vesicular NADPH-dependent cyt-c reductase. Protein concentration is 11.6 µg/ml.	26
3B.	Time course of reduction of cyt-c by liver rough ER vesicular NADPH-dependent cyt-c reductase. Protein concentration is 11.6 µg/ml.	27
4 .	I-V curve of an unmodified BLM. The BLM was made in symmetrical 50 mM choline chloride solution. DC voltage steps were applied and current was measured in each case	28
5.	Gramicidin channels incorporated in BLM. The bathing solution is symmetrical 3 M KCl.	30
6.	Schemes of the protocols for fusion of ER vesicles with a BLM (A) and recording of calcium channels (B)	31

7.	Typical observations of current jumps when ER vesicles fuse with a BLM (A). The bathing solution is symmetrical 50 mM choline chloride. Fusion is initiated by raising the <i>cis</i> choline chloride concentration to 250 mM and adding ER vesicles to the <i>cis</i> side (10 µg/ml). The current returns to zero when the <i>trans</i> choline chloride concentration is raised to 250 mM (B).	32
8.	I-V relationship of a BLM reconstituted with liver ER vesicles. Vesicles were fused with a BLM in asymmetrical choline chloride solution (<i>cis</i> - 250 mM, <i>trans</i> - 50mM). The solution also contained 2.5 mM calcium chloride. After fusion, dc voltage steps were applied and current measured in each case.	34
9.	I-V curves of BLMs reconstituted with liver ER vesicles. Vesicles were made to fuse with a BLM in asymmetrical choline chloride solution (cis - 250 mM, trans - 50mM). The solution also contained 2.5 mM calcium chloride. After fusion, dc voltage steps were applied and current measured in each case. Each symbol represents a different membrane.	35
10.	I-V curve of a BLM reconstituted with liver ER vesicles. Fusion was induced in asymmetrical choline chloride (cis - 250 mM, $trans$ 50 mM) solution. After fusion, dc voltage steps were applied and current measured. I-V curve (∇ - ∇) in asymmetrical choline chloride (cis - 250 mM, $trans$ 50 mM) solutions and (\Box - \Box) in symmetrical choline chloride (cis , $trans$ - 250 mM).	36
11.	Modified schemes of the protocols for fusion of ER vesicles with a BLM (A) and recording of calcium channels (B)	38
12.	I-V curve of a BLM reconstituted with liver ER vesicles. The BLM was made in 11 mM PIPES + 10 mM Ca(OH) ₂ . Vesicles were made to fuse by adding choline chloride to the <i>cis</i> side (final concentration - 200 mM). After fusion, dc voltage steps were applied and current measured.	39

13.	I-V curve of a BLM reconstituted with liver ER vesicles. The BLM was made in 5.5 mM PIPES + 5 mM Ca(OH) ₂ . Vesicles were made to fuse by adding potassium acetate to the <i>cis</i> side (final concentration - 200 mM). After fusion, dc voltage steps were applied and current measured.	40
14.	I-V curve of a BLM reconstituted with liver ER vesicles. The BLM was made in 5.6 mM PIPES + 5 mM Ca(OH) ₂ . After the BLM had completely thinned, potassium chloride was added to the <i>cis</i> side (final concentration - 300 mM) and ER vesicles (10 μg/ml) were added to the <i>cis</i> side. After fusion, dc voltage steps were applied and current measured	41
15 .	I-V curves of BLMs reconstituted with liver ER vesicles. The BLMs were made in symmetrical 20 mM HEPES + 4.5 mM Ca(OH) ₂ solution. KCl solution was added to <i>cis</i> as osmoticant to induce fusion (final concentration - 400 mM). ER vesicles (10 μg/ml) were added to the <i>cis</i> side. After fusion, dc voltage steps were applied and current measured. Each symbol represents a different membrane.	42
16.	A typical reconstitution experiment to reveal any InsP ₃ -gated Ca++ channel present in the liver rough ER membrane. The BLM was made in symmetrical 6.25 mM PIPES + 5.6 mM Ca(OH) ₂ . Potassium acetate was added to the <i>cis</i> side as an osmoticant. The <i>cis</i> chamber was perfused after the fusion event with 25 mM PIPES + 38 mM BTP + 1 mM EGTA + 0.5 mM Ca(OH) ₂ (0.25 µM free Ca++) and the <i>trans</i> calcium level was raised to 23 mM. No calcium channels were observed when the membrane was	
	challenged with 10µM InsP $_3$. \Box - \Box , conductance of the reconstituted membrane; Δ - Δ , conductance after perfusion of <i>cis</i> chamber with 0.25 µM free calcium solution;	
	\lozenge - \lozenge , <i>cis</i> chamber contains 0.25µM free calcium + 50 mM potassium acetate; ∇ - ∇ , <i>cis</i> chamber contains 0.25 µM free calcium + 100 mM potassium acetate.	43
17.	Histogram of P _{CI} - / P _K + values. Each value was obtained by reversal potential measurement in bi-ionic condition and calculated using the Goldman-Hodgkin-Katz equation	45

