

ENCYCLOPEDIA OF

CHEMICAL

TECHNOLOGY

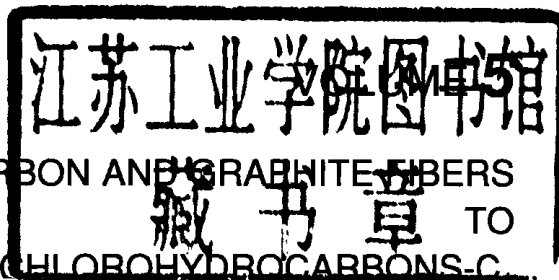
VOLUME 5

FOURTH EDITION

KIRK-OTHMER

ENCYCLOPEDIA OF CHEMICAL TECHNOLOGY

FOURTH EDITION



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EXECUTIVE EDITOR
Jacqueline I. Kroschwitz

EDITOR
Mary Howe-Grant

EDITORIAL STAFF FOR VOLUME 5

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CONTRIBUTORS TO VOLUME 5

Marvin O. Bagby, *USDA National Center for Agricultural Utilization Research, Peoria, Illinois, Survey (under Carboxylic acids)*

O. A. Battista, *Research Services Corporation, Fort Worth, Texas, Cellulose*

Noelle R. Bertoni, *Southern Regional Research Center, New Orleans, Louisiana, Cellulose*

- R. C. Buchanan**, *University of Illinois at Urbana-Champaign, Urbana, Ceramics as electrical materials*
- Richard Burt**, *Tantalum Mining Corporation, Lac Du Bonnet, Manitoba, Canada, Cesium and cesium compounds*
- David Butts**, *Great Salt Lake Minerals and Chemical Corporation, Ogden, Utah, Chemicals from brine*
- Robert R. Cantrell**, *Union Camp Corporation, Savannah, Georgia, Branched-chain acids; Fatty acids from tall oil (both under Carboxylic acids)*
- David W. Cawfield**, *Olin Chemicals, Charleston, Tennessee, Chlorous acid, chlorites, and chlorine dioxide (under Chlorine oxygen acids and salts)*
- Peter Cervoni**, *Lederle Laboratories, Pearl River, New York, Cardiovascular agents*
- Peter S. Chan**, *Lederle Laboratories, Pearl River, New York, Cardiovascular agents*
- J. Peter Clark**, *Epstein and Sons International, Inc., Chicago, Illinois, Chemurgy*
- David L. Crandall**, *Lederle Laboratories, Pearl River, New York, Cardiovascular agents*
- John A. Cuculo**, *North Carolina State University, Raleigh, North Carolina, Cellulose*
- Suresh B. Damle**, *PPG Industries, Inc., Monroeville, Pennsylvania, Carbonic and carbonochloridic esters*
- R. W. Daniels**, *Union Camp Corporation, Savannah, Georgia, Manufacture (under Carboxylic acids)*
- Eckehard Volker Dehmlow**, *Universität Bielefeld, Germany, Catalysis, phase-transfer*
- Terrence Doyle**, *Bristol-Myers Squibb Company, Wallingford, Connecticut, Chemotherapeutics, anticancer*
- Budd L. Duncan**, *Olin Chemicals, Charleston, Tennessee, Chloric acid and chlorates (under Chlorine oxygen acids and salts)*
- Frances Duneczky**, *Atlas Refinery Inc., Newark, New Jersey, Castor oil*
- Kevin Ewsuk**, *Sandia National Laboratories, Albuquerque, New Mexico, Processing (under Ceramics)*
- Richard Fengl**, *Tennessee Eastman, Kingsport, Tennessee, Cellulose esters, inorganic; Cellulose esters, organic*
- Alfred D. French**, *Southern Regional Research Center, New Orleans, Louisiana, Cellulose*
- Bruce C. Gates**, *University of Delaware Center for Catalytic Science and Technology, Newark, Delaware, Catalysis*
- Steven Gedon**, *Tennessee Eastman, Kingsport, Tennessee, Cellulose esters, organic*
- Jill Glass**, *Sandia National Laboratories, Albuquerque, New Mexico, Mechanical properties and behavior (under Ceramics)*
- Derek G. Gray**, *Pulp and Paper Research Institute of Canada, Montreal, Canada, Cellulose*
- N. R. Greening**, *Portland Cement Association, Skokie, Illinois, Cement*
- B. L. Harris**, *Consultant, Glenarm, Maryland, Chemicals in war*
- Richard Helmuth**, *Portland Cement Association, Skokie, Illinois, Cement*

