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# SADTLER

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# SPECTRA

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NUCLEAR MAGNETIC RESONANCE SPECTRA



*Sadtler*  
Division

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## SADTLER NUCLEAR MAGNETIC RESONANCE SPECTRA

Volumes 95-99

With this 1991 supplement of 2,000 spectra to the Sadler Standard Nuclear Magnetic Resonance Spectra collection, the catalog now contains 54,000 proton NMR spectra.

### INTRODUCTION

Protons give rise to signals of varying position, intensity and complexity. The actual position of the signal, the so-called chemical shift, is dependent on the immediate electronic environment of a proton and therefore is an index of its chemical nature. The intensity, which can be obtained by integration, is a measure of the number of protons of that kind which cause the signal. The resonances of protons are often complex with multiplicity arising from nuclear spin interaction between the proton yielding the signal and other protons nearby. It is known that coupling between protons which are two, three and four or more bonds apart (geminal, vicinal and long range coupling respectively) is peculiarly dependent on the geometric relation between the protons in question. Consequently, signal multiplicity characterizes the local molecular geometry of a proton.

In practice, the chemical shift and intensity parameters are readily obtained by inspection of the scanned and integrated spectra. However, the elucidation of spin multiplicity requires more time and in many cases can only be determined unambiguously by spin decoupling experiments or calculation.

### ASSIGNMENTS

The center positions of signals in the spectrum (the chemical shifts  $\delta$ ), have been expressed in terms of a dimensionless field-independent scale on which the frequency of TMS is taken as zero.

$$\delta = \frac{v_{\text{sample}} - v_{\text{TMS}}}{80 \text{ MHz}} \times \text{million}$$

$v$  = resonance frequency of protons in the substance

The chemical shift  $\delta$  is then cited in parts per million (p.p.m.). Chemical shifts are considered to be accurate to  $\pm 3$  Hz or  $\pm 0.05$  p.p.m.

Whenever possible, the chemical shift assignments have been made by first order analysis. In the case of non-first order multiplets, the shift was taken as the estimated weighted center. Alternative assignments have been noted; however, if a chemical shift could not be attributed with reasonable certainty due to overlapping, then no precise assignment was attempted. In several cases, protons have been assigned within a range or about a certain value of chemical shift.

Low intensity multiplets have been enlarged and separately positioned on the chart.

#### INSTRUMENTATION & TECHNIQUE

All spectra have been scanned in the Sadtler Research Laboratories at 80 MHz on a Varian CFT-20 Fourier Transform NMR spectrometer.

Samples are dissolved in deuteriochloroform as the solvent of choice together with added tetramethylsilane (TMS) as an internal standard. Solute concentrations are roughly 5% to 20%. 3-trimethylsilyltetradeca deuterio propionate, sodium salt (TSP) is used as an internal standard when deuterium oxide ( $D_2O$ ) is used as the solvent.

For those samples which are found to be insoluble in deuteriochloroform, other solvents were used; such as dimethyl sulfoxide, polysol-D (DMSO- $D_6$ /CDCl $_3$ ) and deuterium oxide.

In certain cases, a second spectrum may have been scanned after the addition of  $D_2O$  or trifluoroacetic acid to the sample solution.

As in the previous volumes of the NMR Standard Spectra collection, coupling constants have been measured for some of the less usual spin-spin coupling patterns and are printed on the spectra page at the end of the assignments section.

Your comments and suggestions, as well as samples whose spectra require replacement, are always welcome. Samples of new 98% pure compounds (accompanied by structure and/or name, physical data, solubility, and journal reference) are continually being sought and it is only because of the generous contributions of those whose names appear as the "source" that we can offer these spectra to scientists. Our sincere thanks is expressed to these donors.

N-[p-(BENZYLOXY)BENZYLIDENE]-p-TOLUIDINE

52001 M

C<sub>21</sub>H<sub>19</sub>NO

MOL. WT. 301.39

PRISM NO.