18.	Reversal potential measurement under near bi-ionic conditions (KCl <i>cis</i> side and KBr <i>trans</i> side) across a BLM	
	reconstituted with liver ER vesicles. O-O, <i>cis</i> chamber contains 5 mM HEPES + 1 mM Ca(OH) ₂ + 402 mM KCl, <i>trans</i> chamber 5 mM HEPES + 1 mM Ca(OH) ₂ + 2 mM KCl;	
	∇ - ∇ , <i>cis</i> same as above, <i>trans</i> 5 mM HEPES + 1 mM	
	Ca(OH) ₂ + 100 mM KBr; Δ - Δ , <i>ci</i> s same as above, <i>trans</i> 5 mM HEPES + 1 mM Ca(OH) ₂ + 193 mM KBr; ; \Box - \Box , <i>cis</i>	
	same as above, <i>trans</i> 5 mM ĤEPES + 1 mM Ca(OH) ₂ + 281 mM KBr.	46
19.	Reversal potential measurement under near bi-ionic condition (KCI cis side and KI trans side) across a BLM	
	reconstituted with liver ER vesicles. In O-O, the <i>cis</i> chamber contains 5mM HEPES + 1 mM Ca(OH) ₂ + 365 mM KCl and the <i>trans</i> chamber 5mM HEPES + 1 mM Ca(OH) ₂ + 2 mM	
	KCI; ∇ - ∇ , <i>cis</i> chamber contained the same solution as above and the <i>trans</i> chamber contained 5mM HEPES + 1 mM Ca(OH) ₂ + 187 mM KI	48
20.	Reversal potential measurement under near bi-ionic condition (KCI <i>cis</i> side and KCNS <i>trans</i> side) across a BLM	
	reconstituted with liver ER vesicles. In O-O, the <i>cis</i> chamber contains 5mM HEPES + 1 mM Ca(OH) ₂ + 274 mM KCl and the <i>trans</i> chamber 5mM HEPES + 1 mM Ca(OH) ₂ + 2 mM	
	KCI; ∇ - ∇ , the <i>cis</i> chamber contained the same solution as above and the <i>trans</i> chamber contained 5mM HEPES + 1 mM Ca(OH) ₂ + 88 mM KCNS.	49
21.	A control experiment for bi-ionic potential measurement (KCl cis side and KCNS trans side) using an unmodified BLM. In	
	O - O, both the <i>cis</i> and <i>trans</i> chambers contain 5 mM	
	HEPES + 1 mM Ca(OH) ₂ + 2 mM KCl, ∇ - ∇ , the <i>cis</i>	
	chamber contains 363 mM KCl and the <i>trans</i> chamber 187	
	mM KCNS	50