- Eivind Hognestad**, *Glenview, Illinois*, Cement
- Michael T. Holbrook**, *Dow Chemical USA, Plaquemine, Louisiana*, Methyl chloride; Methylene chloride; Chloroform; Carbon tetrachloride (all under Chlorocarbons and chlorohydrocarbons)
- William L. Howard**, *Consultant, Freeport, Texas*, Chelating agents
- Deborah Illman**, *University of Washington, Seattle*, Chemometrics
- Robert W. Johnson**, *Union Camp Corporation, Savannah, Georgia*, Branched-chain acids, Fatty acids from tall oil; Manufacture (all under Carboxylic acids)
- Jerry J. Kaczur**, *Olin Chemicals, Charleston, Tennessee*, Chlorous acid, chlorites, and chlorine dioxide (under Chlorine oxygen acids and salts)
- Herbert D. Kaesz**, *University of California, Los Angeles*, Carbonyls
- Subhash Karkare**, *Amgen, Inc., Thousand Oaks, California*, Cell culture technology
- M. J. Keenan**, *Exxon Chemical Company, Baton Rouge, Louisiana*, Trialkylacetic acids (under Carboxylic acids)
- Barry Kilbourn**, *Molycorp Inc., White Plains, New York*, Cerium and cerium compounds
- Steven H. Kosmatka**, *Portland Cement Association, Skokie, Illinois*, Cement
- M. A. Krevalis**, *Exxon Chemical Company, Baton Rouge, Louisiana*, Trialkylacetic acids (under Carboxylic acids)
- David Lang**, *Portland Cement Association, Skokie, Illinois*, Cement
- Richard Loehman**, *Sandia National Laboratories, Albuquerque, New Mexico*, Ceramic processing; Electronic properties and material structure; Glass structure and properties; Mechanical properties and behavior; Nonlinear optical and electrooptic ceramics, editor; Overview (all under Ceramics)
- D. A. Lomas**, *UOP Research Center, Des Plaines, Illinois*, Regeneration, FCC (under Catalysts)
- Thomas G. Majewicz**, *Aqualon Company, Wilmington, Delaware*, Cellulose ethers
- Sudhir K. Mendiratta**, *Olin Chemicals, Charleston, Tennessee*, Chloric acid and chlorates (under Chlorine oxygen acids and salts)
- F. M. Miller**, *Portland Cement Association, Skokie, Illinois*, Cement
- R. Steve Morrow**, *Coca-Cola USA, Atlanta, Georgia*, Carbonated beverages
- Frank C. Naughton**, *CasChem/CAMREX, Bayonne, New Jersey*, Castor oil
- Robert Nielsen**, *Montgomery, Texas*, Supported (under Catalysts)
- Carol Phifer**, *Sandia National Laboratories, Albuquerque, New Mexico*, Glass structure and properties (under Ceramics)
- Mark Phillips**, *Sandia National Laboratories, Albuquerque, New Mexico*, Nonlinear optical and electrooptic ceramics (under Ceramics)
- Ronald Pierantozzi**, *Air Products and Chemicals, Inc., Allentown, Pennsylvania*, Carbon dioxide; Carbon monoxide
- Thomas J. Podlas**, *Aqualon Company, Wilmington, Delaware*, Cellulose ethers
- Christine M. Quinn**, *Coca-Cola USA, Atlanta, Georgia*, Carbonated beverages
- Daniel J. Reed**, *Dow Chemical USA, Freeport, Texas*, Survey (under Chlorocarbons and chlorohydrocarbons)