79001

SOLVENT

CDCl<sub>3</sub>

SOURCE OF SAMPLE

Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

REFERENCE

TMS

ASSIGNMENTS

a 2.37

b 5.12

c 7.06

d ca7.18

e ca7.40

f 7.87

g 8.39

h 2.22, 6.60 imps

i

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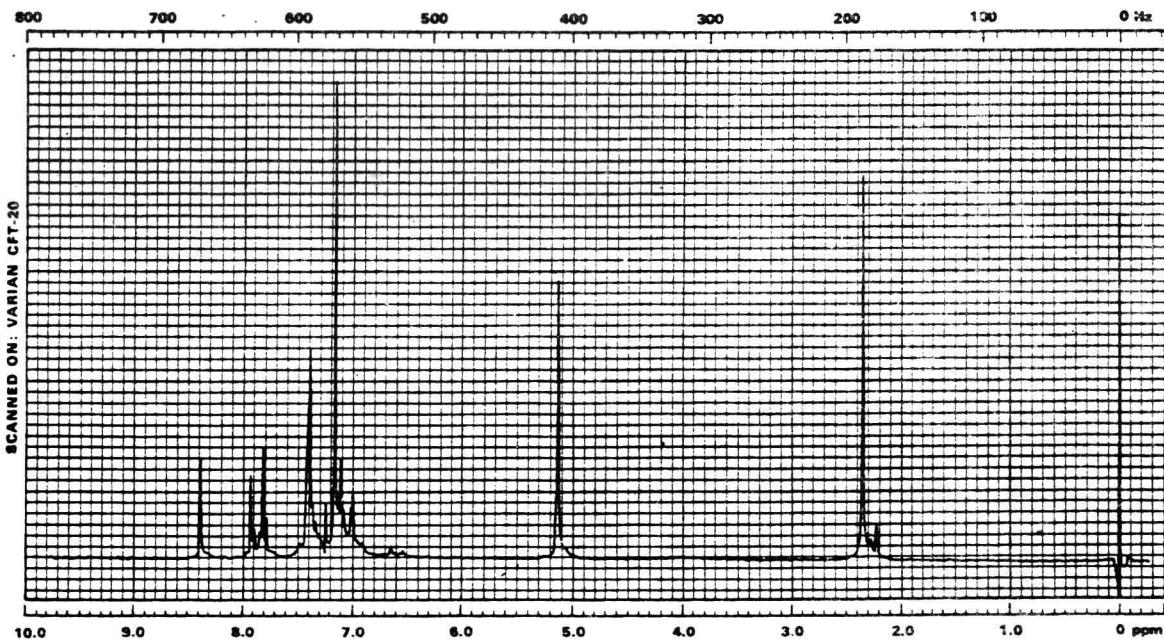
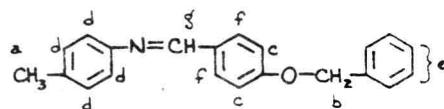
x

y

z

1

2



**52002 M N-[p-(BENZYLOXY)BENZYLIDENE]-3,4-XYLIDINE**

C<sub>22</sub>H<sub>21</sub>NO

MOL. WT. 315.42

PRISM NO.

79002

SOLVENT

CDCl<sub>3</sub>

REFERENCE

TMS

**SOURCE OF SAMPLE**

Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

**ASSIGNMENTS**

a 2.29  
b 5.14  
c 6.80-7.25  
d ca7.40  
e 7.86  
f 8.39  
g 2.18, 6.50 imps

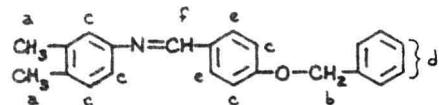
h 7.26 CHCl<sub>3</sub>  
i  
j  
k  
l  
m  
n

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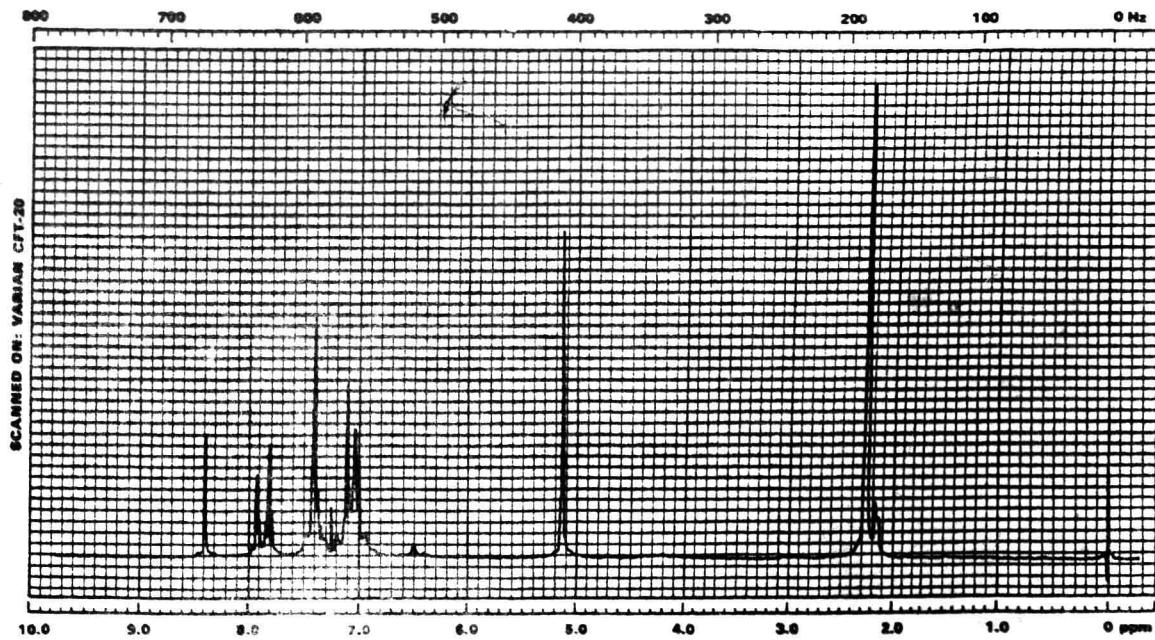
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E