22.	A control experiment for bi-ionic potential measurement (KCl cis side and KI trans side) using an unmodified BLM. In ∇ - ∇ , the six and the trans chambers contain FmM HERES + 1 mM	
	the <i>cis</i> and the <i>trans</i> chambers contain 5mM HEPES + 1 mM Ca(OH) ₂ + 2 mM KCl; O-O, the <i>cis</i> chamber contains 363 mM KCl and the <i>trans</i> chamber 214 mM Kl.	51
23.	Decrease in macroscopic conductance after addition of $ZnCl_2$: I-V curve of a BLM reconstituted with liver ER vesicles. The BLM was made in symmetrical 5.5 mM HEPES + 1 mM Ca(OH) ₂ . Fusion was initiated by adding KCI solution to the <i>cis</i> side to a final concentration of 363 mM. After fusion, the osmotic gradient was eliminated by adding KCI solution to the <i>trans</i> side. O - O, no $ZnCl_2$; ∇ - ∇ , 0.875 mM $ZnCl_2$ to <i>cis</i> and <i>trans</i> sides; Δ - Δ , 1.75 mM $ZnCl_2$ to <i>cis</i> and <i>trans</i> sides; \Box - \Box , 3.5 mM $ZnCl_2$ to <i>cis</i> and <i>trans</i> sides;	55
24.	Inhibition of macroscopic chloride conductance by DIDS. After reconstituting a BLM with liver ER vesicles, different concentrations of DIDS were added to the cis side and the conductance of the membrane was determined	58
25 .	Effect of DIDS and $ZnCl_2$ on macroscopic chloride conductance of a BLM reconstituted with liver ER vesicles. The BLM was made in symmetrical 5mM HEPES + 1 mM $Ca(OH)_2$ + 2 mM KCI. Fusion of the vesicles with the BLM was initiated by adding 363 mM KCI to the <i>cis</i> side. After reconstitution, the osmotic gradient was eliminated by raising the <i>trans</i> KCI concentration. O-O, after reconstitution; ∇ - ∇ ,18 μ M DIDS to both <i>cis</i> and <i>trans</i> sides; Δ - Δ , 36 μ M DIDS to both <i>cis</i> and <i>trans</i> sides; \Box - \Box , 0.5 mM $ZnCl_2$ to both <i>cis</i> and <i>trans</i> sides; \Diamond - \Diamond , 2 mM $ZnCl_2$ to both <i>cis</i> and	60
	trans sides	60
26.	Inhibition of conductance by Zinc (Zn++) and DIDS: Conductance of a BLM reconstituted with liver vesicles (○ - ○ 1.4nS); Conductance after the addition of 2.7 mM ZnCl ₂ to the <i>cis</i> and <i>trans</i> sides (● - ● 1.04 nS); Conductance after addition of DIDS to the <i>cis</i> side (∇ - ∇ 0.192 nS)	61

Figure		Page
27	Single channel activity observed in a planar BLM reconstituted with liver rough ER vesicles. The bathing solution is symmetrical 363 mM KCl. The holding voltages are given near each trace. The channel activity could be eliminated by addition of DIDS.	63
28	Single channel activity observed in a planar BLM reconstituted with liver rough ER vesicles. The bathing solution is symmetrical 363 mM KCl. The holding voltages are given near each trace. The channel activity could be eliminated by addition of ZnCl ₂ .	64

INTRODUCTION

All cells contain an endoplasmic reticulum (ER). Though highly convoluted, the ER membrane is thought to form a single continuous sheet, enclosing a single sac. The ER plays a central role in the biosynthesis of macromolecules used to construct other cellular organelles. Lipids, proteins and complex carbohydrates destined for transportation to the Golgi apparatus, to the plasma membrane, to the lysosome, or to the cell exterior are all synthesized in association with the ER. Two functionally distinct regions of the ER can be easily identified in some cells: the rough ER and the smooth ER. The rough ER is studded with ribosomes on the cytoplasmic side of the membrane. Numerous morphological investigations have demonstrated rough ER and smooth ER to be in direct physical continuity; rough ER is thought to give rise to smooth ER to be process of cisternal "budding." Physical disruption of the palest panager at a process which can be isolated as the microsomal fraction by differential charrifugation.