- D. W. Robinson**, *UOP Research Center, Des Plaines, Illinois*, Regeneration, noble metals (under Catalysts)
- R. D. Roseman**, *University of Illinois at Urbana-Champaign, Urbana*, Ceramics as electrical materials
- E. T. Sauer**, *The Procter & Gamble Company, Cincinnati, Ohio*, Economic aspects (under Carboxylic acids)
- Robert Schwartz**, *Sandia National Laboratories, Albuquerque, New Mexico*, Electronic properties and material structure (under Ceramics)
- David E. Smith**, *FMC Corporation, Princeton, New Jersey*, Carbon disulfide
- Robert W. Timmerman**, *FMC Corporation, Princeton, New Jersey*, Carbon disulfide
- L. L. Upson**, *UOP Research Center, Des Plaines, Illinois*, Regeneration, FCC (under Catalysts)
- Joseph G. Venner**, *BASF Structural Materials, Inc., Charlotte, North Carolina*, Carbon and graphite fibers
- Dolatrai Vyas**, *Bristol-Myers Squibb Company, Wallingford, Connecticut*, Chemotherapeutics, anticancer
- Frank S. Wagner**, *Strem Chemicals Inc., Newbury Port, Massachusetts*, Carbonyls
- David A. Wilson**, *Designed Chemicals R&D, Freeport, Texas*, Chelating agents
- John A. Wojtowicz**, *Olin Corporation, Cheshire, Connecticut*, Dichlorine monoxide, hypochlorous acid, and hypochlorites (under Chlorine oxygen acids and salts); Chloramines and bromamines

NOTE ON CHEMICAL ABSTRACTS SERVICE REGISTRY NUMBERS AND NOMENCLATURE

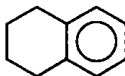
Chemical Abstracts Service (CAS) Registry Numbers are unique numerical identifiers assigned to substances recorded in the CAS Registry System. They appear in brackets in the *Chemical Abstracts* (CA) substance and formula indexes following the names of compounds. A single compound may have synonyms in the chemical literature. A simple compound like phenethylamine can be named β -phenylethylamine or, as in *Chemical Abstracts*, benzenethanamine. The usefulness of the *Encyclopedia* depends on accessibility through the most common correct name of a substance. Because of this diversity in nomenclature careful attention has been given to the problem in order to assist the reader as much as possible, especially in locating the systematic CA index name by means of the Registry Number. For this purpose, the reader may refer to the CAS Registry Handbook—Number Section which lists in numerical order the Registry Number with the *Chemical Abstracts* index name and the molecular formula; eg, **458-88-8**, Piperidine, 2-propyl-, (S)-, $C_8H_{17}N$; in the *Encyclopedia* this compound would be found under its common name, coniine [**458-88-8**]. Alternatively, this information can be retrieved electronically from CAS Online. In many cases molecular formulas have also been provided in the *Encyclopedia* text to facilitate electronic searching. The Registry Number is a valuable link for the reader in retrieving additional published information on substances and also as a point of access for on-line data bases.

In all cases, the CAS Registry Numbers have been given for title compounds in articles and for all compounds in the index. All specific substances indexed in *Chemical Abstracts* since 1965 are included in the CAS Registry System as are a large number of substances derived from a variety of reference works. The CAS Registry System identifies a substance on the basis of an unambiguous computer-language description of its molecular structure including stereochemical detail. The Registry Number is a machine-checkable number (like a Social Security number) assigned in sequential order to each substance as it enters the registry system. The value of the number lies in the fact that it is a concise and unique means of substance identification, which is independent of, and therefore bridges, many systems of chemical nomenclature. For polymers, one Registry Number may

be used for the entire family; eg, polyoxyethylene (20) sorbitan monolaurate has the same number as all of its polyoxyethylene homologues.