**$\alpha$ -CHLORO- $\alpha$ -TOLUNITRILE**

**52003 M**

$C_8H_6ClN$

MOL. WT. 151.60

PRISM NO. 79003

SOLVENT  $CDCl_3$

**SOURCE OF SAMPLE**

Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

REFERENCE TMS

**ASSIGNMENTS**

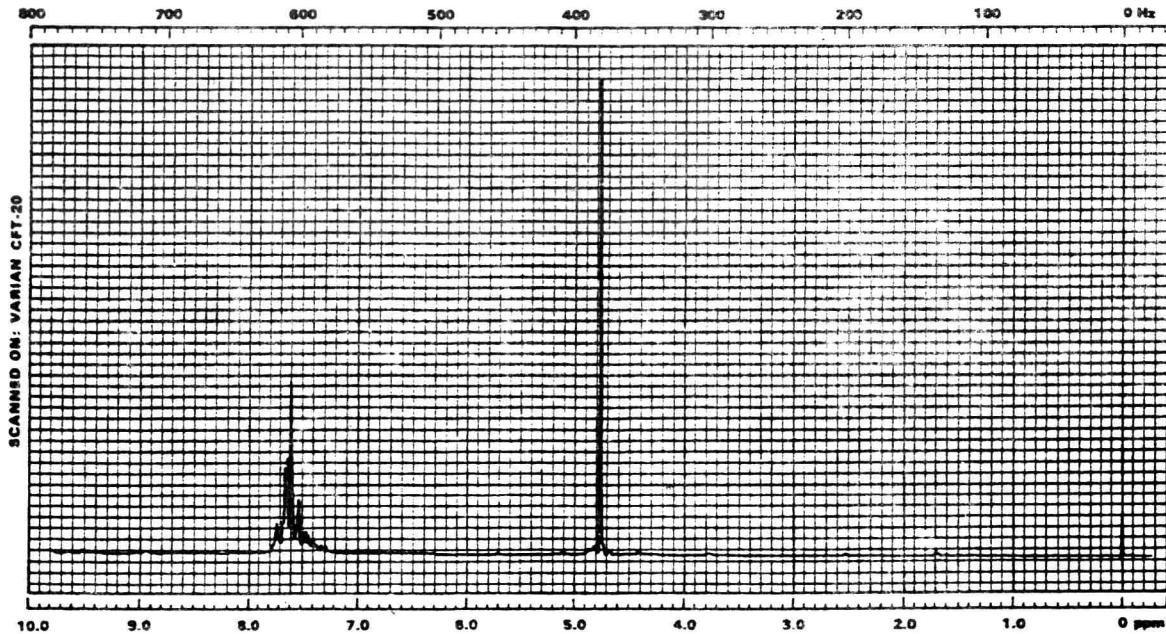
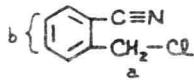
a 4.79  
b 7.25-7.80  
c  
d  
e  
f  
g

h  
i  
j  
k  
l  
m  
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**52004 M**      o-(HEXADECYLSULFONYL)ANILINE

C<sub>22</sub>H<sub>39</sub>NO<sub>2</sub>S

MOL. WT. 381.63

PRISM NO.

79004

SOLVENT

CDCl<sub>3</sub>

REFERENCE

TMS

**SOURCE OF SAMPLE**  
Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

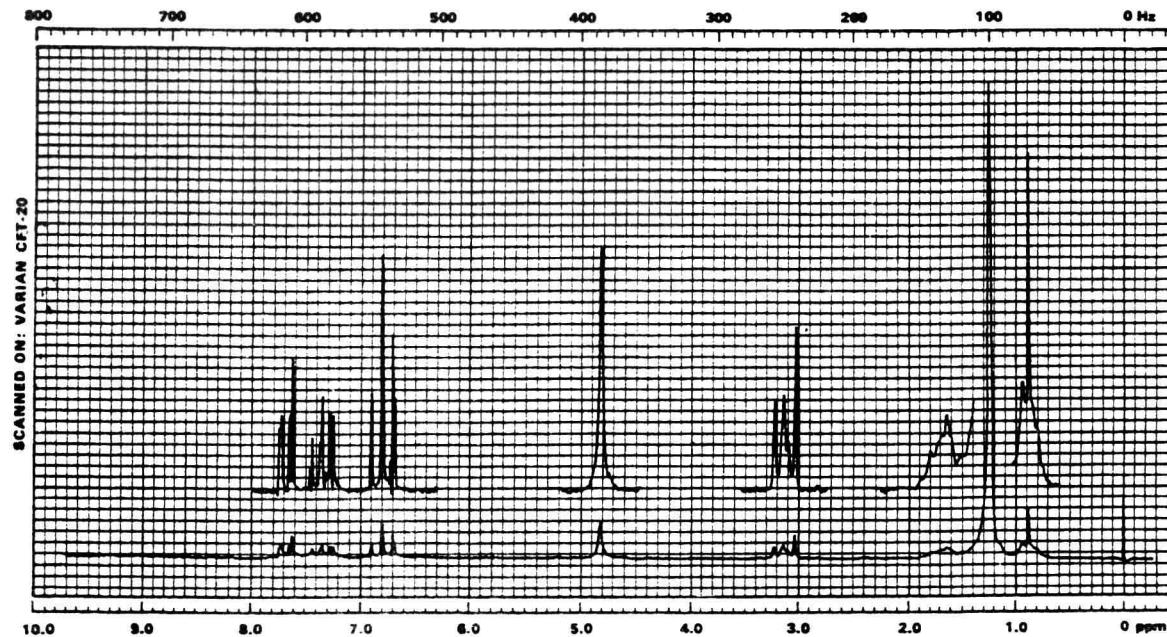
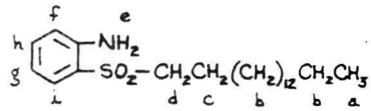
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**ASSIGNMENTS**