Several permeation systems for ions and small solutes are present within the reticulum structures of cells. Three transport proteins, T₁, T₂ and T₃ are required to enable glucose-6-phosphate, phosphate (and pyrophosphate), and glucose to respectively cross the ER membrane (1). Other biologically relevant solutes and ions that cross the ER membrane include D-glucose, L-glucose, L-leucine, choline⁺, K⁺, Na⁺ and Cl⁻ (2). Meissner et al. (2) found that there are two types of liver microsomes (designated as types A and B) with differing permeabilities to glucose and other small molecules. About 70 percent of the

microsomes (type A) are permeable to D-glucose, L-glucose, 2-deoxy-Dglucose, D-mannose, D-mannitol, uridine, glycine, L-leucine, choline+, TRIS+, Rb+, K+, Na+, and Cl-. All of the above solutes, except Cl-, pass with a comparatively slow rate in the remaining 30 percent type B vesicles. Type A and B vesicles are similar in that both are essentially impermeable to sucrose, yet permeable to Cl⁻. By making membrane potential measurements with a fluorescent dye probe, Meissner et al. found that a significant fraction of ER vesicles were more permeable to TRIS+ than to Ca++ or Mg++. They also made another important observation that, despite their preferential permeability to K⁺. a majority of liver microsomes lack an efficient ion-conducting structure for K⁺. such as the K⁺, Na⁺ channel which renders above two-thirds of the SR vesicles highly permeable to K⁺. Treatment with the anion transport inhibitor 4,4'diisothiocyanostilbene-2,2'-disulfonic acid (DIDS) lowered the permeability of type A vesicles to several uncharged and negatively charged solutes, including D-glucose and gluconate. Based on their results, they suggested that a large fraction of liver microsomes is rendered permeable to various biologically relevant solutes and ions, perhaps through the presence of one or more channels with a maximum diameter of approximately 7-8 A° which select(s) against solutes on the basis of their size and charge.

The role of ER in sequestering Ca⁺⁺ has been well recognized. Out of the two major intracellular organelles-i.e., the mitochondria and the ER, now there is general agreement, largely through the application of electron probe x-ray microanalysis to fast frozen tissue, that mitochondria contain little Ca⁺⁺, compatible with the regulation of mitochondrial enzymes but can sequester massive amounts, should the cytosolic Ca⁺⁺ begin to rise and that, despite its relatively small Ca⁺⁺-binding capacity, the ER looks the stronger candidate for a high affinity physiologically relevant Ca⁺⁺ store (3). A Ca⁺⁺-ATPase exists in

the ER membrane. The rat liver microsomal Ca⁺⁺-ATPase has been purified (4). Its molecular weight is 107 kDa and antiserum raised against the 100 kDa sarcoplasmic reticulum (SR) Ca⁺⁺-ATPase cross-reacted with it. A major Ca⁺⁺-binding protein, calreticulin (analogous to calsequestrin), has been shown to be present in the smooth muscle SR and liver ER (5). In this connection it should be borne in mind that SR, a specialized derivative of ER has long been known to be the intracellular Ca⁺⁺ store in skeletal and cardiac muscle.

The rise to prominence of the ER has brought new ideas about Ca++ mobilization and, together with studies on the SR, a clear picture is beginning to emerge about the Ca⁺⁺ sequestration and release processes and their control in these systems. A major step in this direction has been the discovery of inositol 1,4,5-trisphosphate (InsP₃) as an intracellular second messenger (6) and its role in releasing Ca++ from the ER of many types of cells (7). An inositol lipid located within the plasma membrane is the precursor used by the receptor mechanism to release InsP₃ to the cytosol, leaving 1,2-diacyl glycerol (DAG) within the plane of the membrane. Conceptually, this theory became very attractive, since, in one step, it provided a link between membrane receptors and release of Ca++ from a major intracellular store. Consistent with its role as a second messenger, the increase in the level of InsP3 was found to precede the onset of Ca++-dependent events in blowfly salivary gland (8) and in neutrophils (9). The transduction unit within the plasma membrane consists of three main components: 1) a receptor that detects the incoming signal; 2) a G protein that serves to couple the receptor to the third component; and 3) a phosphodiesterase responsible for cleaving the lipid precursor.

Ca⁺⁺ is constantly cycling due to passive efflux and active influx across the ER membrane, and all the available evidence points to InsP₃ acting to stimulate the passive efflux component while having no effect on the pump.