Cross-references are inserted in the index for many common names and for some systematic names. Trademark names appear in the index. Names that are incorrect, misleading, or ambiguous are avoided. Formulas are given very frequently in the text to help in identifying compounds. The spelling and form used, even for industrial names, follow American chemical usage, but not always the usage of *Chemical Abstracts* (eg, *coniine* is used instead of *(S)-2-propylpiperidine*, *aniline* instead of *benzenamine*, and *acrylic acid* instead of *2-propenoic acid*).

There are variations in representation of rings in different disciplines. The dye industry does not designate aromaticity or double bonds in rings. All double bonds and aromaticity are shown in the *Encyclopedia* as a matter of course. For example, tetralin has an aromatic ring and a saturated ring and its structure



appears in the *Encyclopedia* with its common name, Registry Number enclosed in brackets, and parenthetical CA index name, ie, tetralin [119-64-2] (1,2,3,4-tetrahydronaphthalene). With names and structural formulas, and especially with CAS Registry Numbers, the aim is to help the reader have a concise means of substance identification.

CONVERSION FACTORS, ABBREVIATIONS, AND UNIT SYMBOLS

SI Units (Adopted 1960)

The International System of Units (abbreviated SI), is being implemented throughout the world. This measurement system is a modernized version of the MKSA (meter, kilogram, second, ampere) system, and its details are published and controlled by an international treaty organization (The International Bureau of Weights and Measures) (1).

SI units are divided into three classes:

BASE UNITS

length	meter [†] (m)
mass	kilogram (kg)
time	second (s)
electric current	ampere (A)
thermodynamic temperature [‡]	kelvin (K)
amount of substance	mole (mol)
luminous intensity	candela (cd)

SUPPLEMENTARY UNITS

plane angle	radian (rad)
solid angle	steradian (sr)

[†]The spellings “metre” and “litre” are preferred by ASTM; however, “-er” is used in the *Encyclopedia*.

[‡]Wide use is made of Celsius temperature (t) defined by

$$t = T - T_0$$

where T is the thermodynamic temperature, expressed in kelvin, and $T_0 = 273.15$ K by definition. A temperature interval may be expressed in degrees Celsius as well as in kelvin.

DERIVED UNITS AND OTHER ACCEPTABLE UNITS

These units are formed by combining base units, supplementary units, and other derived units (2–4). Those derived units having special names and symbols are marked with an asterisk in the list below.

Quantity	Unit	Symbol	Acceptable equivalent
*absorbed dose	gray	Gy	J/kg
acceleration	meter per second squared	m/s ²	
*activity (of a radionuclide)	becquerel	Bq	1/s
area	square kilometer	km ²	
	square hectometer	hm ²	ha (hectare)
	square meter	m ²	
concentration (of amount of substance)	mole per cubic meter	mol/m ³	
current density	ampere per square meter	A/m ²	
density, mass density	kilogram per cubic meter	kg/m ³	g/L; mg/cm ³
dipole moment (quantity)	coulomb meter	C·m	
*dose equivalent	sievert	Sv	J/kg
*electric capacitance	farad	F	C/V
*electric charge, quantity of electricity	coulomb	C	A·s
electric charge density	coulomb per cubic meter	C/m ³	
*electric conductance	siemens	S	A/V
electric field strength	volt per meter	V/m	
electric flux density	coulomb per square meter	C/m ²	
*electric potential, potential difference, electromotive force	volt	V	W/A
*electric resistance	ohm	Ω	V/A
*energy, work, quantity of heat	megajoule	MJ	
	kilojoule	kJ	
	joule	J	N·m
	electronvolt [†]	eV [†]	
	kilowatt-hour [†]	kW·h [†]	
energy density	joule per cubic meter	J/m ³	
*force	kilonewton	kN	
	newton	N	kg·m/s ²

[†]This non-SI unit is recognized by the CIPM as having to be retained because of practical importance or use in specialized fields (1).