a	0.90	h	7.35	o		v	
b	1.05-1.55	i	7.68	p		w	
c	1.68	j		q		x	
d	3.13	k		r		y	
e	4.81	l		s		z	
f	6.76	m		t		1	
g	6.80	n		u		2	



## 4-ETHOXYMETANILYL FLUORIDE

52005 M

 $C_8H_{10}FNO_3S$ 

MOL. WT. 219.24

PRISM NO. 79005

SOLVENT  $CDCl_3$ 

## SOURCE OF SAMPLE

Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

REFERENCE TMS

## ASSIGNMENTS

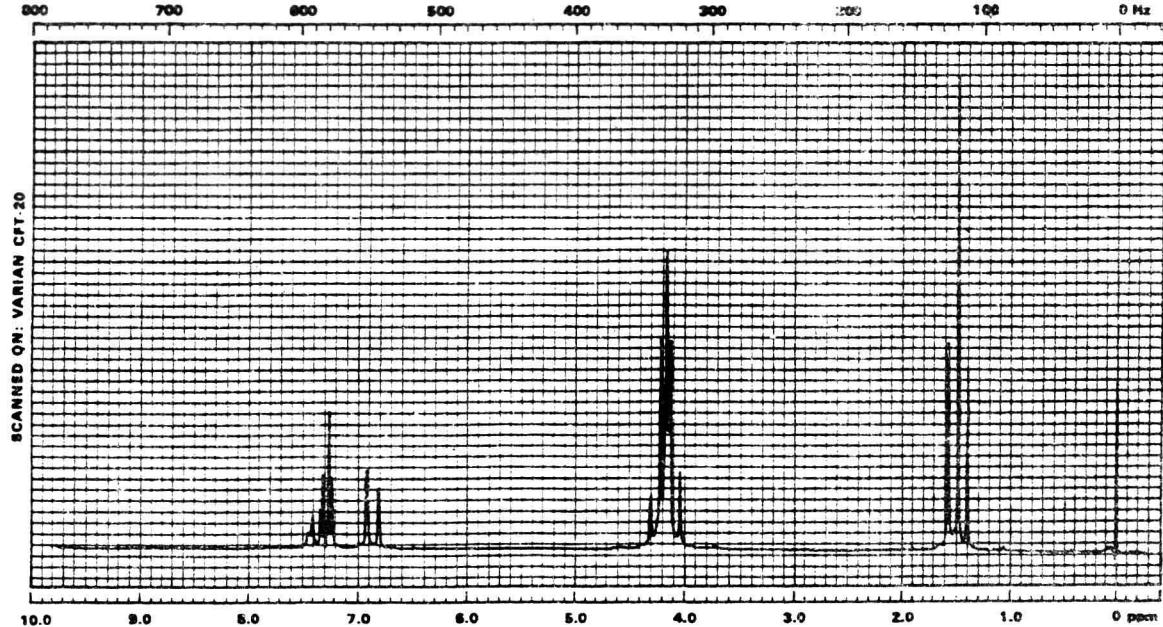
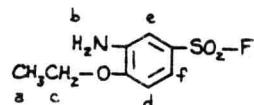
a 1.50  
 b 4.18  
 c 4.19  
 d 6.86  
 e 7.28  
 f 7.37  
 g

h \_\_\_\_\_  
 i \_\_\_\_\_  
 j \_\_\_\_\_  
 k \_\_\_\_\_  
 l \_\_\_\_\_  
 m \_\_\_\_\_  
 n \_\_\_\_\_

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s	z
t	1
u	2



**52006 M 2,6-DIAMINO-3-(PHENYLAZO)PYRIDINE, 1,2-ETHANEDISULFONATE (1:1)**

C<sub>11</sub>H<sub>11</sub>N<sub>5</sub> · C<sub>2</sub>H<sub>6</sub>O<sub>6</sub>S<sub>2</sub>

MOL. WT. 403.44

PRISM NO.

79006

SOLVENT

Polysol

**SOURCE OF SAMPLE**

Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

REFERENCE

TMS

**ASSIGNMENTS**

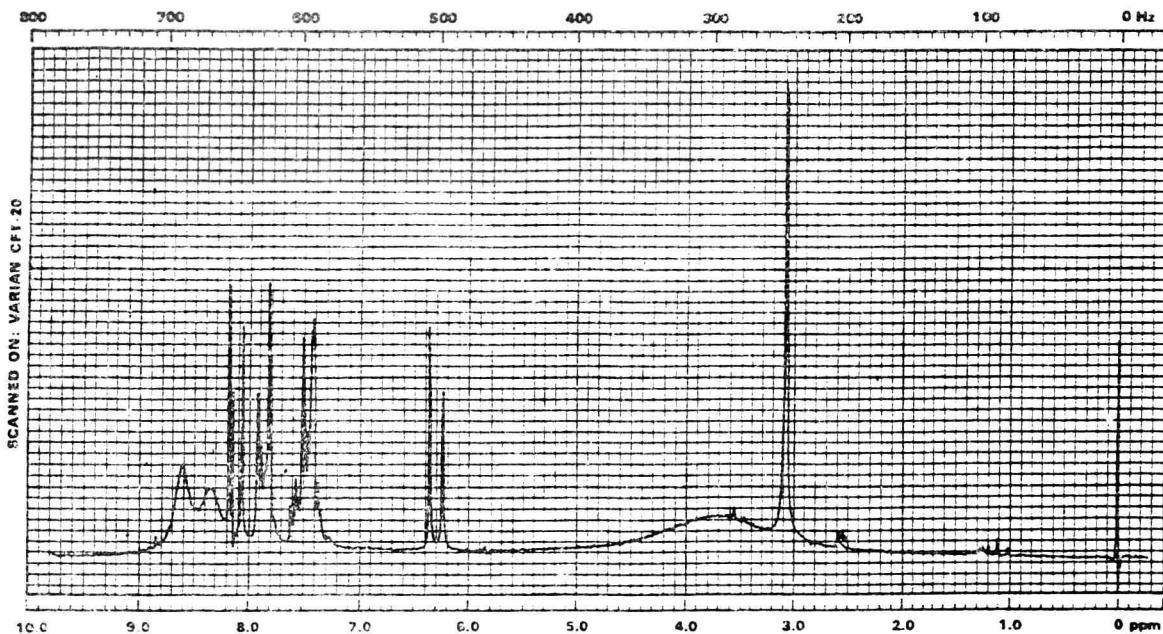
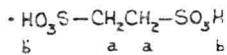
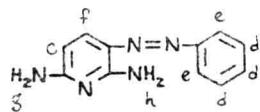
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a	3.10	h	8.60/g	o		v	
b	3.00-5.50*	i	1.10, 1.28 imps	p		w	
c	6.31	j	2.55 DMSO-d <sub>6</sub>	q		x	
d	7.30-7.65	k	8.19 CHCl <sub>3</sub>	r		y	
e	7.80	l		s		z	
f	8.11	m		t		1	
g	8.35/h	n		u		2	

\*Plus H<sub>2</sub>O



***o*-(CHLOROMETHYL)ANISOLE**

**52007 M**

C<sub>8</sub>H<sub>9</sub>ClO

MOL. WT. 156.61

PRISM NO.

79007

SOLVENT

CDCl<sub>3</sub>

REFERENCE

TMS

**SOURCE OF SAMPLE**

Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

**ASSIGNMENTS**

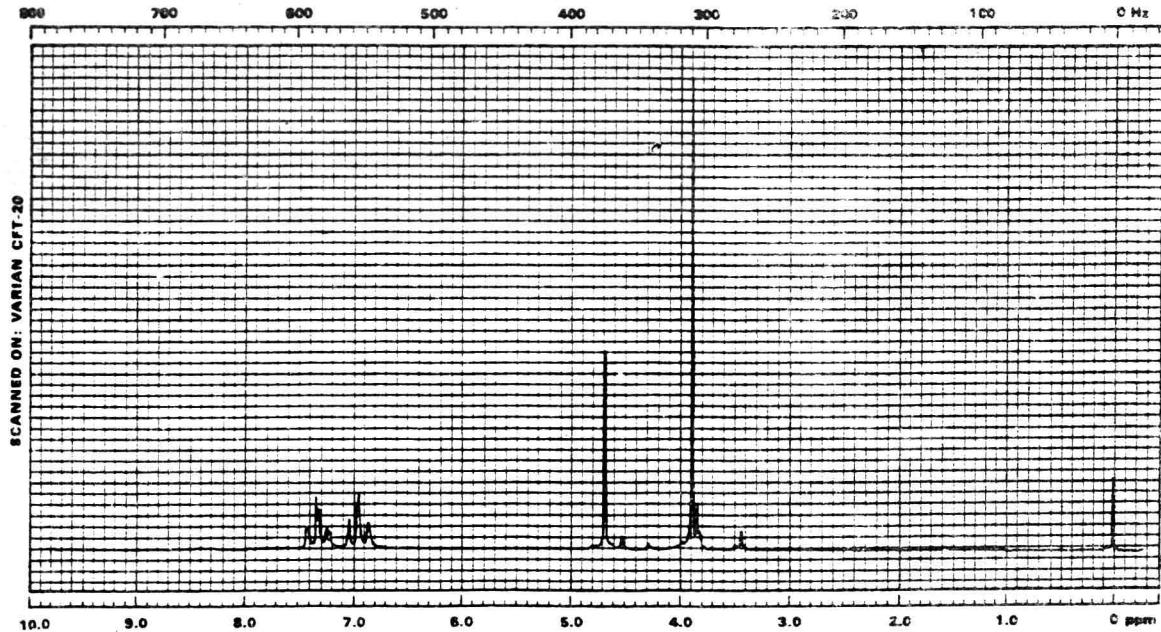
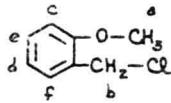
a 3.90  
b 4.69  
c 6.91  
d 6.97  
e 7.33  
f 7.37  
g 3.42, 3.85 imps