Quantity	Unit	Symbol	Acceptable equivalent
*frequency	megahertz	MHz	
	hertz	Hz	1/s
heat capacity, entropy	joule per kelvin	J/K	
heat capacity (specific), specific entropy	joule per kilogram kelvin	J/(kg·K)	
heat transfer coefficient	watt per square meter kelvin	W/(m ² ·K)	
*illuminance	lux	lx	lm/m ²
*inductance	henry	H	Wb/A
linear density	kilogram per meter	kg/m	
luminance	candela per square meter	cd/m ²	
*luminous flux	lumen	lm	cd·sr
magnetic field strength	ampere per meter	A/m	
*magnetic flux	weber	Wb	V·s
*magnetic flux density	tesla	T	Wb/m ²
molar energy	joule per mole	J/mol	
molar entropy, molar heat capacity	joule per mole kelvin	J/(mol·K)	
moment of force, torque	newton meter	N·m	
momentum	kilogram meter per second	kg·m/s	
permeability	henry per meter	H/m	
permittivity	farad per meter	F/m	
*power, heat flow rate, radiant flux	kilowatt watt	kW W	J/s
power density, heat flux density, irradiance	watt per square meter	W/m ²	
*pressure, stress	megapascal kilopascal pascal	MPa kPa Pa	N/m ²
sound level	decibel	dB	
specific energy	joule per kilogram	J/kg	
specific volume	cubic meter per kilogram	m ³ /kg	
surface tension	newton per meter	N/m	
thermal conductivity	watt per meter kelvin	W/(m·K)	
velocity	meter per second kilometer per hour	m/s km/h	
viscosity, dynamic	pascal second millipascal second	Pa·s mPa·s	
viscosity, kinematic	square meter per second square millimeter per second	m ² /s mm ² /s	

Quantity	Unit	Symbol	Acceptable equivalent
volume	cubic meter	m ³	
	cubic decimeter	dm ³	L (liter) (5)
	cubic centimeter	cm ³	mL
wave number	1 per meter	m ⁻¹	
	1 per centimeter	cm ⁻¹	

In addition, there are 16 prefixes used to indicate order of magnitude, as follows:

Multiplication factor	Prefix	Symbol	Note
10 ¹⁸	exa	E	
10 ¹⁵	peta	P	
10 ¹²	tera	T	
10 ⁹	giga	G	
10 ⁶	mega	M	
10 ³	kilo	k	
10 ²	hecto	h ^a	^a Although hecto, deka, deci, and centi are SI prefixes, their use should be avoided except for SI unit-multiples for area and volume and nontechnical use of centimeter, as for body and clothing measurement.
10	deka	da ^a	
10 ⁻¹	deci	d ^a	
10 ⁻²	centi	c ^a	
10 ⁻³	milli	m	
10 ⁻⁶	micro	μ	
10 ⁻⁹	nano	n	
10 ⁻¹²	pico	p	
10 ⁻¹⁵	femto	f	
10 ⁻¹⁸	atto	a	

For a complete description of SI and its use the reader is referred to ASTM E 380 (4) and the article UNITS AND CONVERSION FACTORS which appears in Vol. 24.

A representative list of conversion factors from non-SI to SI units is presented herewith. Factors are given to four significant figures. Exact relationships are followed by a dagger. A more complete list is given in the latest editions of ASTM E 380 (4) and ANSI Z210.1 (6).

Conversion Factors to SI Units

To convert from	To	Multiply by
acre	square meter (m ²)	4.047 × 10 ³
angstrom	meter (m)	1.0 × 10 ^{-10†}
are	square meter (m ²)	1.0 × 10 ^{2†}
astronomical unit	meter (m)	1.496 × 10 ¹¹

[†]Exact.