h 4.51 imp  
i \_\_\_\_\_  
j \_\_\_\_\_  
k \_\_\_\_\_  
l \_\_\_\_\_  
m \_\_\_\_\_  
n \_\_\_\_\_

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**52008 M p-UREIDOBENZOIC ACID**

C<sub>8</sub>H<sub>8</sub>N<sub>2</sub>O<sub>3</sub>

MOL. WT. 180.16

PRISM NO.

79008

SOLVENT

Polysol

**SOURCE OF SAMPLE**

Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

**REFERENCE**

TMS

**ASSIGNMENTS**

a 5.00-7.30\*      h \_\_\_\_\_  
b 7.49      i \_\_\_\_\_  
c 7.84      j \_\_\_\_\_  
d 8.80      k \_\_\_\_\_  
e 6.70-7.20 imps      l \_\_\_\_\_  
f \_\_\_\_\_      m \_\_\_\_\_  
g \_\_\_\_\_      n \_\_\_\_\_

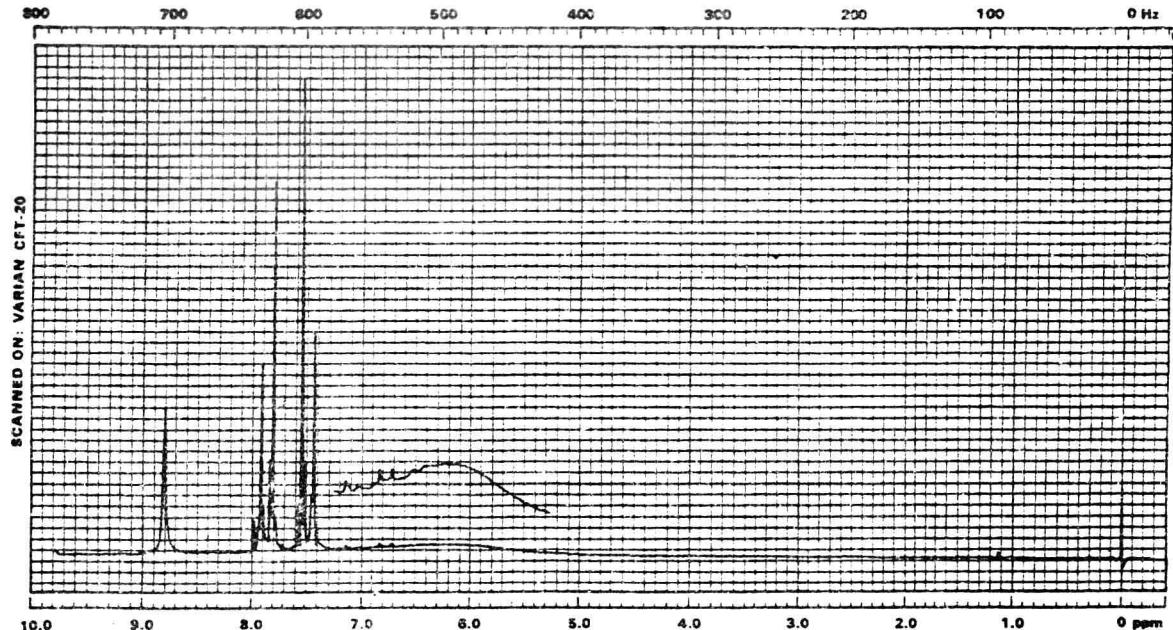
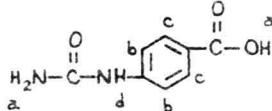
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s \_\_\_\_\_ z \_\_\_\_\_  
t \_\_\_\_\_ 1 \_\_\_\_\_  
u \_\_\_\_\_ 2 \_\_\_\_\_

\*Plus H<sub>2</sub>O



m-ACETYLBENZENESULFONIC ACID, SODIUM SALT

52009 M

C<sub>8</sub>H<sub>7</sub>NaO<sub>4</sub>S

MOL. WT. 222.20

PRISM NO.

79009

SOLVENT

Polysol

SOURCE OF SAMPLE

Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

REFERENCE

TMS

ASSIGNMENTS

a 2.60  
b 7.49  
c 7.95  
d 8.05  
e 8.38  
f 3.53 H<sub>2</sub>O  
g

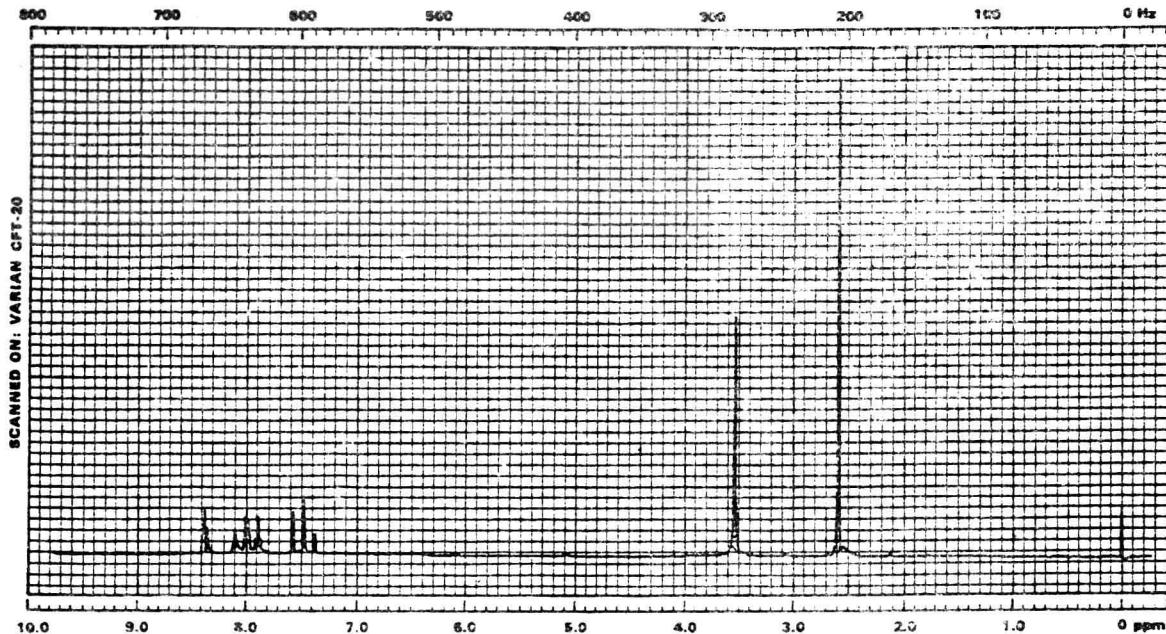
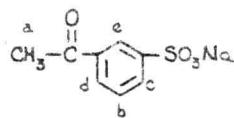
h \_\_\_\_\_  
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52010 M

2-METHYL-5-(TRIFLUOROMETHYL)BENZOTHIAZOLE

C<sub>9</sub>H<sub>6</sub>F<sub>3</sub>NS

MOL. WT. 217.21

PRISM NO.

79010

SOLVENT

CDCl<sub>3</sub>

REFERENCE

TMS

SOURCE OF SAMPLE

Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

ASSIGNMENTS

a 2.88  
b 7.56  
c 7.90  
d 8.21  
e  
f  
g

h  
i  
j  
k  
l  
m  
n

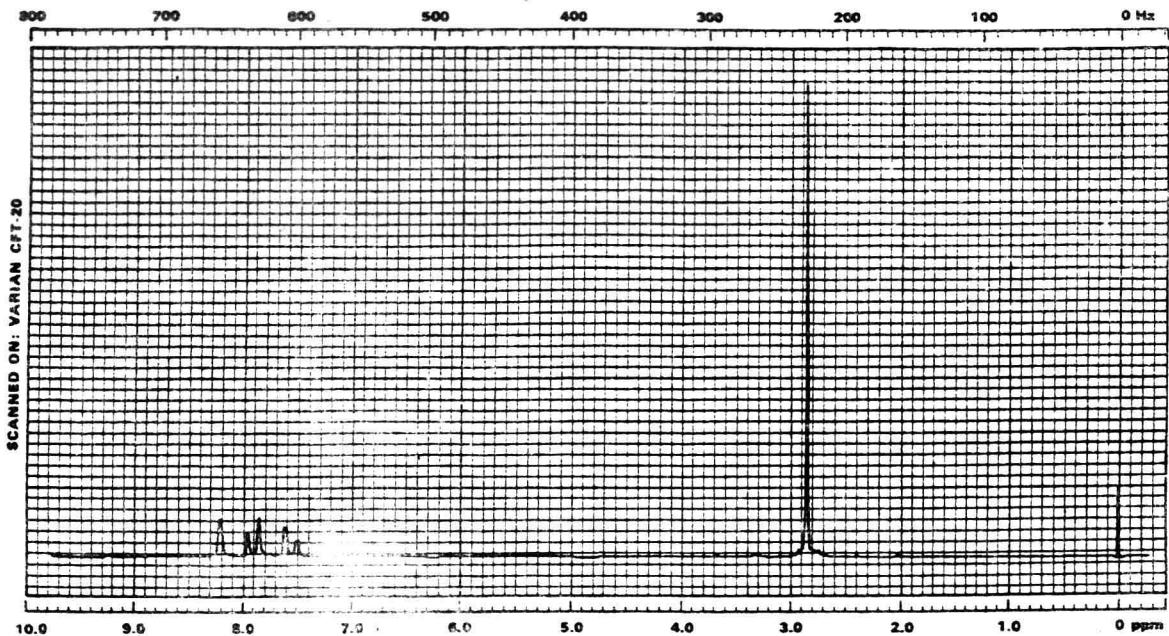
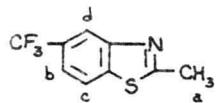
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## 3-AMINO-4-(HEXADECYLTHIO)BENZOIC ACID, METHYL ESTER