To convert from	To	Multiply by
atmosphere, standard	pascal (Pa)	1.013×10^5
bar	pascal (Pa)	$1.0 \times 10^{5\dagger}$
barn	square meter (m ²)	$1.0 \times 10^{-28\dagger}$
barrel (42 U.S. liquid gallons)	cubic meter (m ³)	0.1590
Bohr magneton (μ_B)	J/T	9.274×10^{-24}
Btu (International Table)	joule (J)	1.055×10^3
Btu (mean)	joule (J)	1.056×10^3
Btu (thermochemical)	joule (J)	1.054×10^3
bushel	cubic meter (m ³)	3.524×10^{-2}
calorie (International Table)	joule (J)	4.187
calorie (mean)	joule (J)	4.190
calorie (thermochemical)	joule (J)	4.184^\dagger
centipoise	pascal second (Pa·s)	$1.0 \times 10^{-3\dagger}$
centistokes	square millimeter per second (mm ² /s)	1.0^\dagger
cfm (cubic foot per minute)	cubic meter per second (m ³ /s)	4.72×10^{-4}
cubic inch	cubic meter (m ³)	1.639×10^{-5}
cubic foot	cubic meter (m ³)	2.832×10^{-2}
cubic yard	cubic meter (m ³)	0.7646
curie	becquerel (Bq)	$3.70 \times 10^{10\dagger}$
debye	coulomb meter (C·m)	3.336×10^{-30}
degree (angle)	radian (rad)	1.745×10^{-2}
denier (international)	kilogram per meter (kg/m)	1.111×10^{-7}
	tex [‡]	0.1111
dram (apothecaries')	kilogram (kg)	3.888×10^{-3}
dram (avoirdupois)	kilogram (kg)	1.772×10^{-3}
dram (U.S. fluid)	cubic meter (m ³)	3.697×10^{-6}
dyne	newton (N)	$1.0 \times 10^{-5\dagger}$
dyne/cm	newton per meter (N/m)	$1.0 \times 10^{-3\dagger}$
electronvolt	joule (J)	1.602×10^{-19}
erg	joule (J)	$1.0 \times 10^{-7\dagger}$
fathom	meter (m)	1.829
fluid ounce (U.S.)	cubic meter (m ³)	2.957×10^{-5}
foot	meter (m)	0.3048^\dagger
footcandle	lux (lx)	10.76
furlong	meter (m)	2.012×10^{-2}
gal	meter per second squared (m/s ²)	$1.0 \times 10^{-2\dagger}$
gallon (U.S. dry)	cubic meter (m ³)	4.405×10^{-3}
gallon (U.S. liquid)	cubic meter (m ³)	3.785×10^{-3}
gallon per minute (gpm)	cubic meter per second (m ³ /s)	6.309×10^{-5}
	cubic meter per hour (m ³ /h)	0.2271

[†]Exact.[‡]See footnote on p. xiv.

To convert from	To	Multiply by
gauss	tesla (T)	1.0×10^{-4}
gilbert	ampere (A)	0.7958
gill (U.S.)	cubic meter (m ³)	1.183×10^{-4}
grade	radian	1.571×10^{-2}
grain	kilogram (kg)	6.480×10^{-5}
gram force per denier	newton per tex (N/tex)	8.826×10^{-2}
hectare	square meter (m ²)	$1.0 \times 10^{4\dagger}$
horsepower (550 ft-lbf/s)	watt (W)	7.457×10^2
horespower (boiler)	watt (W)	9.810×10^3
horsepower (electric)	watt (W)	$7.46 \times 10^{2\dagger}$
hundredweight (long)	kilogram (kg)	50.80
hundredweight (short)	kilogram (kg)	45.36
inch	meter (m)	$2.54 \times 10^{-2\dagger}$
inch of mercury (32°F)	pascal (Pa)	3.386×10^3
inch of water (39.2°F)	pascal (Pa)	2.491×10^2
kilogram-force	newton (N)	9.807
kilowatt hour	megajoule (MJ)	3.6 [†]
kip	newton (N)	4.448×10^3
knot (international)	meter per second (m/s)	0.5144
lambert	candela per square meter (cd/m ²)	3.183×10^3
league (British nautical)	meter (m)	5.559×10^3
league (statute)	meter (m)	4.828×10^3
light year	meter (m)	9.461×10^{15}
liter (for fluids only)	cubic meter (m ³)	$1.0 \times 10^{-3\dagger}$
maxwell	weber (Wb)	$1.0 \times 10^{-8\dagger}$
micron	meter (m)	$1.0 \times 10^{-6\dagger}$
mil	meter (m)	$2.54 \times 10^{-5\dagger}$
mile (statute)	meter (m)	1.609×10^3
mile (U.S. nautical)	meter (m)	$1.852 \times 10^{3\dagger}$
mile per hour	meter per second (m/s)	0.4470
millibar	pascal (Pa)	1.0×10^2
millimeter of mercury (0°C)	pascal (Pa)	$1.333 \times 10^{2\dagger}$
minute (angular)	radian	2.909×10^{-4}
myriagram	kilogram (kg)	10
myriameter	kilometer (km)	10
oersted	ampere per meter (A/m)	79.58
ounce (avoirdupois)	kilogram (kg)	2.835×10^{-2}
ounce (troy)	kilogram (kg)	3.110×10^{-2}
ounce (U.S. fluid)	cubic meter (m ³)	2.957×10^{-5}
ounce-force	newton (N)	0.2780
peck (U.S.)	cubic meter (m ³)	8.810×10^{-3}
pennyweight	kilogram (kg)	1.555×10^{-3}
pint (U.S. dry)	cubic meter (m ³)	5.506×10^{-4}

[†]Exact.

To convert from	To	Multiply by
pint (U.S. liquid)	cubic meter (m ³)	4.732×10^{-4}
poise (absolute viscosity)	pascal second (Pa·s)	0.10^{\dagger}
1 b) pound (avoirdupois)	kilogram (kg)	0.4536
pound (troy)	kilogram (kg)	0.3732
poundal	newton (N)	0.1383
pound-force	newton (N)	4.448
pound force per square inch (psi)	pascal (Pa)	6.895×10^3
quart (U.S. dry)	cubic meter (m ³)	1.101×10^{-3}
quart (U.S. liquid)	cubic meter (m ³)	9.464×10^{-4}
quintal	kilogram (kg)	$1.0 \times 10^{2\dagger}$
rad	gray (Gy)	$1.0 \times 10^{-2\dagger}$
rod	meter (m)	5.029
roentgen	coulomb per kilogram (C/kg)	2.58×10^{-4}
second (angle)	radian (rad)	$4.848 \times 10^{-6\dagger}$
section	square meter (m ²)	2.590×10^6
slug	kilogram (kg)	14.59
spherical candle power	lumen (lm)	12.57
square inch	square meter (m ²)	6.452×10^{-4}
square foot	square meter (m ²)	9.290×10^{-2}
square mile	square meter (m ²)	2.590×10^6
square yard	square meter (m ²)	0.8361
stere	cubic meter (m ³)	1.0^{\dagger}
stokes (kinematic viscosity)	square meter per second (m ² /s)	$1.0 \times 10^{-4\dagger}$
tex	kilogram per meter (kg/m)	$1.0 \times 10^{-6\dagger}$
ton (long, 2240 pounds)	kilogram (kg)	1.016×10^3
ton (metric) (tonne)	kilogram (kg)	$1.0 \times 10^{3\dagger}$
ton (short, 2000 pounds)	kilogram (kg)	9.072×10^2
torr	pascal (Pa)	1.333×10^2
unit pole	weber (Wb)	1.257×10^{-7}
yard	meter (m)	0.9144^{\dagger}

[†]Exact.

Abbreviations and Unit Symbols

Following is a list of common abbreviations and unit symbols used in the *Encyclopedia*. In general they agree with those listed in *American National Standard Abbreviations for Use on Drawings and in Text (ANSI Y1.1)* (6) and *American National Standard Letter Symbols for Units in Science and Technology (ANSI Y10)* (6). Also included is a list of acronyms for a number of private and government organizations as well as common industrial solvents, polymers, and other chemicals.