52011 M

 $C_{24}H_{41}NO_2S$ 

MOL. WT. 407.66

PRISM NO.

79011

SOLVENT

 $CDCl_3$ 

## SOURCE OF SAMPLE

Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin

REFERENCE

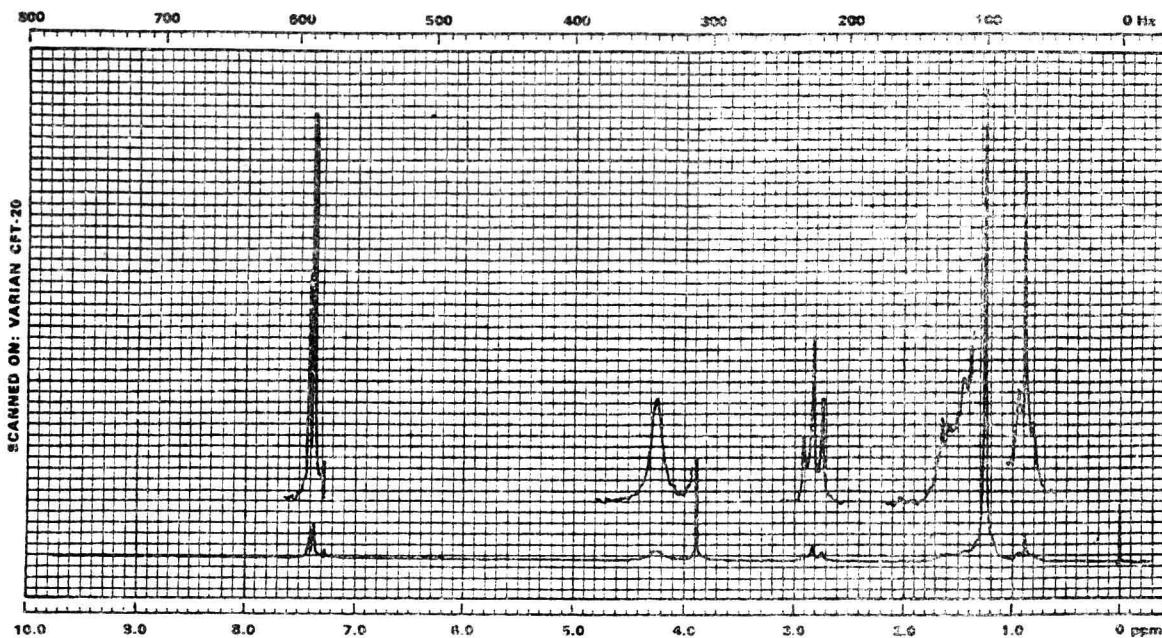
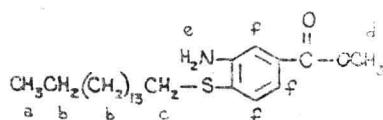
TMS

## ASSIGNMENTS

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a	0.90	h		o		v	
b	1.05-1.90	i		p		w	
c	2.81	j		q		x	
d	3.89	k		r		y	
e	4.23	l		s		z	
f	ca7.39	m		t		1	
g	7.28 $CHCl_3$	n		u		2	



52012 M

## 1-METHYL-3-(1-NAPHTHYL)-1-PHENYLUREA

$$\text{C}_{18}\text{H}_{16}\text{N}_2\text{O}$$

MOL. WT. 276.34

**PRISM NO.**

79013

## SOLVENT

CDCl<sub>3</sub>

## REFERENCE

TMS

**SOURCE OF SAMPLE**

**Aldrich Chemical Company, Inc.,  
Milwaukee, Wisconsin**

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## **ASSIGNMENTS**

a	3.40
b	6.67
c	7.20-7.65
d	7.76
e	7.97
f	
g	

h  
i  
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l  
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